

**ANNUAL REPORTS ON
NMR SPECTROSCOPY**

Volume 10B

ANNUAL REPORTS ON
NMR SPECTROSCOPY

This Page Intentionally Left Blank

ANNUAL REPORTS ON NMR SPECTROSCOPY

Edited by

G. A. WEBB

Department of Chemical Physics, University of Surrey, Guildford, Surrey, England

VOLUME 10B

1980



ACADEMIC PRESS

A Subsidiary of Harcourt Brace Jovanovich, Publishers

London • New York • Toronto • Sydney • San Francisco

ACADEMIC PRESS INC. (LONDON) LTD.
24-28 Oval Road,
London, NW1 7DX

U.S. Edition Published by

ACADEMIC PRESS INC.
111 Fifth Avenue
New York, New York 10003

Copyright © 1980 by ACADEMIC PRESS INC. (LONDON) LTD.

All Rights Reserved

No part of this book may be reproduced in any form by photostat, microfilm, or
any other means, without written permission from the publishers

British Library Cataloguing in Publication Data

Annual reports on NMR spectroscopy.

Vol. 10B

1. Nuclear magnetic resonance spectroscopy

I. Webb, Graham Alan

541'.28 QD96.N8 68-17678

ISBN 0-12-505348-7

ISSN 0066-4103

Printed in Great Britain by J. W. Arrowsmith Ltd.
Bristol BS3 2NT

CONTENTS

PREFACE	vii
-------------------	-----

Fluorine-19 Nuclear Magnetic Resonance Spectroscopy (1976–1978)

V. WRAY

*Gesellschaft für Biotechnologische Forschung mbH
Mascheroder Weg, D-3300 Braunschweig
West Germany*

Contents of Volume 10A	vi
Cumulative indexes of subjects and authors, Volumes 1–10 .	509

CONTENTS OF VOLUME 10A

Nuclear Shielding of the Transition Metals

R. GARTH KIDD

**NMR Chemical Shift Calculations and Stereochemical
Structures of Synthetic Polymers**

ISAO ANDO AND TETSUO ASAKURA

Biological ^{31}P NMR Spectroscopy

I. K. O'NEILL AND C. P. RICHARDS

PREFACE

The widespread application of the NMR spectroscopy of "other nuclei" is clearly reflected in the contributions to these volumes. As with some earlier members of this series of reports, the present volume is split into two parts.

Volume 10B is the more specific one, dealing with the ^{19}F NMR parameters of various series of compounds. It comprises extensive tabulations of ^{19}F NMR data and serves to update reports in earlier volumes of this series.

The range of topics covered in volume 10A indicates some of the numerous areas of science which are dependent upon NMR as a primary investigative tool. The areas covered include transition metal NMR, ^{13}C NMR applications to synthetic polymers, and some uses of ^{31}P NMR in biochemistry.

All the authors writing for volume 10 are new to Annual Reports and, in welcoming their contributions, I wish to thank all of them for their efforts and patience both in the preparation of their manuscripts and of this volume.

*University of Surrey,
Guildford, Surrey,
England*

G. A. WEBB
December 1979

This Page Intentionally Left Blank

Fluorine-19 Nuclear Magnetic Resonance Spectroscopy (1976–1978)

V. WRAY

*Gesellschaft für Biotechnologische Forschung mbH., Mascheroder Weg,
D-3300 Braunschweig, West Germany*

I. Introduction	3
II. Fluorine bonded to a carbon of a saturated aliphatic hydrocarbon	7
A. Fluorine nuclei in a CF ₃ group	9
1. CF ₃ group bonded to carbon	9
2. CF ₃ group bonded to iodine	138
3. CF ₃ group bonded to nitrogen	138
4. CF ₃ group bonded to oxygen	141
5. CF ₃ group bonded to phosphorus	145
6. CF ₃ group bonded to sulphur	152
7. CF ₃ group bonded to any other element	164
B. Fluorine nuclei in a CF ₂ group	167
C. Fluorine nuclei in a CF group	183
D. Miscellaneous studies	201
III. Fluorine bonded to a carbon of an olefin	203
IV. Fluorine bonded to a carbon of an alicyclic non-aromatic system	219
A. Data	221
B. Miscellaneous studies	274
V. Fluorine bonded to a carbon of an aromatic system, including condensed alicyclic aromatic systems	275
A. Fluorine nuclei in monofluorobenzene derivatives	277
B. Fluorine nuclei in difluorobenzene derivatives	319
C. Fluorine nuclei in trifluorobenzene derivatives	329
D. Fluorine nuclei in tetrafluorobenzene derivatives	340
E. Fluorine nuclei in pentafluorobenzene derivatives	360
F. Miscellaneous studies	376
VI. Fluorine bonded to a carbon of a heterocyclic system	377
A. Nitrogen heterocycles	379
B. Oxygen heterocycles	388
C. Sulphur heterocycles	402
D. Selenium heterocycles	403
E. Mixed systems	403
VII. Fluorine bonded to a carbon of a carbonium ionic or carbanionic system	407

VIII. Fluorine bonded to an element other than carbon	419
Fluorine bonded to arsenic	421
Fluorine bonded to gold	422
Fluorine bonded to boron	422
Fluorine bonded to bismuth	424
Fluorine bonded to bromine	425
Fluorine bonded to chlorine	425
Fluorine bonded to germanium	425
Fluorine bonded to hydrogen	425
Fluorine bonded to iodine	425
Fluorine bonded to iridium	426
Fluorine bonded to molybdenum	426
Fluorine bonded to nitrogen	427
Fluorine bonded to niobium	427
Fluorine bonded to osmium	428
Fluorine bonded to phosphorus	428
Fluorine bonded to platinum	441
Fluorine bonded to sulphur	441
Fluorine bonded to antimony	449
Fluorine bonded to selenium	451
Fluorine bonded to silicon	451
Fluorine bonded to tin	460
Fluorine bonded to tantalum	461
Fluorine bonded to tellurium	464
Fluorine bonded to titanium	467
Fluorine bonded to uranium	467
Fluorine bonded to vanadium	468
Fluorine bonded to tungsten	468
Fluorine bonded to xenon	469
Miscellaneous studies	470
IX. Fluorine NMR studies in the solid state, excluding polymers	471
X. Fluorine NMR studies of polymers	475
XI. Fluorine NMR studies using liquid crystal solutions	479
XII. Biological applications of F-19 NMR	483
XIII. Theoretical studies	484
XIV. Index of reviews	486
References	489

I. INTRODUCTION

This Page Intentionally Left Blank

As in previous years the major objective has been to collect the scattered information on F-19 NMR parameters in a form that can readily be retrieved and used. The large volume of data has necessitated a change in policy regarding the presentation of this information. The present reviewer hopes that the use of tables classified according to the type of molecule and internal ordering according to molecular formula will ease the retrieval and comparison of data. Such an approach supplements previous data collections and will hopefully ease computation of such data in the future. The data have been compiled from the author's own records and from the literature abstracts of Current Contents (Physical Sciences), Nuclear Magnetic Resonance Literature (Preston Publications, Illinois), and CA Selects: Nuclear Magnetic Resonance (Chemical Aspects).

All chemical shifts have been converted to the δ scale, with CFCl_3 as reference and positive shifts to high frequency of the reference, using the conversion factors shown in Table I. The original reference has been indicated if it was not CFCl_3 , since several authors persist in not indicating their sign convention or change their convention from paper to paper; some errors may be present in the tables as a consequence. It would be preferable if all journals insisted on the sign convention recommended by IUPAC* and used here.

TABLE I
Reference compound shifts

Index	Compound	Shift from CFCl_3 (ppm)	Index	Compound	Shift from CFCl_3 (ppm)
(a)	$\text{CF}_3 \cdot \text{CO}_2\text{H}$	-78.5(0)	(j)	$\text{CF}_2 \cdot \text{CF}_2 \cdot \text{CF}_2 \cdot \text{CF}_2$	-136.0(0)
(b)	C_6F_6	-162.9(0)	(k)	$p\text{F} \cdot \text{C}_6\text{H}_4 \cdot \text{F}$	-120.0(0)
(c)	$\text{C}_6\text{H}_5 \cdot \text{CF}_3$	-63.9(0)	(l)	$\text{CF}_3 \cdot \text{CO}_2\text{CH}_3$	-74.2(1)
(d)	$\text{C}_6\text{H}_5\text{F}$	-113.15	(m)	$\text{CF}_3 \cdot \text{C}(\text{OH})_2 \cdot \text{CF}_3$	-92.7(7)
(e)	$\text{CFCl}_2 \cdot \text{CFCl}_2$	-67.3(0)	(n)	$\text{CF}_3 \cdot \text{CCl}_3$	-82.2(0)
(f)	$\text{CF}_2 \cdot \text{CF}_2 \cdot \text{CCl}_2 \cdot \text{CCl}_2$	-114.1(0)	(o)	$\text{C}_6\text{H}_5 \cdot \text{SO}_2\text{F}$	+65.5(0)
(g)	CF_2Cl_2	-6.9(0)	(p)	1-fluoronaphthalene	-123.0(0)
(h)	F_2	+422.9(0)	(q)	2-fluoronaphthalene	-117.0(0)
(i)	$\text{CF}_3 \cdot \text{CO} \cdot \text{CF}_3$	-84.6(0)			

The major part of this work consists of tables containing chemical shift and spin-spin coupling constant information for compounds in the liquid state or in solution. The remainder covers studies in solids and liquid crystals, together with biological applications and theoretical studies involving F-19 NMR. To keep this presentation manageable these latter sections either only give the relevant literature references (solids, polymers,

* "Recommendations for the presentation of NMR data for publication in chemical journals. B. Conventions relating to spectra from other nuclei", *Pure Appl. Chem.*, 1976, **45**, 217.

and liquid crystal tables) or brief descriptions with literature references (biological applications and theoretical studies). Finally a bibliography of review articles containing F-19 NMR information is presented.

The tables are arranged according to the type of carbon (or element) to which the fluorine nucleus is bound. The appearance of a compound, with more than one type of fluorine-bonded carbon, in a particular table is according to which fluorine has its chemical shift reported and/or comes higher in the list of tables found above. Thus a molecule with a trifluoromethyl group and a fluoroaromatic residue will appear in Section II provided that the shift of the trifluoromethyl group is reported, but in the fluoroaromatic table (Section V) if only the shift of the fluorine bound to the aromatic residue is reported. Similarly where only coupling constant information is reported the same priority exists. Chemical shift information takes priority over coupling constant information. The only exception to the above is Section VII where an attempt has been made to include all the data for carbonium ions and carbanions irrespective of whether fluorine is directly attached to the charged carbon or not.

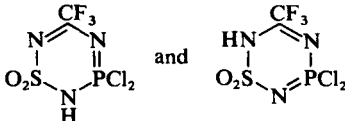
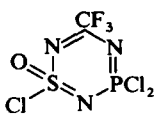
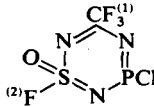
Each table, apart from those in Sections IX to XI consists of (1) the molecular formula of the compound, (2) the structure of the compound, (3) the F-19 chemical shift data, (4) the F-19 spin-spin coupling constant data, and (5) the literature reference number with the reference compound index (from Table I) given in parentheses, where applicable. In Sections IX to XI only columns (1), (2), and (5) are included. Each table is arranged according to the index of the molecular formulae, and the convention adopted for priority of the elements is C, F, H, and then the other elements in alphabetical order. Only the relevant nuclei in the molecule are numbered [column (2)]. In cases where only one number is given but the system shows two chemical shifts, for example the AB spectra from CF₂ groups, these are distinguished by letters following the number.

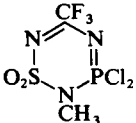
In all cases fluorine-fluorine and fluorine-hydrogen coupling constants are indicated in column (4) by two numbers separated by a hyphen [e.g. (1-2) indicates a $J(\text{F-F})$ or $J(\text{F-H})$ between nuclei numbered (1) and (2) in the structural formula in column (2)], or by numbers followed by letters for AB systems [e.g. (2A-2B) indicates a $J(\text{F-F})$ between fluorine nuclei of an AB system numbered (2) in the structural formula in column (2)]. Other couplings are represented by a number for the fluorine (or F) and an element symbol separated by a hyphen, with the higher atomic number nucleus having priority. The element involved in the coupling may be numbered, but the coupled nuclei, other than fluorine, are always denoted by the element symbol followed by the number [e.g. a coupling between phosphorus and fluorine may appear in column (4) as (P-z), (P-F), (Px-z), (Px-F), $^yJ(\text{P-z})$, $^yJ(\text{P-F})$, $^yJ(\text{Px-z})$, or $^yJ(\text{Px-F})$, where x , y , and z are integers]. The above should be self-evident from the tables.

II. FLUORINE BONDED TO A CARBON
OF A SATURATED
ALIPHATIC HYDROCARBON

This Page Intentionally Left Blank

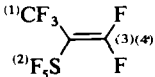
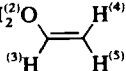
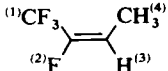
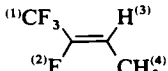
A. Fluorine nuclei in a CF₃ group

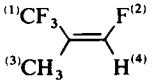
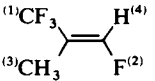
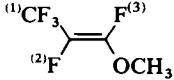
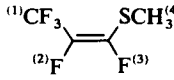
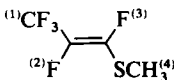
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
<i>1. CF₃ group bonded to carbon</i>				
C ₂ F ₃ HBr ₂	CF ₃ ⁽¹⁾ ·CH ⁽²⁾ Br ₂	-75.1	(1-2) 6	23(a)
C ₂ F ₃ HCl ₂ N ₃ O ₂ PS		-73.6	(P-F) 6	30
C ₂ F ₃ H ₂ NO	CF ₃ ⁽¹⁾ ·CH ⁽²⁾ =NOH	-67.5	(1-2) 5	32(a)
C ₂ F ₃ H ₃ O	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ OH (neat)		(1-2) 8.94 (variation with solvent studied)	35
C ₂ F ₃ BrCl ₂	CF ₃ ·CBrCl ₂	-80.3		23(a)
C ₂ F ₃ Br ₂ Cl	CF ₃ ·CBr ₂ Cl	-79.1		23(a)
C ₂ F ₃ Br ₃	CF ₃ ·CBr ₃	-78		23(a)
C ₂ F ₃ Cl ₃	CF ₃ ·CCl ₃	-81.5		23(a)
C ₂ F ₃ Cl ₃ N ₃ OPS		-75.8	(P-F) 7.3	30
C ₂ F ₄ BrCl	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ BrCl	(1) -83.7 (2) -76.9	(1-2) 8	23(a)
C ₂ F ₄ Br ₂	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ Br ₂	(1) -81.8 (2) -77.2	(1-2) 10	23(a)
C ₂ F ₄ Cl ₂ N ₃ OPS		(1) -75.97 (2) +77.2	(P-1) 8.5 (P-2) 6.8	30

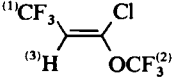
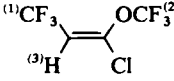
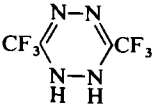
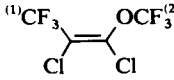
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
C ₂ F ₇ NOS	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·NSOF ₂ ⁽³⁾	(1) -87.6 (3) -50.4	(2) -86.8	(1-3) 1.25 (2-3) 9.4		42
C ₂ F ₇ P	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·PF ₂ ⁽³⁾	(1) -83.4 (3) -106.4	(2) -139.4	(1-2) 2.0 (2-3) 9.0 (P-2) 101.0	(1-3) 6.4 (P-1) 7.4 (P-3) 1259.3	50
C ₂ F ₈ H ₂ S	CF ₃ ⁽¹⁾ ·CH ₂ ⁽⁴⁾ ·SF ₄ ⁽²⁾ F ⁽³⁾	(1) -65.4 (3) +75.5	(2) +68.2	(1-2) 10.1 (2-3) 138 (2-4) 7.3	(1-4) 8.8	52
C ₂ F ₈ O ₃ S	CF ₃ ⁽¹⁾ ·C(O)·OO·SF ₄ ⁽²⁾ F ⁽³⁾	(1) -171.8 (3) +57.9	(2) +56.7	(2-3) 154.8		18
C ₂ F ₈ Si	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·SiF ₃ ⁽³⁾	(1) -86.5 (3) -150.2	(2) -134.4	(1-2) -3.59 (2-3) +4.92	(1-3) -2.81 (Si-3) +275.3	21
C ₂ F ₉ HS	CF ₃ ⁽¹⁾ ·CH ⁽⁵⁾ F ⁽²⁾ ·SF ₄ ⁽³⁾ F ⁽⁴⁾	(1) -76.3 (3) +51.6	(2) -172.1 (4) +68.2	(1-2) 11.0 (1-5) 5.4 (2-4) 2.5 (3-4) 144.0	(1-3) 11.0 (2-3) 2.5 (2-5) 43.5 (3-5) 5.0	52
C ₂ F ₉ BrS	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ Br·SF ₄ ⁽³⁾ F ⁽⁴⁾	(1) -80.0 (3) +50.2	(2) -100.3 (4) +63.8	(3-4) 146.6		53
C ₂ F ₁₀ O ₃ S	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ (OSO ₂ F ⁽³⁾)·SF ₄ ⁽⁴⁾ F ⁽⁵⁾	(1) -79.9 (3) +51.2 (5) +62.3	(2) -110.3 (4) +45.3	(1-2) 12.0 (2-3) 2.8 (2-5) 4.2 (4-5) 142	(1-3) 2.8 (2-4) 12.0 (3-4) 9.5	52
C ₃ F ₃ HBrN	CF ₃ ⁽¹⁾ ·CH ⁽²⁾ Br·CN	-71.5		(1-2) 7.0		62
C ₃ F ₃ H ₂ N	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ ·CN	-66.8		(1-2) 9.2		62
C ₃ F ₃ H ₃ Cl ₂ N ₃ O ₂ PS		-74.2		(P-F) 6.3		30

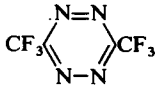
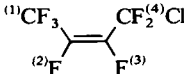
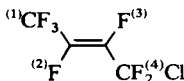
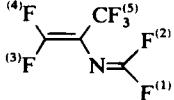
$C_3F_2H_3Cl_2N_3O_2PS$		-68.7		(1-2) 1.5	(P-1) 7.6	30
$C_3F_3Br_2N$	$CF_3 \cdot CBr_2 \cdot CN$	-78.1				62
$C_3F_3Cl_2N$	$CF_3 \cdot CCl_2 \cdot CN$	-77.1				62
$C_3F_3Cl_2NO$	$CF_3 \cdot CO \cdot N = CCl_2$	-74.6				63
C_3F_3NO	$CF_3 \cdot CO \cdot CN$	-80.63				63
C_3F_3NOS	$CF_3 \cdot CO \cdot NCS$	-77.0				63
$C_3F_3NO_2$	$CF_3 \cdot CO \cdot NCO$	-78.8				63
C_3F_4HBr		(1) -60.8	(2) -56.2	(1-2) 17.2 (2-3) 26.9	(1-3) 6.7	68(a)
C_3F_4HBr		(1) -60.65	(2) -57.5	(1-2) 13.4 (2-3) 10.8	(1-3) 6.4	68(a)
$C_3F_4H_3ClS$	$CF_3^{(1)} \cdot CF^{(2)}Cl \cdot SCH_3^{(3)}$	(1) -81.2	(2) -103.2	(1-2) 9.2 (2-3) 1.2	(1-3) 1.0	69(a)
$C_3F_4O_3S$	$CF_3^{(1)} \cdot C(SO_2F^{(2)})=C=O$	(1) -54.1	(2) +79.6	(1-2) 5		67
$C_3F_5H_3S$	$CF_1^{(3)} \cdot CF_2^{(2)} \cdot SCH_3^{(3)}$	(1) -84.0	(2) -95.9	(1-2) 3.4 (2-3) 1.4	(1-3) 0.8	69(a)
$C_3F_5NO_3S$	$CF_2^{(1)} \cdot CF^{(2)}(CN) \cdot OSO_2F^{(3)}$	(1) -74.3 (3) +54.1	(2) -115.4	(1-2) 10.5	(2-3) 5.0	62
C_3F_6HCl	$(CF_3^{(1)})_2CH^{(2)}Cl$	-73.1		(1-2) 6.7		62
$C_3F_6HCl_2OP$	$[(CF_3^{(1)})_2CH^{(2)} \cdot O]PCl_2$	-76.5		(1-2) 5.3		72
$C_3F_6H_2$	$(CF_3^{(1)})_2CH_2^{(2)}$	-63.6		(1-2) 9.7		62
$C_3F_6H_3OP$	$(CF_3^{(1)})_2C(OH)PH_2^{(2)}$	-73.8		(1-2) 2.3		73

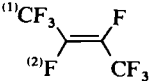
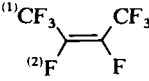
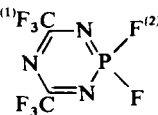
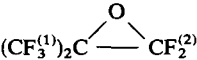
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
C ₃ F ₆ N ₂ OS	CF ₃ ⁽¹⁾ ·CO·N=CF ⁽²⁾ (N=SF ₂ ⁽³⁾)	(1) -79.59 (3) +44.63	(2) -16.27	(2-3) 12.6		63
C ₃ F ₆ OS	CF ₃ ⁽¹⁾ ·CO·SCF ₃ ⁽²⁾	(1) -75.6	(2) -31.4			28
C ₃ F ₇ HO ₂ S	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·SO ₂ H	(1) -72.64	(2) -187.21	(1-2) 8		75
C ₃ F ₇ Cl	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ Cl	(1) -81.40	(2) -144.60	(1-2) 6.6		74
C ₃ F ₇ ClO ₂ S	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·SO ₂ Cl	(1) -70.73	(2) -156.95	(1-2) 8		75(a)
C ₃ F ₇ ClO ₃ S	(CF ₃ ⁽¹⁾) ₂ CCl·OSO ₂ F ⁽²⁾	(1) -78.2	(2) +52.0	(1-2) 2.5		62
C ₃ F ₇ NOS	CF ₃ ⁽¹⁾ ·CO·N=SF ⁽³⁾ ·CF ₃ ⁽²⁾	(1) -76.3 (3) -47.1	(2) -70.6	(1-3) 1.9 (2-3) 0.7		38
C ₃ F ₈ HOP	(CF ₃ ⁽¹⁾) ₂ CH·OPF ₂ ⁽²⁾	(1) -77.0	(2) +47.2	(P-2) 1370		72
C ₃ F ₈ BrN	(CF ₃ ⁽¹⁾) ₂ CBrNF ₂ ⁽²⁾	(1) -78.40	(2) +44.60	(1-2) 12		74
C ₃ F ₈ ClN	(CF ₃ ⁽¹⁾) ₂ CCl·NF ₂ ⁽²⁾	(1) -71.40	(2) +37.30	(1-2) 11		74
C ₃ F ₈ ClNS	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾⁽³⁾ ·N=S(CF ₃ ⁽⁴⁾)Cl	(1) -86.5 (3) -91.4	(2) -88.5 (4) -68.4	(1-2,3) 0 (2-3) 198 (2-4) 6.2 (3-4) 3.5 (1-4) 0.4		38
C ₃ F ₈ S ₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·SS·CF ₃ ⁽³⁾	(1) -84.6 (3) -48.3	(2) -96.4	(2-3) 5.4		47(a)
C ₃ F ₈ S ₃	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·SSS·CF ₃ ⁽³⁾	(1) -84.2 (3) -47.3	(2) -96.8			47(a)
C ₃ F ₉ NOS	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·NSOF ₂ ⁽³⁾	(1) -81.2 (3) +52.1	(2) -144.7	(1-2) 4.4 (1-3) 2.2 (2-3) 11.1		42
C ₃ F ₉ NS	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾⁽³⁾ ·N=S(CF ₃ ⁽⁴⁾)F ⁽⁵⁾	(1) -87.2 (3) -90.2 (5) -17.7	(2) -87.4 (4) -73.0	(1-5) 2.6 (2-5) 22.4 (3-5) 20.6 (2-4) 2.8 (3-4) 1.8 (2-3) 197 ¹ J(1-C) 286 ² J(2,3-C) 45 ¹ J(4-C) 319 ² J(5-C) 44		38

C ₃ F ₁₀ HCIS	CF ₃ ⁽¹⁾ ·CH(CF ₂ ⁽²⁾ Cl)·SF ₅ ⁽³⁾	(1) -61.1 (3) +71.9	(2) -45.5, -51.5		52
C ₃ F ₁₀ O ₃ S	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CO·OO·SF ₅ ⁽³⁾	(1) -82.9 (3) +56.4	(2) -119.8	(1-2) 2.0	18
C ₃ F ₁₀ S		(1) -57.0 (3) -67.6	(2) +72.3 (4) -57.0		52
C ₃ F ₁₀ Si	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·SiF ₃ ⁽⁴⁾	(1) -83.0 (3) -131.3	(2) -127.2 (4) -148.8	(1-2) +0.3 (1-3) +9.10 (1-4) -1.76 (2-3) 0.62 (2-4) -3.57 (3-4) +4.82 (Si-4) +275.7	21
C ₃ F ₁₁ HS	(CF ₃ ⁽¹⁾) ₂ CH·SF ₅ ⁽²⁾	(1) -62.8	(2) +70.8		52
C ₃ F ₁₁ NO ₃ S ₂	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·N=SF ₂ ⁽³⁾ F ⁽⁴⁾ (OSO ₂ F ⁽⁵⁾)	(1) -67.6 (3) +69.1 (ax) (5) +36.9	(2) -144.1 (4) +67.5 (eq)	(1-2) 4.0 (1-3) 13.6 (1-4) 7.6 (3-4) 14.6 (4-5) 2.1	78
C ₃ F ₁₁ NS	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·N=SF ₄ ⁽³⁾	(1) -80.1 (3) +81.8	(2) -148.1	(1-2) 3.8 (1-3) 2.0 (2-3) 19.4	42
C ₃ F ₁₂ O ₃ S ₂	(CF ₃ ⁽¹⁾) ₂ C(OSO ₂ F ⁽²⁾)SF ₄ ⁽³⁾ F ⁽⁴⁾	(1) -66.6 (3) +58.8	(2) +51.0 (4) +65.8	(1-2) 3.0 (1-3) 12.4 (2-3) 2.9 (3-4) 131	52
C ₄ F ₃ H ₅ O	CF ₃ ·CH ₂ ·OCH=CH ₂	-78.67 (shifts for several solvents reported)			61(i)
C ₄ F ₃ H ₅ O	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ O 			(1-2) 8.205 (1-3) 0.723 (1-4) 0.162 (1-5) 0.151	82
C ₄ F ₄ H ₄		(1) -70.2	(2) -131.4	(1-2) 9.0 (1-3) 1.0 (1-4) 2.5 (2-3) 21.8 (2-4) 2.4	56(a)
C ₄ F ₄ H ₄		(1) -75.9	(2) -140.9	(1-2) 11.6 (1-3) 0.9 (2-3) 33.0 (2-4) 2.9	56(a)

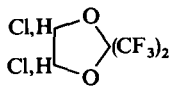
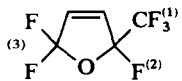
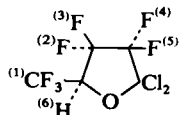
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
C ₄ F ₄ H ₄		(1) -67.1	(2) -122.3	(1-2) 22.0 (2-4) 80.5	(2-3) 4.7	56(a)
C ₄ F ₄ H ₄		(1) -70.3	(2) -142.6	(1-2) 7.0 (2-4) 80.0	(2-3) 3.0	56(a)
C ₄ F ₄ H ₃ I	(2 <i>R</i> , 3 <i>S</i>)-CF ₃ ⁽¹⁾ ·CF ⁽²⁾ H ⁽³⁾ ·CH ⁽⁴⁾ I·CH ₃ ⁽⁵⁾	(1) -76.1	(2) -201.1	(1-2) 10.5 (1-3) 6.1 (1-5) ~0.9	(2-3) 44.4 (2-4) 23.4 (2-5) ~0.9	56(a)
	(2 <i>R</i> , 3 <i>R</i>)-isomer	(1) -75.9	(2) -185.5	(1-2) 10.3 (2-3) 43.0 (1-4) 1.0	(1-3) 5.5 (2-4) 17.0 (1-5) 1.0	56(a)
C ₄ F ₄ H ₃ I	(2 <i>S</i> , 3 <i>S</i>)-CF ₃ ⁽¹⁾ ·CH ⁽⁵⁾ (CH ₃ ⁽⁴⁾)·CF ⁽²⁾ H ⁽³⁾ I	(1) -71.4	(2) -153.6	(1-2) 7.1 (2-3) 51.0	(1-5) ~7.0 (2-5) 27.5	56(a)
	(2 <i>R</i> , 3 <i>R</i>)-isomer	(1) -70.5	(2) -144.5	(1-2) 10.1 (2-3) 48.6	(1-5) 7.6 (2-5) 10.1	56(a)
C ₄ F ₄ O ₂	CF ₃ ⁽¹⁾ ·C(COF ⁽²⁾)=C=O	(1) -56.0	(2) +33.3	(1-2) 11.5		67
C ₄ F ₅ HO ₂	CF ₃ ⁽¹⁾ ·CH ⁽³⁾ (COF ⁽²⁾) ₂	(1) -66.0	(2) +46.4	(1-2) 10	(1-3) 7	67
C ₄ F ₅ H ₃ O		(1) -66.6 (3) -110.3	(2) -194.3			100
C ₄ F ₅ H ₃ S		(1) -68.1 (3) -141.5	(2) -107.3	(1-2) 10.5 (2-3) 19.0 (3-4) 0.9	(1-3) 11.0 (2-4) 0.7	69(a)
C ₄ F ₅ H ₃ S		(1) -69.7 (3) -127.4	(2) -162.5	(1-2) 12.1 (2-3) 143.5 (3-4) 1.9	(1-3) 21.7 (2-4) 0.8	69(a)

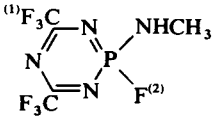
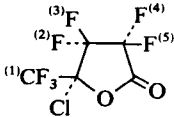
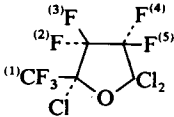
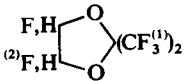
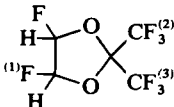
C ₄ F ₅ H ₇ Si	CF ₃ ⁽¹⁾ ·Si(CF ⁽²⁾ H ₂ ⁽³⁾) ₂ CH ₃ ⁽⁴⁾	(1) -63·02 (2) -278·42	(1-2) 3·2 (1-4) 1·0	(2-3) 46·3	45(a)
C ₄ F ₆ HClO		(1) -60·4 (2) -59·0	(1-3) 6·5		86
C ₄ F ₆ HClO		(1) -60·0 (2) -57·8	(1-2) 1·8 (1-3) 6·8		86
C ₄ F ₆ HCl ₃ O	CF ₃ ⁽¹⁾ ·CH ⁽³⁾ Cl·CCl ₂ ·OCF ₃ ⁽²⁾	(1) -68·8 (2) -56·2	(1-3) 7·7		86
C ₄ F ₆ H ₂ ClI	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ I·CF ₂ ⁽³⁾⁽⁴⁾ ·CH ₂ Cl	(1) -73·6 (2) -143·7 (3,4) -105·6	(1-2) 12·2 (1-4) 8·0	(1-3) 12·2	57
C ₄ F ₆ H ₂ N ₄		-70·1			87(a)
C ₄ F ₆ H ₄	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ (CH ₃ ⁽⁵⁾)·CF ₂ ^(3,4) H ⁽⁶⁾	(1) -80·8 (2) -178·1 (3) -131·2 (4) -136·6	(1-2) 6·8 (1-4) 5·7 (2-3) 4·8 (2-5) 22·4 (3-4) 310·5 (3-6) 53·2 (4-6) 54·4	(1-3) 9·1 (1-5) ~1·0 (2-4) 11·4 (2-6) 4·0 (3-5) ~1·5 (4-5) ~1·5	83
C ₄ F ₆ H ₄	CF ₃ ⁽¹⁾ ·CH ⁽⁴⁾ (CF ₂ ⁽²⁾ H ⁽⁵⁾)·CF ⁽³⁾ H ₂ ⁽⁶⁾	(1) -66·7 (2) -124·0 (3) -243·5	(1-4) (1-2) (1-3) 7·9 (2-5) ~50 (3-6) ~50		83
C ₄ F ₆ H ₃ OP	(CF ₃ ⁽¹⁾) ₂ C(OH)·PH ⁽²⁾ CH ₃ ⁽³⁾	-71·0	(1-2) 1·8 (P-1) 12·7	(1-3) 0·7	73
C ₄ F ₆ BrCl ₃	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ Br·CF ₂ ⁽³⁾⁽⁴⁾ ·CCl ₃	(1) -76·0 (2) -133·3 (3) -104·7 (4) -93·0	(1-2) 9·5 (1-4) 7·5 (2-4) 7·5	(1-3) 18·5 (2-3) 7·5 (3-4) 254	57
C ₄ F ₆ Cl ₂ O		(1) (2) -61·0, -65·2	(1-2) 2		

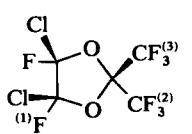
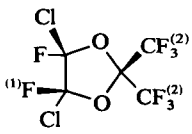
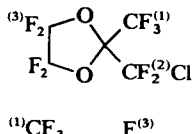
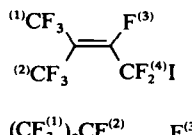
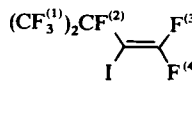
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₄ F ₆ CoO ₃ P	CF ₃ ⁽¹⁾ ·Co(PF ₃ ⁽²⁾)(CO) ₃	(1) +10·50 (2) -16·14	(1-2) 6 (P-1) 59 (P-2) 1371	22(c)
C ₄ F ₆ N ₄		-68·0		87(a)
C ₄ F ₆ Ni	Ni(CF ₃ ·C≡C·CF ₃)	-57·0		88
C ₄ F ₆ O ₃ S	CF ₃ ⁽¹⁾ ·CO·O·CO·SCF ₃ ⁽²⁾	(1) -75·0 (2) -40·7		28
C ₄ F ₇ H ₃	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CH ₃ ⁽³⁾	(1) -79·8 (2) -178·2	(1-2) 7·1 (1-3) 1·2 (2-3) 22·1	83
C ₄ F ₇ H ₃	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ (CF ₂ ⁽³⁾ H ⁽⁵⁾)(CF ⁽⁴⁾ H ₂ ⁽⁶⁾)	(1) -76·7 (2) -192·7 (3) -135·6 (4) -247·6	(1-2) (1-3) (1-4) 7·9 (1-5) ~1·0 (1-6) ~1·1 (2-3) (2-4) ~7 (2-5) 5·6 (2-6) 18·2 (3-5) 46·7 (3-4) 3·5 (3-6) (4-5) ~1·1 (4-6) 53·1	83
C ₄ F ₇ H ₃	CF ₃ ⁽¹⁾ ·CH ⁽⁴⁾ (CF ₂ ⁽²⁾ H ⁽³⁾) ₂	(1) -64·5 (2) -122·6	(1-3) (1-2) 8·1 (2-3) 12·9 (2-4) 52·3	83
C ₄ F ₇ Cl		(1) -67·5 (2) -139·5 (3) -146·5 (4) -59·5	(1-2) 9·59 (1-3) 8·46 (1-4) 16·36 (2-3) 8·5 (2-4) 5·08 (3-4) 12·69	98(a)
C ₄ F ₇ Cl		(1) -71·5 (2) -155·3 (3) -161·5 (4) -60·9	(1-2) 20·30 (1-3) 7·33 (1-4) 1·0 (2-3) 137·6 (2-4) 11·28 (3-4) 29·33	98(a)
C ₄ F ₇ N		(-100°C) (1) -54·5 (2) -39·0 (3) -84·5 (4) -86·8 (5) -65·0	(1-2) 68·0 (1-3) 1·8 (2-3) 24·0 (1-4) 5·1 (2-4) 5·1 (1-5) 0·5 (2-5) 2·0 (3-4) 25·0 (3-5) 8·6 (4-5) 24·8	85

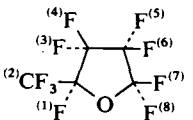
C ₄ F ₈	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ⁽³⁾ =CF ₂ ⁽⁴⁾	(1) -87.7 (3) -192	(2) -124.5 (4) -92.3, -108.5		93
C ₄ F ₈		(1) -72.5	(2) -164.5		93
C ₄ F ₈		(1) -70.1	(2) -147.3		93
C ₄ F ₈ HCl	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ H ⁽⁴⁾ ·CF ₂ ⁽³⁾ CF ₂ Cl	(1) -76.9 (3) -121.5	(2) -214.7	(2-4) 44.0 ¹ J(3-C) 284.0	98
C ₄ F ₈ H ₂	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ (CF ₂ ⁽³⁾ H ⁽⁴⁾) ₂	(1) -74.9 (3) -134.6	(2) -198.0	(1-2) 7.5 (1-3) 8.8 (2-3) 8.2 (3-4) 53.3 (2-4) ~6.5	83
C ₄ F ₈ H ₂	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ⁽³⁾ H ₂ ⁽⁴⁾	(1) -76.5 (3) -243.0	(2) -190.7	(1-2) 6.9 (1-3) 9.4 (1-4) ~1.0 (2-3) 12.4 (2-4) 16.5 (3-4) 46.4	83
C ₄ F ₈ H ₆ NP	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·PF ₃ ⁽³⁾ ·N(CH ₃) ₂	(1) -82.3 (3a) -50.9	(2) -117.8 (3e) -65.9	(3a-3e) 73 (P-2) 113 (P-3a) 910 (P-3e) 1040	71
C ₄ F ₈ N ₂ OS	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾⁽³⁾ ·NS·CF ₃ ⁽⁴⁾ ·NCO	(1) -86.9 (3) -86.9	(2) -83.8 (4) -71.9	(1-2) (1-3) 0 (1-4) 0.5 (2-3) 195 (2-4) 8.3 (3-4) 4.2	38
C ₄ F ₈ N ₂ O ₂ S	(CF ₃ ⁽¹⁾ ·CO·N=) ₂ SF ₂ ⁽²⁾	(1) -77.1	(2) +53.8	(1-2) 0.5	37
C ₄ F ₈ N ₂ S ₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·N=S=N·C(S)CF ₃ ⁽³⁾	(1) -81.9 (3) -51.3	(2) -92.4	(1-2) 0 (1-3) 1.8 (2-3) 4.3	38
C ₄ F ₈ N ₃ P		(1) -75.2	(2) -46.2	(P-1) 6.2 (P-2) 1000	95
C ₄ F ₈ O		(1) -69.9	(2) -109.8		96

Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
C ₄ F ₈ O	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CO·F ⁽³⁾	(1) -73.0 (3) +32.3	(2) -179.9			96
C ₄ F ₈ O	CF ₃ ⁽¹⁾ ·CO·CF ₂ ⁽²⁾ ·CF ₃ ⁽³⁾	(1) -82.1 (3) -84.6	(2) -120.4			97
C ₄ F ₉ H	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ₂ ⁽³⁾ H ⁽⁴⁾	(1) -74.3 (3) -133.1	(2) -191.9	(1-2) 7.2 (2-3) 7.4 (3-4) 52.5	(1-3) 9.1 (2-4) 5.4	83
C ₄ F ₉ H	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ H	(1) -83.7 (3) -133.7	(2) -130.9 (4) -141.3			93
C ₄ F ₉ HN ₂ OS	(CF ₃ ⁽¹⁾⁽²⁾) ₂ C=N·S(O)CF ₃ ⁽³⁾ (=NH)	(1) -62.9 (3) -54.9	(2) -67.5			101
C ₄ F ₉ CSO	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ₂ ⁽³⁾ ·O ⁻ Cs ⁺	(1) -73.1 (3) -19.4	(2) -180.3			96
C ₄ F ₉ IZn	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₃ ·ZnI	(1) -84.0 (2) -125.5, -127.5, -130				93
C ₄ F ₁₀	(CF ₃ ⁽¹⁾) ₃ CF ⁽²⁾	(1) -75.1	(2) -189.4	(1-2) 6.6		83
C ₄ F ₁₀ ClN	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·NF ⁽³⁾ (CF ₂ ⁽⁴⁾ CF ₂ ⁽⁵⁾ Cl)	(1) -85.0 (3) -93.5 (5) -72.0	(2) -110.5 (4) -108.9	(1-2) 0.9 (1-4) 1.6 (3-5) 19.5	(1-3) 15.9 (2-5) 2.7 (4-5) 1.8	99(a)
C ₄ F ₁₀ CINS	(CF ₃ ⁽¹⁾⁽²⁾) ₂ CF ⁽³⁾ ·NSCl·CF ₃ ⁽⁴⁾	(1) -78.5 (3) -145	(2) -80.1 (4) -69.1	(1-3) (2-3)(3-4) ~4.7 (1-4) 0.8 (1-2) 8.4	(2-4) 0	38
C ₄ F ₁₀ O ₂	(CF ₃ ⁽¹⁾) ₃ CO·OF ⁽²⁾	(1) -69.5	(2) -295.0	(1-2) 6.0		105
C ₄ F ₁₀ S ₂	(CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·S) ₂	(1) -84.8	(2) -96.6			47(a)
C ₄ F ₁₁ NOS	(CF ₃ ⁽¹⁾⁽²⁾) ₂ C=N·S(O)F ₂ ⁽³⁾ ·CF ₃ ⁽⁴⁾	(1) -67.1 (3) +70.8	(2) -68.6 (4) -66.3	(3-4) 22		101
C ₄ F ₁₂ N ₂ S	(CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·N=) ₂ SF ₂ ⁽³⁾	(1) -87.5 (3) -63.3	(2) -87.1	(1-3) 1.2 (2-3) 9.55		42

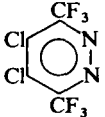
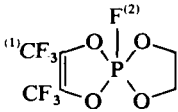
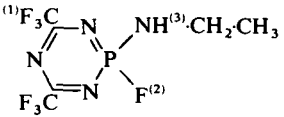
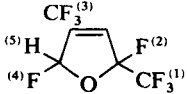
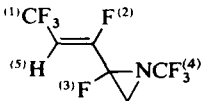
C ₄ F ₁₃ NOS	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·N=SF ₂ ⁽³⁾ F ⁽⁴⁾ (OCF ₃ ⁽⁵⁾)	(1) -69.2 (2) -144.6 (3) +58.1 (ax)(4) +51.2 (eq) (5) -83.3	(1-2) 4.2 (1-3) 12.9 (1-4) 7.1 (3-5) 10.1 (4-5) 7.2 (3-4) 148	78
C ₅ F ₃ H ₃ O ₃	CF ₃ C(CO ₂ CH ₃)=C=O	-55.8		67
C ₅ F ₃ H ₇ O ₂	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ ·CH ₂ ·CO ₂ CH ₃	-68.0	(1-2) 10	117
C ₅ F ₅ H ₅ O ₃	CH ₃ O·CF ₂ ⁽²⁾ ·CH ⁽³⁾ (CF ₃ ⁽¹⁾)·CO ₂ H	(1) -65.1 (2) -75.2 (centre AB)	(1-2) 10.5 (1-3) 7.5 (2-3) 7.5	67
C ₅ F ₅ H ₉ N ₂ OSSi	CF ₃ ⁽¹⁾ ·CO·N=SF ₂ ⁽²⁾ =N·Si(CH ₃) ₃	(1) -77.0 (2) +65.8		37
C ₅ F ₆ HCl ₃ O ₃	(CF ₃) ₂ C(OH)O·CO·CCl ₃	-89.28		122
C ₅ F ₆ H ₂ Cl ₂ O ₂		-79.0		120
C ₅ F ₆ H ₂ Cl ₂ O ₃	(CF ₃) ₂ C(OH)O·CO·CHCl ₂	-89.15		122
C ₅ F ₆ H ₂ O		(1) -83.45 (2) -121.05 (3A) -96.47 (3B) -91.50	(2-3A)(2-3B)10 (3A-3B) 160	121(a)
C ₅ F ₆ H ₂ O ₅	HO ₂ C·CF ₂ ⁽³⁾ ·O·CF ⁽²⁾ (CF ₃ ⁽¹⁾)·CO ₂ H	(1) -83.56 (2) -131.56 (3A) -74.47 (3B) -81.90	(1-2) 3 (2-3A) 15 (2-3B) 8 (3A-3B) 152	121(c)
C ₅ F ₆ H ₃ ClO ₃	(CF ₃ ⁽¹⁾) ₂ C(OH)·O·CO·CH ₂ Cl	(1) -88.94		122
C ₅ F ₆ H ₃ N	CF ₃ ⁽¹⁾ ·CH ₂ ⁽³⁾ ·CH ⁽⁴⁾ (CF ₃ ⁽²⁾)·CN	(1) -66.2 (2) -70.7	(1-2) 1.3 (1-3) 9.8 (2-4) 7.5	117
C ₅ F ₆ H ₄ O ₂ S	(CF ₃) ₂ C(OH)·S·CO·CH ₃	-87.51		122
C ₅ F ₆ H ₄ O ₃	(CF ₃) ₂ C(OH)·O·CO·CH ₃	-88.56		122
C ₅ F ₆ H ₇ OP	(CF ₃ ⁽¹⁾) ₂ CH ⁽²⁾ ·O·P(CH ₃) ₂	-77.5	(1-2) 6.5 (P-1) 6.5	124
C ₅ F ₇ HCl ₂ O		(1) -74.0 (2) -108.2 (3) -128.4 (4) -123.2 (5) -119.7	(1-2) 2.0 (1-3) 13.9 (1-4) 2.1 (2-3) 250 (4-5) 226 (1-6) 5.4	125

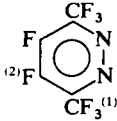
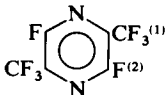
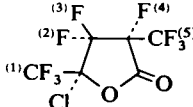
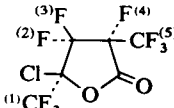
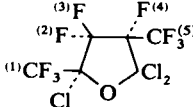
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_3F_7H_3O_3$	$(CF_3^{(1)})_2C(OH) \cdot O \cdot CO \cdot CH_2F$	(1) -89.10		122
$C_3F_7H_3S$	$(CF_3^{(1)})_2CF^{(2)} \cdot S \cdot CH=CH_2$	(1) -76.7 (2) -162.4	(1-2) 10.7	69
$C_3F_7H_4ClS$	$(CF_3^{(1)})_2CF^{(2)} \cdot S \cdot CH_2^{(3)} \cdot CH_2Cl$	(1) -76.9 (2) -164.3	(1-2) 10.3 (2-3) 2.0	69
$C_3F_7H_4N_4P$		(1) -74.7 (2) -44.0	(P-1) 5.5 (P-2) 948	95
$C_3F_7H_5NOS$	$(CF_3^{(1)})_2CF^{(2)} \cdot SON(CH_3)_2$	(1) -70.81, -72.90 (2) -180.21		75
$C_3F_7ClO_2$		(1) -76.8 (2) -121.8 (3) -127.8 (4) -116.5 (5) -129.9	(1-3) 16.7 (1-2) 2.8 (1-5) 2.8 (2-4) 8.6 (3-4) 7.4 (2-3) 244 (4-5) 292	125
$C_3F_7Cl_3O$		(1) -73.4 (2) -104.5 (3) -125.3 (4) -115.9 (5) -119.1	(1-3) 17.4 (1-5) 4.5 (3-5) 8.0 (3-4) 3.8 (2-4) 11.7 (2-5) 4.2 (4-5) 228 (2-3) 244	125
$C_3F_8H_2O$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CO \cdot CH_2^{(4)} \cdot CF_3^{(3)}$	(1) -82.4 (2) -122.2 (3) -63.5	(1-2) 1.0 (2-3) 1.2 (2-4) 0.8 (3-4) 9.2	128
$C_3F_8H_2O_2$		(1) -81.4 (2) -129.4		120
$C_3F_8H_2O_2$		(1) -139.2 (2)(3) -82.36, -80.66		120

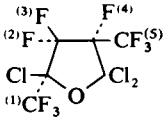
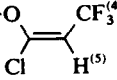
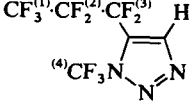
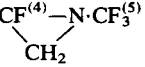
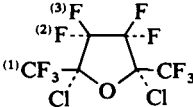
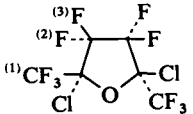
C ₅ F ₈ H ₆ N ₂ S	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾⁽³⁾ ·N=S(CF ₃ ⁽⁴⁾)N(CH ₃ ⁽⁵⁾) ₂	(1) -86.5 (3) -79.3	(2) -74.4 (4) -69.5	(2-3) 185 (4-5) 1.2	38
C ₅ F ₈ Cl ₂ O ₂	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ Cl·CF ₂ ⁽³⁾ ·O ₂ CCF ₂ ⁽⁴⁾ Cl	(1) -78.6 (3) -84.7	(2) -140.4 (4) -65.8	(1-2) 6.1 (2-3) 7.8	127
C ₅ F ₈ Cl ₂ O ₂		(1) -58.5 (2)(3) -80.2, -80.6			120
C ₅ F ₈ Cl ₂ O ₂		(1) -50.6 (2) -80.5			120
C ₅ F ₉ H ₃ N ₂ OS	(CF ₃ ⁽¹⁾⁽²⁾) ₂ C=N·S(O)CF ₃ ⁽³⁾ (=NCH ₃)	(1) -64.8 (3) -52.7	(2) -74.5		101
C ₅ F ₉ H ₃ O	(CF ₃ ⁽¹⁾) ₃ COCH ₃ ⁽²⁾	-71.0		(1-2) 1	130
C ₅ F ₉ ClO ₂	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ Cl·CF ₂ ⁽³⁾ ·O ₂ CCF ₃ ⁽⁴⁾	(1) -78.5 (3) -84.3	(2) -140.4 (4) -76.1	(1-2) 6.3 (2-3) 7.2	127
C ₅ F ₉ ClO ₂		(1) -79.4 (3) -81.7	(2) -68.4		120
C ₅ F ₉ I		(1) -60.8 (3) -101.6	(2) -57.1 (4) -54.7	(1-2) 9.2 (2-3) 9.2 (3-4) 14.2	129(a)
C ₅ F ₉ I		(1) -76.2 (3) -62.0	(2) -158.5 (4) -56.6	(1-2) 6.8 (2-3) 30.0 (3-4) 6.8	129(a)

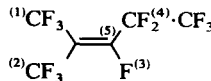
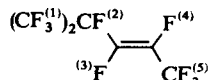
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_5F_{10}H_6NP$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot PF_3^{(4)} \cdot N(CH_3)_2$	(1) -81.6 (3) -115.8 (4e) -67.7	(2) -125.9 (4a) -50.5	(1-3) 10.8 (3-4a) 6.0 (4a-4e) 77 (P-1) 5.2 (P-3) 114 (P-4a) 900 (P-4e) 1000		71
$C_5F_{10}N_2OS$	$(CF_3^{(1)})_2CF^{(2)} \cdot NS \cdot CF_3^{(3)} \cdot NCO$	(1) -80.7 (3) -73.7	(2) -142.9	(2-3) 5.6		38
$C_5F_{10}N_2S_2$	$(CF_3^{(1)})_2CF^{(2)} \cdot NS \cdot NC(S)CF_3^{(3)}$	(1) -74.9 (3) -51.5	(2) -133.6	(1-2) 5.1 (2-3) 7.2	(1-2) 2.2	38
$C_5F_{10}O$		(1) -126.5 (3,4) -128.9 (6) -135.3	(2) -80.8 (5) -126.9 (7,8) -83.6	(5-6) 253		125
$C_5F_{10}O$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CO \cdot CF_3^{(4)}$	(1) -84.0 (3) -120.1	(2) -123.4 (4) -79.8			97
$C_5F_{11}I$	$(CF_3^{(1)})_2CF^{(2)} \cdot CF^{(3)}I \cdot CF_3^{(4)}$	(1) -70.7, -71.2 (2) -167.8 (4) -74.4	(3) -144.8			129(a)
$C_5F_{11}N$	$(CF_3^{(1)})_2CF^{(2)} \cdot CF^{(3)}=NCF_3^{(4)}$	(1) -74.5 (3) -17.3	(2) -191 (4) -59.9			132
$C_5F_{12}HOP$	$(CF_3^{(1)})_2CH^{(3)} \cdot O \cdot P(CF_3^{(2)})_2$	(1) -78.8	(2) -65.2	(1-3) 4.5 (P-2) 100		72
$C_5F_{12}ClN$	$(CF_3^{(1)} \cdot CF_2^{(2)})(CF_3^{(3)})N \cdot CF_2^{(4)} \cdot CF_2^{(5)}Cl$	(1) -85.3 (3) -52.3 (5) -71.3	(2) -94.2 (4) -91.0	(1-3) 7.4 (2-3) 15.6 (3-4) 16.1 (2-5) 13.0 (2-5) 9.6		99(a)
$C_5F_{12}OS$	$(CF_3^{(1)})_3COSCF_3^{(2)}$	(1) -71.0	(2) -53.8			130
$C_5F_{12}O_2$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot O \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot OCF_3^{(4)}$	(1) -90.7 (3) -93.9	(2) -91.9, 91.7 (4) -59.4	(3-4) 10		104
$C_5F_{13}CoO_2P$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot Co(PF_3^{(4)})_2(CO)_2$	(1) -79.67 (3) -58.21	(2) -113.76 (4) -14.88	(3-4) 6 (P-3) 65 (P-4) 1348		22(c)

C ₅ F ₁₃ N	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·N(CF ₃ ⁽⁴⁾) ₂	(1) -84.2 (3) -93.5	(2) -128.0 (4) -55.1	(1-3) 10.4 (3-4) 15.8	(2-4) 8.2	99(a)
C ₅ F ₁₃ NO	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·N(CF ₃ ⁽⁴⁾)OCF ₃ ⁽⁵⁾	(1) -83.4 (3) -101.5 (5) -66.3	(2) -128.4 (4) -66.3			133
C ₅ F ₁₃ P	(CF ₃ ⁽¹⁾) ₂ CF ₂ ⁽²⁾ ·P(CF ₃ ⁽³⁾) ₂	(1) -75.0 (3) -49.2	(2) -187.2	(1-2) 10.1 (2-3) 8.7 (P-2) 56.0	(1-3) 6.1 (P-1) 15.8 (P-3) 88.8	50
C ₅ F ₁₄ Ge	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·Ge(CF ₃ ⁽³⁾) ₃	(1) -85.26 (3) -50.8	(2) -116.9	(1-3) 2.7		45(a)
C ₅ F ₁₅ NS	(CF ₃ ⁽¹⁾) ₂ CF ₂ ⁽²⁾ ·N=SF ₂ ⁽³⁾ (CF ₃ ⁽⁴⁾) ₂	(1) -78.3 (3) +57.6	(2) -144.1 (4) -65.1	(1-2) 4.5 (1-4) 0.4	(1-3) 13.9 (3-4) 12.8	78
C ₆ F ₃ H ₇ O ₂	CF ₃ ⁽¹⁾ ·CH ⁽²⁾ =CH ⁽³⁾ ·CH ₂ ⁽⁴⁾ ·CO ₂ CH ₃ (<i>E</i>)	-65.2		(1-2)(1-3) 6.4, 2.7 (1-4) 2.1		117
C ₆ F ₃ H ₇ O ₂	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ ·CH=CH·CO ₂ CH ₃ (<i>E</i>)	-66.22		(1-2) 10.5		117
C ₆ F ₃ H ₉	CF ₃ ⁽¹⁾ ·CH=CH ⁽²⁾ ·CH ₂ ⁽³⁾ ·CH ₂ ·CH ₃ (<i>cis</i>) (<i>trans</i>)	-58.9 -64.8				117
C ₆ F ₃ H ₉	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ ·CH=CH·CH ₂ ·CH ₃ (<i>cis</i>) (<i>trans</i>)	-67.1 -67.5		(1-2) 2 (1-2) 10.4 (1-2) 10.5	(1-3) 6.5	117
C ₆ F ₃ H ₉ N ₂ OSi	CF ₃ ·CO·N(CN)Si(CH ₃) ₃	-76.84				63
C ₆ F ₃ H ₉ N ₂ Si	(CH ₃) ₃ SiN=C(CF ₃)CN	-77.2				63
C ₆ F ₃ H ₉ O	CF ₃ ⁽¹⁾ ·CH ⁽²⁾ (CH ₃)·CO·CH ₂ ·CH ₃	-69.5		(1-2) 8.5		137
C ₆ F ₃ H ₉ O ₂	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ ·(CH ₂) ₂ ·CO ₂ CH ₃	-67.04		(1-2) 10.2		117
C ₆ F ₃ H ₁₀ NO	CF ₃ ⁽¹⁾ ·CH ⁽²⁾ =N ⁺ (O ⁻)C(CH ₃) ₃	-65.7		(1-2) 6		32(a)
C ₆ F ₃ CoO ₄	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·Co(CO) ₄	(1) -82.36 (2) -56.88				22(c)
C ₆ F ₆ H ₂ N ₂		-68.5				87(a)

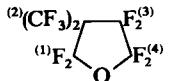
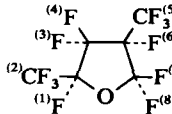
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_6F_6H_6O_2$	$CF_3^{(1)} \cdot CH_2^{(3)} \cdot CH^{(4)}(CF_3^{(2)}) \cdot CO_2CH_3$	(1) -66.9	(2) -69.7	(1-3) 10.4	(2-4) 8.5	117
$C_6F_6H_6O_2S$	$(CF_3)_2C(OH)S \cdot CO \cdot CH_2 \cdot CH_3$	-87.52				122
$C_6F_6Cl_2N_2$		-66.0				87(a)
$C_6F_7H_4O_4P$		(1) -67.0	(2) -74.3	(P-2) 1003		141
$C_6F_7H_6N_4P$		(1) -74.9	(2) -41.5	(P-1) 6.0 (2-3) 3	(P-2) 960	95
$C_6F_7H_6N_4P$	As above with $NH \cdot CH_2 \cdot CH_3$ replaced by $N(CH_3)_2$	(1) -74.9	(2) -44.8	(P-1) 5.5 (2-3) 1.1	(P-2) 961	95
$C_6F_8H_2O$		(1) -82.71 (3) -93.0	(2) -111.86 (4) -112.6	(2-5) 16	(4-5) 60	121(a)
$C_6F_8H_3N$		(1)(4) -59.7 (3) -171.3	(2) -110.3	(2-5) 30	(1-5) 7.5	142
$C_6F_8H_{10}NP$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot PF_3^{(3)}N(CH_2 \cdot CH_3)_2$	(1) -82.5 (3a) -51.5	(2) -119.9 (3e) -66.9	(2-3a) 16.0 (P-2) 116 (P-3e) 1050	(3a-3e) 70 (P-3a) 897	71

$C_6F_8N_2$		(1) -66.7	(2) -145.3			87(a)
$C_6F_8N_2$		(1) -69.5	(2) -82.0			87(a)
C_6F_9H	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot C \equiv CH^{(5)}$	(1) -81.6 (3) -124.0	(2) -126.2 (4) -99.6	(4-5) 5.5		144
$C_6F_9HBr_2$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CBr=CHBr$	(1) -81.6 (3) -121.4	(2) -126.2 (4) -106.1			144
$C_6F_9H_2Br_3$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CBr_2 \cdot CH_2Br$	(1) -81.6 (3) -115.4	(2) -126.2 (4) -102.5			144
$C_6F_9H_3$	$(CF_3)_3C \cdot CH=CH_2$	-67.6				145
$C_6F_9H_3O_2$	$(CF_3)_3C \cdot O \cdot CO \cdot CH_3$	-70.1				130
$C_6F_9ClO_2$		(1) -75.9 (3) -125.3 (5) -74.7	(2) -115.3 (4) -174.9	(1-3) 25.2 (4-5) 9.5 (2-5) 2.3 (2-4) 18.1	(1-2) 2.5 (3-5) 15.6 (2-3) 260	125
$C_6F_9ClO_2$		(1) -77.1 (3) -108.9 (5) -74.9	(2) -118.7 (4) -181.1	(1-2) 18.0 (1-3) 2.5 (4-5) 11.5 (2-3) 262	(1-4) 7.1 (3-5) 15.0 (2-5) 2.5	125
$C_6F_9Cl_3O$		(1) -76.4 (3) -118.5 (5) -71.0	(2) -110.9 (4) -153.5	(1-3) 17.5 (3-5) 14.0 (2-4) 14.4 (4-5) 5.5	(1-2) 2.5 (3-4) 1.0 (2-5) 3.7 (2-3) 248	125

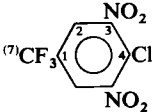
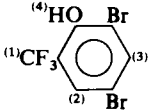
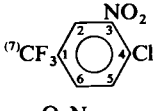
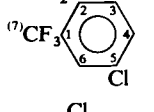
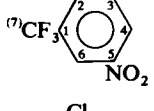
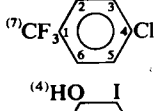
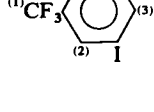
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_9Cl_3O$		(1) -77.4 (2) -120.7 (3) -93.3 (4) -163.1 (5) -71.3	(1-2) 16.8 (1-4) 13.2 (1-3) 1.8 (3-5) 14.5 (3-4) 1.7 (2-4) 8.1 (2-5) 2.1 (4-5) 5.7 (2-3) 254	125
C_6F_{10}	$CF_3^{(1)}CF_2^{(2)} \cdot C(CF_3^{(3)})=C=CF_2^{(4)}$	(1) -87.7 (2) -116.5 (3) -65.7 (4) -95.7	(1-3) 3.0 (1-2) 1.2 (3-4) 3.0 (2-3) 8.1 (2-4) 4.2 (1-4) ~1	146
$C_6F_{10}HClO$		(1) -83.4 (2) -131.3 (3) -86.4 (4) -62.0	(4-5) 7.5	86
$C_6F_{10}HN_3$		(1) -80.7 (2)(3) -105.7, -125.1 (3) -57.4		142
$C_6F_{10}H_3N$	 (50:50 mixture of two enantiomeric pairs)	(1) -74.9 (2) -213.5 (3) -122.8 (AB system) -124.9 (AB system) (4) -188.4, -189.4 (5) -59.7	(2-6) 45	142
$C_6F_{10}H_6N_2S$	$(CF_3^{(1)})_2CF_2^{(2)} \cdot N=S(CF_3^{(3)})N(CH_3^{(4)})_2$	(1) ~-80.8 (2) -135.8 (3) -70.6	(1-2) 5.4 (3-4) 1.3	38
$C_6F_{10}Cl_2O$		(1) -77.8 (2) -107.8 (3) -128.2	(2-3) 230	125
$C_6F_{10}Cl_2O$		(1) -76.9 (2)(3) -120.4		125

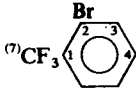
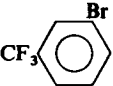
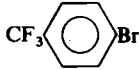
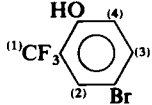
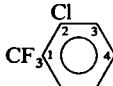
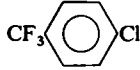
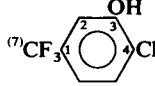
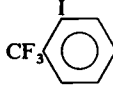
C ₆ F ₁₀ CoO ₃ P	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·Co(PF ₃ ⁽⁴⁾)(CO) ₃	(1) -79.75 (3) -50.78	(2) -114.36 (4) -14.27	(3-4) 6 (P-4) 1373	(P-3) 62	22(c)
C ₆ F ₁₀ O	CF ₃ ⁽¹⁾ ·C≡C·O·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₃ ⁽⁴⁾	(1) -52.8 (3) -131.2	(2) -88.7 (4) -84.3			86
C ₆ F ₁₀ O ₃	(CF ₃ ⁽¹⁾) ₂ CF ₂ ⁽²⁾ ·O ₂ C·CF ₂ ⁽³⁾ ·CO·F ⁽⁴⁾	(1) -79.2 (3) -111.5	(2) -142.3 (4) +20.3	(1-2) 2.4 (P-4) 1373	(3-4) 10.1	147
C ₆ F ₁₁ HO	(CF ₃ ⁽¹⁾) ₂ CH ⁽⁴⁾ ·CO·CF ₂ ⁽²⁾ ·CF ₃ ⁽³⁾	(1) -63.4 (3) -81.9	(2) -123.6	(1-2) 2.9 (1-4) 6.8 (2-4) 0.4	(1-3) 1.1 (2-3) 0.3	128
C ₆ F ₁₁ H ₃ N	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ⁽⁴⁾ -N·CF ₃ ⁽⁵⁾ CH ₂	(1) -81.8 (2)(3) -124.0, -126.8 (4) -189.7	(5) -60.6			142
C ₆ F ₁₁ ClO	CF ₂ ⁽¹⁾ Cl·CF ₂ ⁽²⁾ ·CF ⁽³⁾ (CF ₃ ⁽⁴⁾)·O·CF ⁽⁵⁾ =CF ₂ ⁽⁶⁾	(1) -66.4 (3) -134.3 (5)(6) -114.0, -121.1, -132.5	(2) -117.6 (4) -77.7			148
C ₆ F ₁₁ ClO ₂	CF ₂ ⁽¹⁾ Cl·CF ⁽²⁾ (CF ₃ ⁽³⁾)·O·CF ₂ ⁽⁴⁾ ·CO·CF ₃ ⁽⁵⁾	(1) -68.8 (3) -79.0	(2) -140.3 (4)(5) -75.1	(2-) 21.5 (4,5-) 4.5		143
C ₆ F ₁₁ ClO ₂	(CF ₃ ⁽¹⁾) ₂ CF ₂ ⁽²⁾ ·O·CF ₂ ⁽³⁾ ·CO·CF ₂ ⁽⁴⁾ Cl	(1) -81.3 (3) -73.5	(2) -145.6 (4) -66.2	(1-2) 1.2 (2-3) 21.5	(1-3) 5.7 (3-4) 7.5	143
C ₆ F ₁₂	⁽¹⁾ CF ₃ 			(1-C1) 276 (3-C5) 292	(2-C2) 276 (4-C5) 33	149
C ₆ F ₁₂	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ 			(1-C1) 288 (2-C3) 25 (4-C4) 231 (5-C4) 42	(2-C1) 26 (3-C3) 231 (5-C5) 274 (4-C5) 31	149

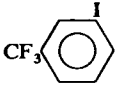
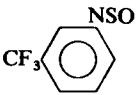
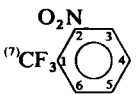
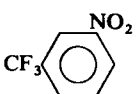
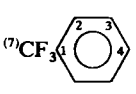
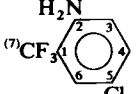
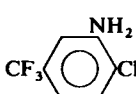
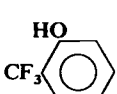
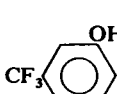
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
C ₆ F ₁₂	$\text{CF}_3^{(1)} \cdot \text{CF}_2^{(2)} \cdot \text{CF}_2^{(3)} \cdot \text{CF}_2^{(4)} \cdot \text{C}(\text{F}^{(7)})=\text{C}(\text{F}^{(6)})\text{F}^{(5)}$	(1) -83.7 (3) -127.0 (5) -193.4 (7) -108.1	(2) -129.0 (4) -120.5 (6) -91.7	(1-3) 10.7 (2-3) 3.5 (2-5) 3.6 (3-7) 4.0 (4-6) 5.8 (5-6) 40.2 (6-7) 51.8	(1-4) 2.5 (2-4) 12.7 (3-5) 7.1 (4-5) 14.1 (4-7) 28.2 (5-7) 117.2	150
C ₆ F ₁₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ⁽⁵⁾ =CF ₂ ⁽⁶⁾	(1) -83.7 (3) -127 (5) -193.4	(2) -129 (4) -120.5 (6) -91.7, -108.1			93
C ₆ F ₁₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ⁽³⁾ =CF·CF ₂ ·CF ₃	(1) -87.1 (3) -158.9	(2) -124.3			93
C ₆ F ₁₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ⁽⁴⁾ =CF ⁽⁵⁾ ·CF ₃ ⁽⁶⁾	(1) -83.7 (3) -121.9 (5) -160.7	(2) -130.9 (4) -160.7 (6) -72.3			93
C ₆ F ₁₂ H ₂ ClO ₂ P	[(CF ₃) ₂ CH·O] ₂ PCl	-77.8				72
C ₆ F ₁₂ H ₃ O ₂ P	[(CF ₃) ₂ C(OH)] ₂ PH	-73.8				73
C ₆ F ₁₂ N ₂ O ₂ S	[(CF ₃ ⁽¹⁾⁽²⁾) ₂ C=N]S(O)CF ₃ ⁽³⁾ [=NC(O)CF ₃ ⁽⁴⁾]	(1) -68.1 (3) -74.5	(2) -63.5 (4) -56.8			101
C ₆ F ₁₂ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ⁽³⁾ (CF ₃ ⁽⁴⁾)·O·CF ⁽⁵⁾ =CF ₂ ⁽⁶⁾	(1)(4) -77.8, -79.7 (2) -123.9 (5)(6) -114.2, -120.6, -133.1	(3) -136.6			148
C ₆ F ₁₂ O	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CO·F ⁽⁵⁾	(1) -70.9 (3) -112.7 (5) +21.0	(2) -184.6 (4) -115.7			151
C ₆ F ₁₂ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ⁽⁴⁾ (CF ₃ ⁽⁵⁾)·CO·F ⁽⁶⁾	(1) -81.6 (3) -117.9 (5) -73.0	(2) -124.6 (4) -180.0 (6) +32.3	(1-3) 10.5 (1-4) 2.4		152

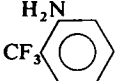
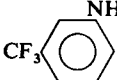
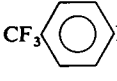
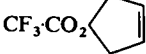
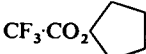
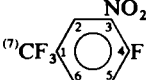
C ₆ F ₁₂ O	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ⁽³⁾ (CF ₃ ⁽⁴⁾)·CO·F ⁽⁵⁾	(1) -72.1 (3) -182.1 (5) +31.9	(2) -176.4 (4) -71.5			152
C ₆ F ₁₂ O		(1) -69.9 (3) -113.5	(2) -63.0 (4) -86.4	(1-2) 12.9 (2-4) 2.0	(2-3) 12.9 (1-3) 1.9	152
C ₆ F ₁₂ O		(1) -124.5 (3) -122.1 (5) -73.1 (7) -73.4	(2) -80.1 (4) -126.4 (6) -182.3 (8) -82.3	(2-4) 10.7 (3-4) 268	(2-3) 3.7 (7-8) 140	152
C ₆ F ₁₂ O	CF ₃ ⁽¹⁾ ·CO·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₃ ⁽⁵⁾	(1) -76.4 (3) -122.4 (5) -83.2	(2) -119.1 (4) -127.1			97
C ₆ F ₁₂ O ₂	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·O·CF ₂ ⁽³⁾ ·CO·CF ₃ ⁽⁴⁾	(1) -81.4 (3) -75.8	(2) -145.4 (4) -75.2	(1-3) 5.5 (2-3) 21.2	(1-2) 2.3 (3-4) 5.6	143
C ₆ F ₁₂ O ₃	CF ₃ ⁽¹⁾ ·CO·OO·C(CF ₃ ⁽²⁾) ₃	(1) -70.0	(2) -69.8			153
C ₆ F ₁₃ H	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₄ ·CF ₂ ⁽³⁾ H	(1) -83.7 -132.5	(2) -126.1, -129.3 (3) -140.9			93
C ₆ F ₁₃ H	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CH ⁽⁵⁾ (CF ₃ ⁽⁴⁾) ₂	(1) -79.4 (3) -110.1	(2) -124.0 (4) -60.7	(3-5)(4-5) 7.0		123
C ₆ F ₁₃ H ₂ O ₂ P	[(CF ₃ ⁽¹⁾) ₂ CH ⁽³⁾ ·O] ₂ PF ⁽²⁾	(1) -77.3	(2) +52.0	(1-3) 6.0	(P-2) 1270	72
C ₆ F ₁₃ Cl	(CF ₃ ⁽¹⁾) ₂ CCl·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₃ ⁽⁴⁾	(1) -66.7 (3) -121.7	(2) -107.5 (4) -80.1			123
C ₆ F ₁₃ ClO ₂	(CF ₃ ⁽¹⁾) ₃ C·OO·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ Cl	(1) -69.42 (3) -71.1	(2) -83.0	(1-2) 2.5 (1-3) ~0	(2-3) 9.0	105
C ₆ F ₁₃ ClO ₂	(CF ₃ ⁽¹⁾) ₃ C·OO·CF ⁽²⁾ Cl·CF ₃ ⁽³⁾	(1) -69.50 (3) -81.0	(2) -83.2	(1-2) 0.4 (1-3) ~0	(2-3) 2.5	105
C ₆ F ₁₃ Cl ₂ N	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·N(CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ Cl) ₂	(1) -81.2 (3) -86.1	(2) -88.7 (4) -68.0			99(a)

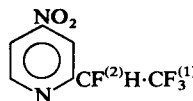
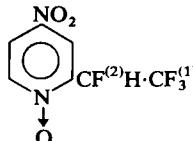
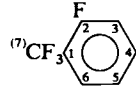
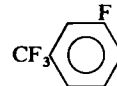
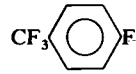
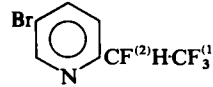
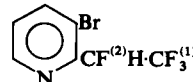
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_6F_{13}I$	$(CF_3^{(1)})_2CF^{(2)} \cdot CF^{(3)}I \cdot CF_2^{(4)(5)} \cdot CF_3^{(6)}$	(1) -69.7 (3) -143.5 (6) -77.5	(2) -168.2 (4)(5) -110.6	(4-5) 284 (5-6) 10.0	(4-6) 10.0	129(a)
$C_6F_{13}O_2P$	$\begin{array}{c} (1)(2) (CF_3)_2C-O \\ \quad \quad \quad \backslash \\ (CF_3)_2C-O-PF^{(3)} \end{array}$	(1) -70.2 (3) -26.8	(2) -67.8	(P-1) 15.0 (1-3) 8.5	(P-3) 1425	155
$C_6F_{14}HN$	$(CF_3^{(1)})_2N \cdot CF_2^{(2)} \cdot CH^{(4)}(CF_3^{(3)})_2$	(1) -54.2 (3) -63.6	(2) -80.0	(1-2) 12.1 (2-3) 10.9 (2-4) 12.8	(1-3) 1.5 (3-4) 7.2	99(a)
$C_6F_{14}HNO_2$	$CF_3^{(1)} \cdot NH^{(4)} \cdot CF_2^{(2)} \cdot OO \cdot C(CF_3^{(3)})_3$	(1) -80.2 (3) -69.2	(2) -68.3	(1-2) 8.3 (3-4) ~4.0	(1-4) 4.0	153
$C_6F_{14}H_2NOP$	$\overline{PF_2^{(1)}(NH_2^{(3)})OC(CF_3)_2C(CF_3)_2O}^{(2)}$	(1) -61.5 At -90 °C (1e) -69.9	(2) -68.4 (1a) -51.8	(P-1) 837 (P-1a) 733 (1a-1e) 67.1	(P-1e) 950 (1a-3) 39.1	156
$C_6F_{14}ClN$	$(CF_3^{(1)})_2N \cdot CF_2^{(2)} \cdot CF^{(3)}(CF_3^{(4)}) \cdot CF_2^{(5)}Cl$	(1) -54.6 (3) -174.3 (5) -62.3	(2) -84.5 (4) -73.5	(1-2) 14.4 (1-4) 2.6	(1-3) 17.7 (1-5) 3.0	99(a)
$C_6F_{14}ClN$	$(CF_3^{(1)})_2N \cdot CF_2^{(2)} \cdot CCl(CF_3^{(3)})_2$	(1) -53.8 (3) -70.1	(2) -77.3	(1-2) 16.2 (2-3) 12.8	(1-3) 3.2	99(a)
$C_6F_{14}O_2$	$CF_3^{(1)} \cdot O \cdot CF^{(2)}(CF_3^{(3)}) \cdot CF(OCF_3) \cdot CF_3$	(1) -53.1, -53.2 (2) -77.0, -78.1 (3) -139.2, -141.0				151
$C_6F_{14}O_2$	$CF_3^{(1)} \cdot CF_2 \cdot O \cdot CF_2^{(2)} \cdot CF_2 \cdot O \cdot CF_2 \cdot CF_3$	(1) -90.5 (2) -91.7				104
$C_6F_{15}H_2O_2P$	$[(CF_3^{(1)})_2 \cdot CH \cdot O]_2PF_3^{(2)}$	(1) -76.0 (2) -65.0		(P-2) 845		157
$C_6F_{15}N$	$(CF_3^{(1)})_2N \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_3^{(5)}$	(1) -55.1 (3) -124.4 (5) -83.8	(2) -92.3 (4) -129.0	(1-2) 15.6 (3-5) 10.4 (2-4) 15.2	(1-3) 8.1 (2-5) 2.4	99(a)

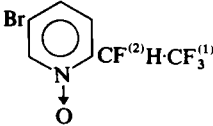
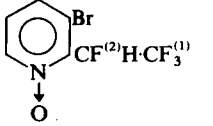
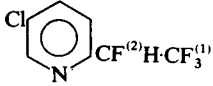
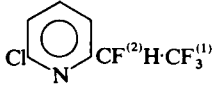
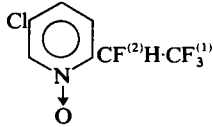
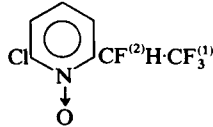
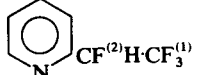
$C_6F_{15}N$	$(CF_3^{(1)})_2N \cdot CF_2^{(2)} \cdot CF^{(3)}(CF_3^{(4)})_2$	(1) -55.5 (3) -184.4	(2) -86.4 (4) -75.6	(1-2)(1-3) 14, 17 (1-4) 2.3 (2-3) 5.9 (2-4) 11.3 (3-4) 6.5	99(a)
$C_6F_{16}N_2S$	$[(CF_3^{(1)})_2CF^{(2)} \cdot N=]_2SF_2^{(3)}$	(1) -81.0 (3) +67.4	(2) -145.8	(1-2) 4.2 (1-3) 2.1 (2-3) 8.6	42
$C_7F_3H_2ClN_2O_4$				(F-C1) 36.4 (F-C2) 3.6 (F-C4) 1.0 (F-C7) 273.5	166
$C_7F_3H_3Br_2O$				(1-2) -0.641 (1-3) -0.578 (1-4) 0.271	167
$C_7F_3H_3ClNO_2$				(F-C1) 34.6 (F-C2) 4.0 (F-C4) 1.2 (F-C6) 3.5 (F-C7) 272.5	166
$C_7F_3H_3ClNO_2$				(F-C1) 34.6 (F-C4) 0.8 (F-C6) 5.6 (F-C7) 273.8	166
$C_7F_3H_3ClNO_2$				(F-C1) 33.0 (F-C2) 1.7 (F-C4) 0.7 (F-C6) 5.6 (F-C7) 273.6	166
$C_7F_3H_3Cl_2$				(F-C1) 32.1 (F-C2) 1.8 (F-C4) 1.3 (F-C6) 5.3 (F-C7) 272.8	166
$C_7F_3H_3I_2O$				(1-2) -0.680 (1-3) -0.600 (1-4) ±0.10	167

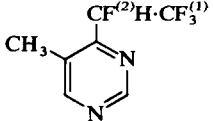
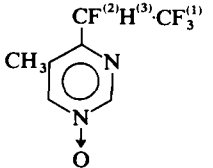
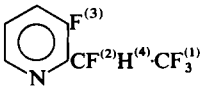
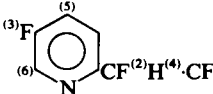
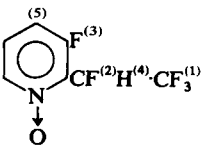
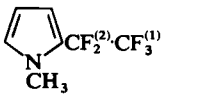
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_3H_4Br$			(F-C1) 31.2 (F-C2) 2.0 (F-C4) 1.0 (F-C6) 5.5 (F-C7) 273.1	166
$C_7F_3H_4Br$			(F-C1) 33.0 (F-C2) 3.9 (F-C4) 1.2 (F-C6) 3.8 (F-C7) 272.6	166
$C_7F_3H_4Br$			(F-C1) 32.9 (F-C2) 3.8 (F-C4) 1.6 (F-C7) 272.0	166
$C_7F_3H_4BrO$			(1-2) -0.648 (1-3) -0.683 (1-4) -0.718 (1-5) -3.069	167
$C_7F_3H_4Cl$			(F-C1) 31.3 (F-C2) 2.0 (F-C4) 1.1 (F-C6) 5.3 (F-C7) 272.6	166
$C_7F_3H_4Cl$			(F-C1) 33.0 (F-C2) 3.8 (F-C4) 1.6 (F-C7) 271.7	166
$C_7F_3H_4ClO$			(F-C1) 33.1 (F-C2) 3.9 (F-C4) 1.5 (F-C6) 3.9 (F-C7) 272.0	166
$C_7F_3H_4I$			(F-C1) 30.8 (F-C2) 2.0 (F-C4) 1.0 (F-C6) 5.6 (F-C7) 273.8	166

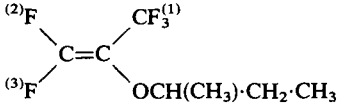
$C_7F_3H_4I$			(F-C1) 32.8 (F-C2) 3.9 (F-C4) 1.3 (F-C6) 3.8 (F-C7) 272.9	166
$C_7F_3H_4NOS$		-63.59		168
$C_7F_3H_4NO_2$			(F-C1) 33.9 (F-C4) 1.0 (F-C6) 5.3 (F-C7) 272.9	166
$C_7F_3H_4NO_2$			(F-C1) 34.0 (F-C2) 4.0 (F-C4) 1.1 (F-C6) 3.6 (F-C7) 272.4	166
$C_7F_3H_5$			(F-C1) 32.2 (F-C2) 3.9 (F-C4) 1.4 (F-C7) 271.5	166
$C_7F_3H_5ClN$			(F-C1) 30.6 (F-C2) 2.0 (F-C4) 1.1 (F-C6) 5.4 (F-C7) 272.3	166
$C_7F_3H_5ClN$			(F-C1) 32.4 (F-C2) 3.9 (F-C4) 1.7 (F-C6) 4.0 (F-C7) 272.0	166
$C_7F_3H_5O$			(F-C1) 30.5 (F-C2) 2.0 (F-C4) 1.2 (F-C6) 5.0 (F-C7) 271.9	166
$C_7F_3H_5O$			(F-C1) 32.4 (F-C2) 3.9 (F-C4) 1.2 (F-C6) 3.9 (F-C7) 272.1	166

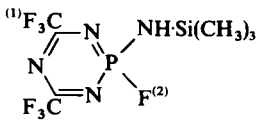
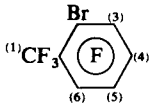
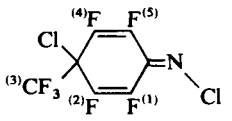
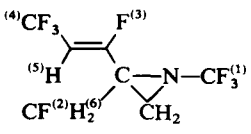
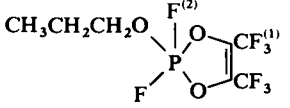
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₇ F ₃ H ₆ N			(F-C1) 29.7 (F-C2) 1.9 (F-C4) 1.1 (F-C6) 5.3 (F-C7) 271.8	166
C ₇ F ₃ H ₆ N			(F-C1) 31.6 (F-C2) 3.9 (F-C4) 1.3 (F-C6) 4.0 (F-C7) 272.0	166
C ₇ F ₃ H ₆ N			(F-C1) 32.5 (F-C2) 3.8 (F-C7) 270.3	166
C ₇ F ₃ H ₇ O ₂		-76.0		117
C ₇ F ₃ H ₉ O	CF ₃ ⁽¹⁾ ·CH ⁽²⁾ ·CO·(CH ₂) ₃ ·CH ₂	-69.5	(1-2) 8.5	137
C ₇ F ₃ H ₉ O ₂		-76.12		117
C ₇ F ₃ H ₁₁	CF ₃ ⁽¹⁾ ·CH ⁽²⁾ =CH ⁽³⁾ ·CH ₂ ⁽⁴⁾ ·(CH ₂) ₂ ·CH ₃ (cis) (trans)	-59.6 -65.3	(1-2) 8.2 (1-2)(1-3) 5.9, 2.0 (1-4) 2.0	117
C ₇ F ₃ H ₁₁	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ ·CH=CH·(CH ₂) ₂ ·CH ₃ (cis) (trans)	-67.5 -68.0	(1-2) 10.7 (1-2) 10.8	117
C ₇ F ₃ H ₁₃	CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ ·(CH ₂) ₄ ·CH ₃	-67.9	(1-2) 10.6	117
C ₇ F ₃ H ₁₄ O ₃ P	CF ₃ ·CH ₂ ·CH ₂ ·P(O)(OCH ₂ CH ₃) ₂	-68.9	(1-2) 9.6 (1-3) 4.2	84
C ₇ F ₄ H ₃ NO ₂			(7-C1) 34.8 (7-C2) 4.0 (7-C4) 1.3 (7-C6) 3.6 (7-C7) 272.0	166

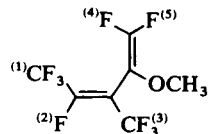
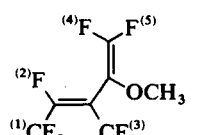
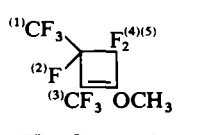
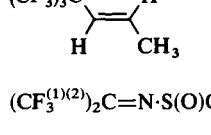
$C_7F_4H_3N_2O_2$		(1) -78.5 (2) -203.9		169
$C_7F_4H_3N_2O_2$		(1) -78.6 (2) -207.0		169
$C_7F_4H_4$			(7-C1) 32.8 (7-C2) 2.3 (7-C4) 1.0 (7-C6) 4.7 (7-C7) 271.5	166
$C_7F_4H_4$	As above	(7) -61.571 (2) -114.763	(2-7) 12.8 (2-C2) -255.5 (2-C7) -1.0 (7-C2) 2.6 (7-C7) -271.8	408
$C_7F_4H_4$			(7-C1) 33.3 (7-C2) 3.9 (7-C4) 1.2 (7-C6) 3.8 (7-C7) 271.7	166
$C_7F_4H_4$			(7-C1) 33.0 (7-C2) 3.8 (7-C4) 1.4 (7-C7) 271.7	166
$C_7F_4H_4BrN$		(1) -76.5 (2) -199.5		169
$C_7F_4H_4BrN$		(1) -79.1 (2) -202.5		169

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_4H_4BrNO$		(1) -73.9 (2) -206.6		169
$C_7F_4H_4BrNO$		(1) -77.9 (2) -205.5		169
$C_7F_4H_4ClN$		(1) -77.5 (2) -201.3		169
$C_7F_4H_4ClN$		(1) -78.5 (2) -202.5		169
$C_7F_4H_4ClNO$		(1) -74.9 (2) -209.1		169
$C_7F_4H_4ClNO$		(1) -78.7 (2) -202.9		169
$C_7F_4H_5N$		(1) -79.5 (2) -202.5		169

$C_7F_4H_5NO$	<i>N</i> -oxide of above	(1) -78.7	(2) -204.5			169
$C_7F_4H_6N_2$		(1) -77.5	(2) -206.0			169
$C_7F_4H_6N_2O$		(1) -77.3	(2) -204.0	(1-2) 13	(1-3) 6	169
$C_7F_4H_8O$	$CF_3^{(1)} \cdot CF^{(2)} = C(CH_3)CO \cdot CH_2 \cdot CH_3$	(1) -68.4	(2) -122			137
$C_7F_5H_4N$		(1) -78.5 (3) -127.5	(2) -205.9	(1-2) 13.4 (1-4) 6.7 (2-4) 44.5	(1-3) 6.7 (2-3) 9.5	169
$C_7F_5H_4N$		(1) -80.1 (3) -128.5	(2) -202.9	(1-2) 12.4 (2-3) 3.4 (3-5) ~7	(1-4) 6.2 (2-4) 44.2 (3-6) ~3	169
$C_7F_5H_4NO$		(1) -77.0 (3) -117.0	(2) -211.5	(1-2) 13.0 (1-4) 6.7 (2-4) 42.2	(1-3) 13.0 (2-3) 26.0 (3-5) 6.0	169
$C_7F_5H_6N$		(1) -83.4	(2) -106	(1-2) 2.8		137
$C_7F_5H_7O_2$	$CF_3^{(1)} \cdot CF^{(2)} = CF^{(3)} \cdot CO_2CH_2 \cdot CH_2 \cdot CH_3$ (<i>E</i>)	(1) -62.5 (3) -150	(2) -151	(1-2) 5.5 (2-3) 120	(1-3) 15	170

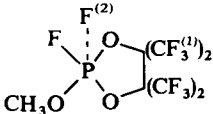
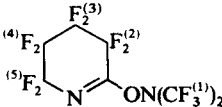
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₇ F ₅ H ₉ O		(1) -66.8 (2) -101.1 (3) -92.7	(1-2) 24.0 (1-3) 9.6 (2-3) 46.4	171(a)
C ₇ F ₅ H ₉ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CH(CH ₃)·CO·CH ₂ ·CH ₃	(1) -83.7 (2) -119.5	(2A-2B) 279	137
C ₇ F ₅ H ₉ OS	CF ₃ ⁽¹⁾ ·CO·CF ₂ ⁽²⁾ ·SC(CH ₃) ₃	(1) -73.6 (2) -83.4	(1-2) 7.9	143
C ₇ F ₅ H ₁₀ ClO	CF ₃ ⁽¹⁾ ·C(OH)(CH ₂ CH ₂ CH ₂ CH ₂)CF ₂ ⁽²⁾ Cl	(1) -74.4 (2) -62.0	(1-2) 11.4	4
C ₇ F ₆ H ₄ N ₂		-67.4, -65.2		87
C ₇ F ₆ H ₆	(CF ₃) ₂ C=C=C(CH ₃) ₂	-63.5		172(a)
C ₇ F ₆ H ₈ O ₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾⁽³⁾ ·CF ⁽⁴⁾ H ⁽⁵⁾ ·CO ₂ CH ₂ CH ₂ CH ₃	(1) -81 (2) -121 (3) -126.5 (4) -206	(2-3) 300 (2-4) 12 (3-4) 14 (4-5) 48	170
C ₇ F ₆ H ₉ O ₃ PS	(CF ₃) ₂ C=C(SCH ₃)P(O)(OCH ₃) ₂	-60.05, -57.72		173(e)
C ₇ F ₆ H ₁₀ ClOP	(CF ₃ ⁽¹⁾⁽²⁾) ₂ CH ⁽³⁾ O·PCl[C(CH ₃) ₃]	(1) -74.1 (2) -73.5	(1-2) 8.0 (1, 2-3) 5.7	72
C ₇ F ₆ H ₁₀ N ₃ OP	(CF ₃) ₂ C=N·P(O)N(CH ₃) ₂ CH ₂ ·CH ₂ ·NCH ₃	-68.3		140
C ₇ F ₆ H ₁₁ O ₃ P	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ H ⁽⁵⁾ ·CF ₂ ⁽³⁾⁽⁴⁾ ·P(O)(OCH ₂ CH ₃) ₂	(1) -75.1 (2) -216.5 (3) -123.47 (4) -129.53	(1-5) ~11 (2-5) 42.0 (3-5) ~11 (4-5) ~11 (3-4) 328 (P-3, 4) ~93	84
C ₇ F ₆ H ₁₁ O ₃ P	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ (CF ₂ ⁽³⁾ H)·P(O)(OCH ₂ CH ₃) ₂	(1) -74.5 (2) -200.1 (3) -133.5	(P-2) 71.0	84
C ₇ F ₇ H ₇	(CF ₃ ⁽¹⁾) ₂ CH ⁽³⁾ ·CF ⁽²⁾ =C(CH ₃ ⁽⁴⁾) ₂	(1) -67.0 (2) -126.5	(1-2) 8.0 (1-3) 8.0 (2-4) 8.0 (2-3) 26.6	172(a)
C ₇ F ₇ H ₈ NO ₂ S	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·S(O)N·CH ₂ ·CH ₂ ·O·CH ₂ ·CH ₂	(1) -70.84, -72.75 (2) -179.32	(1-2) 10	75(a)

$C_7F_7H_8NO_3S$	$(CF_3^{(1)})_2CF^{(2)} \cdot SO_2 \cdot \overbrace{N \cdot CH_2 \cdot CH_2 \cdot O \cdot CH_2 \cdot CH_2}$	(1) -72.13	(2) -170.36	(1-2) 8	75(a)
$C_7F_7H_{10}N_3S$	$(CF_3^{(1)})_2CF^{(2)} \cdot N = \overbrace{S \cdot N(CH_3)CH_2 \cdot CH_2 \cdot NCH_3}$	(1) -80.6	(2) -130.0		140
$C_7F_7H_{10}N_4PSi$		(1) -75.2	(2) -29.2	(P-1) 6.0 (P-2) 970	95
$C_7F_7H_{10}OP$	$(CF_3^{(1)})_2CH^{(3)}O \cdot PF^{(2)} \cdot C(CH_3)_3$	(1) -74.0	(2) -115	(1-3) 5.7 (P-2) 1130	72
C_7F_7Br		(1) -52.6 (3) -122.4 (4)(5) -146.6, -152.8 (6) -133.2			430(b)
$C_7F_7Cl_2N$		(1) -144.3 (2) -134.7 (3) -72.7 (4) -140.6 (5) -146.0		(2-3) 12 (2-4) 4.5 (3-4) 12 (4-5) 5	174(a)
$C_7F_7Cl_2N$	As above	(1) -144.7 (2) -135.0 (3) -73.1 (4) -140.9 (5) -146.3		(2-3) 12.0 (2-4) 4.2 (3-4) 12.0 (4-5) 5.4	175(a)
$C_7F_8H_5N$		(1) -59.4 (2) -222.8 (3) -101.8 (4) -59.2		(3-4) 18 (4-5) 8 (2-6) 49	142
$C_7F_8H_6$	$(CH_3)_2C \cdot CF^{(2)}(CF_3^{(1)}) \cdot CF_2^{(3)} \cdot CF_2^{(4)}$	(1) -75.9 (2)(3) -128.4, -129.8 (4) -194.0			176
$C_7F_8H_7O_3P$		(1) -66.0 (2) -59.2		(P-2) 907	141

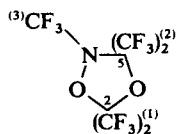
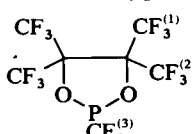
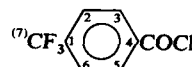
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_8H_{10}N_3P$	$(CF_3^{(1)})_2C=N \cdot PF_2^{(2)} \cdot N(CH_3)CH_2 \cdot CH_2 \cdot NCH_3$	(1) -69.2 (2) -75.2	(P-2) 1067	140
$C_7F_9H_3O$		(1)(3) -64.4, -72.9 (2) -109.6 (4)(5) -58.8, -49.0	(1-2)(2-3) 21, 6 (4-5) 56 (1-3) ~2	146(a)
$C_7F_9H_3O$		(1)(3) -60.3, -70.3 (2) } -98.5, -116.5 (4)(5) } other peak not detected	(1-3) 11.5	146(a)
$C_7F_9H_3O$		(1)(3) -62.2, -77.0 (2) -172.0 (4)(5) -116.7, -118.2	(4-5) 210	146(a)
$C_7F_9H_5$		-67.9		145(a)
$C_7F_9H_9N_2OSSi$	$(CF_3^{(1)(2)})_2C=N \cdot S(O)CF_3^{(3)} [=NSi(CH_3)_3]$	(1) -63.1 (2) -66.8 (3) -55.1		101
$C_7F_9H_9OSi$	$(CF_3^{(1)})_3COSi(CH_3^{(2)})_3$	-74.2	(1-2) 0.2	130
$C_7F_{10}H_4O$	$CF_3^{(1)} \cdot CF_2^{(2)(3)} \cdot CH(CF_3^{(4)}) \cdot C(OCH_3)=CF_2^{(5)(6)}$	(1)(4) -64.5, -86.6 (2)(3) -121.25, -118.55 (5)(6) -98.5, -112.7	(2-3) 286 (5-6) 68	146(a)
$C_7F_{10}H_{10}NP$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot PF_3^{(4)} N(CH_2CH_3)_2$	(1) -81.9 (2) -126.5 (3) -116.8 (4a) -53.7 (4e) -65.9	(1-3) 11.2 (3-4a) 5.0 (4a-4e) 76 (P-1) 5.4 (P-3) 116 (P-4a) 900 (P-4e) 1003	71

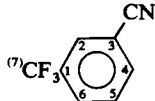
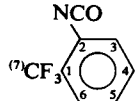
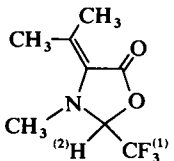
C ₇ F ₁₀ O		(1)(2) -63.6 (3) -70.4 (4) +29.7	(3-4) 10	177
C ₇ F ₁₁ H ₄ N		(1) -59.4 (2) -220.8 (3)(4) -119.8, -125.8 (5) -81.5	(2-6) 48	142
C ₇ F ₁₁ Cl ₃ O		(1) -78.2 (4) -109.0 (5) -117.0 (6) -152.6 (7) -70.9	(1-4) 4.1 (1-5) 4.9 (5-7) 14.1 (4-7) 3.6 (6-7) 5.2 (4-6) 14.7 (4-5) 254	125
C ₇ F ₁₁ Cl ₃ O		(1) -77.8 (4) -118.5 (5) -90.5 (6) -162.0 (7) -71.5	(5-7) 15.0 (4-7) 2.3 (6-7) 5.7 (1-6) 3.3 (4-5) 258	125
C ₇ F ₁₁ Cl ₃ O		(1) -80.6 (2) -108.0 (3) -117.1 (4) -154.1 (7) -76.6	(1-3) 16.3 (1-2) 12.2 (4-7) 17.9 (2-4) 14.7 (2-3) 254	125
C ₇ F ₁₁ Cl ₃ O		(1) -81.0 (2) -117.9 (3) -91.3 (4) -161.4 (5) -115.2 (6) -120.2 (7) -77.5	(1-2) 15.8 (1-4) 10.4 (1-3) 2.5 (3-5) 34.3 (3-6) 19.4 (2-7) 3.6 (4-5) 3.6 (4-6) 4.8 (2-3) 255 (5-6) 300	125
C ₇ F ₁₂		(1)(3)(4) -63.0, -65.0, -72.4 (2)(5)(6) -71.3, -73.0, -105.8		146
C ₇ F ₁₂		(1) -69.3 (2) -65.9 (3) -115.3 (4) -95.3	(1-3) 9.8 (1-2) 3.0 (1-4) ~2 (2-4) 9.6 (2-3) 3.0 (3-4) ~6	146

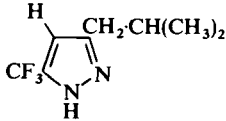
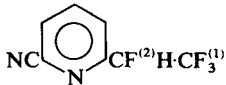
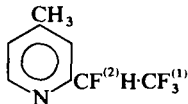
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₇ F ₁₂		(1)(3)(4) -62.0, -63.5, -70.5 (2)(5)(6) -71.6, -72.8, -103.6		146
C ₇ F ₁₂ H ₃ O ₂ P	[(CF ₃) ₂ C·OH] ₂ PCH ₃	-70.0		73
C ₇ F ₁₂ H ₄ N ₂ O	(CF ₃ ⁽¹⁾) ₂ C(OCH ₃)·NH·N=C(CF ₃ ⁽²⁾⁽³⁾) ₂	(1) -75.4 (2)(3) -67.3, -64.6		178
C ₇ F ₁₂ H ₅ O ₂ P	[(CF ₃) ₂ CH·O] ₂ PCH ₃	-76.5		124
C ₇ F ₁₂ H ₅ O ₃ P	[(CF ₃ ⁽¹⁾) ₂ CH ⁽²⁾ ·O] ₂ P(O)CH ₃	-76.8	(1-2) 5.6	124
C ₇ F ₁₂ O ₃	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·OCO·(CF ₂ ⁽³⁾⁽⁴⁾) ₂ ·CO·F ⁽⁵⁾	(1) -79.2 (2) -142.2 (3)(4) -118.7, -119.2 (5) +23.8	(1-2) 2.0	147
C ₇ F ₁₂ O ₄	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·OCO·CF ₂ ⁽³⁾ ·O·CF ₂ ⁽⁴⁾ ·CO·F ⁽⁵⁾	(1) -79.0 (2) -142.5 (3)(4) -76.7, -77.3 (5) +13.0	(1-2) 2.0	147
C ₇ F ₁₃ H		(1) -75.7 (2) -180.8 (3)(4) -62.0, -62.5		146(a)
C ₇ F ₁₃ ClO	CF ₃ ⁽¹⁾ ·CO·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₃ ·CF ₂ ⁽⁴⁾ Cl	(1) -76.2 (2) -120.6 (3) -122.4 (4) -69.4		97
C ₇ F ₁₃ ClO ₂	CF ₂ ⁽¹⁾ Cl·CF ₂ ⁽²⁾ ·CF ⁽³⁾ (CF ₃ ⁽⁴⁾)·O·CF ⁽⁵⁾ (CF ₃ ⁽⁶⁾)·CO·F ⁽⁷⁾	(1) -67.7 (2) -116.3, -117.2 (3) -135.3, -136.2 (4)(6) -76.2, -77.4, -80.6, -81.0 (5) -124.8, -125.3 (7) +23.1, +23.0		148

C ₇ F ₁₄	$\text{CF}_3^{(1)}\cdot\text{CF}_2^{(2)}\cdot\text{CF}_2^{(3)}\cdot\text{CF}_2^{(4)}\cdot\text{CF}_2^{(5)}\cdot\text{C}(\text{F}^{(6)})=\text{C}(\text{F}^{(7)})\text{F}^{(8)}$	(1) -83.7 (2) -129.5 (3) -125.9 (4) -126.9 (5) -121.1 (6) -193.0 (7) -91.5 (8) -108.1	(1-3) 10.1 (1-4) 2.5 (2-3) 2.5 (2-4) 14.7 (2-5) 3.4 (3-6) 3.7 (4-5) 14.7 (4-6) 6.2 (4-7) 28.2 (4-8) 3.8 (6-8) 117.5 (6-7) 40.6 (7-8) 53	150
C ₇ F ₁₄ H ₂	(CF ₃ ⁽¹⁾) ₂ CH ⁽⁶⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ H ⁽⁷⁾	(1) -61.6 (2) -110.9 (3) -128.2 (4) -121.5 (5) -136.2	(1-6)(2-6) 7.2 (4-7) 6.3 (5-7) 51.0	123
C ₇ F ₁₄ H ₃ O ₃ P		(1) -68.7 (2) -64.0	(P-2) 903	155
C ₇ F ₁₄ H ₅ O ₂ P	[(CF ₃ ⁽¹⁾) ₂ CH·O] ₂ PF ₂ ⁽²⁾ ·CH ₃	(1) -73.0 (2) -35.5	(P-2) 846	180
C ₇ F ₁₄ N ₂ O		(1) -70.3 (2) -122.7 (3)(4) -136.9, -137.8 (5) -93.9		179(a)
C ₇ F ₁₄ O	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ₂ ⁽³⁾ ·O·C(CF ₃ ⁽⁴⁾)=CF ₂ ⁽⁵⁾	(1) -74.5 (2) -187.8 (3) -78.7 (4) -66.7 (5) -88.9, -81.1		96
C ₇ F ₁₄ O	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CO·F ⁽⁶⁾	(1) -71.0 (2) -184.5 (3) -112.8 (4) -120.2 (5) -118.2 (6) +20.9		151
C ₇ F ₁₄ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ⁽⁵⁾ (CF ₃ ⁽⁶⁾)·CO·F ⁽⁷⁾	(1) -81.6 (2) -126.5 (3) -120.9 (4) -117.4 (5) -180.1 (6) -72.7 (7) +32.3	(1-3) 10.3 (1-4) 2.6	152
C ₇ F ₁₄ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ⁽⁴⁾ (CF ₂ ⁽⁵⁾ CF ₃ ⁽⁶⁾)·CO·F ⁽⁷⁾	(1) -80.7 (2) -123.8 (3) -115.8 (4) -178.9 (5A) -117.5 (5B) -120.9 (6) -80.5 (7) +33.7	(1-3) 11.0 (1-4) 2.0 (4-6) 13.9 (5A-5B) 290	152

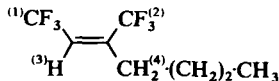
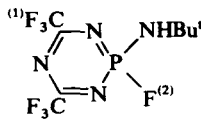
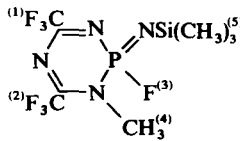
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₇ F ₁₄ O		(1) -124.9 (2) -80.3 (3) -121.7 (4) -126.2 (5) -121.2 (6) -81.6 (7) -182.3 (8) -72.6 (9) -77.8	(2-4) 12.4 (2-3) 3.6 (3-4) 270 (6-7) 15.5 (7-8) 11.3 (8-9) 142	152
C ₇ F ₁₄ O		(1) -128.8 (2) -81.7 (3) -121.9 (4) -113.6 (5) -121.4 (6) -81.5 (7) -184.2 (8) -87.2 (9) -71.7	(3-4) 272 (8-9) 142	152
C ₇ F ₁₄ O		(1) -121.6 (2) -81.3 (3) -122.3 (4) -127.6 (5) -120.8 (6) -126.1 (7) -73.1 (8) -182.3 (9) -75.0 (10) -78.3	(1-2) 12.5 (2-5) 2.6 (2-6) 2.6 (3-4) 296 (5-6) 268 (9-10) 144	152
C ₇ F ₁₄ O		(1) -126.5 (2) -81.2 (3) -123.7 (4) -116.0 (5) -121.9 (6) -74.8 (7) -186.8 (8) -81.9 (9) -73.5	(1-2) 11.8 (4-5) 266 (8-9) 140	152
C ₇ F ₁₄ O		(1) -92.1 (2) -79.8 (3) -125.1 (4) -125.1 (5) -117.4 (6) -125.5 (7) -81.2 (8) -183.4 (9) -73.5 (10) -82.6	(1-2) 130 (5-7) 10.5 (7-8) 2.0 (9-10) 144	152
C ₇ F ₁₄ O		(1) -80.9 (2) -93.4 (3) -119.2 (4) -130.2 (5) -125.0 (6) -140.1 (7) -117.9 (8) -81.4 (9) -184.4 (10) -72.3 (11) -83.1	(1-2) 160 (3-4) 290 (5-6) 274 (10-11) 170	152

$C_7F_{14}O$	$CF_3^{(1)} \cdot CO \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_2 \cdot CF_2^{(4)} \cdot CF_3^{(5)}$	(1) -76.6 (2) -118.4 (3) -122.8 (4) -127.4 (5) -82.6		97
$C_7F_{14}O_3$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CO \cdot OO \cdot C(CF_3^{(3)})_3$	(1) -83.8 (2) -120.0 (3) -69.7	(1-2) 8.3 (1-3) ~1.0	153
$C_7F_{15}NO_2$		(1)(2) -73.37, -76.70 (3) -64.58	(1-2) 4.2 (1-3) or (2-3) 9.18 (1-C2) 37.5 (2-C5) 36.1 $^1J(1-C) \ ^1J(2-C)$ 288.2, 288.3 $^1J(3-C)$ 270	181
$C_7F_{15}O_2P$		(1) -66.7 (2) -65.8 (3) -65.0	(1-2) 6.2 (P-2) 18.8 (P-3) 75.0	181
C_7F_{16}	$(CF_3^{(1)})_2 CF_2^{(2)} \cdot CF^{(3)}(CF_3^{(4)}) \cdot CF_2^{(5)} \cdot CF_3^{(6)}$	(1)(4) -72.1 (2)(3) -178.1, -182.1 (5) -115.5 (6) -82.7		129(a)
$C_7F_{16}N_2OS$	$(CF_3^{(1)})_2 C=N \cdot S(O)CF_3^{(2)} [=NCF^{(3)}(CF_3^{(4)})_2]$	(1)(4) -65.1, -74.9, -79.6 (2) -52.8 (3) -144.7	(2-3) 0.6 (3-4) 3.4	101
$C_7F_{16}O_3$	$CF_3^{(1)} \cdot O \cdot (CF_2^{(2)} CF_2^{(3)} O)_2 \cdot CF_2 \cdot CF_3^{(4)}$	(1) -59.2 (2) -93.7 (3) -91.7, -91.6 (4) -90.5	(1-2) 9	104
$C_7F_{17}NO$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot N(CF_3^{(4)}) O \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot CF_3^{(7)}$	(1) -83.5 (2) -128 (3) -100.7 (4) -66.7 (5) -91 (6) -131 (7) -83.5		133
$C_7F_{17}NO$	$(CF_3^{(1)(2)})_2 CF^{(3)} \cdot N(CF_3^{(4)}) O \cdot CF^{(5)}(CF_3^{(6)})_2$	(1) -73.2 (2) -76.7 (3) -154.3 (4) -64.5 (5) -136.0 (6) -77.9		133
$C_8F_3H_4ClO$			(F-C1) 33.0 (F-C2) 3.8 (F-C4) 1.3 (F-C7) 272.7	166

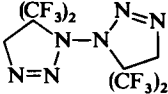
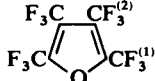
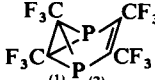
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_3H_4N$			(F-C1) 33.5 (F-C2) 4.0 (F-C4) 1.1 (F-C6) 3.7 (F-C7) 272.3	166
$C_8F_3H_4NO$			(F-C1) 30.8 (F-C4) 1.3 (F-C6) 5.2 (F-C7) 272.3	166
$C_8F_3H_5O_2$			(F-C1) 32.1 (F-C2) 3.8 (F-C4) 1.4 (F-C7) 272.5	166
$C_8F_3H_7$	$CF_3^{(1)} \cdot C \equiv C \cdot C \equiv C \cdot CH^{(2)}(CH_3)_2$	-51.1	(1-2) 1.2	188(a)
$C_8F_3H_7O$	$CF_3^{(1)} \cdot CH_2^{(2)} \cdot O \cdot C_6H_5$	-73.77	(1-2) 8.3	184(e)
$C_8F_3H_7O$	As above	-75.04	(1-2) 8.0	189(e)
$C_8F_3H_7S$	$CF_3^{(1)} \cdot CH_2^{(2)} \cdot SC_6H_5$	-66.5	(1-2) 9	184(e)
$C_8F_3H_7S$	As above	-68.2	(1-2) 10	189(e)
$C_8F_3H_8N$	$CF_3 \cdot CH_2 \cdot NH \cdot C_6H_5$	-73.62	(1-2) 8.2	189(e)
$C_8F_3H_{10}NO_2$			(1-2) 4	190
$C_8F_3H_{11}N_2$	$CF_3 \cdot C \equiv C \cdot CH_2 \cdot C(=NNH_2) \cdot CH(CH_3)_2$	-70.12		188(a)

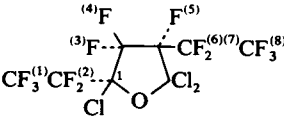
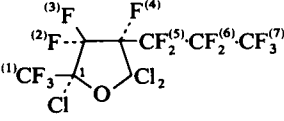
$C_8F_3H_{11}N_2$		-61.60			188(a)
$C_8F_3H_{13}$	$CF_3^{(1)} \cdot CH^{(2)} = CH \cdot CH_2 \cdot (CH_2)_3 \cdot CH_3$ (cis) (trans)	-60.6 -64.7		(1-2) ~6	117
$C_8F_3H_{13}$	$CF_3^{(1)} \cdot CH_2^{(2)} \cdot CH = CH \cdot (CH_2)_3 \cdot CH_3$ (cis) (trans)	-66.9 -67.3		(1-2) 10.7 (1-2) 10.7	117
$C_8F_3H_{15}$	$CF_3^{(1)} \cdot CH_2^{(2)} \cdot (CH_2)_5 \cdot CH_3$	-67.2		(1-2) 10.4	117
$C_8F_4H_4N_2$		(1) -78.7 (2) -204.9			169
$C_8F_4H_4N_2O$	N-oxide of above	(1) -77.5 (2) -208.5			169
$C_8F_4H_6$	$CF_3^{(1)} \cdot CF^{(2)}H^{(3)} \cdot C_6H_5$	(1) -79.7 (2) -197.0		(1-2) 12.8 (1-3) 6.5 (2-3) 44.5	176
$C_8F_4H_7N$		(1) -80.5 (2) -204.5			169
$C_8F_4H_7NO$	N-oxide of above	(1) -79.0 (2) -207.5			169
$C_8F_4H_8O$	$CF_3^{(1)} \cdot CF^{(2)} = \overline{C \cdot CO \cdot (CH_2)_3 \cdot CH_2}$ (two isomers present)	(1) -67.4 to -68.7 (2) -125.3, -129.9		(1-2) 6.3, 8.5	137
$C_8F_5H_5ClNOS$	$CF_3^{(1)} \cdot CF_2^{(2)(3)} \cdot SO \cdot NH \cdot C_6H_4Cl(p)$	(1) -79.1 (2) -120.9 (3) -123.5		(2-3) 249	191(a)
$C_8F_5H_6NOS$	$CF_3^{(1)} \cdot CF_2^{(2)(3)} \cdot SO \cdot NH \cdot C_6H_5$	(1) -79.5 (2) -121.2 (3) -124.0		(2-3) 251	191(a)
$C_8F_5H_9O$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot \overline{CH^{(3)} \cdot CO \cdot (CH_2)_3 \cdot CH_2}$	(1) -83.3 (2) -118.7		(2A-2B) 276.7 (2-3) 15.2	137

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_5H_{10}NO$	$CF_3^{(1)} \cdot CF^{(2)} = CF^{(3)} \cdot CO \cdot N(CH_2CH_3)_2$ (Z) (E)	(1) -67 (2) -154 (3) -127.3 (1) -67 (2) -163 (3) -140.5	(1-2) 12 (1-3) 5 (2-3) 7 (1-2) 11 (1-3) 21 (2-3) 142	170
$C_8F_5H_{11}O$		(1) -66.2 (2) -100.9 (3) -92.7	(1-2) 24.2 (1-3) 9.6 (2-3) 45.6	171(a)
$C_8F_6H_4$			(F-C1) 33.5 (F-C2) 4.0 (F-C4) 1.1 (F-C6) 3.7 (F-C7) 271.6	166 431
$C_8F_6H_4$			(7-C1) 33.8 (7-C2) 4.0 (7-C4) ~1 (7-C7) 272.1	431
$C_8F_6H_4Cl_2S$		-57.70, -54.80		173(e)
$C_8F_6H_5N$		-66.5	(1-2) 8.5	169
$C_8F_6H_5NO_2$		(1) -57.4 (2) -55.0	(1-2) 8.2	193
$C_8F_6H_6N_2$	$(CF_3^{(1)} \cdot CH_2^{(2)} \cdot CH \cdot CN)_2$	Isomer [a] -64.85 [b] -64.97	(1-2) 9.9 (1-2) 9.8	117
$C_8F_6H_6S$		-57.61, -54.90		173(e)

C ₈ F ₆ H ₈ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ⁽³⁾ =C(CH ₃)CO·CH ₂ ·CH ₃ (<i>cis</i> and <i>trans</i>)	(1) -83, -84 (2)(3) -116, -117	(1-2)(1-3) 2·3, 8·5 and 2·3, 7	137
C ₈ F ₆ H ₈ O ₄	(CF ₃ ⁽¹⁾ ·CH ₂ ⁽²⁾ ·CH·CO ₂ H) ₂	Isomer [a] -65·0 [b] -65·08	(1-2) 10·7 (1-2) 11·0	117
C ₈ F ₆ H ₉ BO ₅	[CF ₃ ⁽¹⁾ ·CO·O] ₂ BH·O 		(1-C1) 284 (1-C2) 40·9	194
C ₈ F ₆ H ₁₀		(1) -58·8 (2) -63·8	(1-3) 9 (1-4) 11 (2-4) 11	117
C ₈ F ₆ H ₁₀ N ₂ O ₂	[CF ₃ ·CO·N(CH ₃)CH ₂] ₂	-75·5		140
C ₈ F ₆ H ₁₀ O ₃	(CF ₃) ₂ C(OH)·O·CO·C(CH ₃) ₃	-88·66		122
C ₈ F ₆ H ₁₂	CF ₃ ⁽¹⁾ ·CH ₂ ⁽³⁾ ·CH ⁽⁴⁾ (CF ₃ ⁽²⁾)·CH ₂ ·(CH ₂) ₂ ·CH ₃	(1) -65·7 (2) -72·7	(1-3) 10·7 (1-4) 2·3 (2-3) 8·4	117
C ₈ F ₆ H ₁₃ OP	(CF ₃ ⁽¹⁾) ₂ CH ⁽²⁾ O·P(CH ₃)C(CH ₃) ₃	-73·8	(1-2) 5·7	72
C ₈ F ₇ H ₉ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CH(CH ₃)·CO·CH ₂ ·CH ₃	(1) -81·5 (2)(3) -114, -124	(1-3) 9·8	137
C ₈ F ₇ H ₁₀ N ₄ P		(1) -75·1 (2) -36·2	(P-1) 5·5 (P-2) 960	95
C ₈ F ₇ H ₁₂ N ₄ PSi	As above with NHBu ⁺ replaced by N(CH ₃ ⁽³⁾)Si(CH ₃) ₃	(1) -75·4 (2) -30·2	(P-1) 6·0 (P-2) 970 (2-3) 0·8	95
C ₈ F ₇ H ₁₂ N ₄ PSi		(1)(2) -68·6, -75·3 (3) -24·4	(P-1)(P-2) 3·0, 5·5 (P-3) 1002 (3-4) 0·8 (3-5) 1·8 (1-3) or (2-3) 0·4	95

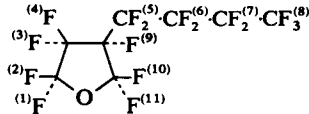
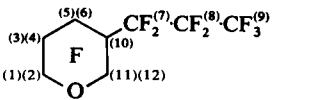
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_8F_7ClO		(1) -78.1 (2) -139.1 (3) -140.5	(1-2) 11.1	4
$C_8F_8H_{10}NO_2P$		(1) -67.6 (2) -57.0	(P-2) 883	141
C_8F_8O		(1) -74.7 (2) -153.4 (3) -161.9 (4) -156.0	(2-3) 16.0 (3-4) 21.0	429
$C_8F_{11}H_3O$		(1) -68.5 (2) -114.0 (3) -60.4	(1-2) 10.2 (1-3) 3.6 (2-3) 4	146(a)
$C_8F_{11}H_3O$		(1)(6) -61.8, -83.8 (2) -124.2 (3) -166.0 (4)(5) -113.4, -117.0	(4-5) 213	146(a)
$C_8F_{11}H_3O$	$(CF_3^{(1)})_2C=C(CF_3^{(2)})C(OCH_3)=CF_2^{(3)(4)}$	(1) -62.5 (2) -61.5 (3) -98.0 (4) -108.0	(3-4) 66	146(a)
$C_8F_{11}H_3O$		(1)(4) -64.0, -85.6 (2) -121.2 (3) -105.8 (5)(6) -107.1, -98.1	(2-3) 11 (5-6) 59 (1-3) or (3-4) 23	146(a)
$C_8F_{11}H_3O$		(1)(4) -59.2, -71.2 (2) -120.4 (3) -102.5 (5)(6) -98.6, -107.0	(2-3) 10 (5-6) 52 (2-1 or 4) 19 (3-1 or 4) 19 (3-5 or 6) 14 (6 or 5-1 or 4) 6	146(a)

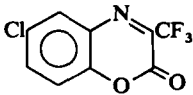
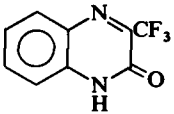
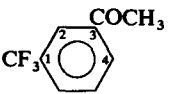
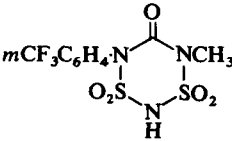
$C_8F_{12}H_4N_2O_2S$	$(CF_3^{(1)})_2C \cdot (SCH_2CO_2H) \cdot NH \cdot N=C(CF_3^{(2)(3)})_2$	(1) -69.5 (2)(3) -66.3, -64.0	(2-3) 6.0	178(a)
$C_8F_{12}H_4N_6$		-72.4, -71.4		178(a)
$C_8F_{12}H_4O$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CH(CF_3^{(4)}) \cdot C(OCH_3)=CF_2^{(5)(6)}$	(1)(4) -64.2, -82.8 (2)(3) -115.1, -128.8 (5)(6) -98.3, -112.2	(5-6) 64.5	146(a)
$C_8F_{12}H_6N_2O$	$(CF_3^{(1)})_2C(OCH_2CH_3)NH \cdot N=C(CF_3^{(2)(3)})_2$	(1) -76.0 (2)(3) -67.3, -64.7	(2-3) 5.7	178(a)
$C_8F_{12}H_6N_2O_2$	$(CF_3^{(1)})_2C(OCH_2CH_2OH)NH \cdot N=C(CF_3^{(2)(3)})_2$	(1) -75.7 (2)(3) -66.8, -64.7	(2-3) 5.7	178(a)
$C_8F_{12}H_6N_2S$	$(CF_3^{(1)})_2C(SCH_2CH_3)NH \cdot N=C(CF_3^{(2)(3)})_2$	(1) -72.2 (2)(3) -68.3, -65.9	(2-3) 6.0	178(a)
$C_8F_{12}O$		(1) -60.73 (2) -67.23 (3) -74.79	(1-2) 2.5 (1-3) 1.25 (2-3) 2.5	196
$C_8F_{12}O$		(1)(2) -57.70, -62.62		196
$C_8F_{12}O$	As above	(1)(2) -60, -65		197
$C_8F_{12}P_2$		(1) -74.1 (2) -69.90	(P-1) 22.6 (P-2) 5.2	198(c)
$C_8F_{12}Pd_2$	$(CF_3C \equiv CCF_3)_2Pd_2$	-56.2		88
$C_8F_{13}H$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot C \equiv CH^{(7)}$	(1) -81.6 (2) -126.6 (3) -122.9 (4) -122.9 (5) -121.6 (6) -99.9	(6-7) 5	144

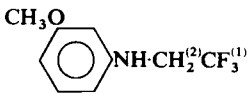
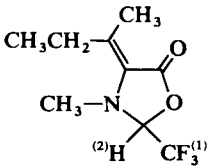
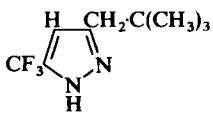
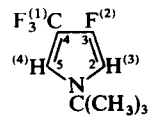
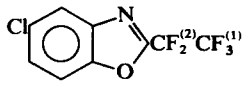
Molecular formula	Structure of compounds	Chemical shifts (pm)	Coupling constants (Hz)	Ref.
$C_8F_{13}HBr_2$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot CBr=CHBr$ (Z)	(1) -81.6 (2) -126.4 (3) -122.2 (4) -122.2 (5) -120.1 (6) -105.9		144
$C_8F_{13}H_2Br_3$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot CBr_2 \cdot CH_2Br$	(1) -81.6 (2) -126.4 (3) -122.0 (4) -122.0 (5) -114.5 (6) -102.5		144
$C_8F_{13}H_6O_2P$	$(CF_3^{(1)})_2CH^{(4)}O \cdot \overbrace{PF^{(3)}(CH_3^{(5)}) \cdot O \cdot C(CF_3^{(2)})_2 \cdot CH_2}$	(1) -72.9, -73.5 (2) -76.4, -77.0 (3) -29.6	(1-4) 5.6 (2-2') 8.0 (3-5) 11.4 (P-3) 832 $^1J(1-C), (2-C)$ 283, 289 $^2J(1-C)$ 35.0 (3-CH ₂) 45.3 (3-CH ₃) 26.5	199
$C_8F_{13}H_8O_2P$	$[(CF_3^{(1)})_2CH^{(3)} \cdot O]_2PF^{(2)}(CH_3^{(4)})_2$	(1) -76.0, -76.5 (2) -13.3	(1-3) 6.0 (2-4) 13.5 (P-2) 726	73
$C_8F_{13}Cl_3O$		(1) -77.0 (2) -119.9 (3) -116.8 (4) -89.2 (5) -161.7 (6) -115.7 (7) -120.3 (8) -81.1	(1-3) 3.0 (3-4) 254 (4-6) 36.7 (4-7) 19.8 (4-8) 9.8 (5-8) 16.6 (6-7) 300	125
$C_8F_{13}Cl_3O$	Groups on C1 above interchanged	(1) -78.1 (3) -106.9 (4) -116.7 (5) -153.1 (8) -80.9	(1-3) 4.3 (1-4) 4.3 (7-8) 16.4 (4-8) 12.2 (3-5) 16.2 (3-4) 256	125
$C_8F_{13}Cl_3O$		(1) -76.5 (2) -107.8 (3) -117.4 (4) -153.6 (5) -113.5 (6) -124.4 (7) -81.0	(1-2) 2.3 (1-3) 18.3 (2-3) 248 (2-4) 15.3 (4-6) 24.3 (5-7) 11.5	125
$C_8F_{13}Cl_3O$	Groups on C1 above interchanged	(1) -77.9 (2) -117.7 (3) -91.5 (4) -161.7 (5) -114.1 (6) -125.1 (7) -81.1	(1-2) 18.3 (1-3) 1.9 (1-4) 12.7 (2-3) 254 (4-7) 1.8 (5-7) 11.5	125

C ₈ F ₁₃ Cl ₃ O		(1) -80.7 (2) -125.9 (1-3) 10.2 (1-4) 2.4 (3) -117.5 (4) -114.4 (3-5) 9.0 (4-6) 27.4 (5) -108.2 (6) -122.1 (5-6) 242 (5-8) 13.8 (7) -111.4 (8) -120.0 (6-8) 8.0 (7-8) 226	125
C ₈ F ₁₄ H ₆ NO ₂ P		(1) -58.1 (2) -67.8 (P-1) 846 (1-3) 2.2 At -20 °C (2) -67.1, -68.2	156
C ₈ F ₁₄ O		[a] (1)(2) -63, -85 (3) -122 [b] (1)(2) -62, -83 (3) -112	197
C ₈ F ₁₄ O ₃	(<i>cis</i> and <i>trans</i> mixture) (CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·O·CO·(CF ₂ ⁽³⁾) ₃ ·CO·F ⁽⁴⁾	(1) -79.2 (2) -142.6 (1-2) 2.1 (3-4) 5.7 (3) -118.0, -123.4 (4) +24.0	147
C ₈ F ₁₅ HO	CF ₃ ⁽¹⁾ ·CO·(CF ₂ ⁽²⁻⁶⁾) ₅ ·CF ₂ ⁽⁷⁾ H	(1) -76.4 (2)-(7) -139.2 -131.1, -124.5, -122.4 -199.2 (ratio 1:1:1:2:1)	132
C ₈ F ₁₅ ClO	CF ₃ ⁽¹⁾ ·CO·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₄ ·CF ₂ ⁽⁴⁾ Cl	(1) -75.0 (2) -119.6 (3) -122.6 (4) -68.8	97
C ₈ F ₁₆		(1) -83.5 (2) -129.1 (3)(4) -124.9 (5) -126.3 (6) -120.7 (7) -192.0 (8) -91.7 (9) -108.5	150 93
C ₈ F ₁₆	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₄ ·CF ⁽³⁾ =CF ⁽⁴⁾ ·CF ₃ ⁽⁵⁾	(1) -83.5 (2) -121.4, -129.4 (3)(4) -160.5 (5) -71.5	93
C ₈ F ₁₆	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₃ ·CF ⁽³⁾ =CF ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CF ₃ ⁽⁶⁾	(1) -83.5 (2)(5) -125.8 (3)(4) -154.5 (6) -87.5	93
C ₈ F ₁₆ H ₆ NP	(CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾) ₂ PF ₂ ⁽⁴⁾ N(CH ₃) ₂	(1) -82.7 (2) -127.4 (3-4) 12.0 (P-3) 114 (3) -116.5 (4) -51.5 (P-4) 940	71

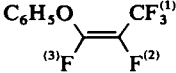
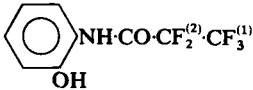
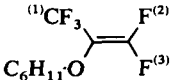
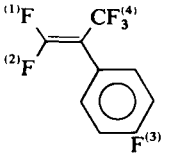
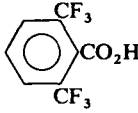
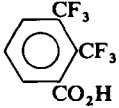
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₈ F ₁₆ O	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₃ ·CF ⁽³⁾ (CF ₃ ⁽⁴⁾)·O·CF ⁽⁵⁾ =CF ₂ ⁽⁶⁾	(1)(4) -77.1, -80.1 (2) -119.5, -120.0, -124.7 (3) -135.0 (5)(6) -114.2, -119.2, -133.9		148
C ₈ F ₁₆ O	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CF ₂ ⁽⁶⁾ ·CO·F ⁽⁷⁾	(1) -71.5 (2) -184.4 (3) -113.0 (4) -120.6 (5) -118.5 (6) -116.8 (7) +25.3		151
C ₈ F ₁₆ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CF ⁽⁶⁾ (CF ₃ ⁽⁷⁾)·CO·F ⁽⁸⁾	(1) -81.6 (2) -126.8 (3) -123.0 (4) -120.4 (5) -117.1 (B) -180.4 (7) -72.5 (8) +32.3	(1-3) 10.4 (1-4) 2.3	152
C ₈ F ₁₆ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ⁽⁵⁾ (CF ₂ ⁽⁶⁾ CF ₃ ⁽⁷⁾)·CO·F ⁽⁸⁾	(1) -81.2 (2) -126.8 (3) -120.6 (4) -115.7 (5) -179.0 (6) -117.4, -120.7 (7) -80.7 (8) +33.7	(1-3) 9.9 (5-7) 12.9 (6A-6B) 294	152
C ₈ F ₁₆ O	(CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾) ₂ CF ⁽⁴⁾ ·CO·F ⁽⁵⁾	(1) -81.4 (2) -124.6 (3) -116.0 (4) -179.0 (5) +33.5		152
C ₈ F ₁₆ O	CF ₃ ⁽¹⁾ ·CO·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₃ ·CF ₂ ⁽⁴⁾ ·CF ₃ ⁽⁵⁾	(1) -76.8 (2) -117.8 (3) -121.8 (4) -126.4 (5) -83.0		97
C ₈ F ₁₆ O		(2) -81.5 (3) -125.3 (7) -73.5 (8) -182.4 (9) -75.6 (10) -79.0	(3-4) 10.2 (1-2) 2.5 (7-8) 6.8 (7-9) 14.0 (8-9) 10.3 (9-10) 140	152
C ₈ F ₁₆ O	Groups (7) and (8) interchanged	(2) -81.5 (3) -125.2 (7) -74.7 (8) -186.6 (9) -83.5 (10) -71.7	(1-2) 2.6 (2-4) 10.2 (5-6) 274 (9-10) 148	152

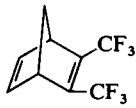
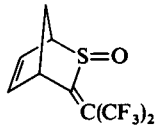
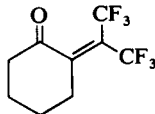
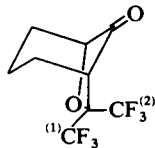
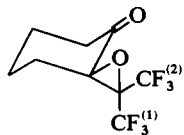
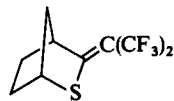
$C_8F_{16}O$		(1) -92.2 (2) -80.0 (3) -126.0 (4) -126.0 (5) -116.9 (6) -121.8 (7) -126.4 (8) -81.6 (9) -183.4 (10) -73.8 (11) -83.6	(1-2) 132 (5-8) 2.7 (6-9) 9.9 (10-11) 142	152
$C_8F_{16}O$		(1) -80.6 (2) -93.2 (3) -118.6 (4) -129.1 (5) -124.7 (6) -139.7 (7) -114.0 (8) -125.6 (9) -80.5 (10) -183.9 (11) -71.8 (12) -79.4	(1-2) 160 (3-4) 298 (5-6) 274 (7-9) 12.2 (11-12) 172	152
$C_8F_{16}O_2$	$CF_3^{(1)} \cdot CF^{(2)}(OCF_3^{(8)}) \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot CO \cdot F^{(7)}$	(1) -77.4 (2) -135.5 (3) -114.7 (4) -118.9 (5) -117.2 (6) -117.2 (7) +25.0 (8) -53.3		151
$C_8F_{16}O_2$	$(CF_3^{(1)})_2CF^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot O \cdot CF^{(5)}(CF_3^{(6)}) \cdot CO \cdot F^{(7)}$	(1) -73.2 (2) -188.2 (3) -119.0 (4) NA (5) -128.9 (6) -81.9 (7) +25.0		151
$C_8F_{16}O_3$	$CF_3^{(1)} \cdot CF^{(2)}(OCF_3^{(3)}) \cdot O \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot CF_2^{(7)} \cdot CO \cdot F^{(8)}$	(1) -80.0 (2) -136.2 (3) -54.3 (4) -113.8 (5) -116.2 (6) -119.6 (7) -118.6 (8) +25.5		200
$C_8F_{16}O_3$	$(CF_3^{(1)})_2CF^{(2)} \cdot O \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot O \cdot CF^{(5)}(CF_3^{(6)}) \cdot CO \cdot F^{(7)}$	(1) -72.5 (2) -189.6 (3) -118.6 (4) NA (5) -128.8 (6) -81.0 (7) +27.8		200
$C_8F_{17}H$	$CF_3^{(1)}(CF_2^{(2)})_4 \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)}H$	(1) -83.5 (2) -125.8 (3) -132 (4) -133.5 (5) -141.5		93
$C_8F_{18}Cl_4N_2S_2$	$(CF_3^{(1)})_2CF^{(2)} \cdot N=SCl_2 \cdot CF_2^{(3)} \cdot [(CF_3)_2CFN=]SCl_2 \cdot CF_2$	(1) -98.2 (2) -144.8 (3) -71.5	(1-2) 3.6 (2-3) 10.6	201

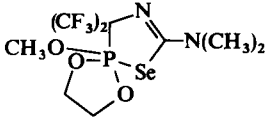
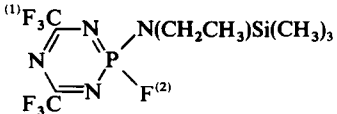
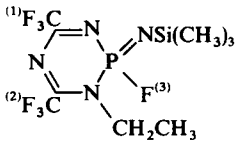
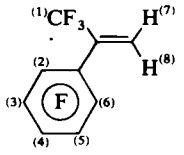
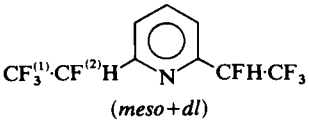
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_{18}N_2OS$	$[(CF_3^{(1)(2)})_2C=N]_2S(O)(CF_2^{(3)})_2$	(1)(2) -70.1, -71.6 (3) -63.2		107
$C_8F_{18}O_3$	$[(CF_3)_3CO]_2O$	-69.8		105
$C_8F_{18}O_3$	$CF_3^{(1)} \cdot CF_2 \cdot O \cdot (CF_2^{(2)} \cdot CF_2 \cdot O)_2 \cdot CF_2 \cdot CF_3$	(1) -90.4 (2) -91.6		104
$C_8F_{18}O_4$	$CF_3^{(1)} \cdot O \cdot (CF_2^{(2)} \cdot CF_2^{(3)} \cdot O)_3 \cdot CF_3$	(1) -59.1 (2) -93.8 (3) -91.7	(1-2) 9	104
$C_8F_{20}N_2S$	$(CF_3^{(1)})_2C=N \cdot C(CF_3^{(2)})_2N=SF_2^{(3)}(CF_3^{(4)})_2$	(1) -66.9 (2) -68.6 (3) +46.6 (4) -64.6	(2-3) 1.3 (3-4) 13.6	78
$C_8F_{22}N_2S_2$	$[-CF_2^{(1)} \cdot SF_2^{(2)} \cdot (NCF^{(3)}(CF_3^{(4)})_2)-]_2$	(1) -96.1 (2) +41.0 (3) -144.5 (4) -69.1	(1-2) 11.8 (2-3) 14.8 (3-4) 3.1	78
$C_9F_3H_3ClNO_2$		-71.6		204(a)
$C_9F_3H_4NO_2$	As above with Cl replaced by H	-71.6		204(a)
$C_9F_3H_5N_2O$		-71.0		204(a)
$C_9F_3H_7O$			(F-C1) 32.8 (F-C2) 3.8 (F-C4) 1.1 (F-C6) 3.7 (F-C7) 272.1	166
$C_9F_3H_8N_3O_5S_2$		-63.26		168

C ₉ F ₃ H ₉	CF ₃ ·C≡C·C≡C·C(CH ₃) ₃	–50·6				188(a)
C ₉ F ₃ H ₁₀ NO		–72·79		(1-2) 9·0		205
C ₉ F ₃ H ₁₂ NO ₂				(1-2) 4·2		190
C ₉ F ₃ H ₁₃ N ₂	CF ₃ ·C(=NNH ₂)·CH ₂ ·C≡C·C(CH ₃) ₃	–70·1				188(a)
C ₉ F ₃ H ₁₃ N ₂		–61·3				188(a)
C ₉ F ₄ H ₆ O	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ H ⁽³⁾ ·CO·C ₆ H ₅	(1) –75	(2) –203	(1-2) 12 (2-3) 48	(1-3) 6·5	170
C ₉ F ₄ H ₁₁ N		(1) –57·9	(2) –167·9	(1-2) 5·6 (1-4) 1·2 (2-3) 3·0 (2-4) 3·9 ¹ J(1-C) 265·4 ² J(1-C) 37·3 ¹ J(2-C) 245·1 ² J(2-C2) 26·2 ² J(2-C4) 12·05 ³ J(2-CF ₃) 2·7		193
C ₉ F ₅ H ₃ ClNO		(1) –84·7	(2) –116·9			204(a)

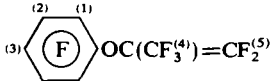
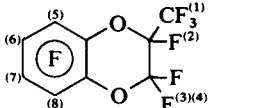
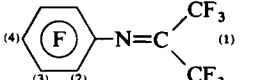
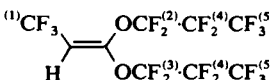
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_9F_5H_4BrO$		(1) -68.3 (3) -92.0	(2) -178.6			207(a)
$C_9F_5H_4BrO$	As above with groups (1) and (2) interchanged	(1) -68.3 (3) -107.7	(2) -185.0			207(a)
$C_9F_5H_4ClO$		(1) -68.5 (3) -92.7	(2) -179.9			207(a)
$C_9F_5H_4ClO$	As above with groups (1) and (2) interchanged	(1) -68.5 (3) -108.6	(2) -186.9			207(a)
$C_9F_5H_4NO$		(1) -85.0 (2) -116.6				204(a)
$C_9F_5H_4NO_3$		(1) -69.3 (3) -93.7	(2) -178.1			207(a)
$C_9F_5H_4NO_3$	As above with groups (1) and (2) interchanged	(1) -69.3 (3) -109.1	(2) -184.5			207(a)
$C_9F_5H_5$	$CF_3^{(1)} \cdot CF^{(2)} = CF^{(3)} \cdot C_6H_5$ (Z) (E)	(1) -65.8 (3) -108.4 (1) -68 (3) -145.8	(2) -155 (2) -169.5	(1-2) 14 (2-3) 8.4	(1-3) 8	170
$C_9F_5H_5ClNO_2$		(1) -84.0 (2) -123.2				204(a)

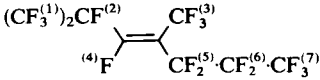
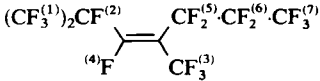
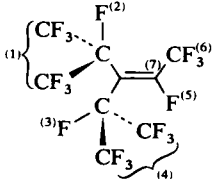
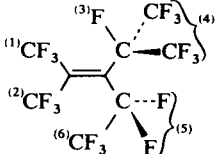
$C_9F_5H_5O$		(1) -68.7 (3) -91.7	(2) -180.6			207(a)
$C_9F_5H_5O$	As above with groups (1) and (2) interchanged	(1) -68.7 (3) -107.5	(2) -187.2			207(a)
$C_9F_5H_6BrO$	$CF_3^{(1)} \cdot C(OH)(C_6H_5)CF_2^{(2)} \cdot Br$	(1) -73.0 (3) -73.5	(2) -56.6 (2) -61.9	(1-2) 11.2 (1-2) 9.5		4 4
$C_9F_5H_6NO_2$		(1) -84.5	(2) -123.8			204(a)
$C_9F_5H_7O$	$CF_3^{(1)} \cdot CF_2^{(2)(3)} \cdot CH^{(4)}(OH) \cdot C_6H_5$	(1) -80.4 (3) -126.8	(2) -119.4	(2-3) 287 (3-4) 14	(2-4) 10	208
$C_9F_5H_{11}O$		(1) -66.4 (3) -92.4	(2) -100.6	(1-2) 24.0 (2-3) 45.6	(1-3) 10.4	171(a)
$C_9F_5H_{14}NO$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CH(CH_3) \cdot CO \cdot N(CH_2CH_3)_2$	(1) -83.4	(2) -119	(2-3) 4.9		137
$C_9F_6H_4$		(1) -75.11 (3) -111.82	(2) -73.25 (4) -60.12	(1-2) 11.03 (1-4) 24.27 (2-4) 11.03	(1-3) <0.5 (2-3) <0.5	326
$C_9F_6H_4O_2$		-60.98				209(a)
$C_9F_6H_4O_2$		-59.56, -56.00				209(a)

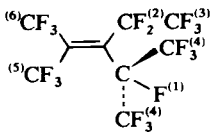
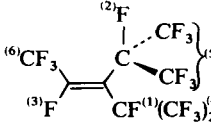
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9F_6H_5NOS$	$(CF_3)_2C=NS(O)C_6H_5$	-65.3		206(a)
$C_9F_6H_6$		-61.9	$^1J(F-C) 270$	210
$C_9F_6H_6OS$		-59.51, -57.70		173(e)
$C_9F_6H_7NO_2S$	$(CF_3^{(1)(2)})_2C(OH)NH \cdot S(O)C_6H_5$	(1) -80.25 (2) -81.25	(1-2) 9.5	206(a)
$C_9F_6H_8O$		-58.7, -59.6	(1-2) 8.4	213
$C_9F_6H_8O_2$		(1)(2) -84.3, -85.7	(1-2) 7.8	213
$C_9F_6H_8O_2$		-65.8, -69.4	(1-2) 9.0	213
$C_9F_6H_8S_2$		-58.71, -55.9		173(e)

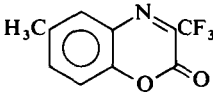
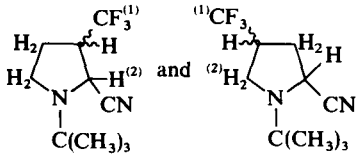
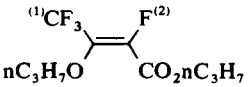
C ₉ F ₆ H ₁₃ N ₂ O ₃ PSe		-66.6 to -65.4				214(a)
C ₉ F ₆ H ₁₄	CF ₃ ⁽¹⁾ ·CH ₂ ⁽³⁾ ·CH(CF ₃ ⁽²⁾)·(CH ₂) ₄ ·CH ₃	(1) -65.3	(2) -72.2	(1-3) 9.6		117
C ₉ F ₇ H ₆ FeIN ₂ O ₂	CF ₃ ·CF ₂ ·CF ₂ ·Fe(CO) ₂ (CH ₃ CN) ₂ I	(1) -81.5 (3) -79.3	(2) -117.7	(1-3) 11.0		215
C ₉ F ₇ H ₆ NO ₂	(CF ₃ ⁽¹⁾) ₂ CH ⁽³⁾ ·CF ⁽²⁾ =C(CN)CO ₂ CH ₂ CH ₃	(1) -63.9	(2) 73.1	(1-2) 8.0 (2-3) 29.2	(1-3) 8.0	172(a)
C ₉ F ₇ H ₆ OP	(CF ₃ ⁽¹⁾) ₂ CH ⁽³⁾ O·PF ⁽²⁾ C ₆ H ₅	(1) -74.8	(2) -102.8	(1-3) 6.0 (P-2) 1116	(P-1) 6.0	72
C ₉ F ₇ H ₁₄ N ₄ PSi		(1) -75.1	(2) -33.4	(P-1) 5.5 (P-2) 975	(P-2) 975	95
C ₉ F ₇ H ₁₄ N ₄ PSi		(1)(2) -67.4, -75.4 (3) -19.5		(P-1)(P-2) 3.3, 5.5 (P-3) 1012		95
C ₉ F ₈ H ₂		(1) -68.11 (3) -163.02	(2) -139.88 (4) -153.65	(1-2) +6.76 (1-4) 0.0 (1-8) 1.415 (2-4) +2.54 (2-6) -4.02 (2-8) 0.0 (3-5) -1.26 (4-8) 0.0	(1-3) 0.0 (1-7) 1.176 (2-3) -21.44 (2-5) +8.65 (2-7) 0.0 (3-4) -19.54 (4-7) 0.0	216
C ₉ F ₈ H ₅ N	 (meso+dl)	{(1) -80.5 (1) -80.7	{(2) -203.9 (2) -204.2			169

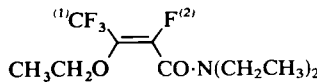
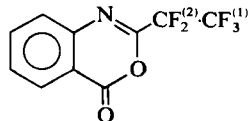
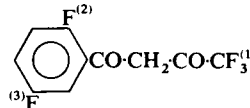
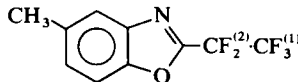
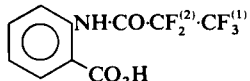
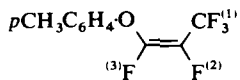
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9F_8H_5N$	<i>N</i> -oxide of above	$\begin{cases} (1) -80.5 & (2) -205.5 \\ (1) -80.6 & (2) -207.1 \end{cases}$		169
$C_9F_8H_6$		$\begin{matrix} (1) -73.8 & (2) -170.0 \\ (3) -168.8 & (4) -74.8 \end{matrix}$	$\begin{matrix} (1-2) 10.8 & (1-3) 6.8 \\ (3-4) 7.6 & (2-4) 16.4 \\ (1-4) 2.6 \end{matrix}$	210
$C_9F_8H_6$		$\begin{matrix} (1) -71.8 & (2) -162.0 \end{matrix}$		210
$C_9F_8H_6$		$\begin{matrix} (1) -73.3 & (2) -164.2 \end{matrix}$		210
$C_9F_9H_5O$		$-84.7, -75.6, -72.9$		217(a)
$C_9F_9H_6OP$	$(CF_3^{(1)})_2CHO \cdot PF_3^{(2)} \cdot C_6H_5$	$\begin{matrix} (1) -75.0 & (2) -55.0 \end{matrix}$	(P-2) 945	180
$C_9F_9H_9O_3$		$\begin{matrix} (1) -86.2 & (2) -86.4 \\ (3) -88.2 \end{matrix}$		217(a)
$C_9F_9H_{16}N_2P$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot PF_2^{(4)} N(CH_2CH_3)_2 NH \cdot CH_2CH_3$	$\begin{matrix} (1) -80.9 & (2) -125.9 \\ (3) -118.9 & (4) -66.5, -55.7 \end{matrix}$	$\begin{matrix} (1-3) 11.3 & (P-3) 105 \\ (P-4) 721, 715 \end{matrix}$	71

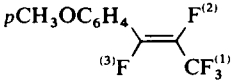
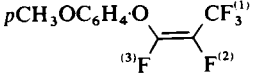
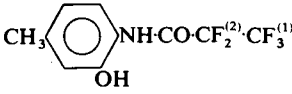
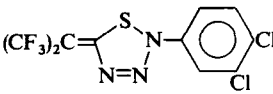
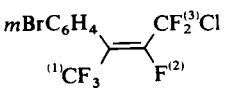
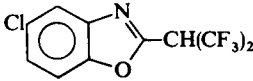
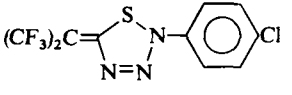
$C_9F_{10}O$		(1)(2) -159.3, -165.0 (3) -162.6 (4) -67.5 (5) -85.8, -95.4		96
$C_9F_{10}O_2$		(1) -78.3 (2) -131.4 (3) -81.0 (4) -92.0 (5-8) -158.7, -159.5, -160.5, -160.5	(3-4) 150	218
$C_9F_{11}N$		(1) -71.6 (2) -154.5 (3) -165.5 (4) -161.5		219(a)
$C_9F_{12}H_8N_2O$	$(CF_3^{(1)})_2C(OCH_2CH_2CH_3)NH \cdot N=C(CF_3^{(2)(3)})_2$	(1) -76.0 (2)(3) -67.5, -65.0		178
$C_9F_{12}H_8N_2O$	As above with nC_3H_7 replaced by iC_3H_7	(1) -76.0 (2)(3) -67.1, -64.5	(2-3) 5.7	178
$C_9F_{12}H_9O_4PSi$	$(CH_3)_3SiO \cdot \overline{P(O) \cdot O \cdot C(CF_3)_2C(CF_3)_2 \cdot O}$	-68.8		155
$C_9F_{12}H_{11}O_4P$	$(CF_3^{(1)}CH_2^{(2)}O)_4PCH_3$		(1-2) 7.9	220
$C_9F_{13}H_7O$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot C(OH)(CH_3)_2$	(1) -80.4 (2) -125.0 (3) -121.2 (4) -121.2 (5) -120.4 (6) -117.9		208
$C_9F_{15}H_3O_3$	$(CF_3^{(1)})_2CF^{(2)} \cdot CF_2^{(3)}O \cdot C(CF_3^{(4)})_2 \cdot CO_2CH_3$	(1)(4) -73.4, -75.1 (2) -186.7 (3) -75.1		96
$C_9F_{16}O_4$	$[(CF_3^{(1)})_2CF^{(2)} \cdot O \cdot CO]_2CF_2^{(3)}$	(1) -79.2 (2) -142.2 (3) -111.4		147
$C_9F_{16}O_4$	$(CF_3^{(1)})_2CF^{(2)} \cdot O \cdot CO \cdot (CF_2^{(3)})_2 \cdot O \cdot (CF_2^{(4)})_2 \cdot CO \cdot F^{(5)}$	(1) -79.2 (2) -142.3 (3)(4) -85.6, -121.3 (5) +23.5	(1-2) 1.9 (4-5) 8.3	147
$C_9F_{17}HO_2$		(1) -59.6 (2)(3) -84.0, -86.3 (4) -130.0 (5) -81.6		86

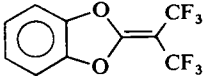
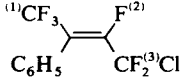
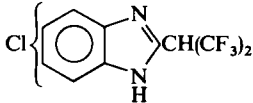
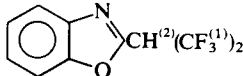
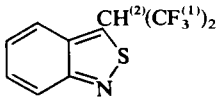
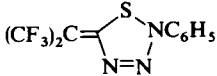
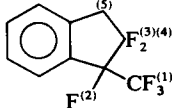
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9F_{17}ClO_3$	$CF_3^{(1)} \cdot CO \cdot CF_2^{(2)} \cdot O \cdot CF^{(3)}(CF_3^{(4)}) \cdot CF_2^{(5)} \cdot O \cdot CF^{(6)}(CF_3^{(7)}) \cdot CF_2^{(8)}Cl$	(1)(2) -75.3 (3) -145.0 (4)(5) -80.6 (6) -141.4 (7) -79.2 (8) -69.0	(1, 2-) 4 (3-) 20 (6-) 19	143
C_9F_{18}		(1) -76.5 (2) -183.6 (3) -55.1 (4) -89.7 (5) -108.8 (6) -127.6 (7) -83.5	(2-3) 42.5 (3-5) 14.5 (3-6) 8.1 (4-5) 37.5 (5-6) 8.3 (5-7) 9.0 (3-6) 18.1 (1-C1) 289 (2-C1) 26 (2-C2) 257 (1-C2) 35 (2-C3) 30 (4-C4) 290 (3-C3) 274 (5-C5) 265 (6-C5) 36 (6-C7) 34 (7-C7) 286	149(a)
C_9F_{18}		(1) -76.5 (2) -183.6 (3) -59.4 (4) -87.5 (5) -103.6 (6) -124.6 (7) -83.5	(2-5) 67.7 (3-4) 28.2 (3-5) 14.1 (3-6) 9.0 (2-6) 33.1 (5-6) 8.3 (5-7) 9.8	149(a)
C_9F_{18}		(1) -75.2 (2) -169.3 (3) -172.5 (4) -75.2 (5) -80.4 (6) -65.8	(2-6) 45.1 (1-3) 26.3 (4-5) 26.3 (1-C1) 289 (2-Cl) 27 or 26 (4-C4) 289 (3-C4) 26 or 27 (5-C5) 284 (5-C6) 39 (6-C6) 277 (6-C7) 45	149(a)
C_9F_{18}		(1) -59.0 (2) -61.2 (3) -161.0 (4) -72.4 (5) -89.9 (6) -73.3	(1-2) 9.6 (1-3) 44.2 (1-C1) 277 (2-C2) 277 (5-C6) 28 (6-C6) 287	149(a)

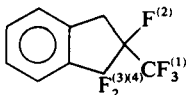
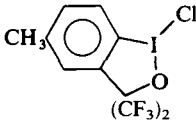
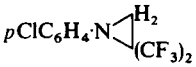
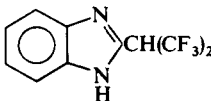
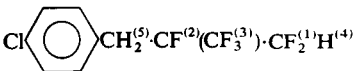
C_9F_{18}	As above	(1) -56.1 (3) -159.8 (5) -87.7	(2) -59.1 (4) -70.3 (6) -71.2	(1-2) 12 (1-4) ~2 (2-6) 13	(1-3) 52 (2-5) 13 (3-4) ~2	221
C_9F_{18}		(1) -156.1 (3) -74.1	(2) -91.5 (6) -56.4	(1-2) 40 (1-4) ~2 (2-6) 20 (3-6) 8	(1-3) 24 (2-3) 0.7 (3-4) 0.7 (5-6) 11	221
C_9F_{18}		(1) -180.8 (3) -78.9 (5) -72.9	(2) -180.2 (4) -73.2 (6) -63.5			221
$C_9F_{18}H_2$	$(CF_3)_3C \cdot CH_2 \cdot C(CF_3)_3$	-66.8				222(a)
$C_9F_{18}H_3Br_2O_3P$	$[(CF_3)_2CHO]_3PBr_2$	-73.0				157
$C_9F_{18}H_3Cl_2O_3P$	$[(CF_3)_2CHO]_3PCl_2$	-77.3				157
$C_9F_{18}H_3O_3P$	$[(CF_3^{(1)})_2CH^{(2)}O]_3P$	-76.0				157
$C_9F_{18}H_3O_3P$	As above	-76.0		(1-2) 5.3		72
$C_9F_{18}H_3O_3PS$	$[(CF_3)_2CHO]_3P(S)$	-76.4				157
$C_9F_{18}H_3O_4P$	$[(CF_3)_2CHO]_3P(O)$	-75.5				157
$C_9F_{18}O_2$	$CF_3^{(1)} \cdot (CF_2^{(2)})_3 \cdot CF^{(3)}(CF_3^{(4)}) \cdot O \cdot CF^{(5)}(CF_3^{(6)}) \cdot CO \cdot F^{(7)}$	(1)(4)(6) -76.1, -77.3, -80.1, -81.1 (2) -117.7, -118.5, -124.5, -128.8 (3) -139.4, -140.8 (5) -123.0, -125.0 (7) +29.2, +27.1				148
$C_9F_{18}O_3$	$(CF_3^{(1)})_2CF^{(2)} \cdot O \cdot CF_2^{(3)} \cdot CF^{(4)}(CF_3^{(5)}) \cdot O \cdot CF_2^{(6)} \cdot CO \cdot CF_3^{(7)}$	(1) -81.5 (3) -81.3 (5) -80.7	(2) -145.0 (4) -146.0 (6)(7) -75.3	(2-3) 21 (6, 7-) 5	(4-6) 20	143
$C_9F_{18}O_3$	$[(CF_3^{(1)})_2CF^{(2)} \cdot O \cdot CF_2^{(3)}]_2CO$	(1) -81.4 (3) -75.0	(2) -145.3	(1-2) 2.1 (2-3) 21.5	(1-3) 5.5	143

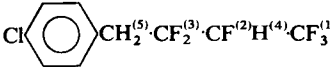
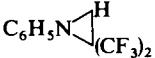
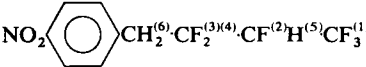
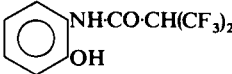
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9F_{19}H$	$(CF_3^{(1)})_3C \cdot CF_2^{(2)}H \cdot C(CF_3)_3$	(1) -64.7	(1-2) 15.9	222(a)
C_9F_{20}	$(CF_3^{(1)})_3C \cdot CF_2^{(2)} \cdot C(CF_3)_3$	(1) -61.4	(1-2) 14.9	222(a)
$C_9F_{20}H_3O_3P$	$[(CF_3^{(1)})_2CHO]_3PF_2^{(2)}$	(1) -79.0 (2) -57.0	(P-2) 790	157
$C_9F_{20}O_4$	$CF_3^{(1)} \cdot O \cdot (CF_2^{(2)} \cdot CF_2^{(3)} \cdot O)_3 \cdot CF_2 \cdot CF_3^{(4)}$	(1) -59.0 (2) -93.7 (3) -91.6 (4) -90.4	(1-2) 9	104
$C_{10}F_3H_6NO_2$		-71.5		204(a)
$C_{10}F_3H_{15}N_2$		$\begin{cases} -72.3 \\ -71.2 \\ -71.0 \end{cases}$	(1-2) 7.5 7.6 9.9	193
$C_{10}F_4H_{10}N_2$		(1) -59.2 (2) -148.3	(1-2) 6.3 (2-4) 5.0	193
$C_{10}F_4H_{10}N_2$	As above but with CF_3 and F interchanged	(1) -58.0 (2) -163.0	(1-2) 8.7 (2-4) 2.6	193
$C_{10}F_4H_{14}O_3$		(1) -66.7 (2) -141.7	(1-2) 20	170
$C_{10}F_4H_{14}O_3$	$CF_3 \cdot CF = C(OnC_3H_7)CO_2nC_3H_7$ (two isomers)	[a] (1) -64.5 (2) -147 [b] (1) -65 (2) -142	(1-2) 11 (1-2) 10	170

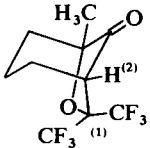
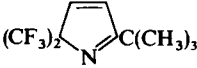
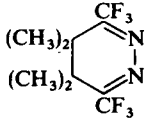
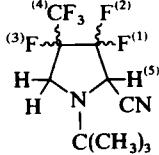
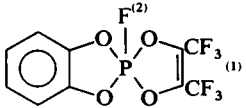
C ₁₀ F ₄ H ₁₅ NO		(1) -66.7	(2) -131	(1-2) 22	170
C ₁₀ F ₄ H ₁₅ NO	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ =C(OCH ₂ CH ₃)CO·N(CH ₂ CH ₃) ₂	(1) -68.3	(2) -126.5	(1-2) 8	170
C ₁₀ F ₄ H ₁₆ O ₃	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ H ⁽³⁾ ·CH ⁽⁴⁾ (OnC ₃ H ₇)CO ₂ nC ₃ H ₇ (two isomers)	[a] (1) -75.7	(2) -212	(1-2) 10 (2-3) 43	(1-3) 6 (2-4) 26
		[b] (1) -63.5	(2) -206	(1-2) 11 (2-3) 45	(1-3) 11 (2-4) 11
C ₁₀ F ₄ H ₁₆ O ₃	CF ₃ ⁽¹⁾ ·CH ⁽³⁾ (OnC ₃ H ₇)·CF ⁽²⁾ H ⁽⁴⁾ ·CO ₂ nC ₃ H ₇	(1) -73.5	(2) -209.3	(1-2) 10 (2-3) 26	(1-3) 6 (2-4) 49
C ₁₀ F ₅ H ₄ NO ₂		(1) -84.1	(2) -121.3		204(a)
C ₁₀ F ₅ H ₅ O ₂		(1) -80.0 (2)(3) -117.5, -118.7		(2-3) 0.8	224(a)
C ₁₀ F ₅ H ₆ NO		(1) -84.9	(2) -116.9		204(a)
C ₁₀ F ₅ H ₆ NO ₃		(1) -84.3	(2) -124.3		204(a)
C ₁₀ F ₅ H ₇ O		(1) -68.5 (3) -91.6	(2) -181.0		207(a)
C ₁₀ F ₅ H ₇ O	As above with groups (1) and (2) interchanged	(1) -68.5 (3) -107.5	(2) -188.1		207(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_5H_7O$		(1) -68.2 (2) -174.3 (3) -147.9	(1-2) 10.5 (1-3) 22 (2-3) 130	100
$C_{10}F_5H_7O_2$		(1) -69.0 (2) -182.3 (3) -92.3		207(a)
$C_{10}F_5H_7O_2$	As above with groups (1) and (2) interchanged	(1) -69.0 (2) -189.3 (3) -108.1		207(a)
$C_{10}F_5H_8NO_2$		(1) -84.0 (2) -123.2		204(a)
$C_{10}F_6H_3Cl_2N_3S$		-57.15		173(e)
$C_{10}F_6H_4BrCl$		(1) -60.6 (2) -109.8 (3) -55.3	(1-2) 24.1 (1-3) 1.3 (2-3) 11.4	4
$C_{10}F_6H_4BrCl$	As above with groups (2) and (3) interchanged	(1) -55.6 (2) -106.0 (3) -57.0	(1-2) 11.3 (1-3) 15.3 (2-3) 11.3	4
$C_{10}F_6H_4ClNO$		-65.8		226(a)
$C_{10}F_6H_4ClNS$	As above with O replaced by S	-66.2		226(a)
$C_{10}F_6H_4ClN_3S$		-57.15, -56.83		173(e)

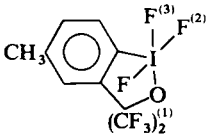
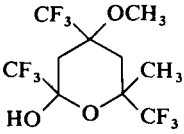
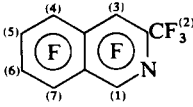
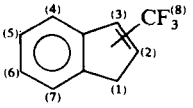
C ₁₀ F ₆ H ₄ O ₂		-57.1				226(a)
C ₁₀ F ₆ H ₅ Cl		(1) -60.7 (3) -55.1	(2) -111.4	(1-2) 24.1 (2-3) 11.2	(1-3) 1.2	4
C ₁₀ F ₆ H ₅ Cl	As above with groups (2) and (3) interchanged	(1) -55.7 (3) -56.9	(2) -107.8	(1-2) 10.9 (2-3) 11.1	(1-3) 15.3	4
C ₁₀ F ₆ H ₅ CIN ₂		-66.7				226(a)
C ₁₀ F ₆ H ₅ NO		-66.5				226(a)
C ₁₀ F ₆ H ₅ NS	As above with O replaced by S	-66.5				226(a)
C ₁₀ F ₆ H ₅ NS	As above	-65.1		(1-2) 6		173(e)
C ₁₀ F ₆ H ₅ NS		-62.32		(1-2) 8		173(e)
C ₁₀ F ₆ H ₅ N ₃ S		-57.20, -56.32				173(e)
C ₁₀ F ₆ H ₆		(1) -77.8 (3) -111.8	(2) -172.5 (4) -114.9	(3-4) 250	(3,4-5) 18.0	176

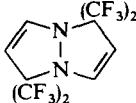
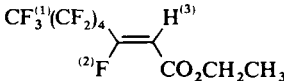
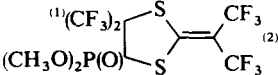
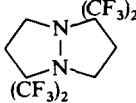
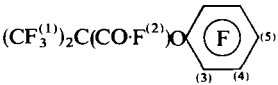
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
C ₁₀ F ₆ H ₆		(1) -78.0 (3) -100.6	(2) -180.5 (4) -111.4	(3-4) 280		176
C ₁₀ F ₆ H ₆ ClO		-76.13				227
C ₁₀ F ₆ H ₆ ClN		-67.5				228(a)
C ₁₀ F ₆ H ₆ N ₂		-67.5				226(a)
C ₁₀ F ₆ H ₆ O ₂ S	(CF ₃) ₂ C(OH)S·CO·C ₆ H ₅	-87.15				122
C ₁₀ F ₆ H ₇ Br	<i>o</i> BrC ₆ H ₄ ·CH ₂ ⁽⁶⁾ ·CF ₂ ⁽³⁾⁽⁴⁾ ·CF ⁽²⁾ H ⁽⁵⁾ ·CF ₃ ⁽¹⁾	(1) -74.8 (3) -107.0	(2) -208.8 (4) -109.5	(1-2) 12.0 (2-3,4) 12.0 (2-5) 44.0 (4-5) 6.0 (4-6) 18.0	(1-3,4) 12.0 (1-5) 6.0 (3-5) 12.0 (3-6) 20.0 (2-6) 2.0	176
C ₁₀ F ₆ H ₇ Cl	As above with Br replaced by Cl	(1) -77.0 (3) -108.0	(2) -216.0 (4) -113.5	(1-2) 12.0 (3-4) 270 (2-5) 43.5 (4-5) 6.0 (4-6) 19.0	(1-3,4) 12.0 (1-5) 6.0 (3-5) 12.0 (3-6) 21.5 (2-6) 2.0	176
C ₁₀ F ₆ H ₇ Cl		(1) -134.6 (3) -77.4	(2) -184.7	(1-3) 8.0 (2-3) 8.0 (3-5) 23.0	(1-4) 52.0 (2-4) 3.8	176

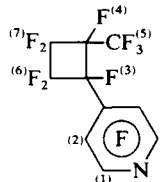
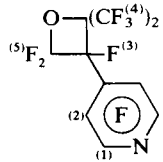
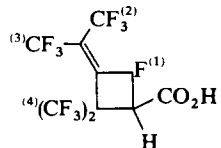
$C_{10}F_6H_7Cl$		(1) -74.9 (3) -111.0	(2) -214.0	(1-2) 12.0 (1-4) 6.0 (2-4) 43.7 (3-4) 12.0, 6.0 (3-5) 17.0, 15.5	(1-3) 12.0 (2-3) 12.0 (2-5) 3.0	176
$C_{10}F_6H_7N$		-67.5				228(a)
$C_{10}F_6H_7NOS$	$(CF_3)_2C=N \cdot S(O) \cdot C_6H_4 \cdot CH_3p$	-65.3				206(a)
$C_{10}F_6H_7NO_2$		(1) -74.8 (3) -108.3	(2) -213.5 (4) -111.7	(1-2) 11.5 (1-4) 11.5 (2-3) 11.5 (2-5) 45.5 (3-4) 278 (3-6) 13.0 (4-5) 13.0	(1-3) 11.5 (1-5) 6.0 (2-4) 11.5 (2-6) 3.0 (3-5) 6.0 (4-6) 14.0	176
$C_{10}F_6H_7NO_2$		-64.5				226(a)
$C_{10}F_6H_8$	$CF_3^{(1)} \cdot CF^{(2)}(CH_3^{(4)}) \cdot CF_2^{(3)} \cdot C_6H_5$	(1) -76.9 (3) -108.1	(2) -172.1	(1-2) 7.0 (2-3) 5.0	(1-3) 11.0 (2-4) 22.5	176
$C_{10}F_6H_8$	$CF_3^{(1)} \cdot CF^{(2)}H^{(4)} \cdot CF_2^{(3)} \cdot CH_2^{(5)} \cdot C_6H_5$	(1) -75.5 (3) -111.3	(2) -214.7	(1-2) 10.5 (1-4) 5.5 (2-4) 43.5 (3-4) 10.5, 5.5 (3-5) 18.0, 16.5 (2-5) 3.0	(1-3) 10.5 (2-3) 10.5	176
$C_{10}F_6H_9NO_2S$	$(CF_3^{(1)(2)})_2C(OH)NH \cdot S(O) \cdot C_6H_4CF_3p$	(1) -80.25	(2) -81.20	(1-2) 9.5		206(a)
$C_{10}F_6H_9NO_2S$	$(CF_3^{(1)(2)})_2C(OCH_3^{(3)})NH \cdot S(O)C_6H_5$	(1) -75.7	(2) -75.0	(1-2) 10.0	(1-3) 1.5	206(a)

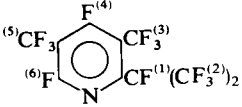
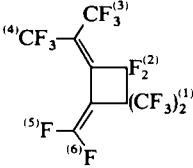
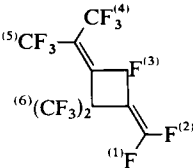
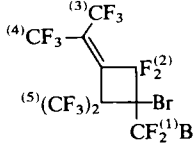
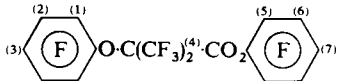
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_6H_{10}O_2$		-82.3, -83.3	(1-2) 7.8	213
$C_{10}F_6H_{11}N$		-69.7		229
$C_{10}F_6H_{12}N_2$		-66.7		87(a)
$C_{10}F_6H_{12}N_2$		(diastereoisomers) [a] (1)(2) -119.7, -110.3 (3) -178.0 (4) ~-77 [b] (1)(2) ~-113.5 (3) -175.0 (4) ~-77 [c] (1)(2) ~-112.5 (3) -177.0 (4) ~-77	(1-2) 243 (1-4) ~13.4 (1-5) ~13.4 (2-4) 3.5 (3-4) 7 (3-4) 7	193
$C_{10}F_6H_{12}O_4$	$(CF_3 \cdot CH_2 \cdot CH \cdot CO_2CH_3)_2$ (two isomers)	[a] -66.32 [b] -66.37		117
$C_{10}F_7H_4O_4P$		(1) -66.0 (2) -66.4	(P-2) 1004	141
$C_{10}F_7H_5O$	$CF_3^{(2)} \cdot C_6H_4 \cdot CO \cdot CF^{(3)(4)}H \cdot CF_3^{(1)}$	(1) -75 (2) -64 (3) -202	(1-3) 1.5 (1-4) 6.5 (3-4) 48	170
$C_{10}F_7H_5O_3$	$(CF_3^{(1)})_2C(OH)O \cdot CO \cdot C_6H_4F$	(1) -88.36		122

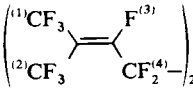
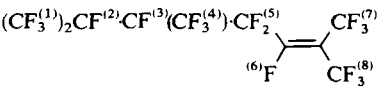
C ₁₀ F ₇ H ₁₈ N ₄ PSi ₂		(1) -75.3	(2) -15.4	(P-1) 5.7	(P-2) 973	95
C ₁₀ F ₈ H ₄	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ =CF ⁽³⁾ ·C ₆ H ₄ CF ₃ ⁽⁴⁾ <i>p</i> (Z) (E)	(1) -65.0 (3) -100.0 (1) -68.4 (3) -148.5	(2) -150.5 (4) -62.2 (2) -166.5 (4) -64.3	(1-2) 13 (2-3) 7.5 (1-2) 12 (2-3) 142	(1-3) 7.5 (1-3) 24	170
C ₁₀ F ₈ H ₅ O ₂ P		(1) -64.7	(2) -45.9	(P-2) 976		141
C ₁₀ F ₈ H ₆	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ H ⁽⁶⁾ ·CF ₂ ⁽³⁾⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·C ₆ H ₅	(1) -75.9 (3) -115.8 (5) -113.1	(2) -212.5 (4) -122.8	(1-6) 6.0 (2-3) 11.0 (3-4) 264	(2-6) 44.0 (2-4) 6.5	176
C ₁₀ F ₈ H ₁₀ NO ₂ P		(1) -67.4	(2) -56.6	(P-2) 876		141
C ₁₀ F ₉ H ₃ O ₂		(1) -63.78	(2) -60.33			209(a)
C ₁₀ F ₉ H ₃ O ₂		(1) -63.78	(2) -60.33			209(a)
C ₁₀ F ₉ H ₃ O ₂		-59.23, -55.83				209(a)

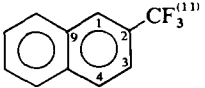
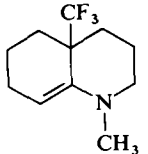
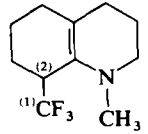
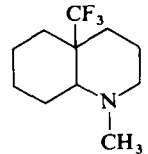
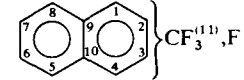
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_9H_6IO$		(1) -74.30 (2) -13.20 (3) -41.57	(1-2) 8 (2-3) 116	227
$C_{10}F_9H_6NOS$	$CF_3^{(1)} \cdot (CF_2^{(2)})_2 \cdot CF_2^{(3)(4)} \cdot SO \cdot NH \cdot C_6H_5$	(1) -81.2 (2) -121.0, -125.7 (3) -117.0 (4) -120.4	(3-4) 248	191
$C_{10}F_9H_{11}O_3$		-80.7, -85.7, -86.0		217(a)
$C_{10}F_9N$		(1) -61.2 (2) -66.3 (3) -132.6 (4) -143.3 (5)(6) -148.5, -145.7 (7) -134.4	(1-3) 30 (1-7) 66 (3-4) 60	231
$C_{10}F_{10}H_{11}NO$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)(5)} \cdot CF^{(6)}H^{(7)} \cdot CO \cdot N(CH_2CH_3)_2$	(1) -79 (2)(3) -124, -121 (4) -117 (5) -121 (6) -194	(1-) 11.2 (4-5) 280 (6-7) 46	232
$C_{10}F_{11}$	 CF ₃ at C2 or C3	(1) -117.3 (2) or (3) -112.9 (4)(7) -145.5, -147.7 (5)(6) -154.3 (8) -67.7		233
$C_{10}F_{12}HNO$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CO \cdot NH$ 	(1) -82.2 (2) -121.2 (3) -128.3 (4) -146.9 (5) -157.1 (6) -164.9		219(a)

$C_{10}F_{12}H_4N_2$		-73.45		234(a)
$C_{10}F_{12}H_6O$	$CF_3^{(1)}(CF_2)_4 \cdot CF=C(CH_2CH_3)CHO \quad (Z + E)$	(1) -81.0		137
$C_{10}F_{12}H_6O_2$		(1) -81.0	(2-3) 30	137
$C_{10}F_{12}H_7O_3PS_2$		(1) -65.73, -63.52 (2) -57.75		173(e)
$C_{10}F_{12}H_8N_2$		-72.90		234(a)
$C_{10}F_{12}H_8N_2O$	$(CF_3^{(1)})_2C(N \cdot CH_2 \cdot CH_2 \cdot O \cdot CH_2 \cdot CH_2) \cdot N=C(CF_3^{(2)(3)})_2$	(1) -68.3 (2)(3) -66.1, -63.6	(2-3) 5.8	178
$C_{10}F_{12}H_{10}N_2S$	$(CF_3^{(1)})_2 \cdot C[S(CH_2)_3CH_3]NH \cdot N=C(CF_3^{(2)(3)})_2$	(1) -70.8 (2)(3) -66.8, -64.3	(2-3) 6.0	178
$C_{10}F_{12}O_2$		(1) -71.6 (2) +39.0 (3)(4) -151.0, -163.2 (5) -157.4		96
$C_{10}F_{13}H_7O$	$CF_3^{(1)} \cdot (CF_2^{(2)})_5 \cdot C(CH_3)_2CHO$	(1) -80.1 (2) -117, -119, -122.8, -126.6		137
$C_{10}F_{13}H_7O$	$CF_3^{(1)} \cdot (CF_2)_5 \cdot CH(CH_2CH_3) \cdot CHO$	(1) -80.5		137
$C_{10}F_{13}H_9O$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot C(OH)(CH_3)CH_2CH_3$	(1) -80.8 (2) -124.1 (3)(4) -119.5, -120.4 (5) -118.3 (6) -116.6		208

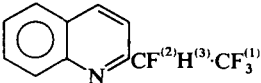
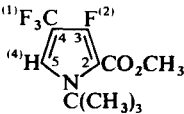
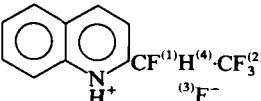
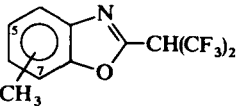
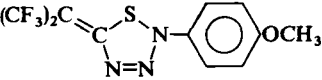
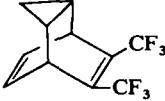
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_{13}H_{10}NOS$	$CF_3^{(1)} \cdot (CF_2^{(2)})_4 \cdot CF_2^{(3)(4)} \cdot SO \cdot NH \cdot (CH_2)_3 \cdot CH_3$	(1) -82.0 (2) -120.3 -122.0, -123.4, -127.1 (3) -121.7 (4) -116.5	(3-4) 251	191(a)
$C_{10}F_{13}N$	 (Two isomers)	[a] (1) -90.6 (2) -138.8 (3)(4) -170.8, -188.4 (5) -73.1 (6)(7) -126.2, -133.0, -126.5 (1:1:2) [b] (1) -92.6 (2) -135.6 (3)(4) -168.4, -193.8 (5) -76.6 (6)(7) -123.9, -131.6, -131.6 (1:1:2)		212
$C_{10}F_{13}NO$		(1) -89.7 (2) -139.7 (3) -153.9 (4) -74.0 (5) -75.0, -76.2		212
$C_{10}F_{14}H_2O_2$		(1) -108.5 (2)(3) -67.5, -65.9 (4) -67.9		235
$C_{10}F_{14}H_{10}NO_2P$	$PF_2^{(1)}(NH^{(3)}) \cdot C(CH_3)_3 \cdot O \cdot \underbrace{C(CF_3)_2 C(CF_3)_2}_{(2)} O$	(1) -58.6 (2) -68.7 At -80° (1a) -51.1 (1e) -67.2	(P-1) 872 (1-3) 14.0 (P-1a) 790 (P-1e) 957 (1a-1e) 69.3 (1a-3) 34.7	156
$C_{10}F_{14}H_{10}NO_2P$	As above with $NH \cdot C(CH_3)_3$ replaced by $N(CH_2CH_3)_2$	At -20° (1) -57.5 (2) -67.4, -69.2	(P-1) 853	156

C ₁₀ F ₁₄ H ₁₀ N ₂ S ₂	$[(CF_3^{(1)})_2CF^{(2)} \cdot S \cdot N(CH_3)CH_2]_2$	(1) -75.1 (2) -167.3			140
C ₁₀ F ₁₅ N		(1) -179.83 (2) -72.72 (3) -52.84 (4) -88.01 (5) -58.06 (6) -55.47	(1-3) 50 (2-3) 3.4 (3-4) 39.0 (4-5) 21.6 (5-6) 21.6 (4-6) 21.6		236
C ₁₀ F ₁₆		(1) -68.5 (2) -104.8 (3)(4)(6) -63.5 to -63.7 (5) -53.9	(1-2) 9.0 (2-3) 9.0 (4-5) 33 (5-6) 33		235
C ₁₀ F ₁₆		(1)(2) -65.98, -65.63 (3) -111.0 (4) -61.5 (5) -68.0 (6) -68.7	(1-2) 20 (3-4) 18.2 (3-6) 10.1 (5-6) 3.4		235
C ₁₀ F ₁₆ H ₁₀ NP	$(CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_3^{(3)})_2 PF_2^{(4)} N(CH_2CH_3)_2$	(1) -81.6 (2) -126.9 (3) -116.9 (4) -52.4	(3-4) 11.5 (P-3) 115 (P-4) 927		71
C ₁₀ F ₁₆ Br ₂		(1) -49.6 (2) -106.3 (3)(4) -68.1, -61.5 (5) -67.8			235
C ₁₀ F ₁₆ O		(1)(2)(5)(6) -151.6, -153.4, -162.1, -163.5 (3)(7) -155.8, -157.7 (4) -71.8			96
C ₁₀ F ₁₇ H	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot CF_2^{(7)} \cdot CF_2^{(8)} \cdot C \equiv CH^{(9)}$	(1) -82.0 (2) to (7) -118.3 (8) -100.3	(8-9) 5		144

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₁₀ F ₁₇ HBr ₂	As above with —C≡CH replaced by —CBr=CHBr (Z)	(1) -81.2 (2) -126.6 (3) to (6) -122.0 (7) -120.4 (8) -105.9		144
C ₁₀ F ₁₇ H ₂ Br ₃	As above with —C≡CH replaced by —CBr ₂ ·CH ₂ Br	(1) -81.2 (2) -125.8 (3) to (6) -121.6 (7) -114.1 (8) -102.5		144
C ₁₀ F ₁₈		(1) -62.0 (2) -59.1 (3) -97.8 (4) -114.1	(1-2) 10.0 (1-3) 32.0	129(a)
C ₁₀ F ₁₈ H ₂ O ₄	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·O·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CO ₂ CH ₂ ⁽⁹⁾ CF ₂ ⁽⁵⁾ CF ₂ ⁽⁶⁾ ·OCF ₂ ⁽⁷⁾ ·CF ₃ ⁽⁸⁾	(1)(8) -87.3 (2)(7) -89.0 (3) -86.6 (4) -121.7 (5) -124.0 (6) -85.8	(5-9) 12.0	237
C ₁₀ F ₁₈ O ₄	[(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·O·CO] ₂ (CF ₂ ⁽³⁾) ₂	(1) -79.1 (2) -142.3 (3) -118.5	(1-2) 2.1	147
C ₁₀ F ₁₈ O ₅	[(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·O·CO·CF ₂ ⁽³⁾] ₂ O	(1) -79.2 (2) -142.8 (3) -77.4	(1-2) 1.9	147
C ₁₀ F ₂₀		(1) -71.2 (2)(3) -178.5, -180.5 (4) -72.6 (5) -105.4 (6) -95.5 (7) -59.5 (8) -61.7	(6-8) 32.5 (7-8) 11.5	129(a)
C ₁₀ F ₂₀	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ⁽³⁾ (CF ₃ ⁽⁴⁾)·C[CF ⁽⁵⁾ ·(CF ₃ ⁽⁶⁾) ₂]=CF ₂ ⁽⁷⁾	(1)(6) -71.0, -72.6 (2)(3)(5) -176.5, -180.5, -186.5 (4) -74.5 (7)(8) -58.5, -62.4		129(a)
C ₁₀ F ₂₀	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·C[CF ⁽⁵⁾ ·(CF ₃ ⁽⁶⁾) ₂]=CF ₂ ⁽⁷⁾	(1)(6) -71.0, -72.5 (2)(5) -180.5 (3)(4) -110.5, -114.9 (7)(8) -58.5, -62.5		129(a)

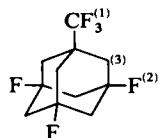
C ₁₁ F ₃ H ₇			(11-C1) 4·5 (11-C3) 3·2 (11-C9)~0·7 (11-C11) 272·1	431
C ₁₁ F ₃ H ₁₅ S	CF ₃ ⁽¹⁾ C(SCH ₂ CH ₃)=CH ⁽²⁾ ·C≡C·C(CH ₃) ₃	-66·44	(1-2) 1·5	188(a)
C ₁₁ F ₃ H ₁₆ N		-67·1		137
C ₁₁ F ₃ H ₁₆ N		-66·3	(1-2) 8·5	137
C ₁₁ F ₃ H ₁₈ N		-58·8		137
C ₁₁ F ₄ H ₆			(1-C1) 248·5 (1-C2) 21·2 (1-C3) 10·0 (1-C5) 2·5 (1-C6) <0·5 (1-C7) 1·9 (1-C8) 6·3 (11-C3) 6·2 (11-C5) 2·5 (11-C11) 272·7	431
	F at C1 CF ₃ at C4			

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
	F at C1 CF ₃ at C5		(1-C1) 256.2 (1-C2) 19.8 (1-C3) 8.1 (1-C4) 4.4 (1-C6) 5.9 (1-C7) 2.2 (1-C8) 5.1 (11-C4) 2.9 (11-C6) <0.5 (11-C8) ~1 (11-C11) 271.4	
	F at C1 CF ₃ at C7		(1-C1) 254.0 (1-C2) 19.5 (1-C3) 8.4 (1-C4) 4.3 (1-C5) 3.1 (1-C6) ~1 (1-C8) 4.6 (1-C9) ~3 (1-C10) 15 (11-C6) 2.9 (11-C7) 32.0 (11-C8) 5.3 (11-C9) ~1.5 (11-C11) 272.1	
	F at C2 CF ₃ at C5		(2-C1) 25.7 (2-C2) 249.2 (2-C3) 21.2 (2-C4) 9.5 (2-C5) 1.2 (2-C7) 0.9 (2-C8) 5.9 (2-C10) 9.6 (11-C5) 3.4 (11-C7) 3.1	
	F at C2 CF ₃ at C6		(2-C1) 25.3 (2-C2) 248.6 (2-C3) 21.1 (2-C4) 9.1 (2-C5) 1.2 (2-C6) 3.2 (2-C8) 5.7 (2-C9) ~6 (2-C10) 9.5 (11-C6) 2.9 (11-C8) 4.8 (11-C11) 273.5	
	F at C2 CF ₃ at C6		(2-C1) 23.4 (2-C2) 248.5 (2-C3) 25.5 (2-C4) 9.6 (2-C6) 2.5 (2-C7) ~0.5 (11-C5) ~1 (11-C7) 5.9 (11-C11) 273.2	

C ₁₁ F ₄ H ₇ N		(1) -77.5 (2) -201.3	(1-2) 13 (2-3) 44	(1-3) 6	169
C ₁₁ F ₄ H ₈ O	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ =C(C ₆ H ₅)CO·CH ₃	(1) -53.5 (2) -100	(1-2) 9.15		137
C ₁₁ F ₄ H ₁₃ NO ₂		(1) -57.9 (2) -142.3	(1-2) 6.5 (2-4) 5.2 ¹ J(1-C) 265.3 ¹ J(2-C) 349.5 ² J(2-C2) 27.1 ³ J(2-CO) 17.6	(1-4) ~1	193
C ₁₁ F ₄ H ₁₃ NO ₂	As above but with CF ₃ and F interchanged	(1) -56.0 (2) -166.0	(1-2) 11.5 ² J(2-C5) 15.5	(2-4) 2.5	193
C ₁₁ F ₅ H ₈ N		(1) -200.5 (2) -76.5 (3) -136.5	(1-2) 12.6 (2-4) 6.0	(1-4) 44.4	169
C ₁₁ F ₆ H ₆ O ₂	(CF ₃ ⁽¹⁾) ₂ CH ⁽²⁾ ·CO·CO·C ₆ H ₅	-63.4	(1-2) 7.9		213
C ₁₁ F ₆ H ₆ O ₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CO·CH ₂ ·CO·C ₆ H ₄ F ⁽³⁾ <i>p</i>	(1) -86.0 (2) -127.0 (3) -105.5			224(a)
C ₁₁ F ₆ H ₇ NO		(5CH ₃) -66.5 (7CH ₃) -66.2			226(a)
C ₁₁ F ₆ H ₇ N ₃ OS		-57.19, -56.48			173(e)
C ₁₁ F ₆ H ₇ N ₃ S	As above with OCH ₃ replaced by CH ₃	-57.74, -56.83			173(e)
C ₁₁ F ₆ H ₈		-60.8	¹ J(F-C) 272		210

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₁₁ F ₆ H ₈ OS	(CF ₃ ⁽¹⁾) ₂ CH ⁽²⁾ ·CS·O·C ₆ H ₄ CH ₃ <i>p</i>	−66·11	(1-2) 6	173(a)
C ₁₁ F ₆ H ₉	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ H ⁽⁴⁾ ·CF ₂ ⁽³⁾ ·CH ₂ ⁽⁵⁾ ·C ₆ H ₄ CH ₃ <i>p</i>	(1) −75·5 (2) −214·0 (3) −111·5	(1-2) 11·0 (1-3) 11·0 (1-4) 5·5 (2-3) 11·0 (2-4) 44·0 (3-4) 11·0, 5·5 (3-5) 20·0, 15·0 (2-5) 3·0	176
C ₁₁ F ₆ H ₉ N	<i>p</i> CH ₃ C ₆ H ₄ ·N $\begin{smallmatrix} \text{H} \\ \diagup \diagdown \\ (\text{CF}_3)_2 \end{smallmatrix}$	−67·25		228(a)
C ₁₁ F ₆ H ₉ NO	As above with CH ₃ replaced by CH ₃ O	−67·15		228(a)
C ₁₁ F ₆ H ₁₀	CF ₃ ⁽¹⁾ ·CF ⁽²⁾ H ⁽⁵⁾ ·CF ₂ ⁽³⁾⁽⁴⁾ ·CH ⁽⁶⁾ (CH ₃)C ₆ H ₅ (diastereoisomers)	[a] (1) −75·0 (2) −214·0 (3) −118·3 (4) −125·7	(1-2) 12·5 (1-3,4) 12·5 (1-5) 6·2 (2-5) 44·7 (3-4) 270 (3-5) 18·0 (4-5) 3·0	176
		[b] (1) −75·5 (2) −206·8 (3)(4) −117·5	(1-2) 11·0 (1-3,4) 11·0 (2-3,4) 11·0 (2-5) 43·0 (3,4-5) 11·0 (1-5) 8·5 (3,4-6) 15·5	
C ₁₁ F ₆ H ₁₀ N ₂	$\begin{smallmatrix} ^{(1)}\text{F}_3\text{C} & \text{CF}_3^{(2)} \\ & / \backslash \\ \text{H} & \text{CN} \\ & / \backslash \\ \text{N} & \text{C}(\text{CH}_3)_3 \end{smallmatrix}$	(1) −58·2 (2) −57·5	(1-2) 6·0	193
C ₁₁ F ₆ H ₁₁ NOS ₂	(CF ₃) ₂ C(SCH ₂ CH ₃)NH·S(O)C ₆ H ₅	−72·2 to −69·8		206(a)
C ₁₁ F ₆ H ₁₁ NO ₂	$\begin{smallmatrix} ^{(1)}\text{F}_3\text{C} & \text{CF}_3^{(2)} \\ & / \backslash \\ ^{(4)}\text{H} & \text{CO}_2\text{H} \\ & / \backslash \\ \text{N} & \text{C}(\text{CH}_3)_3 \end{smallmatrix}$	(1) −57·3 (2) −55·4	(1-2) 6·7 (1-4) ~1	193
C ₁₁ F ₆ H ₁₁ NO ₂ S	(CF ₃) ₂ C(OC ₆ H ₅)NH·S(O)CH ₂ CH ₃	−75·65		206(a)
C ₁₁ F ₆ H ₁₁ NO ₂ S	(CF ₃) ₂ C(OC ₆ H ₄ CH ₃ <i>p</i>)NHS(O)CH ₃	−75·7, −74·95		206(a)

C₁₁F₆H₁₂

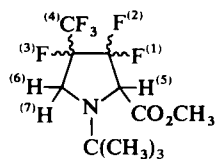


(1) -79.16 (2) -143.70

(1-3) ~2

239(c)(j)

C₁₁F₆H₁₅NO₂



(diastereoisomers)

[a] (1)(2) -120.25, -106.15
(3) -179.7 (4) -76.7

(1-2) 249.6 (1-3) ~0
(1-4) 14.4 (1-5) 23.3
(1-6,7) 3.5 (2-3) ~0
(2-5) ~7.8
(3-6)(3-7) 27.5, 20.5
(3-4) 7.0

193

[b] (1)(2) -118.92, -116.08
(3) -177.1 (4) -76.7

(1-2) 248.0 (1-5) 12.5
(2-3) ~0 (2-4) 12.0
(2-5) 15.2 (3-6,7) 32.0

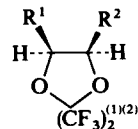
[c] (1)(2) -117.96, -112.04
(3) -183.0 (4) -77.8

(1-2) 248

[d] (1)(2) -126.79, -113.21
(3) -74.8

(1-2) 247

C₁₁F₆H₁₆O



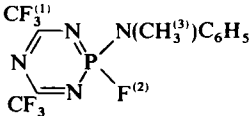
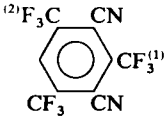
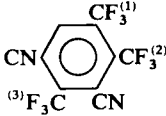
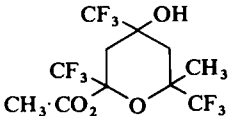
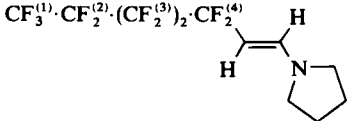
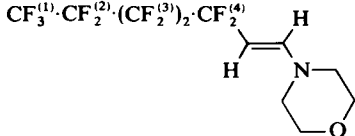
R¹ = CH₃
CH₂CH₃
(CH₂)₂CH₃
R² = (CH₂)₄CH₃
(CH₂)₃CH₃
(CH₂)₂CH₃

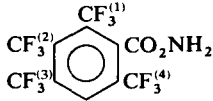
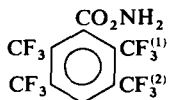
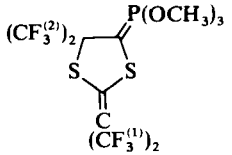
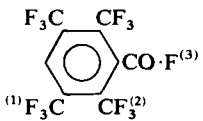
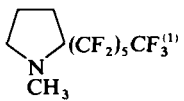
(1) -78.88 (2) -78.88
-78.79 -78.95
-78.78 -78.90

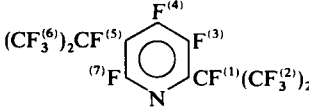
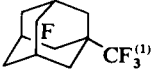
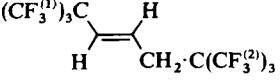
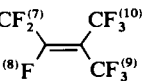
(1-2) 9.1
9.14
9.14

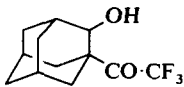
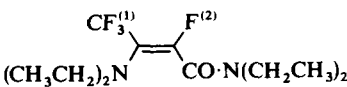
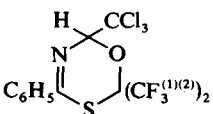
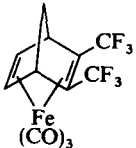
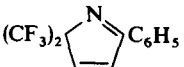
240(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{11}F_6H_{16}O$	 $R^1 = CH_3$ $R^2 = (CH_2)_4CH_3$ CH_2CH_3 $(CH_2)_3CH_3$ $(CH_2)_2CH_3$ $(CH_2)_2CH_3$	(1) -79.65 (2) -79.72 -79.60 -79.62 -81.26 -81.26	(1-2) 8.41 8.4 8.4	240(b)
$C_{11}F_7H_5D_2O_3$	$C_6H_5 \cdot CO \cdot CO \cdot CF_2^{(3)} \cdot CF_2^{(2)} \cdot CF_3^{(3)} \cdot D_2O$	(1) -82.9 (2) -120.9 (3) -125.9	(1-2) 10	241(b)
$C_{11}F_7H_5O_2$	$C_6H_5 \cdot CO \cdot CO \cdot CF_2^{(3)} \cdot CF_2^{(2)} \cdot CF_3^{(1)}$	(1) -82.4 (2) -120.0 (3) -126.9	(1-2) 9.5	241(b)
$C_{11}F_7H_5O_2$	 $R^1 = CH_3$ $R^2 = (CH_2)_4CH_3$ CH_2CH_3 $(CH_2)_3CH_3$ $(CH_2)_2CH_3$ $(CH_2)_2CH_3$	(1) -79.3 (2) -135.7 (3)(4) -134.3		224(a)
$C_{11}F_7H_6BrO$	$C_6H_5 \cdot CHBr \cdot CO \cdot CF_2^{(3)} \cdot CF_2^{(2)} \cdot CF_3^{(1)}$	(1) -81.5 (2) -118.1 (3) -126.5	(2-3) 10	241(b)
$C_{11}F_7H_7O$	$C_6H_5 \cdot CH_2 \cdot CO \cdot CF_2^{(3)} \cdot CF_2^{(2)} \cdot CF_3^{(1)}$	(1) -81.3 (2) -120.3 (3) -126.5	(1-2) 9.5	241(b)
$C_{11}F_7H_7O_2$	$C_6H_5 \cdot CO \cdot CH_2 \cdot O \cdot CF_2^{(2)}(CF_3^{(1)})_2$	(1) -85.0 (2) -153.0	(1-2) 2.6	213
$C_{11}F_7H_7O_3$	$C_6H_5 \cdot CO \cdot CO \cdot CF_2^{(3)} \cdot CF_2^{(2)} \cdot CF_3^{(1)} \cdot H_2O$	(1) -82.8 (2) -120.7 (3) -125.9	(1-2) 10	241(b)
$C_{11}F_7H_8NO_3S_2$	 $R^1 = CH_3$ $R^2 = (CH_2)_4CH_3$ CH_2CH_3 $(CH_2)_3CH_3$ $(CH_2)_2CH_3$ $(CH_2)_2CH_3$	(1) -65.10	(1-2) 8	173(e)

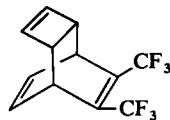
$C_{11}F_7H_8N_4P$		(1) -75.0 (2) -44.9	(P-1) 5.5 (2-3) 1.1	(P-2) 960	95
$C_{11}F_7H_{18}N_4PSi$	As above with $N(CH_3)C_6H_5$ replaced by $N[Si(CH_3^{(3)})_3]C(CH_3)_3$	(1) -75.2 (2) -19.5	(P-1) 6.0 (2-3) 0.3	(P-2) 967	95
$C_{11}F_9HN_2$		(1) -59.96 (2) -64.90			209(a)
$C_{11}F_9HN_2$		(1) -58.46 (2) -58.89 (3) -56.38			242(a)
$C_{11}F_9H_{11}O_4$		-79.2, -86.3, -88.1			217(a)
$C_{11}F_{11}H_{10}N$		(1) -80.0 (2) -124.8 (3) -120.8 (4) -102.2			243
$C_{11}F_{11}H_{10}NO$		(1) -80.0 (2) -125.2 (3) -121.0 (4) -102.4			243

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{11}F_{11}H_{12}N$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_2 \cdot CF_2^{(4)} \begin{array}{c} H \\ \backslash \\ C \\ / \\ H \end{array} \begin{array}{c} H \\ \backslash \\ C \\ / \\ N(CH_2 \cdot CH_3)_2 \end{array}$	(1) -80.0 (2) -124.9 (3) -121.0 (4) -102.4		243
$C_{11}F_{12}H_3NO_2$		(1) -53.85 (2) -55.12 (3) -58.98 (4) -60.61		209(a)
$C_{11}F_{12}H_3NO_2$		(1)(2) -59.93, -55.33		242(a)
$C_{11}F_{12}H_5MoO_3Sb$	$[(\eta^5-C_5H_5)Mo(CO)_3CF_3^{(3)} \cdot CF_2^{(2)} \cdot CF_3^{(1)}]SbF_6$	(1) -80.12 (2) -106.85 (3) +106.5	(1-2) 2.4 (1-3) 8.0 (2-3) 16.1	244(a)
$C_{11}F_{12}H_8O$	$CF_3^{(1)} \cdot (CF_2)_4 \cdot CF=C(CH_3) \cdot CO \cdot CH_2 \cdot CH_3$	(1) -81.5		137
$C_{11}F_{12}H_8O_2$	$CF_3^{(1)} \cdot (CF_2)_4 \cdot CF=C(CH_3) \cdot CO_2CH_2CH_3$	(1) -82.5		137
$C_{11}F_{12}H_9NO$	$CF_3 \cdot (CF_2)_4 \cdot CF=C(CH_3) \cdot CO \cdot N(CH_3)_2$	(1) -82		137
$C_{11}F_{12}H_9O_3PS_2$		(1) -57.72 (2) -66.51	(P-2) 1.3	173(e)
$C_{11}F_{13}HO$		(1) -59.76 (2) -55.61 (3) +56.60		242(a)(o)
$C_{11}F_{13}H_8N$		(1) -80.5		137

C ₁₁ F ₁₃ H ₉ FeN ₃ O ₂ P	[CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·Fe(CO) ₂ (CH ₃ CN) ₃]PF ₆ ⁽⁴⁾	(1) -76.9 (2) -117.4 (3) -79.4 (4) -73.3	(1-3) 11.7	215
C ₁₁ F ₁₃ H ₉ O	CF ₃ ⁽¹⁾ (CF ₂ ⁽²⁾) ₅ ·CH(CH ₃)·CO·CH ₂ ·CH ₃	(1) -81.6 (2) -115.2, -121.2, -123.3, -126.6		137
C ₁₁ F ₁₃ H ₁₁ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CF ₂ ⁽⁶⁾ ·C(OH)(CH ₂ CH ₃) ₂	(1) -80.4 (2) -124.5 (3) -120.5 (4) -120.5 (5) -117.1 (6) -108.0		208
C ₁₁ F ₁₇ N		(1) -186.5 (2) -76.86 (3) -144.7 (4) -108.6 (5) -180.7 (6) -77.61 (7) -56.31	(1-2) 6 (1-3) 51 (2-3) 5 (3-7) 25 (4-5) ~44 (4-6) 11.5 (5-6) 5.5 (5-7) ~44 (6-7) 13.5	236
C ₁₁ F ₁₈		(1) -70.2 Rest: -125.3, -135.4, -234.0 (2:2:1)		245(n)
C ₁₁ F ₁₈ H ₄		(1) -67.9 (2) -67.4		145(a)
C ₁₁ F ₁₈ H ₇ O ₃ P	[(CF ₃ ⁽¹⁾) ₂ CH ⁽³⁾ O] ₂ P(CH ₃)CH ₂ C(CF ₃ ⁽²⁾) ₂ O	(1) -74.5 (2) -78.0	(1-3) 5.0 ¹ J((1-C)(2-C))285, 284 ² J(1-C) 34.4 ² J(2-C) 33.0	199
	(at -60°)	(1) -74.1, -74.4, -74.5, -74.9 (2) -77.3, -78.7	(1-3) 5.2, 6.0 (2-2') 8.8	
C ₁₁ F ₁₈ H ₉ O ₃ P	[(CF ₃ ⁽¹⁾) ₂ CHO] ₃ P(CH ₃) ₂	(1a) -73.0 (1e) -71.3		180
C ₁₁ F ₂₀ O ₄	(CF ₂ ⁽³⁾⁽⁴⁾) ₃ [CO ₂ CF ⁽²⁾ (CF ₃ ⁽¹⁾) ₂] ₂	(1) -79.3 (2) -142.7 (3)(4) -118.0, -123.2	(1-2) 1.9	147
C ₁₁ F ₂₂	(CF ₃ ⁽¹⁾⁽²⁾) ₂ CF ⁽³⁾ ·CF ⁽⁴⁾ (CF ₂ ⁽⁵⁾ ·CF ₃ ⁽⁶⁾)·CF ₂ ⁽⁷⁾ 	(1)(2) -70.2, -73.0 (3)(4) -173.7, -179.0 (5) -113.0 (AB) (6) -82.0 (7) -102.7(AB) (9) -61.9 (10) -59.6	(5A-5B) 302 (7A-7B) 290 (7-9) 10.0 (8-9) 31.0 (9-10) 10.0	129(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{11}F_{24}H_3O_3P$	$[(CF_3^{(1)})_2CHO]_3P(CF_3^{(2)})_2$	(1) -74.0 (2) -61.7		180
$C_{11}F_{24}O_5$	$CF_3^{(1)}O(CF_2^{(2)}CF_2^{(3)}O)_4CF_2CF_3^{(4)}$	(1) -58.9 (2) -93.5 (3) -91.5 (4) -90.3	(1-2) 10	104
$C_{12}F_3H_{15}O_2$		-73	$^2J(F-C)$ 35	247
$C_{12}F_3H_{18}N$	$CF_3C[N(CH_2CH_3)_2]=CH \cdot C \equiv C \cdot CH(CH_3)_2$	-65.69		188(a)
$C_{12}F_3H_{20}$	$(CF_3^{(1)} \cdot CH_2^{(2)} \cdot CH \cdot CH_2 \cdot CH_2 \cdot CH_3)_2$	-64.8	(1-2) 10.5	117
$C_{12}F_4H_{20}N_2O$		(1) -58 (2) -121	(1-2) 20	170
$C_{12}F_4H_{20}N_2O$	$CF_3^{(1)} \cdot CF^{(2)}=C[N(CH_2CH_3)_2]CO \cdot N(CH_2CH_3)_2$	(1) -60 (2) -104	(1-2) 5.6	170
$C_{12}F_6H_6Cl_3NOS$		(1)(2) -76.6, -74.7	(1-2) 9.5	248(a)
$C_{12}F_6H_6FeO_3$		-55.7		210
$C_{12}F_6H_7N$		-69.1		229

C₁₂F₆H₈

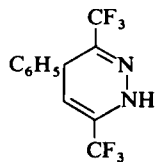


-60.6

¹J(F-C) 275

210

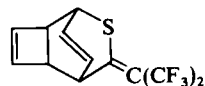
C₁₂F₆H₈N₂



-70.1, -68.5

87

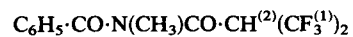
C₁₂F₆H₈S



-57.1, -53.6

173(e)

C₁₂F₆H₉NO₂

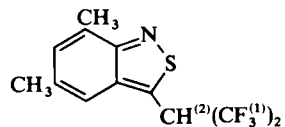


-63.72

(1-2) 7.5

173(e)

C₁₂F₆H₉NS

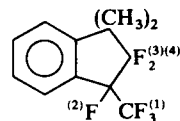


-66.32

(1-2) 8

173(e)

C₁₂F₆H₁₀

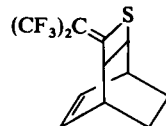


(1) -73.8 (2) -153.7
(3) -120.1 (4) -123.0

(1-2) 14.0 (1-3) 10.0
(1-4) 5.0 (2-3) 10.0
(3-4) 250 (2-4) 2.0

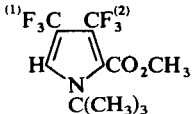
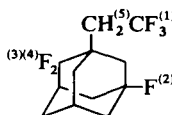
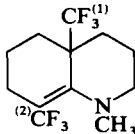
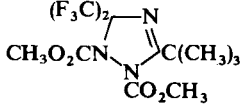
176

C₁₂F₆H₁₀S

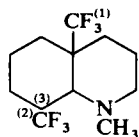


-56.9, -53.4

173(e)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}F_6H_{12}$	$CF_3^{(1)} \cdot CF^{(2)}H^{(5)} \cdot CF_2^{(3)(4)} \cdot CH^{(6)}(CH_2CH_3) \cdot C_6H_5$	[a] (1) -74.9 (2) -215.5 (3) -116.5 (4) -121.0	(1-2) 11.0 (1-3,4) 11.0 (3-4) 260 (2-5) 43.0 (3-5) 19.5 (4-5) 1.8 (1-5) 5.5	176
	(diastereoisomers)	[b] (1) -73.5 (2) -215.5 (3)(4) -114.5	(1-2) 11.0 (1-3,4) 11.0 (2-3,4) 11.0 (2-5) 44.0 (3,4-5) 11.0 (1-5) 5.5	
$C_{12}F_6H_{13}NO_2$		(1) -57.8 (2) -56.3	(1-2) 6.35	193
$C_{12}F_6H_{13}NO_2S$	$(CF_3)_2C(OCH_2CH_3)NH \cdot S(O)C_6H_4CH_3p$	-75.7		206(a)
$C_{12}F_6H_{14}$		(1) -97.91 (2) -144.38 (3) -115.58 (4) -109.99	(1-5) 13.0 (3-4) 236	239
$C_{12}F_6H_{15}N$		(1)(2) -62.5, -68.4		137
$C_{12}F_6H_{15}N_3O_4$		-74.40		234(a)
$C_{12}F_6H_{16}O_4$	$(CF_3^{(1)} \cdot CH_2^{(2)} \cdot C(CH_3)CO_2CH_3)_2$	-64.6	(1-2) 10.4	117

C₁₂F₆H₁₇N



(1) -61.7

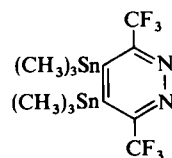
(2) -60.4

(1-2) 11

(2-3) 11

137

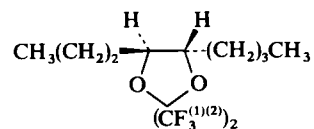
C₁₂F₆H₁₈N₂Sn₂



-62.0

87(a)

C₁₂F₆H₁₈O



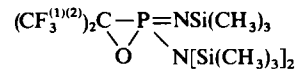
(1) -79.61

(2) -79.63

(1-2) 8.4

240(a)

C₁₂F₆H₂₇N₂OPSi₃



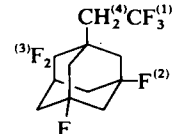
(1) -66.0

(2) -68.0

(1-2) 8.9 (P-1) 3.7

250

C₁₂F₇H₁₃



(1) -97.58

(2) -138.85

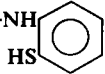
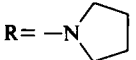
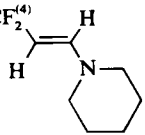
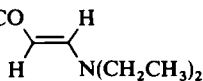
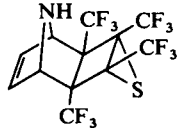
(1-4) 13

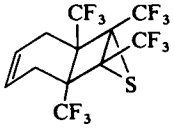
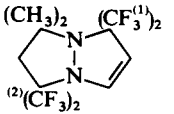
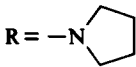
239(a)

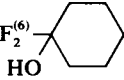
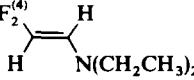
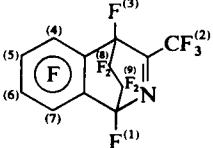
Fluorine nuclei in a CF₃ group

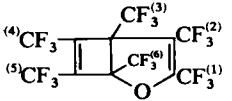
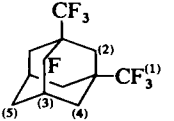
91

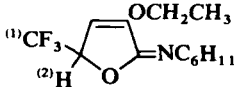
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}F_8H_{15}NO_2$	 (diastereoisomers)	[a] (1) -182.0 (2) -175.4 (3)(4) -77.1, -78.2 [b] (1) -166.3 (2) -178.8 (3) -73.8 (4) -76.6 [c] (1) -178.0 (2) -186 (3) -79.6 (4) -80.0 [d] (1) -174.0 (2) -178.8 (3) -75.8 (4) -77.3	(1-3) 17.6 (1-4) 5.3 (1-5) 6.8 (2-3) 4.9 (2-6) 27.8 (2-7) 23.2 (2-4) 19.4 (3-4) 2.8 (1-3) 13.4 (1-5) 26.3 (2-4) 19.0 (3-4) 6.7 (1-3) 19.7 (1-5) 6.4 (2-4) 9.9 (3-4) 9.9 (1-3) 9.2 (1-5) 26.6 (2-4) 18.4 (2-6) 29.0 (2-7) 25.0 (3-4) 9.2	193
$C_{12}F_9H_5$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot C \equiv C \cdot C_6H_5$	(1) -81.0 (2) -125.0 (3) -124.0 (4) -97.5		251
$C_{12}F_9N_3$		-58.86		242(a)
$C_{12}F_{11}H_4NS$		(1)(2) -55.86, -49.5	(1-2) 9.1 (1-3)(2-3) 1.2, 1.3 (1-4)(2-4) 0.5, 2.6	173(e)
$C_{12}F_{11}H_5$		(1) -81.0 (2) -127.5 (3) -125.5 (4) -117.0 (5) -169.5 (6) -145.0	(1-2) 140.0	251
$C_{12}F_{11}H_6NO_2S$		(1)(5) -82.0, -83.0 (2) -130.0 (3) -79.6, -82.7 (4) -123.0	(3A-3B) 140	252

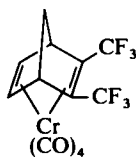
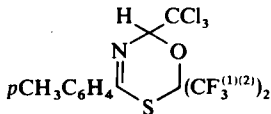
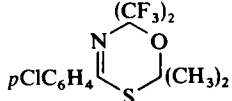
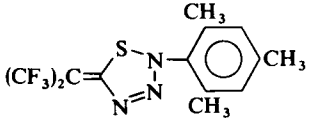
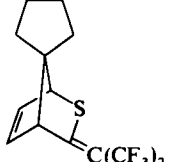
$C_{12}F_{11}H_6NO_2S$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot O \cdot CF^{(4)}(CF_3^{(5)}) \cdot CO \cdot NH$ 	(1)(5) -80.8, -81.6 (2) -129.0 (3) -80.1, -84.3 (4) -131.0	(3A-3B) 144	252
$C_{12}F_{11}H_{10}NO$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_2 \cdot CF_2^{(4)} \cdot CO$  $R = -N$ 	(1) -82.1 (2) -127.1 (3) -123.4 (4) -120.9		243
$C_{12}F_{11}H_{10}NO_2$	As above with $R = -N$ 	(1) -82.1 (2) -127.4 (3) -123.7 (4) -121.2		243
$C_{12}F_{11}H_{12}N$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_2 \cdot CF_2^{(4)}$ 	(1) -80.0 (2) -124.9 (3) -121.1 (4) -102.0		243
$C_{12}F_{11}H_{12}NO$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_2 \cdot CF_2^{(4)} \cdot CO$ 	(1) -81.6 (2) -126.6 (3) -123.4 (4) -119.9		243
$C_{12}F_{11}H_{16}NO$	$(^3CF_3) \cdot CF_2^{(2)} \cdot CF_3^{(1)}$ $(^4CF_3) \cdot O^- \cdot ^+NH(CH_2CH_3)_3$	(1) -81.2 (2) -116.1 (3) -50.8 (4) -56.1	(2-3) 20.4 (2-4) 1.7 (1-3) 1.7 (3-4) 10.3	128
$C_{12}F_{12}H_5NS$		-60.98, -61.94		253(c)
$C_{12}F_{12}H_6N_2S$	$(CF_3^{(1)})_2C(SC_6H_5)NH \cdot N = C(CF_3^{(2)(3)})_2$	(1) -70.2 (2)(3) -66.5, -64.0	(2-3) 5.7	178

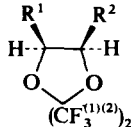
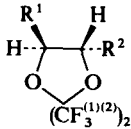
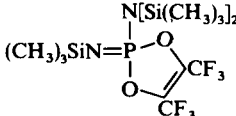
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₁₂ F ₁₂ H ₆ S		-59.5, -67.9		253(c)
C ₁₂ F ₁₂ H ₇ O ₂ P	[(CF ₃ ⁽¹⁾) ₂ CH ⁽²⁾ O] ₂ PC ₆ H ₅	-74.5	(1-2) 5.0	72
C ₁₂ F ₁₂ H ₁₀ N ₂		(1) -70.95 (2) -72.80		234(a) 254
C ₁₂ F ₁₂ H ₁₂ N ₂	As above with reduced double bond	-73.1, -70.7		254(a)
C ₁₂ F ₁₂ H ₁₄ N ₂ S	(CF ₃ ⁽¹⁾) ₂ C(SnC ₆ H ₁₃)NH·N=C(CF ₃ ⁽²⁾⁽³⁾) ₂	(1) -70.4 (2)(3) -66.3, -63.9		178
C ₁₂ F ₁₃ H ₆ NOS	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₄ ·CF ₂ ⁽³⁾⁽⁴⁾ ·SO·NH·C ₆ H ₅	(1) -81.7 (2) -120.2, -121.5, -122.3, -126.0 (3) -120.5 (4) -116.4	(3-4) 250	191(a)
C ₁₂ F ₁₃ H ₆ NO ₃ S ₂	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₄ ·CF ₂ ⁽³⁾⁽⁴⁾ ·SO·NH·SO ₂ C ₆ H ₅	(1) -81.2 (2) -120.1, -121.9, -122.5, -126.0 (3) -117.6 (4) -115.5	(3-4) 237	191(a)
C ₁₂ F ₁₃ H ₉ O	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₅ ·CH·CO·(CH ₂) ₃ ·CH ₂	(1) -80.4 (2) -111.6, -118.2, -120.7, -124.9		137
C ₁₂ F ₁₃ H ₁₀ N	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₃ ·CF ₂ ⁽⁴⁾  R = 	(1) -79.6 (2) -123.8 (3) -120.4 (4) -101.3		243

C ₁₂ F ₁₃ H ₁₀ NO	As above with R = 	(1) -80.0 (2) -123.8 (3) -120.4 (4) -102.9	243
C ₁₂ F ₁₃ H ₁₀ NO	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₅ ·CH·CO·(CH ₂) ₂ ·N(CH ₃)·CH ₂	(1) -81 (2) -114, -119, -123	137
C ₁₂ F ₁₃ H ₁₁ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CF ₂ ⁽⁶⁾ · 	(1) -80.0 (2) -124.1 (3)(4) -120.0, -121.6 (5) -119.5 (6) -117.0	208
C ₁₂ F ₁₃ H ₁₂ N	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₃ ·CF ₂ ⁽⁴⁾ · 	(1) -80.0 (2) -123.5 (3) -119.6 (4) -101.7	243
C ₁₂ F ₁₃ H ₁₂ NOS	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₄ ·CF ₂ ⁽³⁾⁽⁴⁾ ·SO·NH·C ₆ H ₁₁	(1) -82.0 (2) -121.2, -123.0, -126.9 (1:1:2) (3) -122.9 (4) -117.7	(3-4) 248 191(a)
C ₁₂ F ₁₃ H ₁₅ FeN ₂ O ₅ P ₂	[CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·Fe(CO) ₂ (CH ₃ CN) ₂ P(OCH ₃) ₃]PF ₆ ⁽⁴⁾	(1) -85.4 (2) -117.4 (3) -79.3 (4) -73.3	(1-3) 11.5 215
C ₁₂ F ₁₃ H ₁₅ O ₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CF ₂ ⁽⁶⁾ ·C(OH)(CH ₃)·CH ₂ ·C(OH)(CH ₃) ₂	(1) -80.4 (2) -124.0 (3)(4) -119.5, -120.4 (5) -119.5 (6) -116.2	208
C ₁₂ F ₁₃ N		(1) -173.1 (2) -73.9 (3) -213.2 (4)(7) -145.8 (5)(6) -151.3 (8) -125.0, -129.4 or (9) -126.2, -127.7	(8A-8B) } 230 (9A-9B) } 235 231
C ₁₂ F ₁₄ H ₇ O ₂ P	[(CF ₃ ⁽¹⁾) ₂ CHO]PF ₂ ⁽²⁾ ·C ₆ H ₅	(1) -75.0 (2) -45.0	(P-2) 846 180
C ₁₂ F ₁₄ H ₁₀ NO ₂ P	(CH ₂ =CH·CH ₂) ₂ N·PF ₂ ⁽²⁾ ·O·C(<u>CF₃)₂C(CF₃)₂</u> ·O (1)	(1) -70.4 (2) -59.6	(P-2) 881 141

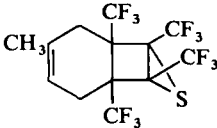
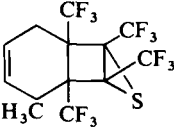
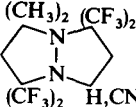
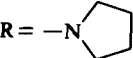
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}F_{14}H_{18}NO_2PSi_2$	$[(CH_3)_3Si]_2N \cdot \overbrace{PF_2^{(2)} \cdot O \cdot C(CF_3)_2 C(CF_3)_2 \cdot O}^{(1)}$	(1) -67.0, -68.2 (2) -38.6 At -130° (2a) -25.1 (2e) -53.8	(P-2) 920 (P-2a) 772 (P-2e) 996 (2a-2e) 76.0	156
$C_{12}F_{16}H_5NO_2P_2$	$C_6H_5PF_2^{(3)} = N \cdot \overbrace{PF_2^{(2)} \cdot O \cdot C(CF_3)_2 C(CF_3)_2 \cdot O}^{(1)}$	(1) -68.9 (2) -43.4 (3) -66.5	(1-2) 2.0 (P2-2) 846 (P3-2) 18.3 (P3-3) 1107	155
$C_{12}F_{17}H_2O_2P$	$[(CF_3)^{(1)}]_2CH^{(5)}O]_2P$ 	(1) -78.3 (2) -139 (3) -167 (3) -153	(1-5) 6.0	72
$C_{12}F_{18}O$		(1) -66.3 (2) -55.9 (3) -66.3 (4)(5) -63.18 (6) -73.5		255(c)
$C_{12}F_{18}O_2$		-52.7, -61.5, -77.5		255(c)
$C_{12}F_{20}$		(1) -57.8 (2) -101.50 (3) -220.63 (4) -113.90 (5) -123.10		256(a) 245
$C_{12}F_{21}NO_2$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2)^{(3)}_5 \cdot N$ 	(1) -81.7 (2) -126.6 (3) -119.5, -122.1, -123.2 (4) -123.9 (5) -138.1		219(a)

C ₁₂ F ₂₂ O ₃	(CF ₃ ⁽¹⁾) ₂ ·CF ⁽²⁾ ·CO·CF ₂ ⁽³⁾ ·CF ₂ ·O·CF ⁽⁵⁾ (CF ₃ ⁽⁴⁾)CO·CF ⁽⁷⁾ (CF ₃ ⁽⁶⁾) ₂	(1) -72.3 (2) -188.7 (3) -118.1 (4) -78.7 (5) -133.8 (6) -72.3 (7) -187.0	200
C ₁₂ F ₂₂ O ₄	(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·CO·CF ⁽³⁾ (CF ₃ ⁽⁴⁾)·O·CF ₂ ·CF ₂ ⁽⁵⁾ ·CF ₂ ·O·CF ⁽⁶⁾ (CF ₃ ⁽⁷⁾)·CO·F ⁽⁸⁾	(1) -72.5 (2) -191.1 (3) -134.3 (4) -79.0 (5) -127.3 (6) -129.1 (7) -81.1 (8) +27.5	200
C ₁₂ F ₂₂ O ₄	[CF ₃ ⁽¹⁾ ·CF ⁽²⁾ (OCF ₃ ⁽³⁾)·CO·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾] ₂	(1) -80.1 (2) -136.4 (3) -54.6 (4) -114.7 (5) -119.0	200
C ₁₂ F ₂₂ O ₅	[(CF ₃ ⁽¹⁾) ₂ CF ⁽²⁾ ·O·CO·(CF ₂ ⁽³⁾ ⁽⁴⁾) ₂] ₂ O	(1) -79.2 (2) -142.3 (3)(4) -85.3, -121.2	(1-2) 1.9 147
C ₁₂ F ₂₃ ClO ₄	CF ₂ ⁽¹⁾ Cl·CF ⁽²⁾ (CF ₃ ⁽³⁾)·O[CF ₂ ⁽⁴⁾ ·CF ⁽⁵⁾ (CF ₃ ⁽⁶⁾)·O] ₂ CF ₂ ⁽⁷⁾ ·CO·CF ₃ ⁽⁸⁾	(1) -69.0 (2) -141.0 (3) -79.2 (4)(6) -80.6 (5) -144.8, -145.4 (7)(8) -75.3	(7,8-) 5 143
C ₁₂ F ₂₄ O ₄	As above with Cl replaced by F	(1) -81.5 (2) -144.9 (4) -81.2 (5) -145.8 (6) -80.7 (7)(8) -75.6	(7,9-) 5.5 143
C ₁₃ F ₃ H ₁₈ NO ₂		-80.5	(1-2) 5.6 259(a)
C ₁₃ F ₃ H ₁₉ N	CF ₃ ·C[N(CH ₂ CH ₃) ₂]=CH·C≡C·C(CH ₃) ₃	-65.9	188(a)
C ₁₃ F ₃ H ₂₁ O ₄	CF ₃ ·C(OC ₃ H ₇)=C(OC ₃ H ₇)CO ₂ C ₃ H ₇	[a] -67.9 [b] -69.2 (ratio 3 : 1)	170
C ₁₃ F ₃ H ₂₂ NSi ₂	<i>m</i> CF ₃ C ₆ H ₄ ·N[Si(CH ₃) ₃] ₂	-62.93	168
C ₁₃ F ₃ H ₂₂ NSn ₂	<i>m</i> CF ₃ C ₆ H ₄ ·N[Sn(CH ₃) ₃] ₂	-62.15	168

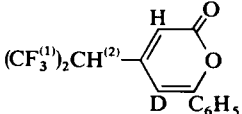
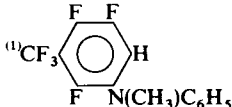
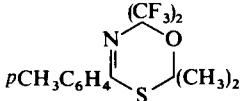
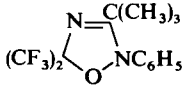
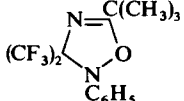
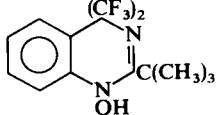
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{13}F_4H_{15}N$	$CF_3^{(1)} \cdot C[N(CH_2CH_3)_2] = CF^{(2)} \cdot C_6H_5$	(1) -61 (2) -115	(1-2) 19.0	170
$C_{13}F_4H_{13}N$	$CF_3^{(1)} \cdot CF^{(2)} = C[N(CH_2CH_3)_2]C_6H_5$	(1) -63 (2) -160	(1-2) 11.5	170
$C_{13}F_6H_6CrO_4$		-55.6		210
$C_{13}F_6H_6MnO_4$	As above with Cr replaced by Mn	-55.7		210
$C_{13}F_6H_8Cl_3NOS$		(1)(2) -76.5, -74.6	(1-2) 9.5	248(a)
$C_{13}F_6H_{10}ClNOS$		-77.1		248(a)
$C_{13}F_6H_{11}NOS$	As above with Cl replaced by H	-77.2		248(a)
$C_{13}F_6H_{11}N_3S$		-57.12, -56.62		173(e)
$C_{13}F_6H_{12}S$		-59.27, -55.0		173(e)

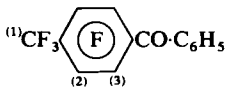
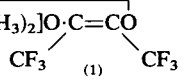
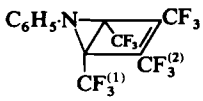
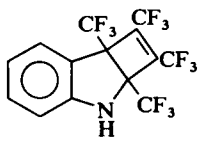
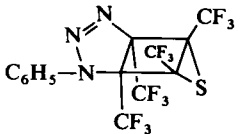
C ₁₃ F ₆ H ₁₃ ClNO ₄ P	$p\text{Cl}\cdot\text{C}_6\text{H}_4\cdot\overline{\text{C}=\text{N}\cdot\text{C}(\text{CF}_3)_2\text{P}(\text{OCH}_3)_3\cdot\text{O}}$	-66·5	(P-F) 4·0	261(a)
C ₁₃ F ₆ H ₁₄ BrNO ₂	(CF ₃ ⁽¹⁾⁽²⁾) ₂ C=CH·CHBr·CO·CO·NH·C ₆ H ₁₁	(1)(2) -64·5, -58·3	(1-2) 8·6	259(a)
C ₁₃ F ₆ H ₁₄ BrNO ₂	(CF ₃) ₂ CH·CH=CBBr·CO·CO·NH·C ₆ H ₁₁	-65·6	(1-2) 8·6	259(a)
C ₁₃ F ₆ H ₁₄ N ₂ O ₂ S	(CF ₃) ₂ C $\overline{\text{NCH}_2\text{CH}_2\text{OCH}_2\text{CH}_2}$ NH·S(O)C ₆ H ₅	-69·1		206(a)
C ₁₃ F ₆ H ₁₅ Br ₂ NO ₂	(CF ₃) ₂ CH·CHBr·CHBr·CO·CO·NH·C ₆ H ₁₁	-65·1		259(a)
C ₁₃ F ₆ H ₁₅ NO ₂	(CF ₃ ⁽¹⁾) ₂ CH ⁽²⁾ ·CH=CH·CO·CO·NH·C ₆ H ₁₁	-67·1	(1-2) 8·6	259(a)
C ₁₃ F ₆ H ₂₀ O				
	$\begin{array}{ll} \text{R}^1 = \text{CH}_3 & \text{R}^2 = (\text{CH}_2)_6\text{CH}_3 \\ \text{CH}_2\cdot\text{CH}_3 & (\text{CH}_2)_5\text{CH}_3 \\ (\text{CH}_1)_2\text{CH}_3 & (\text{CH}_2)_4\text{CH}_3 \\ (\text{CH}_2)_3\text{CH}_3 & (\text{CH}_2)_3\text{CH}_3 \end{array}$	$\begin{array}{l} (1)(2) -78\cdot89, -78\cdot89 \\ -78\cdot81, -78\cdot96 \\ -78\cdot77, -78\cdot87 \\ -78\cdot79, -78\cdot87 \end{array}$	$\begin{array}{l} (1-2) 9\cdot1 \\ 9\cdot11 \\ 9\cdot09 \\ 9\cdot16 \end{array}$	240(a)
C ₁₃ F ₆ H ₂₀ O				
	$\begin{array}{ll} \text{R}^1 = \text{CH}_3 & \text{R}^2 = (\text{CH}_2)_6\text{CH}_3 \\ \text{CH}_2\text{CH}_3 & (\text{CH}_2)_5\text{CH}_3 \\ (\text{CH}_2)_2\text{CH}_3 & (\text{CH}_2)_4\text{CH}_3 \\ (\text{CH}_2)_3\text{CH}_3 & (\text{CH}_2)_3\text{CH}_3 \end{array}$	$\begin{array}{l} (1)(2) -79\cdot65, -79\cdot73 \\ -79\cdot63, -79\cdot63 \\ -79\cdot60, -79\cdot60 \\ -79\cdot59, -79\cdot59 \end{array}$	$\begin{array}{l} (1-2) 8\cdot44 \\ 8\cdot4 \\ 8\cdot4 \\ 8\cdot4 \end{array}$	240(a)
C ₁₃ F ₆ H ₂₇ N ₂ O ₂ PSi ₃		-67·5		250

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{13}F_{11}H$		(1) -55.9 (3)(10)(6)(7) -135.2, -136.6, -138.5 (4)(8)(5)(9) -148.0, -150.9, -152.5, -154.3		430
$C_{13}F_{11}H_{12}NO$		(1) -81.6 (2) -127.1 (3) -123.4 (4) -120.9		243
$C_{13}F_{12}$		(2) -58.3 (3) -134.9 (4) -149.4 (5) -147.9 (6) -136.4 (7) -139.9 (8) -161.4 (9) -150.9		262(b)
$C_{13}F_{12}$		(2)a(6)a -138.7 (2)b -114.7 (3)a(5)a -135.9 (4)b -128.3 (5)b(8)a,b -161.3 (6)b -123.2 (7)a,b -137.5 (9)a -149.8 (9)b -150.5 (4)a(3)b -58.1		262(b)
$C_{13}F_{12}H_6S$		(1) -61.38 (2) -60.30		253(c)
$C_{13}F_{12}H_8N_2$	$(CF_3^{(1)})_2C \cdot NH(CH_2 \cdot C_6H_5) \cdot N=C(CF_3^{(2)(3)})_2$	(1) -74.7 (2)(3) -66.3, -64.3	(2-3) 5.7	178(a)

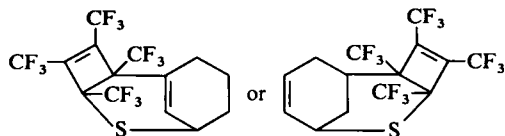
$C_{13}F_{12}H_8N_2S$	$(CF_3^{(1)})_2C \cdot (SCH_2C_6H_5) \cdot NH \cdot N=C(CF_3^{(2)(3)})_2$	(1) -70.7 (2)(3) -66.7, -64.5	(2-3) 5.7	178(a)
$C_{13}F_{12}H_8S$		-58.5, -59.5, -67.5 (ratio 1 : 1 : 2)		253(c)
$C_{13}F_{12}H_8S$		-59.1, -60.3, -61.5, -67.5		253(c)
$C_{13}F_{12}H_{11}N_3$		-72.5, -71.2, -68.0, -66.7		254(a)
$C_{13}F_{13}H_7$	$C_6H_5CH_2C(CF_3^{(4)})_2CF_2^{(3)} \cdot CF_2^{(2)}CF_3^{(1)}$	(1) -63.6 (2)(3) -106.4, -124.4 (4) -81.7		293(a)
$C_{13}F_{13}H_7O$	$C_6H_5 \cdot CH^{(7)}(OH) \cdot CF_2^{(6)} \cdot CF_2^{(5)} \cdot CF_2^{(4)} \cdot CF_2^{(3)} \cdot CF_2^{(2)} \cdot CF_3^{(1)}$	(1) -80.8 (2) -125.0 (3) -121.2 (4) -121.2 (5) -118.3 (6A) -116.0 (6B) -124.7	(6A-6B) 308 (6A-7) 6.5 (6B-7) 17.5	208
$C_{13}F_{13}H_7O_2$	$oHOC_6H_4 \cdot CH^{(7)}(OH) \cdot CF_2^{(6)} \cdot CF_2^{(5)} \cdot CF_2^{(4)} \cdot CF_2^{(3)} \cdot CF_2^{(2)} \cdot CF_3^{(1)}$	(1) -80.8 (2) -126.2 (3) -122.5 (4) -122.5 (5) -122.5 (6A) -116.3 (6B) -124.5	(6A-6B) 300 (6A-7) 7 (6B-7) 17	208
$C_{13}F_{13}H_{10}NO$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_3 \cdot CF_2^{(4)} \cdot CO$  	(1) -81.6 (2) -127.1 (3) -123.2 (4) -121.6		243

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{13}F_{13}H_{10}NO_2$	As above with $R = -N \begin{array}{c} \diagup \diagdown \\ \diagdown \diagup \end{array} O$	(1) -82.6 (2) -127.1 (3) -122.9 (4) -120.9		243
$C_{13}F_{13}H_{12}N$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_3 \cdot CF_2^{(4)} \begin{array}{c} H \\ \diagup \diagdown \\ H \end{array} \begin{array}{c} H \\ \diagdown \diagup \\ N \end{array} \begin{array}{c} \diagup \diagdown \\ \diagdown \diagup \end{array}$	(1) -80.0 (2) -123.8 (3) -120.0 (4) -102.1		243
$C_{13}F_{13}H_{12}NO$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_3 \cdot CF_2^{(4)} \begin{array}{c} CO \\ \diagup \diagdown \\ H \end{array} \begin{array}{c} H \\ \diagdown \diagup \\ N \end{array} \begin{array}{c} \diagup \diagdown \\ \diagdown \diagup \end{array} (CH_2CH_3)_2$	(1) -81.6 (2) -127.1 (3) -123.7 (4) -121.6		243
$C_{13}F_{13}H_{15}O$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_3^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot C(OH)(iC_3H_7)_2$	(1) -80.5 (2) -124.5 (3) -120.5 (4) -120.5 (5) -117.1 (6) -108.0		208
$C_{13}F_{14}H_{18}NO_2PSi$	$\overbrace{PF_2^{(2)} \{ NC(CH_3)_3 (Si(CH_3)_3) \} OC(CF_3)_2 C(CF_3)_2 O}^{(i)}$ (at -33°)	(1) -67.1, -68.5 (2a) -35.1 (2e) -47.4	(2a-2e) 70.5 (P-2a) 906 (P-2e) 925	156
$C_{13}F_{18}H_{12}O_3P_2$	$(CF_3^{(1)})_2 CH^{(4)} O \cdot P [OC(CF_3^{(3)})_2 P(CH_3)_2] (CH_3) \cdot CH_2 \cdot C(CF_3^{(2)})_2 O$	(1) -72.4, -73.1 (2) -76.5, -78.3 (3) -67.0, -67.7	(2-2') 9.0 (1-4) 6.0 (P-3) 3.0 $^1J(1-C)(2-C)(3-C)$ 282, 285, 289 $^2J(1-C)$ 35.3 $^2J(2-C)$ 32.7 $^3J(3-C)$ 29.1	199
$C_{13}F_{23}H_3O_4$	$(CF_3)_2 CF \cdot CF_2 \cdot O \cdot C(CF_3)_2 \cdot CF_2 \cdot O \cdot C(CF_3)_2 \cdot CO_2CH_3$	(1) -73.0 (2) -186.3 (3) -74.2 (4) -70.6 (5) -71.3 (6) -75.1		96
$C_{13}F_{24}H_5O_4P$	$[(CF_3^{(1)})_2 CHO]_3 \overbrace{PCH_2C(CF_3^{(2)})_2 O}$	(1) -74.3 (2) -77.3		124
$C_{13}F_{14}H_5O_4P$	$[(CF_3)_2 CHO]_2 \overbrace{P(CH_3)OC(CF_3)_2 C(CF_3)_2 O}$	-74.3		124
$C_{13}F_{24}H_7O_4P$	$[(CF_3^{(1)})_2 CH^{(2)} O]_4 P(CH_3)$	(1a) -75.3 (1e) -74.8	(1-2) 6.0	180

$C_{13}F_{28}O_6$	$CF_3^{(1)} \cdot O \cdot (CF_2^{(2)} \cdot CF_2^{(3)} \cdot O)_5 \cdot CF_2 \cdot CF_3^{(4)}$	(1) -59.0 (2) -93.6 (3) -91.5 (4) -90.3	104
$C_{14}F_6H_2O_{10}Os_3$	$HOs_3(CO)_{10}(CF_3^{(1)} \cdot C=CH \cdot CF_3^{(2)})$	(1) -57.93 (2) -59.51	266
$C_{14}F_6H_7DO_2$		-63.2 (1-2) 7.9	267(a)
$C_{14}F_6H_8O_2$	As above with D replaced by H	-64.4 (1-2) 7.7	267(a)
$C_{14}F_6H_9N$		(1) -58.7 Rest -123.5, -139.1, -141.0	451
$C_{14}F_6H_{10}O_2$	$(CF_3)_2C=C=C(C_6H_5)(CO_2CH_2CH_3)$	-61.0	172
$C_{14}F_6H_{13}NOS$		-77.3	248(a)
$C_{14}F_6H_{14}N_2O$		-79.1	268(a)
$C_{14}F_6H_{14}N_2O$		-74.1	268(a)
$C_{14}F_6H_{14}NO_2$		-76.9	268(a)
$C_{14}F_6H_{16}N_2OS$	$(CF_3)_2C[N \cdot CH_2 \cdot (CH_2)_3 \cdot CH_2]NH \cdot SO \cdot C_6H_5$	-68.9	206(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_6H_{16}N_2O_2S$	$(CF_3)_2C[N \cdot (CH_2)_2 \cdot O \cdot CH_2 \cdot CH_2]NH \cdot SO \cdot C_6H_4CH_3p$	-69.25		206(a)
$C_{14}F_6H_{24}$	$[CF_3^{(1)} \cdot CH_2^{(2)} \cdot CH \cdot (CH_2)_3 \cdot CH_3]_2$	-65.5	(1-2) 11.7	117
$C_{14}F_7H_5O$		(1) -56.5 (2)(3) -140.3	(1-2) 22.0	269
$C_{14}F_7H_7O$	As above with CO replaced by CHO	(1) -56.2 (2)(3) -141.6	(1-2) 22.6	269
$C_{14}F_7H_{14}N$	$pCF_3^{(2)}C_6H_4 \cdot CF^{(3)}=C[N(CH_2CH_3)_2]CF_3^{(1)}$	(1) -57 (2) -60 (3) -110	(1-3) 1.7	170
$C_{14}F_7H_{15}NO_2P$	$C_6H_5 \cdot PF^{(2)}[N(CH_2CH_3)_2]O \cdot C(=O)CF_3$ 	(1) -65.1, -66.6 (2) -40.3	(P-2) 821	141
$C_{14}F_{12}H_5N$		(1) -63.5 (2) -66.7		270(c)
$C_{14}F_{12}H_5N$		-63.1, -63.9, -58.7, -53.1		270(c)
$C_{14}F_{12}H_5N_3S$		-60.3, -61.3, -61.9, -56.5		270(c)

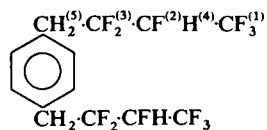
C₁₄F₁₂H₈S



−60·72, −62·15, −64·09, −67·38

253(c)

C₁₄F₁₂H₁₀

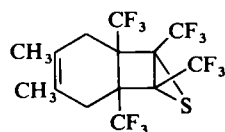


(1) −74·5 (2) −214·5
(3) −110·5

(1-2) 12·0 (1-3) 12·0
(1-4) 6·0 (2-3) 12·0
(2-4) 45·0 (2-5) 3·0
(3-4) 12·0, 6·0
(3-5) 18·0, 15·0

176

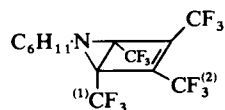
C₁₄F₁₂H₁₀S



−58·7, −67·3

253(c)

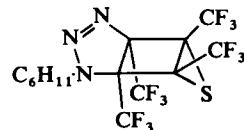
C₁₄F₁₂H₁₁N



(1) −62·38 (2) −61·18

270(c)

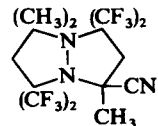
C₁₄F₁₂H₁₁N₃S



−59·3, −61·5, −62·3, −57·1

270(c)

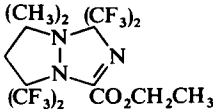
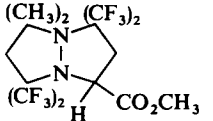
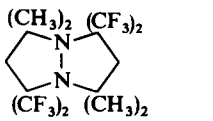
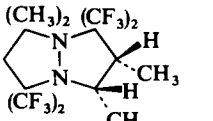
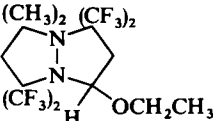
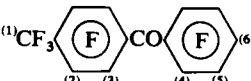
C₁₄F₁₂H₁₃N₃

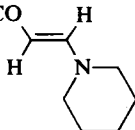
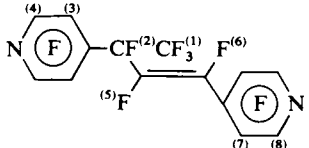
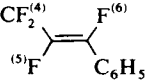


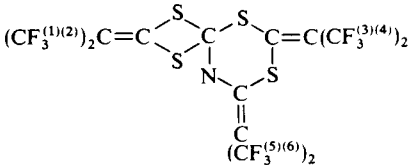
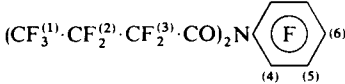
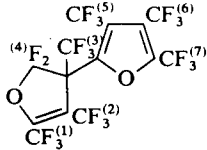
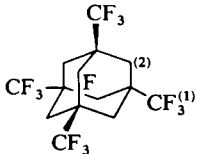
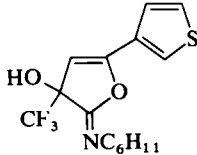
−70·3, −70·5, −65·6, −64·4

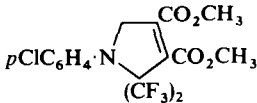
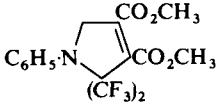
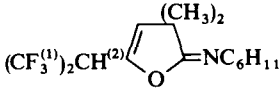
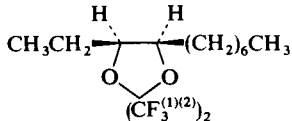
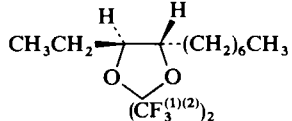
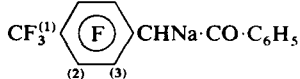
254(a)

Fluorine nuclei in a CF₃ group

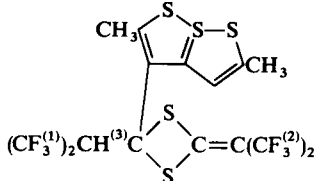
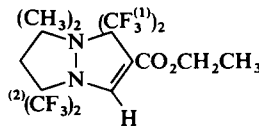
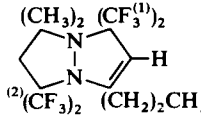
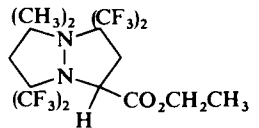
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_{12}H_{13}N_3O_2$		-74.3, -71.3		234(a)
$C_{14}F_{12}H_{13}O_4P$	$(CF_3^{(1)} \cdot CH_2^{(2)} \cdot O)_4PC_6H_5$		(1-2) 8.3	220
$C_{14}F_{12}H_{14}N_2O_2$		-73.05, -72.05, -67.8, -66.95		254(a)
$C_{14}F_{12}H_{16}N_2$		-68.10		234(a)
$C_{14}F_{12}H_{16}N_2$		-72.2, -71.4, -68.2, -61.6		254(a)
$C_{14}F_{12}H_{16}N_2O_2$		-73.1, -72.0, -68.5, -67.5		254(a)
$C_{14}F_{12}O$		(1) -56.9 (2) -139.6 (3)(4) -141.4 (5) -161.2 (6) -146.7	(1-2) 20.8 (5-6) 19.5	269

C ₁₄ F ₁₂ O	As above with CO replaced by CHO	(1) -56.5 (2) -140.8 (3)(4) -142.8 (5) -163.8 (6) -155.9	(1-2) 21.9 (5-6) 20.2	269
C ₁₄ F ₁₃ H ₅	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₃ ·CF ₂ ⁽⁴⁾ ·C≡C·C ₆ H ₅	(1) -81.5 (2) -127.5 (3) -123.1 (4) -97.5		251
C ₁₄ F ₁₃ H ₉ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CF ₂ ⁽⁶⁾ ·C(OH)(CH ₃)·C ₆ H ₅	(1) -80.5 (2) -126.2 (3)(4) -122.0, -123.0 (5) -118.5 (6) -116.5		208
C ₁₄ F ₁₃ H ₁₂ NO	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₃ ·CF ₂ ⁽⁴⁾ ·CO 	(1) -82.1 (2) -127.1 (3) -123.7 (4) -120.9		243
C ₁₄ F ₁₃ H ₁₇ O	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·CF ₂ ⁽⁶⁾ ·C(OH)(CH ₂ (CH ₃)·CH ₂ ·C(OH)(CH ₃)·CH ₂ ·CH ₃	(1) -81.6 (2) -124.5 (3)(4) -120.8, -121.6 (5) -117.9 (6) -116.2		208
C ₁₄ F ₁₄ N ₂		(1) -78.2 (2) -167.0 (3)(7) -139.5 (4)(8) -88.0 (5)(6) -141.8, -148.6		212
C ₁₄ F ₁₅ H ₅	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₃ ·CF ₂ ⁽⁴⁾ · 	(1) -81.5 (2) -127.5 (3) -123.1 (4) -117.5 (5) -170.0 (6) -145.5	(5-6) 139.0	251
C ₁₄ F ₁₇ H ₆ NOS	CF ₃ ⁽¹⁾ ·(CF ₂ ⁽²⁾) ₆ ·CF ₂ ⁽³⁾⁽⁴⁾ ·SO·NH·C ₆ H ₅	(1) -82.0 (2) -120.5, -121.5, -126.2 (3) -121.1 (4) -116.7	(3-4) 250	191(a)

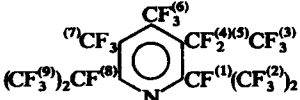
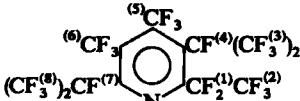
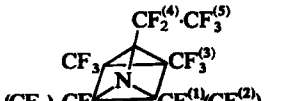
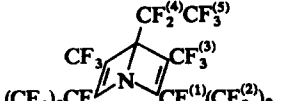
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_{18}H_3NS_4$		(1)(2) $\left\{ \begin{array}{l} -59.46, -54.90 \\ (3)(4) \end{array} \right.$ (3)(4) $\left\{ \begin{array}{l} -57.61, -56.30 \\ (5)(6) \end{array} \right.$ (5)(6) $\left\{ \begin{array}{l} -58.48 \end{array} \right.$	(1-2) $\left\{ \begin{array}{l} 8.7 \\ (3-4) \end{array} \right.$ (3-4) $\left\{ \begin{array}{l} 9.3 \\ (5-6) \end{array} \right.$	173(e)
$C_{14}F_{19}NO_2$		(1) -82.6 (2) -126.8 (3) -115.2 (4) -141.5 (5) -163.2 (6) -150.5		219(a)
$C_{14}F_{20}O_2$		(1) -74.3 (2) -60.8 (3) -63.3 (4) -61.5, -66.3 (5) -57.3 (6) -59.3 (7) -64.5	(4A-4B) 150	177
$C_{14}F_{24}$		(1) -54.1 (2) -98.2		245(a)
$C_{15}F_3H_{16}NSO_2$		-80.8		259(a)
$C_{15}F_3H_{22}N_2OSi_2$	$mCF_3C_6H_4 \cdot N[Si(CH_3)_3]CO \cdot N(CH_3)[Si(CH_3)_3]$	-62.93		168
$C_{15}F_6H_{11}NOS_2$	$(CF_3)^{(1)(2)}_2C(SC_6H_5)NH \cdot SO \cdot C_6H_5$	(1)(2) -70.4, -69.2	(1-2) 10.0	206(a)

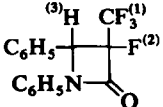
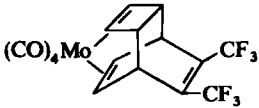
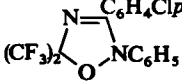
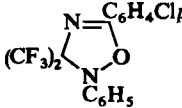
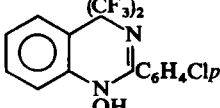
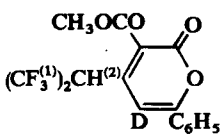
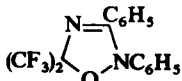
C ₁₅ F ₆ H ₁₂ ClNO ₄		-68.5		228(a)
C ₁₅ F ₆ H ₁₂ N ₂ OS	(CF ₃ ⁽¹⁾⁽²⁾) ₂ C(NH·CH ₂ ·C ₆ H ₅)NH·SO·C ₆ H ₅	(1)(2) -74.4, -72.9	(1-2) 10.5	206(a)
C ₁₅ F ₆ H ₁₃ NO ₄		-68.9		228(a)
C ₁₅ F ₆ H ₁₆ Br ₂ NO ₄ P	C ₆ H ₅ ·CHBr·CHBr·C≡N·C(CF ₃) ₂ ·P(OCH ₃) ₂ ·O	-66.0	(P-F) 4.0	261(a)
C ₁₅ F ₆ H ₁₆ NO ₄ P	As above with CHBr·CHBr replaced by CH=CH	-66.5	(P-F) 4.0	261(a)
C ₁₅ F ₆ H ₁₈ BrNO ₂	(CF ₃ ⁽¹⁾⁽²⁾) ₂ C=CH·CBr=C(OCH ₂ CH ₃)CO·NH·C ₆ H ₁₁	(1)(2) -64.0, -60.4	(1-2) 7.5	259(a)
C ₁₅ F ₆ H ₁₉ NO		-66.5	(1-2) 8	259(a)
C ₁₅ F ₆ H ₁₉ NO ₂	(CF ₃ ⁽¹⁾⁽²⁾) ₂ C=CH·CH=C(OCH ₂ CH ₃)CO·NH·C ₆ H ₁₁	(1)(2) -62.0, -56.2	(1-2) 8.4	259(a)
C ₁₅ F ₆ H ₂₄ O		(1)(2) -78.81, -78.90	(1-2) 9.01	240(b)
C ₁₅ F ₆ H ₂₄ O		(1)(2) -79.63, -79.65	(1-2) 8.4	240(b)
C ₁₅ F ₇ H ₆ NaO		(1) -53.8 (2) -146.5 (2) -142.2		272(b)

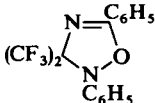
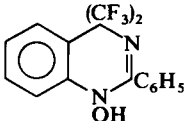
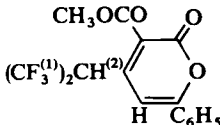
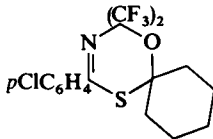
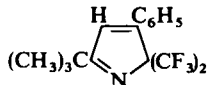
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{15}F_7H_7O$	As above with Na replaced by H	(1) -55.4 (2) -142.2 (3) -140.0	(1-2) 22.1	272(b)
$C_{15}F_7H_{17}NO_2P$	$pCH_3C_6H_4 \cdot \overbrace{PF^{(2)}[N(CH_2CH_3)_2]OC=CO}^{CF_3 \quad (1) \quad CF_3}$	(1) -65.8, -67.5 (2) -41.5	(P-2) 823	141
$C_{15}F_9H_6NO$	$CF_3^{(1)} \cdot CF^{(2)}H^{(3)} \cdot CO \cdot N(C_6H_5)C_6F_5$		(1-3) 5.7 (2-3) 45.9	263(b)
$C_{15}F_{11}H_{20}N$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_2 \cdot CF_2^{(4)} \begin{array}{c} H \\ \backslash \\ C \\ / \\ H \end{array} \begin{array}{c} H \\ \backslash \\ N[CH_2CH(CH_3)_2]_2 \\ / \end{array}$	(1) -80.0 (2) -124.8 (3) -120.8 (4) -102.4		243
$C_{15}F_{12}HN$	$\begin{array}{cc} (1)CF_3 & (6) \\ \text{---} & \text{---} \\ (2) & (3) \end{array} \text{---} CH(CN) \text{---} \begin{array}{cc} & (4) \\ \text{---} & \text{---} \\ (5) & \end{array}$	(1) -55.7 (2) -139.3 (3) -137.4 (4) -139.3 (5) -160.8 (6) -151.9		273
$C_{15}F_{12}HNaO$	$\begin{array}{cc} (1)CF_3 & (3) \\ \text{---} & \text{---} \\ (2) & \end{array} \text{---} CHNa \text{---} CO \text{---} C_6F_5$	(1) -54.4 (2) -146.0 (3) -140.3	(1-2) 21.7	272(b)
$C_{15}F_{12}H_2O$	As above with Na replaced by H	(1) -56.5 (2) -141.3 (3) -139.9	(1-2) 22.5	272(b)
$C_{15}F_{12}H_8OS_2$	$pCH_3C_6H_4 \cdot O \begin{array}{c} S \\ \backslash \\ C \\ / \\ S \end{array} \begin{array}{c} S \\ \backslash \\ C=C(CF_3^{(2)})_2 \\ / \\ S \end{array} \begin{array}{c} (CF_3^{(1)})_2CH \\ \backslash \end{array}$	(1) -61.13 (2) -58.53		173(e)
$C_{15}F_{12}H_8OS_2$	$pHOC_6H_4 \cdot C(CF_3^{(1)})_2CH \begin{array}{c} S \\ \backslash \\ C \\ / \\ S \end{array} \begin{array}{c} S \\ \backslash \\ C=C(CF_3^{(2)})_2 \\ / \\ S \end{array}$	(1) -64.74 (2) -57.79		173(e)

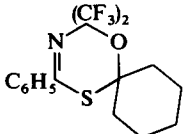
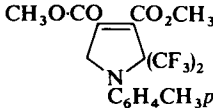
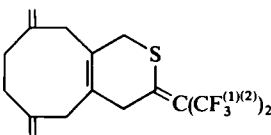
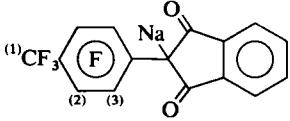
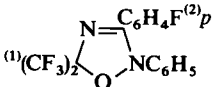
$C_{15}F_{12}H_8S_5$		(1) -62.91 (2) -57.56 (1-3) 7	173(e)
$C_{15}F_{12}H_{14}N_2O_2$		(1) -68.50 (2) -73.15	234(a)
$C_{15}F_{12}H_{14}N_2O_2$	As above with substituents on double bond interchanged	(1) -69.50 (2) -70.25	234(a)
$C_{15}F_{12}H_{16}N_2$		(1) -71.0 (2) -69.8	254(a)
$C_{15}F_{12}H_{16}N_2$	As above with substituents on double bond interchanged	(1) -68.0 (2) -72.5	254(a)
$C_{15}F_{12}H_{16}N_2O_2$		-72.9, -71.7, -67.7, -66.9	254(a)
$C_{15}F_{12}H_{27}N_2O_2PSi_3$	$[(CH_3)_3Si]_2P \overbrace{[=NSi(CH_3)_3OC(CF_3)_2C(CF_3)_2O]}^{(1)(2)}$	(1)(2) -68.4, -68.9 (1-2) 7.8	250
$C_{15}F_{12}LiN$		(1) -53.6 (2) -147.4 (3) -147.4 (4) -141.7 (5) -166.5 (6) -166.5	273
$C_{15}F_{12}NNa$	As above with Li replaced by Na	(1) -52.7 (2) -147.9 (3) -147.9 (4) -141.4 (5) -167.2 (6) -167.2	273

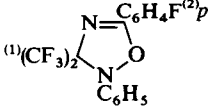
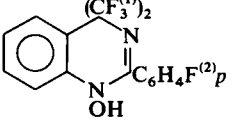
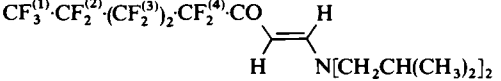
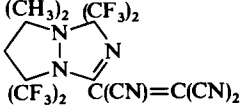
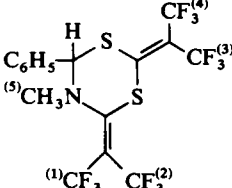
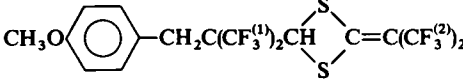
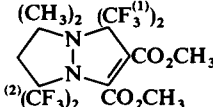
Molecular formula	Structure of compounds	Chemical shifts (pm)	Coupling constants (Hz)	Ref.
$C_{15}F_{13}H_5$		(1) -62.7 (2) -57.9 (3) -68.0 (4) -90.9	(1-2) 10.0 (1-3) 10.0 (2-3) 4.0	172(a)
$C_{15}F_{13}H_{19}O$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)} \cdot CF_2^{(6)} \cdot C(OH)(nC_4H_9)_2$	(1) -80.4 (2) -124.5 (3)(4) -120.0, -120.8 (4) -117.9 (5) -115.8		208
$C_{15}F_{13}H_{21}FeN_2O_5P_2$	$[CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot Fe(CO)_2(CH_3CN)_2P(OCH_2CH_3)_3]PF_6^{(4)}$	(1) -85.6 (2) -117.2 (3) -78.9 (4) -72.9	(1-3) 11.0	215
$C_{15}F_{14}H_2O$		(1) -57.5 (2) -139.5	(1-2) 21.9	269
$C_{15}F_{14}H_{18}NO_2P$	$PF_2^{(2)} \cdot \overbrace{(N \cdot C(CH_3)_2 \cdot CH_2)_3 \cdot C(CH_3)_2}^{(1)} \cdot O \cdot \underbrace{C(CF_3)_2 C(CF_2)_2}_{(1)} \cdot O$	(1) -67.8 (2) -38.3 At -80° (2a) -31.3 (2e) -47.4	(P-2) 910 (2a-2e) 77.6 (P-2a) 832 (P-2e) 1001	156
$C_{15}F_{14}O$		(1) -56.9 (2)(3) -138.8 to -141.0	(1-2) 21.9	269
$C_{15}F_{18}H_{12}O_3$	$[(CF_3^{(1)})_2 CF_2^{(2)} \cdot O \cdot CF_2^{(3)}]_2 C(OH)C_6H_{11}$	(1) -80.8 (2) -145.6 (3) -73.2	(1-3) 2 (2-3) 23	143
$C_{15}F_{19}H_8O_3P$	$[(CF_3^{(1)})_2 CHO]_3 PF_6^{(2)} C_6H_5$	(1) -73.8 (2) -48.8	(P-2) 860	180
$C_{15}F_{25}N$		(1) -178.12 (2) -71.75 (3) -151.30 (4) -69.04 (5) -51.43 (6) -53.61 (7) -108.30 (8) -80.33	(1-4) 47 (3-5) 51 (6-7) 23.3	236 274

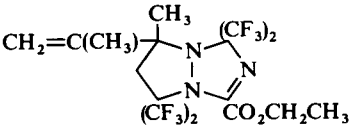
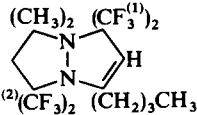
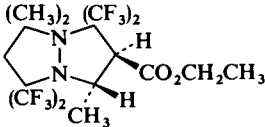
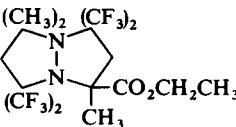
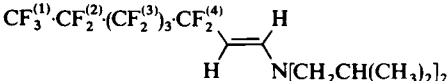
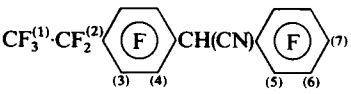
C ₁₅ F ₂₅ N		(1) -178.4 (2) -72.2 (3) -71.3 (4)(5) -80.6, -90.1 (6) -51.9 (7) -51.9 (8) -179.6 (9) -72.2	(1-3) 37 (3-6) 10 (4-6) 40	(1-4) 100 (4-5) 300 (7-8) 44	274
C ₁₅ F ₂₅ N		(1) -108.3 (2) -78.8 (3) -69.1 (4) -153.1 (5) -52.1 (6) -52.1 (7) -179.9 (8) -72.4	(1-3) 23 (6-7) 50	(4-5) 48 (7-8) 3.2	274
C ₁₅ F ₂₅ N		(1) -181.0 (2) -73.4 (3) -61.2 (4) -121.2 (5) -82.5			274
C ₁₅ F ₂₅ N		(1) -114.5, -119.3 (2) -82.5 (3)(6) -62.0 (4) -186.7 (5)(8) -73.4 (7) -183.6			274
C ₁₅ F ₂₅ N		(1) -188.2 (2) -77.0, -78.6 (3) -64.5 (4) -118.2 (5) -84.4	(3-4) 6		274
C ₁₅ F ₂₈ HNO ₂	CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·(CF ₂ ⁽³⁾) ₃ ·CF ₂ ⁽⁴⁾ ·CO·NH·CO·CF ₂ ⁽⁴⁾ (CF ₂ ⁽³⁾) ₄ ·CF ₂ ⁽²⁾ ·CF ₃ ⁽¹⁾	(1) -82.4 (2) -127.4 (3) -122.5 to -123.8 (4) -120.2			219
C ₁₅ F ₂₈ O ₄	[(CF ₂ ⁽¹⁾) ₂ CF ⁽²⁾ ·CO·CF ⁽³⁾ (CF ₃ ⁽⁴⁾)·O·CF ₂ ⁽⁵⁾] ₂ CF ₂ ⁽⁶⁾	(1) -72.4 (2) -172.9 (3) -113.4 (4) -78.8 (5) NA (6) -125.9			200
C ₁₅ F ₃₀ H ₅ O ₅ P	[(CF ₃) ₂ CHO] ₅ P	-73.0			157
C ₁₅ F ₃₂ O ₇	CF ₃ ⁽¹⁾ ·O·(CF ₂ ⁽²⁾ CF ₂ ⁽³⁾ O) ₆ ·CF ₂ ·CF ₃ ⁽⁴⁾	(1) -59.1 (2) -93.7 (3) -91.6 (4) -90.4	(1-2) 9		104

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{16}F_4H_6NO$		(1) -78.5 (2) -182.0	(1-2) 11.2 (2-3) 3.4	276(a)
$C_{16}F_6H_8MoO_4$		-60.5		210
$C_{16}F_6H_9ClN_2O$		-78.3		268(a)
$C_{16}F_6H_9ClN_2O$		-73.9		268(a)
$C_{16}F_6H_9ClN_2O$		-76.2		268(a)
$C_{16}F_6H_9DO_4$		-62.8	(1-2) 8.1	267(a)
$C_{16}F_6H_{10}N_2O$		-78.2		268(a)

C ₁₆ F ₆ H ₁₀ N ₂ O		-74.1		268(a)
C ₁₆ F ₆ H ₁₀ N ₂ O		-76.2		268(a)
C ₁₆ F ₆ H ₁₀ O ₄		-62.6	(1-2) 8.1	267(a)
C ₁₆ F ₆ H ₁₄ ClNOS		-77.3		248(a)
C ₁₆ F ₆ H ₁₄ N ₂ OS	(CF ₃ ⁽¹⁾⁽²⁾) ₂ C(NH·C ₆ H ₅)NH·SO·C ₆ H ₄ CH ₃ <i>p</i>	(1)(2) -74.1, -72.55	(1-2) 10.5	206(a)
C ₁₆ F ₆ H ₁₄ N ₂ OS	(CF ₃) ₂ C(NH·CH ₂ ·C ₆ H ₅)NH·SO·C ₆ H ₅	-75.7 to -74.2		206(a)
C ₁₆ F ₆ H ₁₅ IrO ₂	(η-C ₅ (CH ₃) ₅)Ir(CO) ₂ (CF ₃ ⁽¹⁾ C≡CCF ₃ ⁽²⁾)	-60.84, -63.44	(1-2) 8	277
C ₁₆ F ₆ H ₁₅ N		-66.6		229(a)
C ₁₆ F ₆ H ₁₅ N	As above with substituents about C=C interchanged	-69.7		229(a)
C ₁₆ F ₆ H ₁₅ N	(CF ₃ ⁽¹⁾) ₂ CH ⁽²⁾ ·N=C[C(CH ₃) ₃]·C≡C·C ₆ H ₅	-70.5	(1-2) 6.5	229(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{16}F_6H_{15}NOS$		-77.4		248(a)
$C_{16}F_6H_{15}NO_4$		-68.6		228(a)
$C_{16}F_6H_{15}NO_5$	As above with $C_6H_4CH_3p$ replaced by $C_6H_4OCH_3p$	-68.75		228(a)
$C_{16}F_6H_{16}S$		-57.1, -53.9	(1-2) 11	173(e)
$C_{16}F_6H_{28}$	$(CF_3 \cdot CH_2 \cdot CH \cdot (CH_2)_4 \cdot CH_3)_2$	-64.8	(1-2) 10.8	117
$C_{16}F_7H_4NaO_2$		(1) -55.1 (3) -134.2	(2) -144.5	272(b)
$C_{16}F_7H_5O_2$	As above with Na replaced by H	(1) -56.1 (3) -138.3	(2) -140.4	272(b)
$C_{16}F_7H_9N_2O$		(1) -77.9 (2) -103.6		268(a)

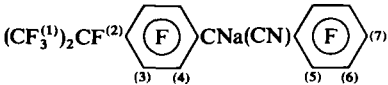
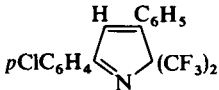
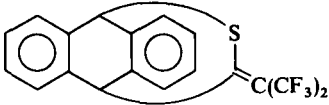
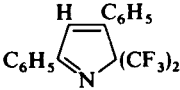
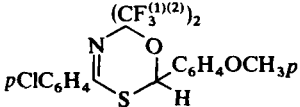
C ₁₆ F ₇ H ₉ N ₂ O		(1) -73.6 (2) -102.3	268(a)
C ₁₆ F ₇ H ₉ N ₂ O		(1) -75.9 (2) -106.6	268(a)
C ₁₆ F ₁₁ H ₂₀ NO		(1) -81.6 (2) -127.4 (3) -123.4 (4) -120.9	243
C ₁₆ F ₁₂ H ₈ N ₆		-73.65, -70.35	234(a)
C ₁₆ F ₁₂ H ₉ NS ₂		(1)(2) -58.1, -53.9 (3)(4) -57.1, -56.0	(1-2) 8 (3-4) 9 (3,4-1 or 2) 4.5 (1 or 2-5) 1.3 173(e)
C ₁₆ F ₁₂ H ₁₀ OS ₂		(1) -64.86 (2) -67.40	173(e)
C ₁₆ F ₁₂ H ₁₄ N ₂ O ₄		(1) -68.30 (2) -70.70	234(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{16}F_{12}H_{15}N_3O_2$		-75.45, -72.15, -70.65, -68.80		234(a)
$C_{16}F_{12}H_{18}N_2$		(1) -71.0 (2) -69.8		254(a)
$C_{16}F_{12}H_{18}N_2$	As above with double-bond substituents interchanged	(1) -68.0 (2) -72.5		254(a)
$C_{16}F_{12}H_{18}N_2O_2$		-70.5, -67.5, -65.1 (1:2:1)		254(a)
$C_{16}F_{12}H_{18}N_2O_2$		-71.0, -70.4, -67.8, -66.5		254(a)
$C_{16}F_{13}H_{15}NO_2P$	$C_6H_5PF^{(2)}[N(CH_2CH_3)_2OC(CF_3)^{(1)}_2C(CF_3)_2O]$	(1) -68.1, -70.9 (2) -39.3	(P-2) 819	141
$C_{16}F_{13}H_{20}N$		(1) -80.0 (2) -123.8 (3) -119.6 (4) -101.7		243
$C_{16}F_{14}HN$		(1) -85.3 (2) -110.1 (3) -137.5 (4) -137.1 (5) -139.3 (6) -160.8 (7) -152.0	(1-2) 2.2 (1-3) 6 (2-3) 31.7 (5-7) 3.5 (6-7) 21.4	273

$C_{16}F_{14}HN$		(1) -55.8 (3) -137.0	(2) -139.3	(1-2) 21.9	273
$C_{16}F_{14}LiN$		(1) -85.8 (3) -146.3 (5) -141.5 (7) -166.4	(2) -107.8 (4) -147.3 (6) -166.4	(2-3) 29 (1-3) 7	273
$C_{16}F_{14}LiN$		(1) -54.3 (3) -143.0	(2) -146.4	(1-2) 21.5	273
$C_{16}F_{14}NNa$	As above with Li replaced by Na	(1) -54.1 (3) -143.4	(2) -146.8	(1-2) 21.2	273
$C_{16}F_{14}NNa$		(1) -85.7 (3) -146.8 (5) -141.3 (7) -166.7	(2) -107.5 (4) -147.7 (6) -166.7	(2-3) 29	273
$C_{16}F_{18}O_4Pd$	$Pd(CO)_4(CF_3 \cdot C \equiv C \cdot CF_3)_3$	-56.2			88
$C_{16}F_{22}$		Isomer [a] [b] [c]	(1)(3) -176.2, -167.6 (2) -141.5 (4) -118.8 (5) -81.6 (1)(3) -174.4, -168.2, -164.7 (2:1:1) (2) -141.6, -138.4 (6:2) (4) -117.9 (5) -81.4 (1)(3) -174.8, -163.8 (1:1) (2) -141.1, -139.6 (2:2) (4) -117.9 (9) -80.6		278(b)
$C_{16}F_{26}MoO_8P_2$		(1) -66.2, -68.5 (2) +18.5		$[^1J(P-F) + ^3J(P-F)]$ 1285	155

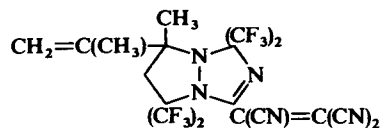
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₁₆ F ₂₇ N		(1) -179.8 (2) -72.7 (3) -152.1 (4) -69.8 (5)(6) -50.7, -52.6 (7) -180.8 (8) -73.3	(1-4) 47 (3-5) 55 (6-7) 55	274
C ₁₇ F ₃ H ₁₈ NO ₂		-80.7		259(a)
C ₁₇ F ₃ H ₂₀ NO ₃	CF ₃ ·C(OH)(CO·NH·C ₆ H ₁₁)·CH ₂ ·CO·C ₆ H ₅	-79.8		259(a)
C ₁₇ F ₆ H ₉ ClN ₂ O ₃ S				248(a)
	R ¹ = <i>p</i> ClC ₆ H ₄ ; R ² = <i>m</i> NO ₂ C ₆ H ₄ ; R ³ = H	(1)(2) -77.4, -76.0	(1-2) 9	
C ₁₇ F ₆ H ₁₀ ClNOS	As above with R ¹ = <i>p</i> ClC ₆ H ₄ ; R ² = C ₆ H ₅ ; R ³ = H	(1)(2) -77.3, -75.8	(1-2) 9	248(a)
C ₁₇ F ₆ H ₁₀ N ₂ O ₃ S	As above with R ¹ = C ₆ H ₅ ; R ² = <i>m</i> NO ₂ C ₆ H ₄ ; R ³ = H	(1)(2) -77.4, -76.1	(1-2) 9	248(a)
C ₁₇ F ₆ H ₁₁ NOS	As above with R ¹ = C ₆ H ₅ ; R ² = C ₆ H ₅ ; R ³ = H	(1)(2) -77.4, -75.8	(1-2) 9	248(a)
C ₁₇ F ₆ H ₁₂ N ₂ O		-78.2		268(a)
C ₁₇ F ₆ H ₁₂ N ₂ O		-73.2		268(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{17}F_{13}H_{10}N$		(1) -82.2		137
$C_{17}F_{13}H_{17}NO_2P$	$p\text{-CH}_3\text{C}_6\text{H}_4\cdot\text{PF}^{(2)}_6[\text{N}(\text{CH}_2\text{CH}_3)_2]\text{OC}(\text{CF}_3)_2\text{C}(\text{CF}_3)_2\text{O}$ (1)	(1) -68.0, -70.7 (2) -39.9	(P-2) 826	141
$C_{17}F_{13}H_{20}NO$	$\text{CF}_3^{(1)}\cdot\text{CF}_2^{(2)}\cdot(\text{CF}_2^{(3)})_3\cdot\text{CF}_2^{(4)}\cdot\text{CO}$ 	(1) -81.2 (2) -127.1 (3) -123.4 (4) -120.9		243
$C_{17}F_{16}HN$	$(\text{CF}_3^{(1)})_2\text{CF}^{(2)}\text{-C}_6\text{H}_4\text{(F)-CH(CN)-C}_6\text{H}_4\text{(F)-CF}_3^{(7)}$ (3) (4) (5) (6) (7)	(1) -74.5 (2) -177.2 (3) -134.5 (4) -136.8 (5) -139.3 (6) -160.9 (7) -152.0	(1-2) 5.8 (1-3) 13.5 (2-3) 46.5 (5-7) 3 (6-7) 21.2	273
$C_{17}F_{16}HN$	$\text{CF}_3^{(1)}\text{CF}_2^{(2)}\text{-C}_6\text{H}_4\text{(F)-CH(CN)-C}_6\text{H}_4\text{(F)-CF}_3^{(7)}$ (3) (4) (5) (6) (7)	(1) -85.1 (2) -110.2 (3) -136.8 (4) -136.8 (5) -136.8 (6) -139.3 (7) -56.2	(1-2) 2.2 (1-3) 6 (2-3) 31.7 (6-7) 22.8	273
$C_{17}F_{16}H_{22}Cl_2\text{OSi}_2$	$\text{CH}_3\cdot\text{SiCl}(\text{CH}_2\cdot\text{CH}_2\cdot\text{CF}_3^{(1)})\text{CH}_2\cdot\text{CH}_2\cdot\text{CF}_2^{(2)}\cdot\text{CF}_2^{(3)}\cdot\text{CF}_2^{(4)}\cdot\text{O}\cdot\text{CF}_2^{(5)}\cdot\text{CF}_2^{(6)}\cdot\text{CH}_2\cdot\text{CH}_2\text{SiCl}(\text{CH}_3)(\text{CH}_2\cdot\text{CH}_2\cdot\text{CF}_3^{(4)})$	(1)(7) -68.9 (2)(6) -116, -119 (3) -127 (4)(5) -83.5, -87.8		131
$C_{17}F_{16}\text{LiN}$	$\text{CF}_3^{(1)}\text{CF}_2^{(2)}\text{-C}_6\text{H}_4\text{(F)-CLi(CN)-C}_6\text{H}_4\text{(F)-CF}_3^{(7)}$ (3) (4) (5) (6) (7)	(1) -85.7 (2) -108.5 (3) -144.9 (4) -142.9 (5) -142.9 (6) -146.2 (7) -54.7	(2-3) 29.5 (6-7) 20.5	273
$C_{17}F_{16}\text{NNa}$	As above with Li replaced by Na	(1) -85.6 (2) -108.2 (3) -145.4 (4) -143.0 (5) -143.0 (6) -146.1 (7) -54.0		273

$C_{17}F_{16}NNa$		(1) -75.0 (2) -174.3 (3) -142.9 (4) -147.1 (5) -141.3 (6) -166.9 (7) -166.9	273
$C_{18}F_6H_{10}ClN$		-67.0	229(a)
$C_{18}F_6H_{10}ClN$	As above with H and C ₆ H ₅ interchanged	-69.0	229(a)
$C_{18}F_6H_{10}ClN$	$(CF_3^{(1)})_2CH^{(2)} \cdot N=C(C_6H_4Clp)C \equiv C \cdot C_6H_5$	-69.8	(1-2) 6.5 229(a)
$C_{18}F_6H_{10}S$		-57.61, -52.50	173(e)
$C_{18}F_6H_{11}N$		-66.9	229(a)
$C_{18}F_6H_{11}N$	As above with groups about C=C interchanged	-69.0	229(a)
$C_{18}F_6H_{11}N$	$(CF_3^{(1)})_2CH^{(2)} \cdot N=C(C_6H_5) \cdot C \equiv C \cdot C_6H_5$	-69.9	(1-2) 6.5 229(a)
$C_{18}F_6H_{12}ClNO_2S$		(1)(2) -77.6, -76.0	(1-2) 9 248(a)
$C_{18}F_6H_{12}N_2O_3S$	As above with Cl replaced by CH ₃ and OCH ₃ replaced by <i>m</i> NO ₂	(1)(2) -77.5, -76.2	(1-2) 9 248(a)
$C_{18}F_6H_{13}NOS$	As above with Cl replaced by CH ₃ and OCH ₃ replaced by H	(1)(2) -77.4, -75.9	(1-2) 9 248(a)
$C_{18}F_6H_{13}NO_2S$	As above with Cl replaced by H	(1)(2) -77.6, -76.0	(1-2) 9 248(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{18}F_6H_{14}N_4O_2S_4$		-61.91		168
$C_{18}F_6H_{14}N_4O_2Se$		-63.42		168
$C_{18}F_6H_{14}O_4$		(1)(2) -54.5, -57.5	(1-2) 8.8	267(a)
$C_{18}F_6H_{15}ClNO_3P$	$pClC_6H_4C \equiv NC(CF_3)_2P(OCH_3)_2C_6H_5O$	-64.9		261(a)
$C_{18}F_6H_{16}NO_3P$	$C_6H_5C \equiv NC(CF_3)_2P(OCH_3)_2C_6H_5O$	-65.4		261(a)
$C_{18}F_7H_{10}N$		(1) -67.2 (2) -103.8		229(a)
$C_{18}F_7H_{10}N$	As above with H and C_6H_5 interchanged	(1) -69.2 (2) -105.7		229(a)
$C_{18}F_7H_{10}N$	$(CF_3^{(1)})_2CH^{(3)} \cdot N = C(C_6H_4F^{(2)}p)C \equiv C \cdot C_6H_5$	(1) -69.9 (2) -104.8	(1-3) 6.5)	229(a)

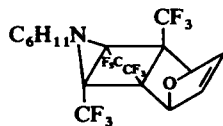
C₁₈F₁₂H₁₀N₆



-74.95, -71.20, -70.25,
-68.30

234(a)

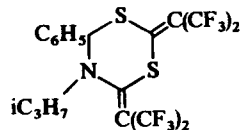
C₁₈F₁₂H₁₁NO



-60.1, -63.7

270(c)

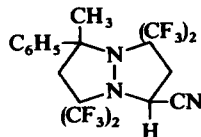
C₁₈F₁₂H₁₃NS₂



-57.67, -56.52, -55.93
-52.46

173(e)

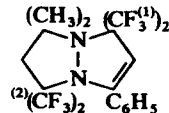
C₁₈F₁₂H₁₃N₃



-71.2, -70.4, -66.3,
(fourth signal obscured by signals
from other compounds)

254(a)

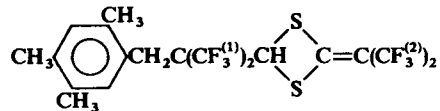
C₁₈F₁₂H₁₄N₂



(1) -70.75 (2) -69.0

254(a)

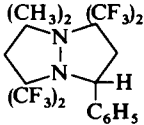
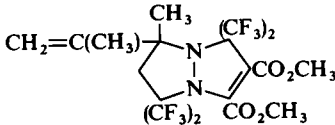
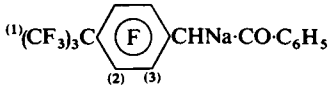
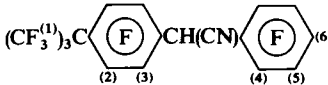
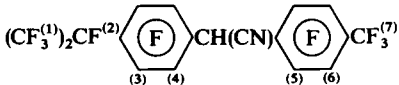
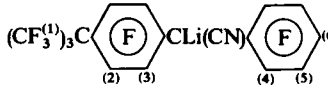
C₁₈F₁₂H₁₄S₂

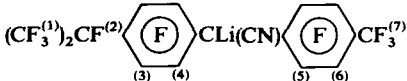
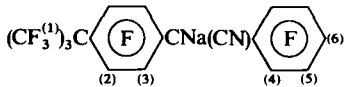
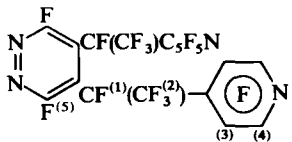
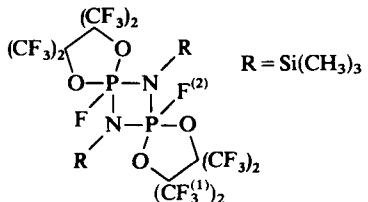
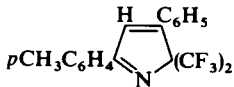


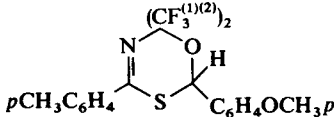
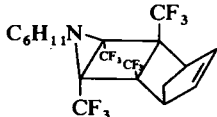
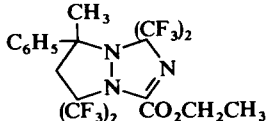
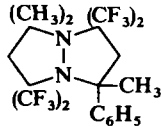
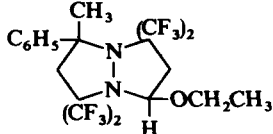
(1) -67.40 (2) -57.83

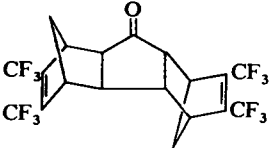
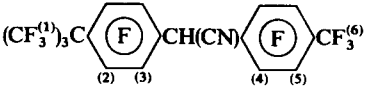
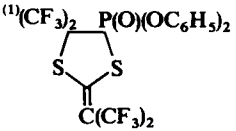
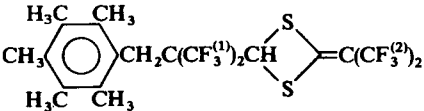
173(e)

Fluorine nuclei in a CF₃ group

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{18}F_{12}H_{16}N_2$		-73.1, -70.2, -67.0 (1:1:2)		254(a)
$C_{18}F_{12}H_{16}N_2O_4$		-71.75, -70.95, -69.55, -65.75		234(a)
$C_{18}F_{13}H_6NaO$		(1) -60.9 (3) -140.7	(2) -134.4 (1-2) 26	272(b)
$C_{18}F_{13}H_7O$	As above with Na replaced by H	(1) -60.4 (3) -138.8	(2) -129.4 (1-2) 26	272(b)
$C_{18}F_{18}HN$		(1) -60.2 (3) -136.3 (5) -161.1	(2) -126.5 (4) -139.5 (6) -152.2 (1-2) 26.4 (4-5) 20.7 (4-6) 3.5	273
$C_{18}F_{18}HN$		(1) -74.1 (3) -134.4 (5) -136.7 (7) -55.9	(2) -177.4 (4) -136.7 (6) -139.1 (1-2) 6 (2-3) 44.5 (1-3) 13.5 (6-7) 22.2	273
$C_{18}F_{18}H_{22}Cl_2O_2Si_2$	$[(CF_3^{(1)} \cdot CH_2 \cdot CH_2)CH_3SiCl(CH_2)_2CF_2^{(2)}CF_2^{(3)}OCF_2^{(4)}]$	(1) -69.0 (3)(4) -88.0, -89.0	(2) -118, -120	131
$C_{18}F_{18}LiN$		(1) -60.9 (3) -146.0 (5) -166.2	(2) -135.2 (4) -141.1 (6) -166.2 (1-2) 25.7 (2-3) 14.5	273

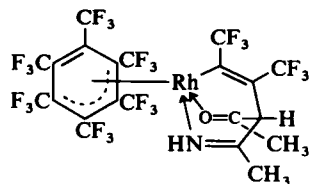
$C_{18}F_{18}LiN$		(1) -74.9 (3) -141.8 (5) -142.7 (7) -53.8	(2) -175.7 (4) -142.7 (6) -146.1	(1-2) 6 (2-3) 39.5	(1-3) 12.5 (6-7) 21.0	273
$C_{18}F_{18}NNa$	As above with Li replaced by Na	(1) -75.0 (3) -142.8 (5) -142.8 (7) -53.9	(2) -175.4 (4) -142.8 (6) -146.4			273
$C_{18}F_{18}NNa$		(1) -61.4 (3) -146.7 (5) -166.5	(2) -135.8 (4) -141.3 (6) -166.5			273
$C_{18}F_{18}N_4$		(1) -158.8 (3) -140.9 (5) -73.7	(2) -75.8 (4) -91.3			212
$C_{18}F_{24}H_9O_4P$	$[(CF_3)_2CHO]_4PC_6H_5$	-72.5				180
$C_{18}F_{26}H_{18}N_2O_4P_2Si_2$		(1) -67.1, -70.3 (2) -69.5		(1-2) 12.3 $^1J(P-F)$ 1018.5 $^3J(P-F)$ 23.7 $[^4J(F-F) + ^5J(F-F)]$ 8		281
$C_{19}F_3H_{24}ClOP_2Rh$	$RhCl[(CF_3^{(1)}CH_2^{(2)})CO][P(CH_3)_2C_6H_5]_2$	-56.05		(1-2) 15.0	(Rh-F) 3.8	279
$C_{19}F_3H_{22}ClOP_2Rh$	As above with CH_2 replaced by $CF_2(2)$	(1) -79.2 (2) -62.3		(P-F) 26		279
$C_{19}F_6H_{13}N$		-66.9				229(a)
$C_{19}F_6H_{13}N$	As above with H and C_6H_5 group interchanged	-69.2				229(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{19}F_6H_{13}N$	$(CF_3^{(1)})_2CH^{(2)} \cdot N=C(pCH_3C_6H_4) \cdot C \equiv C \cdot C_6H_5$	-69.9	(1-2) 6.5	229(a)
$C_{19}F_6H_{15}NO_2S$		-77.7, -76.2	(1-2) 9	248(a)
$C_{19}F_6H_{18}NO_3P$	$pCH_3C_6H_4C \equiv N - C(CF_3)_2P(OCH_3)_2C_6H_5O$	-64.9		261(a)
$C_{19}F_6H_{18}NO_4P$	As above with pCH_3 replaced by pCH_3O	-65.5		261(a)
$C_{19}F_{12}H_{13}N$		-60.4, -62.1		270(c)
$C_{19}F_{12}H_{15}CoO$	$[\eta-C_5(CH_3)_5]Co[C_4(CF_3)_4CO]$	-51.84, -54.40		277
$C_{19}F_{12}H_{15}N_3O_2$		-75.30, -72.15, -71.20, -69.00		234(a)
$C_{19}F_{12}H_{15}ORh$	$[\eta-C_5(CH_3)_5]Rh[C_4(CF_3)_4CO]$	-52.34, -55.35		277
$C_{19}F_{12}H_{18}N_2$		-70.1, -69.6, -66.0, -65.0	(F-H) 11.0, 11.0	254(a)
$C_{19}F_{12}H_{18}N_2O$		-72.65, -72.2, -67.35, -63.9	(F-H) 10.5, 12.0	254(a)

C ₁₉ F ₁₂ O		-60.3, -60.4	(F-F) 7	210
C ₁₉ F ₂₀ HN		(1) -60.1 (2) -126.5 (3) -135.7 (4) -137.1 (5) -137.1 (6) -139.2	(1-2) 26.5 (5-6) 23.0	273
C ₁₉ F ₂₀ LiN	As above with H replaced by Li	(1) -60.5 (2) -134.0 (3) -142.0 (4) -142.3 (5) -146.1 (6) -53.9	(1-2) 25.9 (2-3) 11.0 (5-6) 21.5	273
C ₁₉ F ₂₀ NNa	As above with H replaced by Na	(1) -61.1 (2) -134.7 (3) -142.5 (4) -142.5 (5) -146.6 (6) -54.0		273
C ₂₀ F ₅ H ₂₄ ClOP ₂ Rh	RhClI(CF ₃ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CH ₂ ⁽³⁾)CO[P(CH ₃) ₂ C ₆ H ₅] ₂	(1) -85.1 (2) -101.3	(2-3) 26.0	279
C ₂₀ F ₆ H ₂₈ NO ₄ P	$p\text{CH}_3\text{C}_6\text{H}_4\text{C}=\text{N}-\text{C}(\text{CF}_3)_2\text{P}[\text{OCH}(\text{CH}_3)_2]_3\text{O}$	-65.8	(P-F) 4.0	261(a)
C ₂₀ F ₆ H ₂₈ NO ₅ P	As above with <i>p</i> CH ₃ replaced by <i>p</i> CH ₃ O	-65.9	(P-F) 4.0	261(a)
C ₂₀ F ₇ H ₂₂ ClOP ₂ Rh	RhClI(CF ₂ ⁽¹⁾ ·CF ₂ ⁽²⁾ ·CF ₃ ⁽³⁾)CO[P(CH ₃) ₂ C ₆ H ₅] ₂	(1) -65.6 (2) -115.3 (3) -79.1	(1-3) 12	279
C ₂₀ F ₁₂ H ₁₁ O ₃ PS ₂		(1) -64.84, -62.93 (2) -57.45		173(e)
C ₂₀ F ₁₂ H ₁₈ S ₂		(1) -64.67 (2) -67.40		173(e)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{20}F_{26}H_{18}N_2O_4P_2$	As $C_{18}F_{26}H_{18}N_2O_4P_2Si_2$ above with $R = C(CH_3)_3$	(1) -66.5 , -69.9 (2) -70.4	(1-2) 12.3 $^1J(P-F)$ 1032.3 $^3J(P-F)$ 32.6 $[^4J(F-F) + ^5J(F-F)]$ 8	281
$C_{21}F_6H_{26}N_2OSn_2$	$[mCF_3C_6H_4N \cdot Sn(CH_3)_3]_2CO$	-62.87		168
$C_{21}F_7H_{24}BrIOP_2Rh$	$RhBrI(CF_3^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CH_2^{(4)})CO[P(CH_3)_2C_6H_5]_2$	(1) -80.8 (2) -126.2 (3) -97.7	(1-3) 10 (3-4) 28.0	279
$C_{21}F_7H_{24}ClIOP_2Rh$	As above with Br replaced by Cl	(1) -80.9 (2) -126.4 (3) -97.5	(1-3) 10 (3-4) 28.0	279
$C_{21}F_{11}H_{16}N$	$CF_3^{(1)} \cdot CF_2^{(2)} \cdot (CF_2^{(3)})_2 \cdot CF_2^{(4)}$ 	(1) -80.0 (2) -125.2 (3) -121.1 (4) -102.1		243
$C_{21}F_{12}H_{10}N_6$		-74.95 , -71.20 , -70.25 -68.30		234(a)
$C_{21}F_{12}H_{16}N_2O_4$		(1)(2) -71.70 , -68.85 , -64.05 (2:1:1)		234(a)
$C_{21}F_{12}H_{22}N_2O$		-73.0 , -72.2 , -67.05 , -63.5		254(a)

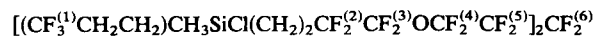
C₂₁F₂₄H₇NORh



−77.35, −74.75, −71.30,
−68.63, −67.06
(ratio 3:1:2:1:1)

284(c)

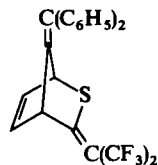
C₂₁F₂₄H₂₂Cl₂O₂Si₂



(1) −68.9 (2) −120
(3)(4) −83.0, −87.0
(5) −126 (6) −123

131

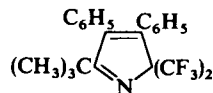
C₂₂F₆H₁₄S



−58.7, −54.8

173(e)

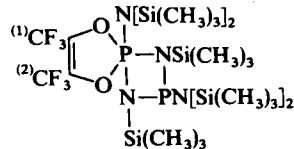
C₂₂F₆H₁₉N



−67.5

229(a)

C₂₂F₆H₅₄N₄O₂P₂Si₆

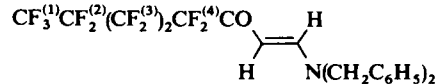


(1)(2) −58.5, −63.6

(1-2) 12.7

250

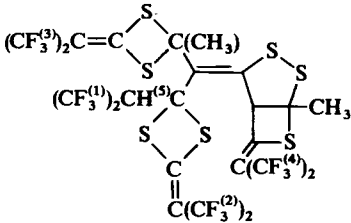
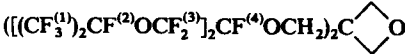
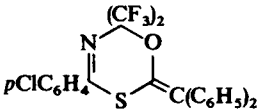
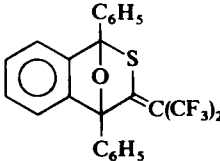
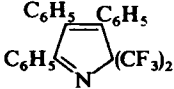
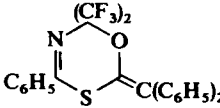
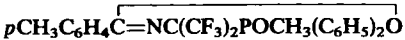
C₂₂F₁₁H₁₆NO

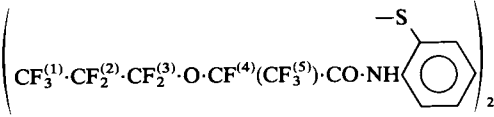
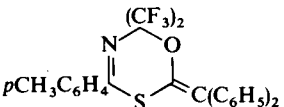
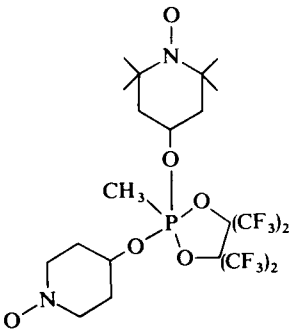
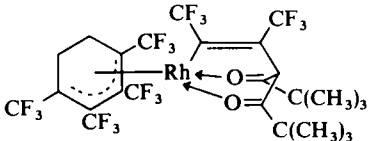


(1) −82.6 (2) −127.8
(3) −124.2 (4) −121.7

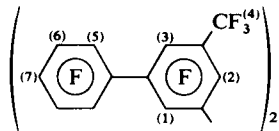
243

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{22}F_{13}H_{16}N$		(1) -80.4 (2) -124.6 (3) -120.4 (4) -102.9		243
$C_{22}F_{18}H_{15}Co$	$[\eta-C_5(CH_3)_5]Co[C_6(CF_3)_6]$	-50.64, -51.36, -56.91 (1:1:1)		277
$C_{22}F_{18}H_{15}Rh$	As above with Co replaced by Rh	-51.75, -51.78, -56.34 (1:1:1)		277
$C_{23}F_6H_{17}ClNO_2P$	$pClC_6H_4C \equiv NC(CF_3)_2P(OCH_3)(C_6H_5)_2O$	-63.6	(P-F) 3.2	261(a)
$C_{23}F_6H_{18}NO_2P$	As above with Cl replaced by H	-63.9	(P-F) 3.2	261(a)
$C_{23}F_6H_{18}NO_5P$	$CH_3OC \equiv NC(CF_3)_2P(OC_6H_5)_3O$	-66.5	(P-F) 3.5	261(a)
$C_{23}F_{12}H_{15}NO_2S_2$		(1)(2) -58.9, -56.7 (3)(4) -55.5, -54.3		173(e)
$C_{23}F_{12}H_{16}N_2$		-70.5, -69.0, -68.65, -67.6		254(a)
$C_{23}F_{12}H_{18}N_2$		[a] -70.4, -69.7, -66.1 (1:1:2) [b] -70.6, -68.9, -66.7, -65.3		254(a)
$C_{23}F_{13}H_{16}NO$		(1) -81.6 (2) -127.1 (3) -123.7 (4) -120.9		243

$C_{23}F_{24}H_8S_7$		(1) -64.37 (2)(3) -57.67, -57.56 (4) -57.74	(1-5) 6	173(e)
$C_{23}F_{38}H_8O_7$		(1) -81.4 (2) -146.0 (3) -78.8 (4) -142.6	(2-3) 21.8	143
$C_{24}F_6H_{14}ClNOS$		-76.9		248(a)
$C_{24}F_6H_{14}OS$		-60.56, -53.5		173(e)
$C_{24}F_6H_{15}N$		-66.7		229(a)
$C_{24}F_6H_{15}NOS$		-76.80		248(a)
$C_{24}F_6H_{20}NO_2P$		-63.6	(P-F) 3.0	261(a)
$C_{24}F_6H_{20}NO_3P$	As above with pCH_3 replaced by pCH_3O	-64.0	(P-F) 3.0	261(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{24}F_{18}H_{16}Ir_2O_2$	$[\eta-C_5(CH_3)_5]Ir_2(CO)_2(CF_3C\equiv CCF_3)_3H$	-47.4, -52.7, -57.92, -58.4 (2:2:1:1)	(3-4) 9 (3-H) 3 (4-H) 8	277
$C_{24}F_{22}H_{10}N_2O_4S_2$		(1)(5) -81.9, -82.5 (2) -130.0 (3) -79.8, -85.3 (4) -133.0	(3A-3B) 142	252
$C_{25}F_6H_{17}NOS$		-76.9		248(a)
$C_{25}F_{12}H_{37}N_2O_6P$		-70.3		286
$C_{25}F_{18}H_{22}O_2Rh$		-74.19, -72.15, -70.45, -67.14 (1:2:1:2)		284(c)
$C_{26}F_8H_{17}N_2O_2P_2Ru$	$[Ru(O_2C\cdot CF_3^{(1)})_2\{PF_2^{(2)}(N(CH_3)_2)_2\}P(C_6H_5)]$	(1) -76.5 (2) +1.7	$[^1J(P-1)+^3J(P-1)]$ 1189	288

C₂₆F₂₄

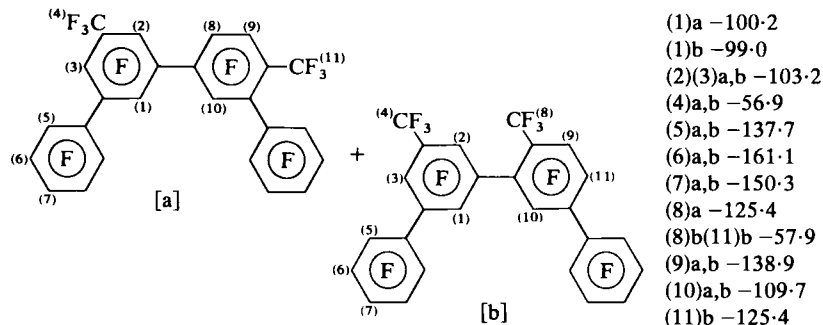


(1) -99.4 (2)(3) -103.3
(4) -57.2 (5) -137.5
(6) -161.1 (7) -150.2

(2-4) = (3-4) 22

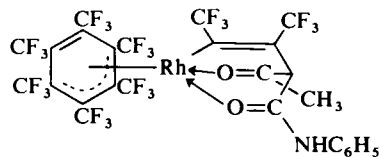
262(b)

C₂₆F₂₄



(1)a -100.2 (8-9)a = (9-11)b = 21
(1)b -99.0 (8-10)a = (10-11)b = 2.5
(2)(3)a,b -103.2
(4)a,b -56.9
(5)a,b -137.7
(6)a,b -161.1
(7)a,b -150.3
(8)a -125.4
(8)b(11)b -57.9
(9)a,b -138.9
(10)a,b -109.7
(11)b -125.4

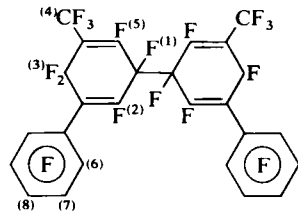
C₂₆F₂₄H₁₀NO₂Rh



-75.8, -74.3, -73.34,
-69.6, -67.68 (3:1:1:2:1)

284(c)

C₂₆F₂₆



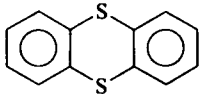
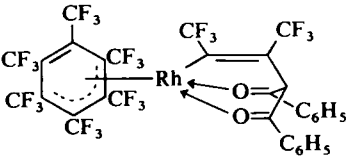
(1) -174.8
(2)(5) -108.8, -103.3
(3) -97.0, -90.4
(4) -58.6 (6) -137.5
(7) -160.1 (8) -148.7

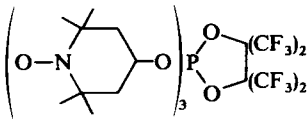
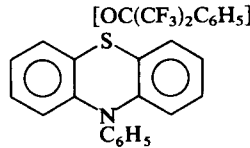
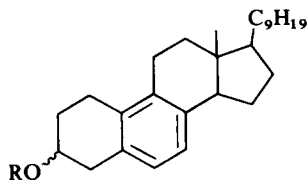
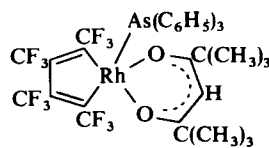
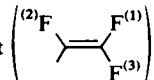
262(b)

C₂₇F₁₃H₂₁As₅FeN₂O₂P [CF₃⁽¹⁾·CF₂⁽²⁾·CF₂⁽³⁾·Fe(CO)₂(CH₃CN)₂As(C₆H₅)₃]PF₆⁽⁴⁾

(1) -78.8 (2) -117.5
(3) -79.6 (4) -73.5

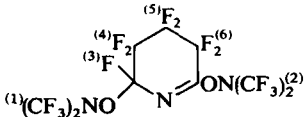
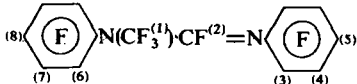
215

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{27}F_{13}H_{21}FeN_2O_2P_2$	As above with As replaced by P	(1) -80.2 (2) -117.2 (3) -79.1 (4) -73.1		215
$C_{27}F_{13}H_{21}FeN_2O_5P_2$	As above with As replaced by P and C_6H_5 by OC_6H_5	(1) -83.1 (2) -117.8 (3) -79.8 (4) -73.5	(1-3) 11.2	215
$C_{28}F_6H_{20}NO_5P$	$C_6H_5OC=NC(CF_3)_2P(OC_6H_5)_3O$	-66.5	(P-F) 3.5	261(a)
$C_{29}F_6H_{22}NO_5P$	$pCH_3OC_6H_4C=NC(CF_3)_2P(OC_6H_5)_3O$	-65.2	(P-F) 3.3	261(a)
$C_{29}F_{12}H_{30}ORh_2$	$[\eta-C_5(CH_3)_5]Rh(CF_3C\equiv CCF_3)_2CO$	-52.30, -55.48 (1:1)		277
$C_{30}F_6H_{22}NO_4P$	$C_6H_5CH=CHC=NC(CF_3)P(OC_6H_5)_3O$	-65.2	(P-F) 3.4	261(a)
$C_{30}F_6H_{22}N_2NiO_4$	$Ni(C_6H_5\cdot CO\cdot CH\cdot CO\cdot CF_3)_2\cdot 2C_5H_5N$	-51.2		289
$C_{30}F_{12}H_{18}O_2S_2$	$[OC(CF_3)_2C_6H_5]_2$ 	-70.1		290
$C_{30}F_{12}H_{18}O_3S$	As above with lower -S- replaced by -O-	-70.1		290
$C_{30}F_{12}H_{30}O_2Rh_2$	$[\eta-C_5(CH_3)_5]Rh_2(CF_3C\equiv CCF_3)_2(CO)_2$	-47.73, -51.51 (1:1)		277
$C_{31}F_{24}H_{11}O_2Rh$		-75.60, -73.75, -69.45, -59.80 (3:2:2:1)		284(c)
$C_{32}F_6H_{26}N_2NiO_4$	$Ni(C_6H_5\cdot CO\cdot CH\cdot CO\cdot CF_3)_2\cdot 2CH_3C_5H_4N(\beta)$	-52.5		289
$C_{32}F_6H_{26}N_2NiO_4$	As above with $CH_3C_5H_4N(\beta)$ replaced by $CH_3C_5H_4N(\gamma)$	-53.0		289

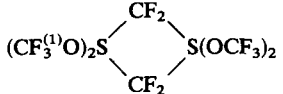
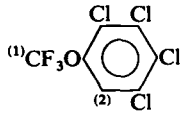
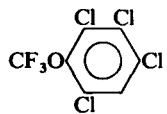
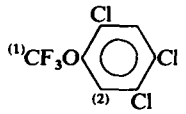
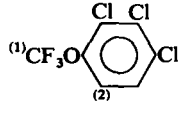
$C_{33}F_{12}H_{51}N_3O_8P$		-69.3			286
$C_{36}F_{12}H_{23}NO_2S$	 [OC(CF ₃) ₂ C ₆ H ₅] ₂	-70.1			290
$C_{37}F_3H_{49}O_3$	 R = C ₆ H ₅ -C(OCH ₃)(CF ₃)-CO (+)	RO  -74.75 RO  -74.55			291
$C_{37}F_{12}H_{34}AsO_2Rh$		-75.38, -70.06			284(c)
$C_{39}F_5H_{30}ClOPt$	<i>cis</i> -[PtCl(CF ₂ ⁽¹⁾ COCF ₃ ⁽²⁾)(P(C ₆ H ₅) ₃) ₂]	(1) -92.5 (2) -75.7	(1-2) 6 (Pt-2) 9 (P-1) <i>trans</i> 36	(Pt-1) 360 (P-1) <i>cis</i> 26	292(c)
$C_{40}F_6H_{30}O_2P_2Pt$	<i>trans</i> -[(C ₆ H ₅) ₃ P] ₂ Pt  (OCOCF ₃ ⁽⁴⁾)	(1) -100.2 (2) -130.1 (3) -148.5 (4) -75.5	(1-2) 90.1 (2-3) 106.2 (P-2) 4.0	(1-3) 28.9 (P-1) 5.0 (P-3) 2.0	334

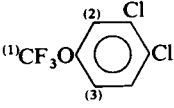
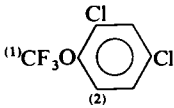
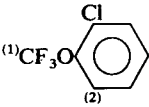
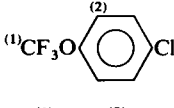
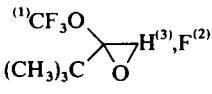
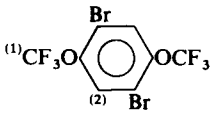
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₄₀ F ₉ H ₃₀ O ₄ P ₃ Ru	[Ru(O ₂ CCF ₃ ⁽¹⁾) ₂ (PF ₃ ⁽²⁾)(P(C ₆ H ₅) ₃) ₂]	(1) -75.5 (2) -8.6	(P-2) 1283	288
C ₄₀ F ₁₁ H ₃₀ O ₄ P ₄ Ru ⁺	[Ru(O ₂ CCF ₃ ⁽¹⁾){PF ₂ ⁽³⁾ (O ₂ CCF ₃ ⁽²⁾)}(PF ₃ ⁽⁴⁾)(P(C ₆ H ₅) ₃) ₂] ⁺	(1) -78.1 (2) -77.2 (3) +3.9 (4) -5.9	(P-3) 1206 (P-4) 1276	288
C ₄₂ F ₈ H ₃₆ NO ₂ P ₃ Ru	[Ru(O ₂ CCF ₃ ⁽¹⁾) ₂ {PF ₂ ⁽²⁾ (N(CH ₃) ₂)}(P(C ₆ H ₅) ₃) ₂]	(1) -77.0 (2) -7.8	(P-2) 1177	288
C ₄₄ F ₂₆ H ₃₂ Fe ₂ N ₄ O ₄ P ₄	{[CF ₁ ⁽³⁾ CF ₂ ⁽²⁾ CF ₂ ⁽³⁾ Fe(CO) ₂ (CH ₃ CN) ₂] ₂ (C ₆ H ₅) ₂ PCH ₂ CH ₂ P(C ₆ H ₅) ₂ }(PF ₆ ⁽⁴⁾) ₂	(1) -80.3 (2) -115.5 (3) -77.7 (4) -71.0	(1-3) 11.3	215
C ₁₈₈ F ₁₈₀ H ₂₃₈ O ₃₁ Si ₂₂	CH ₂ =CHSi(CH ₃) ₂ O[Si(CH ₃)(CH ₂ CH ₂ CF ₃ ⁽¹⁾)(CH ₂) ₂ CF ₂ ⁽²⁾ CF ₂ ⁽³⁾ OCF ₂ ⁽⁴⁾ CF ₂ OCF ₂ CF ₂ (CH ₂) ₂ Si(CH ₃) ·(CH ₂ CH ₂ CF ₃)O] ₁₀ Si(CH ₃) ₂ CH=CH ₂	(1) -88.4 (2) -119.8 (3)(4) -90.0		131
<i>2. CF₃ group bonded to iodine</i>				
CF ₃ ClINO ₃	CF ₃ ·ICl(NO ₃)	-28.4		9
CF ₃ Cl ₂ I	CF ₃ ·ICl ₂	-30.9		9
CF ₃ I	CF ₃ I	-7.6		9
CF ₃ IN ₂ O ₆	CF ₃ ·I(NO ₃) ₂	-25.4		9
CF ₄ INO ₃	CF ₃ ⁽¹⁾ ·IF ⁽²⁾ (NO ₃)	(1) -29.3 (2) -165.2	(1-2) 8.3	9
CF ₅ I	CF ₃ ⁽¹⁾ ·IF ₂ ⁽²⁾	(1) -33.7 (2) -175.7	(1-2) 6.5	9
<i>3. CF₃ group bonded to nitrogen</i>				
CF ₃ H ₂ N	CF ₃ ⁽¹⁾ ·NH ₂ ⁽²⁾	-48.9	(1-2) 10.4	13
C ₂ F ₃ N	CF ₃ ·NC	-51.2		31

C ₂ F ₅ NO	CF ₃ ⁽¹⁾ ·N·CF ₂ ⁽²⁾ ·O	(1) -67.9 (2B) -108.1	(2A) -91.8	(1-2A) 3.1 (2A-2B) 1.8(?) ¹ J(1-C) 293 ¹ J(2A-C) 295 ¹ J(2B-C) 280	(1-2B) 15.8	39
C ₂ F ₁₀ HNO ₂ S	• CF ₃ ⁽¹⁾ ·NH ⁽⁵⁾ ·CF ₂ ⁽²⁾ ·OO·SF ₄ ⁽³⁾ F ⁽⁴⁾	(1) -55.9 (3) +56.1	(2) -70.0 (4) +59.2	(1-2) 8.5 (3-4) 154.5	(1-5) 4.2	39
C ₃ F ₈ HNO ₂	CF ₃ ⁽¹⁾ ·NH ⁽⁴⁾ ·CF ₂ ⁽²⁾ ·OO·CF ₃ ⁽³⁾	(1) -55.9 (3) -68.7	(2) -70.2	(1-2) 8.5 (1-4) 4.25	(2-3) 4.0	39
C ₃ F ₉ NS	(CF ₃ ⁽¹⁾) ₂ N·SCF ₃ ⁽²⁾	(1) -56.9	(2) -52.6	(1-2) 3.45		38
C ₃ F ₉ N ₃ O	(CF ₃ ⁽¹⁾) ₂ N·N(CF ₃ ⁽²⁾)NO	(1) -59.3	(2) -62.5			76
C ₃ F ₉ N ₃ O ₂	(CF ₃ ⁽¹⁾) ₂ N·N(CF ₃ ⁽²⁾)NO ₂	(1) -61.0	(2) -59.8			76
C ₄ F ₆ H ₅ NO	(CF ₃) ₂ N·CH ₂ ·CH ₂ ·OH	-59.7				91
C ₄ F ₁₀ ClN	(CF ₃ ⁽¹⁾) ₂ N·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ Cl	(1) -54.7 (3) -72.7	(2) -94.5	(1-2) 16.1 (2-3) ≤1.7	(1-3) 8.2	99(a)
C ₅ F ₃ H ₁₀ N	CF ₃ ·N(CH ₂ ·CH ₃) ₂	-60.5				111
C ₅ F ₇ H ₄ N	⁽¹⁾ (CF ₃) ₂ N ⁽²⁾ F ⁽³⁾ H CH ₃	(1) -60.5	(2) -97.9	(1-2) 4.0	(2-3) 25.2	126(a)
C ₅ F ₇ H ₄ N	⁽¹⁾ (CF ₃) ₂ N ⁽²⁾ F ⁽⁴⁾ CH ₃ ⁽³⁾ H	(1) -59.0	(2) -92.9	(1-2) 3.5 (2-4) 3.5	(2-3) 6.0	126(a)
C ₅ F ₇ H ₄ N	⁽¹⁾ (CF ₃) ₂ N ⁽⁴⁾ CH ₃ ⁽²⁾ F ⁽³⁾ H	(1) -58.5	(2) -127.8	(2-3) 77.4	(2-4) 5.7	126(a)
C ₅ F ₇ H ₄ N	⁽¹⁾ (CF ₃) ₂ N ⁽⁴⁾ CH ₃ ⁽³⁾ H ⁽²⁾ F	(1) -58.5	(2) -130.3	(2-3) 78.8	(2-4) 5.2	126(a)

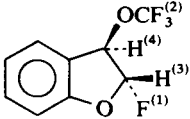
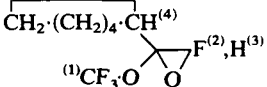
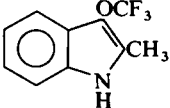
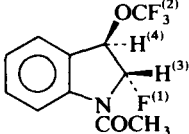
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_3F_7H_3BrN$	(1 <i>R</i> , 2 <i>S</i>)- $CH_3^{(5)} \cdot CH^{(4)}Br \cdot CF^{(2)}H^{(3)} \cdot N(CF_3^{(1)})_2$	(1) -56.7	(2) -149.5	(1-2) 6.3 (2-4) 6.3	(2-3) 45.0	126(a)
	(1 <i>R</i> , 2 <i>R</i>)-isomer	(1) -56.1	(2) -145.3	(1-2) 6.7 (2-5) 1.7	(2-3) 46.2	126(a)
$C_3F_7H_3BrN$	(1 <i>R</i> , 2 <i>S</i>)- $CF^{(2)}BrH^{(3)} \cdot CH^{(4)}(CH_3) \cdot N(CF_3^{(1)})_2$	(1) -56.2	(2) -143.2	(1-2) 3.8 (2-4) 12.6	(2-3) 48.0	126(a)
	(1 <i>R</i> , 2 <i>R</i>)-isomer	(1) -55.8	(2) -141.7	(1-2) 3.5 (2-4) 7.0	(2-3) 51.0	126(a)
$C_3F_{11}Cl_2N$	$CF_3^{(1)} \cdot N(CF_2^{(2)} \cdot CF_2^{(3)}Cl)_2$	(1) -51.5 (3) -70.3	(2) -89.8	(1-2) 15.9 (2-3) 6.4	(1-3) 10.6	99(a)
$C_3F_{12}ClN$	$(CF_3^{(1)})_2N \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)}Cl$	(1) -54.5 (3) -121.3	(2) -91.2 (4) -70.4	(1-2) 15.6 (2-4) 14.1	(1-3) 8.4	99(a)
$C_6F_{12}H_3ClN_2O_2$	$(CF_3^{(1)(2)})_2NO \cdot CHCl \cdot CH_2 \cdot ON(CF_3^{(3)})_2$	(1) -67.6 (3) -70.0	(2) -69.2			133
$C_6F_{13}H_3N_2O_2$	$(CF_3^{(1)(2)})_2NO \cdot CHF^{(3)} \cdot CH_2 \cdot ON(CF_3^{(4)})_2$	(1) -68.65 (3) -136.0	(2) -69.8 (4) -69.9			133
$C_6F_{15}HN_2O_2$	$(CF_3^{(1)(2)})_2NO \cdot CHF^{(3)} \cdot CF_2^{(4)} \cdot ON(CF_3^{(5)})_2$	(1) -69.0 (3) -143.8 (5) -68.6	(2) -70.1 (4) -91.3			133
$C_9F_{19}N_3O_2$		(1)(2) -69.1, -70.1 (3) -97.5 (4)(5) -134.5 to -137.7 (6) -122.2				179(a)
$C_{14}F_{14}N_2$		(1) -55.9 (3) -149.9 (5) -160.4 (7) -160.4	(2) -40.9 (4) -163.4 (6) -144.0 (8) -148.5	(1-2) 16.5		271(b)

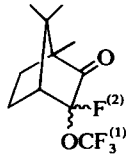
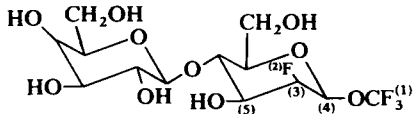
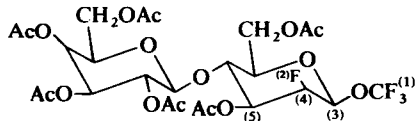
4. CF₃ group bonded to oxygen

CF ₃ HO	CF ₃ ·OH	−54·5	(F-C) 256	12
C ₄ F ₁₀ O ₂	CF ₃ ⁽¹⁾ ·O·CF ₂ ⁽²⁾ ·CF ₂ ·OCF ₃	(1) −59·4 (2) −94·0	(1-2) 10	104
C ₄ F ₁₂ O ₂ S	(CF ₃ ⁽¹⁾ O) ₂ S(CF ₃ ⁽²⁾) ₂	(1) −73·6 (2) −56·2	(1-2) 7·2	107
C ₄ F ₁₂ O ₃ S	(CF ₃ ⁽¹⁾ O) ₂ S(O)(CF ₃ ⁽²⁾) ₂	(1) −74·1 (2) −67·4	(1-2) 9·6	107
C ₅ F ₃ H ₁₁ O ₃ SSi	CF ₃ O·SO ₂ ·CH ₂ ·Si(CH ₃) ₃	−74·3		80
C ₆ F ₁₄ O ₃	CF ₃ ⁽¹⁾ ·O·(CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·O) ₂ ·CF ₃	(1) −59·3 (2) −93·8 (3) −91·8	(1-2) 9	104
C ₆ F ₁₆ O ₄ S ₂		(1) −72·1 (2) −96·2	(1-2) 12·1	107
C ₇ F ₃ HCl ₄ O		−57·0	(1-2) 1·3	165
C ₇ F ₃ HCl ₄ O		−56·6		165
C ₇ F ₃ H ₂ Cl ₃ O		−59·2	(1-2) 1·3	165
C ₇ F ₃ H ₂ Cl ₃ O		−58·6	(1-2) 1·3	165

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_3H_3Cl_2O$		-58.7	(1-2)(1-3) 0.7	165
$C_7F_3H_3Cl_2O$		-58.8	(1-2) 1.3	165
$C_7F_3H_4ClO$		-58.8	(1-2) ~1	165
$C_7F_3H_4ClO$		-58.4	(1-2) 1.3	165
$C_7F_4H_{10}O$	$CF_3^{(1)}O \cdot CF^{(2)}H \cdot CO \cdot C(CH_3)_3$	(1) -61.5 (2) -137.5		134
$C_7F_4H_{10}O_2$		(1) -53.6 (2) -163.1	(1-2) 8.8 (2-3) 85.0	134
$C_8F_6H_2Br_2O_2$		-58.6	(1-2) 1.2	165
$C_8F_6H_2Br_2O_2$		-59.9		165

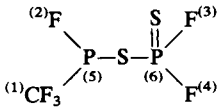
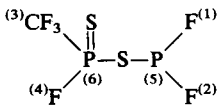
$C_8F_6H_2Br_2O_2$		-58.6	(1-2) 1.3	165
$C_8F_6H_2Cl_2O_2$		-58.9	(1-2) 1.4	165
$C_8F_6H_2Cl_2O_2$		-58.8		165
$C_8F_6H_2Cl_2O_2$		-59.2	(1-2) 1.2	165
$C_8F_6H_2Cl_2O_2$		(1) -58.6 (2) -56.6	(1-3) 1.2	165
$C_8F_8O_3$		(1) -69.5 (2) -153.9 (3) -161.9 (4) -156.1	(2-3) 15.6 (3-4) 20.7	429
$C_9F_3H_5O_2$		-61		495
$C_9F_3H_6NO$		-61		495

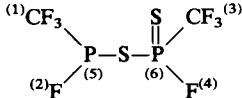
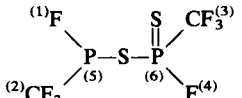
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_9F_4H_6O_2$		(1) -138	(2) -59	(1-3) 60	(1-4) 15	495
$C_9F_4H_6O_2$	As above with groups (2) and (4) interchanged	(1) -140	(2) -60	(1-3) 62 (1-2) 3.5	(1-4) 17	495
$C_9F_4H_6O_2$	$CF_3^{(1)} \cdot O \cdot CF^{(2)}H^{(3)} \cdot CO \cdot$ 	(1) -60.2	(2) -135.5	(1-2) 4.9 (2-4) 0.9	(2-3) 56	134
$C_9F_4H_{12}O$	$CF_3^{(1)} \cdot O \cdot CF^{(2)}H^{(3)} \cdot CO \cdot \overbrace{CH^{(4)} \cdot (CH_2)_4 \cdot CH_2}$	(1) -60.7	(2) -138.2	(1-2) 4.6 (2-4) 1.6	(2-3) 56	134
$C_9F_4H_{12}O$		(1) -54.8	(2) -163.0	(1-2) 3.9 (2-4) 1.7	(2-3) 84.0	134
$C_{10}F_3H_8NO$		-60				495
$C_{10}F_4H_8O$	$CF_3^{(1)} \cdot O \cdot CF^{(2)}H^{(3)} \cdot CO \cdot CH_2^{(4)} \cdot C_6H_5$	(1) -60.0	(2) -137.5	(1-2) 4.9 (2-4) 1.5	(2-3) 56	134
$C_{11}F_4H_9NO_2$		(1) -139	(2) -59	(1-3) 70		495
$C_{11}F_4H_9NO_2$	As above with groups (2) and (4) interchanged	(1) -154	(2) -61	(1-3) 70		495

$C_{11}F_4H_{14}O_2$		[a] (1) -52.8 (2) -113 [b] (1) -53.7 (2) -115.4	(1-2) 9.4 (1-2) 9.5	134
$C_{12}F_{26}O_6$	$CF_3^{(1)} \cdot O \cdot (CF_2^{(2)} \cdot CF_2^{(3)} \cdot O)_5 \cdot CF_3$	(1) -58.9 (2) -93.5 (3) -91.5	(1-2) 10	104
$C_{13}F_4H_{19}O_9$		(1) -74.8 (2) -238.1	(2-3) 50 (2-4) 18 (2-5) 32	260
$C_{14}F_{30}O_7$	$CF_3^{(1)} \cdot O \cdot (CF_2^{(2)} \cdot CF_2^{(3)} \cdot O)_6 \cdot CF_3$	(1) -59.0 (2) -93.6 (3) -91.4	(1-2) 9	104
$C_{25}F_4H_{32}O_{16}$		(1) -59.5 (2) -219.9	(2-3) 16 (2-4) 51 (2-5) 27	260
$C_{25}F_4H_{32}O_{16}$	As above with OCF_3 axial and F equatorial	(1) -58.9 (2) -202.0	(2-3) 12 (2-4) 48	260

5. CF_3 group bonded to phosphorus

CF_3Br_2P	$CF_3 \cdot PBr_2$	-67.0	(F-C) 323.6 (P-F) 67.9	6
CF_3Br_4P	$CF_3 \cdot PBr_4$	-66.8	(P-F) 142	8
CF_3Cl_2P	$CF_3 \cdot PCl_2$	-72.8	(F-C) 323.0 (P-F) 79.5	6
CF_3Cl_4P	$CF_3 \cdot PCl_4$		(P-F) 151	8
CF_3I_2P	$CF_3 \cdot PI_2$	-57.9	(F-C) 324.5 (P-F) 48.4	6
CF_4HPS_2	$(CF_3^{(1)})F^{(2)}PS_2H$	(1) -75.0 (2) -54.1	(1-2) 3.1 (P-1) 132 (P-2) 1169	14

Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
CF ₃ Cl ₂ P	CF ₃ ⁽¹⁾ ·PF ₂ Cl ₂			(P-1) 155		8
CF ₃ P	CF ₃ ⁽¹⁾ ·PF ₂ ⁽²⁾	(1) -84.3	(2) -106.7	(1-2) 5.92 (2-C) 21.8 (P-2) 1240	(1-C) 318.4 (P-1) 88.4	6
CF ₆ P ₂ S ₂		(1) -64.1 (3) -16.1	(2) -167.1 (4) -18.0	(1-2) 4.2 (1-5) 72.6 (2-4) 1.3 (2-6) 1.2 (3-5) 0.5 (4-5) 16.3	(1-4) 1.3 (2-3) 8.8 (2-5) 1154.0 (3-4) 82.5 (3-6) 1224.0 (4-6) 1203.0	14
CF ₆ P ₂ S ₂		(1) -59.3 (3) -80.1	(2) -63.7 (4) -56.3	(1-2) 92.5 (1-5) 1370.0 (2-4) 7.0 (2-6) 23.0 (3-6) 136.0 (4-6) 1174.0	(1-4) 6.0 (1-6) 9.0 (2-5) 1291.0 (3-4) 2.6 (4-5) 22.0	14
CF ₇ P	CF ₃ ⁽¹⁾ ·PF ₄			(P-1) 170		8
C ₂ F ₃ H ₃ ClOP	CF ₃ ·PCI·OCH ₃	-72.3		(P-F) 73.3		6
C ₂ F ₃ H ₃ IOP	CF ₃ ⁽¹⁾ ·PI·OCH ₃ ⁽²⁾	-65.5		(1-2) 0.8 (P-1) 61.9		6
C ₂ F ₃ H ₄ OP	CF ₃ ⁽¹⁾ ·PH ⁽²⁾ ·OCH ₃ ⁽³⁾	-61.4		(1-2) 12.0 (P-1) 60.3	(1-3) 1.1	6
C ₂ F ₄ H ₃ OP	CF ₃ ⁽¹⁾ ·PF ⁽²⁾ ·OCH ₃ ⁽³⁾	(1) -78.1	(2) -119.4	(1-2) 5.17 (2-3) 1.9 (P-2) 1138	(1-3) 0.76 (P-1) 82.6	6
C ₂ F ₆ HP	(CF ₃ ⁽¹⁾) ₂ PH ⁽²⁾	-48.9		(1-2) 10 (P-1) 70		6
C ₂ F ₆ BrP	(CF ₃) ₂ PBr	-59.8		(P-F) 80.7		6
C ₂ F ₆ Br ₃ P	(CF ₃) ₂ PBr ₃	-81.4		(P-F) 182		8
C ₂ F ₆ ClOP	(CF ₃) ₂ P(O)Cl	-72.3		(P-F) 129.8		44

C ₂ F ₆ ClP	(CF ₃) ₂ PCl	-63.3		(P-F) 85.1	44, 6	
C ₂ F ₆ Cl ₃ P	(CF ₃) ₂ PCl ₃			(P-F) 190	8	
C ₂ F ₆ IP	(CF ₃) ₂ PI	-55.0		(P-F) 72	6	
C ₂ F ₇ Br ₂ P	(CF ₃ ⁽¹⁾) ₂ PF ⁽²⁾ Br ₂	(1) -78.6	(2) -101.5	(P-1) 179 (1-2) 13.0	8	
C ₂ F ₇ Cl ₂ P	(CF ₃ ⁽¹⁾) ₂ PFCl ₂			(P-1) 183	8	
C ₂ F ₇ P	(CF ₃ ⁽¹⁾) ₂ PF ⁽²⁾	(1) -68.3	(2) -221.6	(1-2) 3.4 (P-2) 1002	6	
C ₂ F ₈ P ₂ S ₂		(1) -63.6 (3) -74.1	(2) -164.8 (4) -59.1	(1-2) 5.0 (2-3) 1.2 (2-6) 5 (3-5) 5.2 (4-6) 1165	(1-5) 74 (2-5) 1156 (3-4) 2.8 (3-6) 134	14
C ₂ F ₈ P ₂ S ₂		(1) -166.4 (3) -74.5	(2) -63.6 (4) -58.9	(1-2) 4.3 (1-5) 1148 (2-5) 75 (3-5) 2.8 (4-6) 1180	(1-3) <0.5 (1-6) 4 (3-4) 2.8 (3-6) 134	14
C ₂ F ₉ P	(CF ₃ ⁽¹⁾) ₂ PF ₃			(P-1) 176	8	
C ₃ F ₃ H ₆ O ₂ P	(CF ₃ ⁽¹⁾ ·P(OCH ₃ ⁽²⁾) ₂	-74.0		(1-2) 0.7 (P-1) 80.4	6	
C ₃ F ₃ H ₆ PS ₂	CF ₃ ⁽¹⁾ ·P(SCH ₃ ⁽²⁾) ₂	-59.0		(1-2) 0.74 (P-1) 65.8	6	
C ₃ F ₃ H ₈ NPS ₂ ⁻	(CH ₃) ₂ NH ₂ ⁺ F ⁽²⁾ (CF ₃ ⁽¹⁾)PS ₂ ⁻	(1) -77.8	(2) -37.4	(1-2) 1 (P-2) 1071	14	
C ₃ F ₅ H ₈ NP	CF ₃ ⁽¹⁾ ·PF ₂ ⁽²⁾ (NH·CH ₃) ₂	(1) -70.9	(2) -66.9	(1-2) 14.7 (P-2) 870	71	
C ₃ F ₆ H ₃ OP	(CF ₃ ⁽¹⁾) ₂ POCH ₃ ⁽²⁾	-66.3		(1-2) 0.4 (P-1) 85.8	6	
C ₃ F ₆ H ₃ PS	(CF ₃ ⁽¹⁾) ₂ PSCH ₃ ⁽²⁾	-57.8		(1-2) 0.5 (P-1) 79	6	

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_3F_6H_6NP$	$CF_3^{(1)} \cdot PF_3^{(2)} \cdot N(CH_3)_2$	(1) -71.2 (2a) -55.5 (2e) -71.2	(1-2a) 17.5 (2a-2e) 76 (P-1) 165 (P-2a) 883 (P-2e) 995	71
$C_6F_6H_6NP$	$CF_3^{(1)} \cdot PF_3^{(2)} \cdot NH \cdot CH_2 \cdot CH_3$	(1) -69.9 (2a) -64.5, -56.1 (2e) -74.8	(1-2a) 16.0 (2a-2e) 70 (P-1) 166 (P-2a) 860, 815 (P-2e) 1000	71
$C_3F_8H_4NP$	$(CF_3^{(1)})_2PF_2^{(2)} \cdot NH \cdot CH_3$	(1) -70.7 (2) -67.3	(1-2) 16.5 (P-1) 157 (P-2) 849	71
$C_3F_9Br_2P$	$(CF_3)_3PBr_2$	-65.6	(P-F) 127	8
$C_3F_9Cl_2P$	$(CF_3)_3PCl_2$		(P-F) 139	8
C_3F_9P	$(CF_3)_3P$	-50.7	(P-F) 85.7	6
$C_3F_{10}P_2S_2$		(1) -54.2 (2) -54.5 (3) -74.0 (4) -59.2	(1-5) 78.5 (1-2) 9.0 (1-4) 1.2 (3-5) 7.0 (3-6) 134.5 (4-5) 10.5 (4-6) 1173.4	14
$C_3F_{10}P_2S_2$		(1) -62.0 (2) -164.7 (3) -68.5 (4) -68.7	(1-2) 3.4 (1-5) 77.8 (2-3) 1.4 (2-5) 1161.2 (2-6) 3.8 (3-4) 7.2 (3-5) 6.0 (3-6) 112.5 (4-5) 2.5	14
$C_3F_{11}P$	$(CF_3^{(1)})_3PF_2$		(P-1) 167	8
$C_4F_6H_4ClOP$	$(CF_3)_2P \cdot O \cdot CH_2 \cdot CH_2Cl$	-64.6	(P-F) 86	90
$C_4F_6H_6ClP$	$(CF_3)_2P(CH_3)_2Cl$		(P-F) 79	8
$C_4F_8H_6NP$	$(CF_3^{(1)})_2PF_2^{(2)} \cdot N(CH_3)_2$	(1) -69.5 (2) -59.9	(1-2) 17.5 (P-1) 158 (P-2) 880	71
$C_4F_9H_3ClP$	$(CF_3^{(1)})_3P(CH_3^{(3)})Cl$	-63.3	(P-F) 101 (1-2) 0.8	8, 102, 103

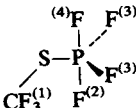
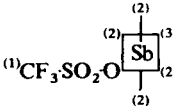
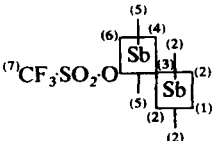
$C_4F_{10}H_3OP$	$(CF_3^{(1)})_2PCF_3^{(2)}(F^{(3)})(OCH_3^{(4)})$	(1) -63.0(eq) (2) -67.0(ax) (3) -51.4	(1-2) 12.0 (1,2-4) 1.2 (P-2) 72.0	(1,2-3) 15.5 (P-1) 133.0 (P-3) 909	106
$C_4F_{10}H_3P$	$(CF_3^{(1)(2)})_3PF^{(3)} \cdot CH_3^{(4)}$	(1) -59.7(eq) (2) -68.2(ax) (3) -13.3	(1-2) 12.5 (P-1) 135.5 (3-4) 8.7	(1-3) 12.5 (P-3) 843 (P-2) 35	102, 8, 103
$C_4F_{10}H_3PS$	$(CF_3^{(1)})_2PCF_3^{(2)}(F^{(3)})(SCH_3)$	(1)(2) -60.3(eq), -59.9(ax)	(1-2) 13.0 (P-eq) 134.0	(P-3) 973 (P-ax) 33.8	106
$C_4F_{12}H_2BP_2^-$	$[(CF_3)_2P]_2BH_2^-$	-52.7			108
$C_4F_{12}H_8B_3P_2^-$	$[(CF_3)_2PBH_3]_2BH_2^-$	-62.8	(P-F) 58.5		108
$C_4F_{12}OP_2$	$[(CF_3)_2P]_2O$	-66.5			44
$C_4F_{12}O_2P_2$	$(CF_3^{(1)})_2P \cdot O \cdot P(O)(CF_3^{(2)})_2$	(1) -65.6 (2) -72.5	(P-1) 93.1 (P-2) 131.9		44
$C_4F_{12}O_3P_2$	$[(CF_3)_2PO]_2O$	-72.03	(P-F) 137.5		44
$C_4F_{12}P_2$	$(CF_3)_4P_2$	-47			44
$C_5F_5H_{12}N_2P$	$CF_3^{(1)} \cdot PF_2^{(2)}[N(CH_3)_2]_2$	(1) -69.5 (2) -59.5	(1-2) 16.5 (P-2) 770	(P-1) 146	71
$C_5F_5H_{12}N_2P$	$CF_3^{(1)} \cdot PF_2^{(2)} \cdot N(CH_3)_2 \cdot NH \cdot CH_2 \cdot CH_3$	(1) -71.6 (2) -71.5, -59.9	(1-2) 14.7 (P-2) 710, 697	(P-1) 148	71
$C_5F_6H_9P$	$(CF_3)_2P(CH_3)_3$		(P-F) 6.3		8
$C_5F_6H_{10}NP$	$CF_3^{(1)} \cdot PF_3^{(2)} \cdot N(CH_2 \cdot CH_3)_2$	(1) -69.9 (2ax) -57.9 (2eq) -69.9	(1-2ax) 16.0 (P-1) 166 (P-2eq) 1004	(2ax-2eq) 72 (P-2ax) 897	71
$C_5F_6H_{10}NP$	$CF_3^{(1)} \cdot PF_3^{(2)} \cdot NH \cdot iC_4H_9$	(1) -69.3 (2ax) -64.5, -56.2 (2eq) -74.4	(1-2ax) 16.7 (P-1) 164 (P-2eq) 1000	(2ax-2eq) 72 (P-2ax) 860, 820	71
$C_5F_6H_{11}BNP$	$(CF_3)_2P \cdot BH_2 \cdot N(CH_3)_3$	-52.0	(P-F) 63		108
$C_5F_6H_{11}BP_2$	$(CF_3)_2P \cdot BH_2 \cdot P(CH_3)_3$	-52.9	(P-F) 63		108

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_3F_6H_{14}B_2NP$	$(CH_3)_3N \cdot BH_2 \cdot P(CF_3)_2 \cdot BH_3$	-61.8	(P-F) 58	108
$C_3F_6H_{14}B_2P_2$	$(CH_3)_3P \cdot BH_2 \cdot P(CF_3)_2 \cdot BH_3$	-63.0	(P-F) 60	108
$C_3F_9H_3NP$	$(CF_3)_3P(CH_3)CN$		(P-F) 69	8
$C_3F_9H_6ClNP$	$(CF_3^{(1)})_3 \cdot PCl \cdot N(CH_3^{(2)})_2$	(30°C) -59.0	(P-1e) 130 (P-1a) 53 (1-2) 0.7 (P-1) 107.0	8 106
$C_3F_9H_6OP$	$(CF_3^{(1)(2)})_3P(CH_3)OCH_3$	(-40°C) (1) -60.0(e) (2) -66.9(a)	(P-1) 108.5 (P-2) 62	102, 8, 103
$C_3F_9H_6O_2P$	$(CF_3^{(1)})_2P(CF_3^{(2)})(OCH_3^{(3)})_2$	(-80°C) (1) -63.6(a) (2) -61.7(e)	(1-2) 13.5 (P-1) 88 (P-2) 108.0 (1,2-3) 0.8	106
$C_3F_9H_6P$	$(CF_3)_3P(CH_3)_2$		(P-Fe) 114 (P-Fa) 23.5	8, 103
$C_3F_9H_6PS$	$(CF_3^{(1)(2)})_3P(CH_3)(SCH_3)$	(-10°C) (1) -58.9(e) (2) -63.5(a)	(P-1) 106.0 (P-2) 33.5	102, 8, 103
$C_3F_{10}H_6NP$	$(CF_3^{(1)})_2P(CF_3^{(2)})(F^{(3)})N(CH_3)_2$	(-40°C) (1) -62.3(e) (2) -59.6(a) (3) -32.3	(1-2) 16.0 (1,2-3) 16.0 (P-1) 130.0 (P-2) 54.0 (P-3) 849	106
$C_6F_8H_4ClCoNO_4P$	$Co(CO)_2NO \cdot P(CF_3)_2(O \cdot CH_2 \cdot CH_2Cl)$	-68.6	(P-F) 95	90
$C_6F_8H_{10}NP$	$CF_3^{(1)} \cdot PF_3^{(2)} \cdot \overbrace{N \cdot (CH_2)_4 \cdot CH_2}$	(1) -68.5 (2a) -57.9 (2e) -69.9	(1-2a) 14.0 (2a-2e) 70 (P-1) 166 (P-2a) 910 (P-2e) 1050	71
$C_6F_8H_{10}NP$	$(CF_3^{(1)})_2 \cdot PF_2^{(2)} \cdot N(CH_2 \cdot CH_3)_2$	(1) -69.3 (2) -65.9	(1-2) 17.0 (P-1) 156 (P-2) 870	71
$C_6F_9H_9NP$	$(CF_3^{(1)(2)})_3P(CH_3^{(3)})N(CH_3^{(4)})_2$	(1) -57.6(e) (2) -65.3(a)	(1-3)(2-3) 1.0 (1-4)(2-4) 1.1 (P-1) 110.0 (P-2) 36.5	102 103, 8
$C_6F_{10}H_9OPSi$	$(CF_3^{(1)})_3P(F^{(2)})OSi(CH_3)_3$	(1) -66.7 (2) -45.5	(1-2) 13.5 (P-1) 120.5 (P-2) 979	106

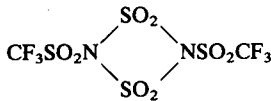
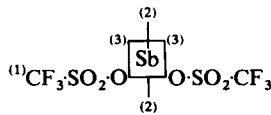
C ₆ F ₁₂ NiO ₄ P	$ \begin{array}{c} {}^{(1)}(\text{CF}_3)_2\text{P}=\text{O} \quad \text{CO} \\ \quad \quad \quad \downarrow \\ {}^{(2)}(\text{CF}_3)_2\text{P} \quad \text{Ni} \\ \quad \quad \quad \quad \quad \downarrow \\ \quad \quad \quad \quad \quad \text{CO} \end{array} $	(1) -72.6 (2) -70.0	(P-1) 130.6 (P-2) 101.3	44
C ₆ F ₁₈ H ₄ B ₂ P ₃	$[(\text{CF}_3^{(1)})_2\text{P}^{(3)}\text{BH}_2\text{P}^{(4)}(\text{CF}_3^{(2)})_2\text{BH}_2\text{P}(\text{CF}_3)_2]^-$	(1) -53.3 (2) -60.0	(P3-1) 64.5 (P3-2) 11.0 (P4-2) 57.0	108
C ₆ F ₁₈ H ₁₀ B ₄ P ₃	$[(\text{CF}_3^{(1)})_2\text{P}^{(3)}(\text{BH}_3)\text{BH}_2]_2\text{P}^{(4)}(\text{CF}_3^{(2)})_2^-$	(1) -62.5 (2) -59.2	(P3-1) 59 (P4-2) 67	108
C ₇ F ₅ H ₁₆ N ₂ P	CF ₃ ⁽¹⁾ ·PF ₂ ⁽²⁾ ·N(CH ₃) ₂ NH·iC ₄ H ₉	(1) -69.9 (2) -68.9, -58.5	(1-2) 14.7 (P-1) 149 (P-2) 720, 697	71
C ₇ F ₆ MnO ₅ P	(CF ₃) ₂ PMn(CO) ₅	-44.0	(P-F) 57.4	135
C ₇ F ₆ O ₅ Pre	(CF ₃) ₂ Pre(CO) ₅	-44.7	(P-F) 59.6	135
C ₇ F ₉ H ₁₂ N ₂ P	(CF ₃ ⁽¹⁾) ₂ PCF ₃ ⁽²⁾ [N(CH ₃) ₂] ₂	(1) -57.0(a) (2) -55.6(e)	(P-1) 50.8 (P-2) 107.0	106
C ₇ F ₁₂ H ₁₃ B ₂ P ₃	(CF ₃ ⁽¹⁾) ₂ P ⁽³⁾ BH ₂ P ⁽⁴⁾ (CF ₃ ⁽²⁾) ₂ BH ₂ P(CH ₃) ₃	(1) -53.5 (2) -60.0	(P3-1) 66.0 (P3-2) 13.4 (P4-2) 66.0	108
C ₈ F ₅ H ₁₆ N ₂ P	CF ₃ ⁽¹⁾ ·PF ₂ ⁽²⁾ ·N·(CH ₂) ₄ ·CH ₂ ·NH·CH ₂ ·CH ₃	(1) -70.4 (2) -69.9, -58.7	(1-2) 14.5 (P-1) 148 (P-2) 725, 695	71
C ₈ F ₆ H ₄ Cl ₂ MnO ₅ P	[MnCl(CO) ₄ L] L = (CF ₃) ₂ P·OCH ₂ ·CH ₂ Cl	-62.7 (<i>trans</i>) -63.6 (<i>cis</i>)	(P-F) 80 (P-F) 83	90
C ₈ F ₆ H ₅ ClMnO ₅ P	[MnH(CO) ₄ L]	-66.6 (<i>trans</i>) -66.3 (<i>cis</i>)	(P-F) 85 (P-F) 86	90
C ₉ F ₅ H ₂₀ N ₂ P	CF ₃ ⁽¹⁾ ·PF ₂ ⁽²⁾ [N(CH ₂ ·CH ₃) ₂] ₂	(1) -70.2 (2) -59.9	(1-2) 15.3 (P-1) 147 (P-2) 780	71
C ₉ F ₆ H ₉ AsMnO ₄ P	<i>cis</i> -(CF ₃) ₂ P·Mn(CO) ₄ As(CH ₃) ₃	-43.7	(P-F) 57.4	135
C ₉ F ₆ H ₉ MnO ₄ P ₂	As above with As replaced by P	-44.0	(P-F) 55.8	135
C ₁₀ F ₆ H ₁₃ Cl ₂ MnNO ₄ P	[MnCl(CO) ₃ N(CH ₃) ₃ L] L = (CF ₃) ₂ P·OCH ₂ ·CH ₂ Cl	-65.2 -68.9	(P-F) 96 (P-F) 81	90
C ₁₀ F ₆ BrO ₈ Pre ₂	Re ₂ (CO) ₈ P(CF ₃) ₂ Br	-53.0	(P-F) 63.2	225
C ₁₀ F ₆ ClO ₈ Pre ₂	As above with Br replaced by Cl	-52.6	(P-F) 65.0	225

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}F_{12}As_8O_8PRe_2$	$Re_2(CO)_8As(CF_3^{(2)})_2P(CF_3^{(1)})_2$	(1) -54.0 (2) -48.7	(P-1) 64.3 (P-2) 1.5	225
$C_{13}F_3H_{10}Hg_2P^{2+}$	$[Hg-Hg-P(CF_3)(C_6H_5)_2]^{2+}$	-53.7	(P-F) 93 $^3J(Hg-F)$ 94 $^4J(Hg-F)$ 46	258
$C_{13}F_3H_{10}P$	$CF_3 \cdot P(C_6H_5)_2$	-54.6	(P-F) 73	258
$C_{26}F_6H_{20}HgP_2^{2+}$	$[CF_3P(C_6H_5)_2HgP(CF_3)(C_6H_5)_2]^{2+}$	-53.3	$^2J(P-F)$ 77 $^4J(P-F)$ 18	258

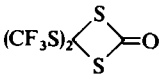
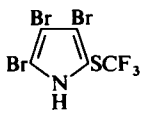
6. CF_3 group bonded to sulphur

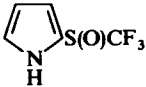
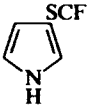
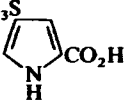
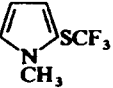
CF_3ClS	$CF_3 \cdot SCl$	-49.4		10
$CF_3N_5O_4S_6$	$CF_3 \cdot SO_2 \cdot NSO_2 \cdot S_4N_4$	-79.2		11
CF_7PS		(1) -37.4 (2) -14.2 (3) -61.4 (4) -5.4	(P-1) 22 (P-2) 923 (P-3) 1057 (P-4) 1058 (1-2) 22 (1-3) 0 (1-4) 22 (2-3) 99 (2-4) 22 (3-4) 116	15
CF_8OSSb^-		(1) -69.9 (2) -100.9 (3) -135.4	(2-3) 101	19
$CF_{13}O_3SSb^-$		(1) -140.4 (2) -109.4 (3) -80.2 (4) -127.7 (5) -90.9 (6) -114.9 (7) -68.6	(2-3) 52 (3-4) 75 (3-5) 90 (4-6) 100 (4-5) 135 (5-6) 135 (1-2) 100	19

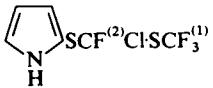
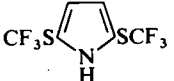
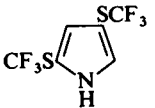
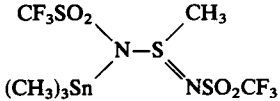
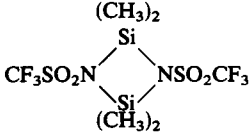
$\text{CF}_{13}\text{O}_3\text{SSb}_2^-$		(1) -131.4 (2) -103.1 (3) -70.9	(1-2) 102	19
$\text{CF}_{18}\text{O}_3\text{SSb}_3^-$		(1) -68.3 (2) -84.9		19
$\text{CF}_{23}\text{O}_3\text{SSb}_4^-$		(1) -67.9 (2) -102.4 (3) -122.6		19
$\text{CF}_{28}\text{O}_3\text{SSb}_5^-$		(1) -67.8 (2) -140.2		19
$\text{C}_2\text{F}_3\text{H}_2\text{NOS}$	$\text{CF}_3\cdot\text{S}\cdot\text{C}(\text{O})\text{NH}_2$	-47		28
$\text{C}_2\text{F}_3\text{H}_3\text{OS}$	$\text{CF}_3^{(1)}\cdot\text{S}(\text{O})\text{CH}_3^{(2)}$	-78.3	(1-2) 1.0	36
$\text{C}_2\text{F}_3\text{H}_3\text{S}_2$	$\text{CF}_3^{(1)}\cdot\text{SS}\cdot\text{CH}_3^{(2)}$	-47.4	(1-2) 0.6	10
$\text{C}_2\text{F}_3\text{BrOS}$	$\text{CF}_3\cdot\text{S}\cdot\text{C}(\text{O})\text{Br}$	-43.0		28
$\text{C}_2\text{F}_3\text{ClS}_2$	$\text{CF}_3\cdot\text{S}\cdot\text{C}(\text{S})\text{Cl}$	-47.7		29
$\text{C}_2\text{F}_3\text{Cl}_3\text{S}_2$	$\text{CF}_3\cdot\text{S}\cdot\text{CCl}_2\cdot\text{SCl}$	-42.0		29
$\text{C}_2\text{F}_4\text{N}_2\text{S}$	$\text{CF}_3^{(1)}\cdot\text{SF}^{(2)}=\text{N}\cdot\text{CN}$	(1) -70.0 (2) -35.2	(1-2) 2.3	38

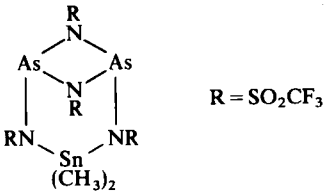
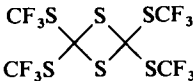
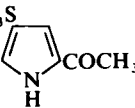
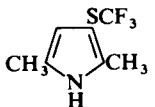
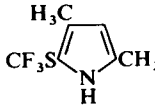
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₂ F ₅ H ₃ OS	CF ₃ ⁽¹⁾ ·SF ₂ ⁽²⁾⁽³⁾ ·OCH ₃	(1) -69.28 (2) +51.54 (3) +50.87	(1-2) 23.96 (1-3) 21.04	40
C ₂ F ₅ H ₃ OS	CF ₃ ⁽¹⁾ (CH ₃ ⁽³⁾)·S(O)F ₂ ⁽²⁾	(1) -69.94 (2) +24.67	(1-2) 21.6 (2-3) 1.2	40
C ₂ F ₆ HNS	(CF ₃) ₂ S=NH	-68.4		48
C ₂ F ₆ H ₂ BNS ₂	(CF ₃ S) ₂ NBH ₂	-52.5		41
C ₂ F ₆ BBr ₂ NS ₂	(CF ₃ S) ₂ NBBr ₂	-51.8		41
C ₂ F ₆ BCl ₂ NS ₂	(CF ₃ S) ₂ NBCl ₂	-52.1		41
C ₂ F ₆ ClNO ₂ S	(CF ₃) ₂ S=NSO ₂ Cl	-57.8		43
C ₂ F ₆ N ₂ O ₈ S ₄		-73.6		11
C ₂ F ₆ OS ₂	CF ₃ ⁽¹⁾ ·S(O)S·CF ₃ ⁽²⁾	(1) -69.5 (2) -34.2	(1-2) 1.5 ¹ J(1-C) 336.9 ¹ J(2-C) 311.3	46
C ₂ F ₆ S ₂	(CF ₃ S) ₂	-46.6	¹ J(F-C) 313.7	46, 2
C ₂ F ₆ S ₃	CF ₃ ·S·S·S·CF ₃	-47.7		47(a)
C ₂ F ₆ S ₄	CF ₃ ·S·S·S·S·CF ₃	-45.9		47(a)
C ₂ F ₈ BNS ₂	(CF ₃ ⁽¹⁾ S) ₂ NBF ₂ ⁽²⁾	(1) -52.9 (2) -113.4	(1-2) 2.22	41
C ₂ F ₈ NOPS	(CF ₃ ⁽¹⁾) ₂ S=NP(O)F ₂ ⁽²⁾	(1) -62.4 (2) -70.2	(P-2) 988	43
C ₂ F ₁₀ O ₆ S ₂ Sb ⁻		(1) -68.8 (2) -78.8 (3) -114.4	(2-3) 130	19
C ₃ F ₃ H ₆ NOS	CF ₃ ·S·NH·CH ₂ ·CH ₂ ·OH	-52.97		65
C ₃ F ₃ NOS	CF ₃ ·S·CO·CN	-44.6		28

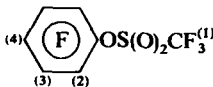
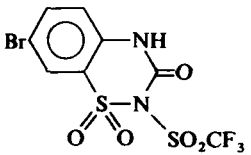
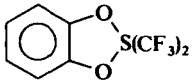
C ₃ F ₃ NOS ₂	CF ₃ ·S·CO·NCS	-41·8			28
C ₃ F ₃ NO ₂ S	CF ₃ ·S·CO·NCO	-46·3			28
C ₃ F ₄ O ₂ S ₂	CF ₃ ⁽¹⁾ ·SS·CO·CO·F ⁽²⁾	(1) -45·1	(2) +16·3		66
C ₃ F ₅ H ₆ NS	CF ₃ ⁽¹⁾ ·SF ₂ ⁽²⁾ ·N(CH ₃) ₂	(1) -63·94	(2) -1·84	(1-2) 12·4	40
C ₃ F ₅ ClS	$ \begin{array}{c} {}^{(1)}\text{CF}_3\text{S} \\ \diagdown \quad \diagup \\ \text{Cl} \quad \text{F}^{(2)} \\ \diagup \quad \diagdown \\ \text{F}^{(3)} \end{array} $	(1) -43·32 (3) -67·05	(2) -72·54	(1-2) 3 (1-3) 2·6	70
C ₃ F ₆ H ₃ NS	(CF ₃ ⁽¹⁾) ₂ S=NCH ₃ ⁽²⁾	-58·0		(1-2) 2·26	48
C ₃ F ₆ N ₂ S	(CF ₃) ₂ S=N·CN	-57·6			43
C ₃ F ₆ OSSe	CF ₃ ⁽¹⁾ ·S·CO·SeCF ₃ ⁽²⁾	(1) -39·1	(2) -31·8		28
C ₃ F ₆ OS ₂	CF ₃ ⁽¹⁾ ·SS·CO·CF ₃	(1) -44·7			66
C ₃ F ₆ OS ₄	CF ₃ ·SS·CO·SS·CF ₃	-46·4			66
C ₃ F ₇ H ₂ NS ₃	(CF ₃ ⁽¹⁾) ₂ S ⁽²⁾ CF ⁽²⁾ ·S·NH ₂	(1) -37·1	(2) -96·4	(1-2) 9·3	29
C ₃ F ₇ ClS ₂	(CF ₃ ⁽¹⁾) ₂ S ⁽²⁾ CF ⁽²⁾ Cl	(1) -39·5	(2) -30·0	(1-2) 9·3	29
C ₃ F ₇ ClS ₃	(CF ₃ ⁽¹⁾) ₂ S ⁽²⁾ CF ⁽²⁾ ·SCl	(1) -37·46	(2) -91·9	(1-2) 9·3	29
C ₃ F ₉ BClNS ₃	(CF ₃ ⁽¹⁾) ₂ S ⁽²⁾ NBCl(SCF ₃ ⁽²⁾)	(1) -51·6	(2) -35·5		41
C ₃ F ₉ NO ₂ S ₂	(CF ₃) ₂ S=NSO ₂ CF ₃	-58·3, -76·2			43
C ₃ F ₉ NS ₃	(CF ₃ S) ₃ N	(-96·1°C) -51·7, -53·3 (ratio 2:1)			77
C ₃ F ₁₀ BNS ₃	(CF ₃ ⁽¹⁾) ₂ S ⁽²⁾ NBF ⁽³⁾ (SCF ₃ ⁽²⁾)	(1) -52·0 (3) -66·6	(2) -33·3	(1-3) 2·25 (2-3) 9·95	41
C ₃ F ₁₇ O ₉ S ₃ Sb ₂ ⁻	$ \begin{array}{c} {}^{(1)}\text{CF}_3\text{SO}_2\text{O} \begin{array}{ c } \hline \text{Sb} \\ \hline \end{array} \text{O} \begin{array}{ c } \hline \text{Sb} \\ \hline \end{array} \text{OSO}_2\text{CF}_3 \\ \begin{array}{c} \text{O} \\ \parallel \\ \text{S} \\ \\ \text{CF}_3 \end{array} \end{array} $	(1) -68·4 (3) -118·4	(2) -79·9	(2-3) 125	19
C ₄ F ₆ H ₅ NOS	(CF ₃ ⁽¹⁾) ₂ S(O)=N·CH ₂ ⁽²⁾ ·CH ₃	-73·1		(1-2) 0·77	48

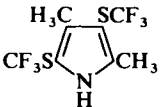
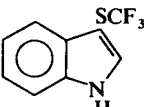
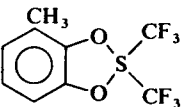
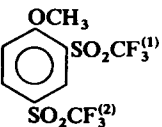
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_4F_6H_5NS$	$(CF_3^{(1)})_2S=N \cdot C H_2^{(2)} \cdot CH_3$	-58.7	(1-2) 2.14	48
$C_4F_6H_6BNS_2$	$(CF_3S)_2N \cdot B(CH_3)_2$	-53.1		41
$C_4F_6OS_4$		-44.7		89
$C_4F_6O_2S_4$	$(CF_3 \cdot SS \cdot C(O))_2$	-44.3		66
$C_4F_7NOS_3$	$(CF_3^{(1)}S)_2CF^{(2)} \cdot SOCN$	(1) -37.16 (2) -89.7	(1-2) 9.3	29
$C_4F_7NS_3$	$(CF_3^{(1)}S)_2CF^{(2)} \cdot SCN$	(1) -37.6 (2) -77.1	(1-2) 9.3	29
$C_4F_7NS_3Se$	$(CF_3^{(1)}S)_2CF^{(2)} \cdot SS_2SeCN$	(1) -37.6 (2) -84.8	(1-2) 9.3	29
$C_4F_7NS_4$	$(CF_3S)_2CF \cdot SSCN$	(1) -36.8 (2) -88.1	(1-2) 9.3	29
$C_4F_8S_2$	$(CF_3^{(1)}S)_2C=CF_2^{(2)}$	(1) -43.68 (2) -52.0	(1-2) 3.2	89
$C_4F_9ClS_4$	$(CF_3S)_3CSCl$	-37.6		29
C_4F_9NOS	$(CF_3)_2S=N \cdot CO \cdot CF_3$	-56.7, -74.7		43
$C_4F_{12}HBN_2S_4$	$[(CF_3^{(1)}S)_2N]_2BH^{(2)}$	-53.1	(1-2) 0.25	41
$C_4F_{12}BNS_4$	$(CF_3^{(1)}S)_2N \cdot B(SCF_3^{(2)})_2$	(1) -52.2 (2) -31.5	(1-2) 1.01	41
$C_4F_{12}N_2S_2$	$[(CF_3)_2S=N]_2$	-60.0		43
$C_4F_{13}BN_2S_4$	$[(CF_3^{(1)}S)_2N]_2BF^{(2)}$	(1) -52.5 (2) -87.2	(1-2) 1.35	41
$C_5F_3HBr_3NS$		-43.92		109
$C_5F_3HI_3NS$	As above with Br replaced by I	-44.25		109

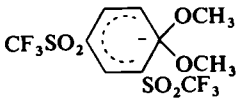
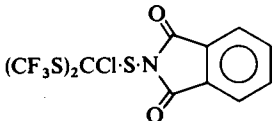
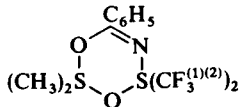
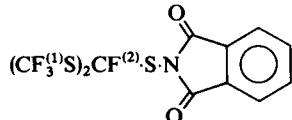
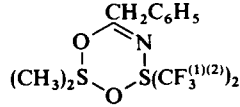
$C_5F_3H_4NOS$		-74.07	109
$C_5F_3H_4NOS$	As above with $S(O)CF_3$ replaced by SO_2CF_3	-79.21	109
$C_5F_3H_4NS$	As above with $S(O)CF_3$ replaced by SCF_3	-46.02	109
$C_5F_3H_4NS$		-46.35	109
$C_5F_3H_8NO_2S$	$CF_3S \cdot NH \cdot CH_2 \cdot CO_2 \cdot CH_2 \cdot CH_3$	-54.27	65
$C_5F_3H_{10}NOS$	$CF_3S \cdot NH \cdot C(CH_3)_2 \cdot CH_2 \cdot OH$	-54.34	65
$C_5F_3H_{12}ClN_2OS$	$CF_3S(O)[N(CH_3)_2]_2Cl$	-80.6	118
$C_5F_3H_{12}ClN_2S$	$CF_3S[N(CH_3)_2]_2Cl$	-73.6	118
$C_5F_3H_{12}NO_2S$	$CF_3SO_2^- N(CH_3)_4^+$	-78.6	119
$C_5F_6H_7NOS$	$(CF_3^{(1)})_2S(O)=NCH^{(2)}(CH_3)_2$	-73.5	48
$C_5F_6H_7NS$	$(CF_3^{(1)})_2S=NCH^{(2)}(CH_3)_2$	-59.8	48
$C_5F_6H_9NSSi$	$(CF_3)_2S=NSi(CH_3)_3$	-69.6	43
$C_6F_3H_4NO_2S$		-45.42	109
$C_6F_3H_4NS$		-43.05	136
$C_6F_3H_5N_2O_4S_2$	$CF_3 \cdot SO_2 \cdot NSO_2 \cdot C_6H_5N$	-79.5	11
$C_6F_3H_6NS$		-46.09	109

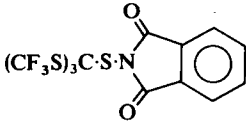
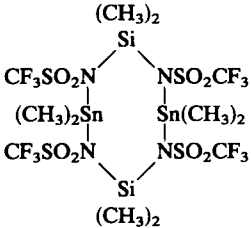
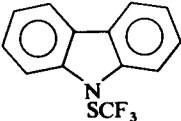
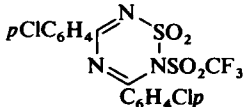
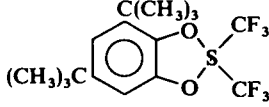
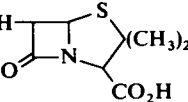
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_3H_{10}Cl_2NS_2$	$CF_3S \cdot CCl_2 \cdot SN(CH_2CH_3)_2$	-41.4		29
$C_6F_3MnO_5S$	$CF_3SMn(CO)_5$	-24.4		135
$C_6F_3O_5ReS$	$CF_3SRe(CO)_5$	-26.7		135
$C_6F_4H_4ClNS_2$		(1) -39.11 (2) -48.52	(1-2) 9.0	109
$C_6F_6H_3NS_2$		-44.82		109
$C_6F_6H_3NS_2$		-45.18, -45.85		109
$C_6F_6H_9N_2O_4S_3Sn$		-77.8		139
$C_6F_6H_{10}N_2S_2$	$[CF_3SN(CH_3)CH_2]_2$	-48.7		140
$C_6F_6H_{12}BBr_2N_2S_2$	$(CF_3S)_2NBBr_2^+N(CH_3)_4^-$	-48.0		41
$C_6F_6H_{12}BCl_2N_2S_2$	$(CF_3S)_2NBCl_2^+N(CH_3)_4^-$	-49.1		41
$C_6F_6H_{12}N_2O_4S_2Si_2$		-79.2		154

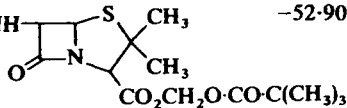
$C_6F_{10}O_2S_2$	$CF_3^{(1)}S \cdot S \cdot CO \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CO \cdot F^{(5)}$	(1) -44.7 (3) -122.2 (5) +23.95	(2) -114.7 (4) -117.3	(2-4) 11.2 (2-5) 6.8 (4-5) 8.4	66
$C_6F_{12}H_6As_2N_4O_8S_4Sn$	 R = SO ₂ CF ₃	-79.1			154
$C_6F_{12}H_6N_2S_2Si$	$[(CF_3)_2S=N]_2Si(CH_3)_2$	-74.4			43
$C_6F_{12}S_6$		-39.7			89
$C_7F_3H_3S$	$CF_3S \cdot (CH_2)_5 \cdot CH_3$	-42.4			36
$C_7F_3H_4N_3O_4S_2$	$CF_3SO_2NSO_2N$ 	-79.3			11
$C_7F_3H_6NOS$	CF_3S 	-45.76			109
$C_7F_3H_8NS$		-45.78			109
$C_7F_3H_8NS$		-46.08			109

Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_7F_3H_{13}OS$	$CF_3S(O)(CH_2)_5CH_3$	-74.9				36
$C_7F_3H_{18}NO_2SSi_2$	$CF_3SO_2 \cdot N[Si(CH_3)_3]_2$	-76.7				154
$C_7F_3H_{18}NO_2SSn_2$	As above with Si replaced by Sn	-78.5				154
$C_7F_3H_{18}NSSi_2$	$CF_3SN[Si(CH_3)_3]_2$	-52.25				65
$C_7F_6H_3Br_2HgNS_2$	$Hg(SCF_3)_2 \cdot 3,5-Br_2C_6H_3N$	-21.60				136
$C_7F_6H_4BrHgNS_2$	$Hg(SCF_3)_2 \cdot 2-BrC_6H_4N$	-22.25				136
$C_7F_6H_4BrHgNS_2$	$Hg(SCF_3)_2 \cdot 3-BrC_6H_4N$	-22.38				136
$C_7F_6H_3BCl_2N_2S_2$	$(CF_3S)_2NB^-\cdot Cl_2 N^+C_5H_5$	-51.0				41
$C_7F_6H_5HgNS_2$	$Hg(SCF_3)_2 \cdot C_6H_5N$	-22.32				136
$C_7F_6H_{10}ClNS_3$	$(CF_3S)_2CCl \cdot SN(CH_2CH_3)_2$	-39.4				29
$C_7F_8O_3S$		(1) -73.4 (3) -161.0	(2) -151.2 (4) -154.1	(1-2) 6.6 (3-4) 20.5	(2-3) 16.5	419
$C_7F_{12}O_2S_4$	$[CF_3^{(1)} \cdot S \cdot CO \cdot CF_2^{(2)}]_2 CF_2^{(3)}$	(1) -46.0 (3) -133.5	(2) -119.2	(2-3) 10.0		66
$C_7F_{14}H_2N_2OS_6$	$[(CF_3^{(1)}S)_2CF^{(2)}S \cdot NH]_2CO$	(1) -37.2 (2) -88.8		(1-2) 9.3		29
$C_8F_3H_4BrN_2O_5S_2$		-73.6				11
$C_8F_6H_4O_2S$		-58.6				192

$C_8F_6H_7NS_2$		-45.33			109
$C_8F_6H_7NS_2$		-45.26, -45.49			109
$C_8F_9H_{10}NS_4$	$(CF_3S)_3C \cdot S \cdot N(CH_2CH_3)_2$	-37.8			29
$C_8F_{12}S_6$	$(CF_3^{(1)}S \cdot CS \cdot C(SCF_3^{(2)})=)_2$	(1) -62.4 (2) -40.7	$^1J(1-C)(2-C)$	313.32, 314.1	89
$C_9F_3H_6NS$		-45.07			109
$C_9F_3H_8NO_2S$	$CF_3S \cdot NH \cdot CH(C_6H_5) \cdot CO_2H$	-53.94			65
$C_9F_3H_{10}NOS$	$CF_3S \cdot NH \cdot CH_2 \cdot CH(OH) \cdot C_6H_5$	-53.31			65
$C_9F_3H_{20}ClN_2OS$	$CF_3S(O)Cl[N(CH_2CH_3)_2]_2$	-81.5			118
$C_9F_3H_{20}ClN_2S$	$CF_3SCl[N(CH_2CH_3)_2]_2$	-77.6			118
$C_9F_6H_6O_2S$		-57.6			192
$C_9F_6H_6O_5S_2$		(1) -76.09 (2) -78.11			211(c)
$C_{10}F_3H_9N_2O_3S$	$CF_3SNH \cdot CO \cdot NH \cdot CH(C_6H_5) \cdot CO_2H$	-52.89			65
$C_{10}F_3H_9N_2O_5S$	$CF_3SO_2NH \cdot CO \cdot NH \cdot CH(C_6H_5) \cdot CO_2H$	-76.62			65

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_6H_9NOS_2$	$CF_3S \cdot NH \cdot CH_2 \cdot CH(C_6H_5) \cdot OSCF_3$	-54.05, -53.06		65
$C_{10}F_6H_9O_6S_2^-$		-76.72, -79.31		211(c)
$C_{11}F_3H_{12}NO_2S$	$CF_3S \cdot NH \cdot CH(C_6H_5) \cdot CO_2CH_2CH_3$	-54.27		65
$C_{11}F_6H_4ClNO_2S_3$		-39.4		29
$C_{11}F_6H_{11}NO_2S_2$		(1) -55.2 (2) -57.3	(1-2) 3.2	192
$C_{11}F_6H_{13}NO_4S_2$	As above with CH_3 replaced by OCH_3	(1) -55.9 (2) -57.6	(1-2) 3.3	192
$C_{11}F_7H_4NO_2S_3$		(1) -37.0 (2) -86.5	(1-2) 9.3	29
$C_{12}F_6H_{13}NO_2S_2$		(1) -56.8 (2) -58.9	(1-2) 3.1	192
$C_{12}F_6H_{13}NO_4S_2$	As above with CH_3 replaced by OCH_3	(1) -56.1 (2) -58.2	(1-2) 3.1	192

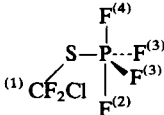
$C_{12}F_9H_4NO_2S_4$		-36.5	29
$C_{12}F_{12}H_{24}N_4O_8Si_2Sn_2$		-79.8	154
$C_{13}F_3H_8NS$		-50.32	109
$C_{14}F_6H_{36}N_2O_4S_4Sn_4$	$[CF_3SO_2N(Sn(CH_3)_3)_2S-]_2$	-75.0	154
$C_{15}F_3H_8Cl_2N_3O_4S_2$		-71.3	11
$C_{15}F_3H_{10}N_3O_4S_2$	As above with Cl replaced by H	-71.3	11
$C_{16}F_6H_{20}O_2S$		-58.1	192
$C_{18}F_3H_{19}N_4O_5S_2$		-53.20	65

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{18}F_3H_{19}N_4O_7S_2$	As above with CF_3 replaced by CF_3SO_2	-76.22		65
$C_{23}F_3H_{28}N_3O_6S_2$	$CF_3S\cdot NH\cdot CH(C_6H_5)\cdot CO\cdot NH$ 	-52.90		65
$C_{24}F_3H_{29}N_4O_7S_2$	As above with $CF_3S\cdot NH$ replaced by $CF_3SNH\cdot CO\cdot NH$	-53.47		65
$C_{24}F_3H_{29}N_4O_9S_2$	As above with $CF_3S\cdot NH$ replaced by $CF_3SO_2NH\cdot CO\cdot NH$	-77.61		65
7. CF_3 group bonded to any other element				
CF_3Br_3Ge	$CF_3\cdot GeBr_3$	-65.2		7(a)
CF_3Br_3Sn	$CF_3\cdot SnBr_3$	-47.7	(Sn-F) 871/835	7(a)
CF_6Se	$CF_3^{(1)}\cdot SeF_2^{(2)}F^{(3)}$	(1) -65.08 (2) -21.93 (3) -59.66	(2-3) 94.4	16
CF_8Si_2	$CF_3^{(1)}\cdot SiF_2^{(2)}\cdot Si_3^{(3)}$	(1) -69.8 (2) -142.4 (3) -123.3	(1-2) +6.09 (1-3) +1.91 (2-3) +13.27	21
$CF_{15}CoP_4$	$CF_3^{(1)}\cdot Co(PF_3^{(2)})_4$	(1) +17.71 (2) -13.79	(1-2) 6 (P-1) 46 (P-2) 1319	22
$C_2F_3H_3Hg$	$CF_3^{(1)}\cdot Hg\cdot CH_3^{(2)}$	-36.3	(1-2) 0.45 (Hg-1) 940	34(a)
$C_2F_6Br_2Ge$	$(CF_3)_2GeBr_2$	-59.4		7(a)
$C_2F_6Br_2Sn$	$(CF_3)_2SnBr_2$	-43.8		7(a)
C_2F_6Hg	$(CF_3)_2Hg$	-40.8	(Hg-F) 1251	45(a)
$C_2F_{12}CoOP_3$	$CF_3Co(PF_3)_3CO$	(1) +15.27 (2) -16.58	(1-2) 4 (P-1) 41 (P-2) 1395	22(c)
C_3F_9HGeO	$(CF_3)_3GeOH$	-56.9		45

C_3F_9BrGe	$(CF_3)_3GeBr$	-55.4				7(a)
$C_3F_9CoO_2P_2$	$CF_3^{(1)}Co(PF_3^{(2)})_2(CO)_2$	(1) +12.89	(2) -16.83	(P-1) 47	(P-2) 1337	22(c)
$C_4F_3H_9Pb$	$CF_3 \cdot Pb(CH_3)_3$	-42.9		(Pb-F) 240.6		34(a)
$C_4F_3H_9Sn$	$CF_3 \cdot Sn(CH_3)_3$	-49.3		(^{119}Sn -F) 270		34(a)
$C_4F_6H_6Ge$	$CF_3^{(1)} \cdot Ge(CF_2^{(2)}H^{(4)})(CF^{(3)}H_2^{(5)})CH_3$	(1) -56.2 (3) -271.0	(2) -132.2	(1-2) ~3.1 (2-3) ~3.1 (3-5) 46.0	(1-3) ~3.1 (2-4) 46.0	45(a)
$C_4F_6H_6Pb$	$(CF_3)_2 \cdot Pb(CH_3)_2$	-39.5		(Pb-F) 379		34(a)
$C_4F_6H_6Si$	$CF_3^{(1)} \cdot Si(CF_2^{(2)}H^{(4)})(CF^{(3)}H_2)CH_3$	(1) -61.97 (3) -279.69	(2) -139.65	(1-3) 3.0 (2-4) 45.0	(1-3) 3.0	45(a)
$C_4F_6H_6Sn$	$(CF_3)_2Sn(CH_3)_2$	-47.0		(^{119}Sn -F) 346.0 (^{117}Sn -F) 330.4		34(a)
$C_4F_7H_5Ge$	$CF_3^{(1)} \cdot Ge(CF_2^{(2)}H^{(3)})_2CH_3$	(1) -55.0	(2) -131.5	(1-2) 3.3	(2-3) 46.5	45(a)
$C_4F_7H_5Ge$	$CF_3^{(1)} \cdot Ge(CF_2^{(2)}H^{(4)})(CF^{(3)}H_2^{(5)})_2$	(1) -53.3 (3) -270.5	(2) -130.3	(1-2) ~3 (2-4) 45.5	(2-3) ~3 (3-5) 46.0	45(a)
$C_4F_8H_4Ge$	$(CF_3^{(1)})_2Ge(CF^{(2)}H_2^{(3)})_2$	(1) -53.7	(2) -271.5	(1-2) 3.4	(2-3) 47.0	45(a)
$C_4F_8H_4Ge$	$CF_3^{(1)} \cdot Ge(CF_2^{(2)}H^{(4)})_2CF^{(3)}H_2^{(5)}$	(1) -51.7 (3) -271.5	(2) -129.0	(1-2) ~3.0 (2-4) 45.6	(1-3) ~3.0 (3-5) 46.0	45(a), 92
$C_4F_9H_3Ge$	$CF_3^{(1)} \cdot Ge(CF_2^{(2)}H^{(3)})_3$	(1) -50.6	(2) -128.2	(1-2) 3.2	(2-3) 46.0	92, 45(a)
$C_4F_9H_3Ge$	$(CF_3^{(1)})_2Ge(CF_2^{(2)}H^{(4)})CF^{(3)}H_2^{(5)}$	(1) -52.2 (3) -271.7	(2) -129.1	(1-2)(1-3) 3.3 (2-3) 3.0 (2-4) 45.5	(3-5) 46.5	92, 45(a)
$C_4F_{10}H_2Ge$	$(CF_3^{(1)})_2Ge \cdot CFH_2$	(1) -52.7				45(a)
$C_4F_{10}H_2Ge$	$(CF_3^{(1)})_2Ge(CF_2^{(2)}H^{(3)})_2$	(1) -50.9	(2) -127.9	(1-2) 3.2	(2-3) 46.0	45(a), 92
$C_4F_{11}HGe$	$(CF_3^{(1)})_3Ge \cdot CF_2^{(2)}H^{(3)}$	(1) -51.30	(2) -127.5	(1-2) 3.0	(2-3) 45.5	45(a), 92
$C_4F_{12}Ge$	$(CF_3)_4Ge$	-50.9				7(a)

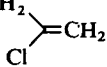
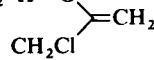
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₄ F ₁₂ Ge	(CF ₃) ₄ Ge	-51.5		45(a), 92
C ₄ F ₁₂ Sn	(CF ₃) ₄ Sn		(¹¹⁹ Sn-F) 531	34
C ₆ F ₃ MnO ₅ Se	CF ₃ Se·Mn(CO) ₅	-22.4		135
C ₆ F ₃ O ₅ ReSe	CF ₃ Se·Re(CO) ₅	-23.0		135
C ₇ F ₃ H ₅ Hg	CF ₃ ·Hg·C ₆ H ₅		(Hg-F) 1000	34
C ₇ F ₆ AsMnO ₅	(CF ₃) ₂ As·Mn(CO) ₅	-40.2		135
C ₇ F ₆ AsO ₅ Re	(CF ₃) ₂ As·Re(CO) ₅	-40.5		135
C ₉ F ₆ H ₉ AsMnO ₄ P	<i>cis</i> -(CF ₃) ₂ As·Mn(CO) ₄ P(CH ₃) ₂	-40.2		135
C ₉ F ₆ H ₉ AsO ₄ Pre	<i>cis</i> -(CF ₃) ₂ As·Re(CO) ₄ P(CH ₃) ₂	-40.2		135
C ₁₀ F ₆ AsBrMnO ₈	Mn(CO) ₈ As(CF ₃) ₂ Br	-46.1		225
C ₁₀ F ₆ AsBrO ₈ Re ₂	Re ₂ (CO) ₈ As(CF ₃) ₂ Br	-48.5		225
C ₁₀ F ₆ AsClMnO ₈	Mn(CO) ₈ As(CF ₃) ₂ Cl	-45.6		225
C ₁₈ F ₃ H ₂₂ BrIOP ₂ Rh	RhBrI(CF ₃ ⁽¹⁾)CO[P(CH ₃ ⁽²⁾) ₂ C ₆ H ₅] ₂	-1.28	(P-F) 17.8 (Rh-F) 14.5	279
C ₁₈ F ₃ H ₂₂ BrIOP ₂ Rh	As above		(P-F) +16.8 (Rh-F) -14.1 (1-2) ≤ 0.1, 0.2	280
C ₁₈ F ₃ H ₂₂ ClIOP ₂ Rh	As above with Br replaced by Cl	-3.63	(P-F) 18.0 (Rh-F) 14.5	279
C ₁₈ F ₃ H ₂₂ ClIOP ₂ Rh	As above with Br replaced by Cl		(P-F) +18.0 (Rh-F) -14.2 (1-2) ≤ 0.1, 0.2	280
C ₁₈ F ₃ H ₂₂ Cl ₂ OP ₂ Rh	As above with Br and I replaced by Cl	-1.75	(P-F) 17.6 (Rh-F) 14.0	279
C ₁₈ F ₃ H ₂₂ I ₂ OP ₂ Rh	As above with Br replaced by I		(P-F) +17.8 (Rh-F) -14.5 (1-2) ≤ 0.1, 0.2	280
C ₂₈ F ₃ H ₂₆ ClIOP ₂ Rh	RhClI(CF ₃)CO[P(CH ₃)(C ₆ H ₅) ₂] ₂	-1.2	(P-F) 18.3 (Rh-F) 13.8	279

B. Fluorine nuclei in a CF₂ group

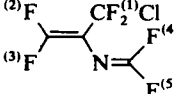
CF ₂ HCl	CF ₂ ⁽¹⁾ H ⁽²⁾ Cl	-72.6	(1-2) 62.6	4
CF ₂ ClN ₃ O ₂ S	CF ₂ Cl·SO ₂ N ₃	-96.26		3
CF ₂ Cl ₂ S	CF ₂ Cl·SCl	-37.4	(F-C) 329	2
CF ₃ ClOS	CF ₂ ⁽¹⁾ Cl·S(O)F ⁽²⁾	(1) -67.7 (2) -16.0	(1A-1B) 134 (1A,1B-2) 1.4	5
CF ₃ ClO ₂ S	CF ₂ ⁽¹⁾ Cl·SO ₂ F ⁽²⁾	(1) -98.43 (2) +98.61		3
CF ₃ ClS	CF ₂ ⁽¹⁾ Cl·SF ⁽²⁾	(1) -43.0 (2) -293.7	(1-2) 6.85 (1-C) 327	2
CF ₃ ClS	CF ₂ ⁽¹⁾ Cl·SF ₂ ⁽²⁾ F ⁽³⁾	(1) -52.8 (2) +47.2 (3) -36.4	(1-2) 25.0 (1-3) 0 (2-3) 42.9	5
CF ₆ CIPS		(1) -37.4 (2) -14.2 (3) -61.4 (4) -5.4	(P-1) 3.7 (P-2) 937 (P-3) 1063 (P-4) 1072 (1-2) 3.7 (1-3) 0 (1-4) 28 (2-3) 100 (2-4) 28 (3-4) 117	15
CF ₈ O ₃ S	CF ₂ ⁽³⁾ OF ⁽⁴⁾ ·OOSF ₄ ⁽¹⁾ F ⁽²⁾	(1)(2) +56.4 (3) -80.2 (4) +156.8	(1-3) 3.8 (2-3) 3.8 (3-4) 33.8	18
CF ₈ S ₂	CF ₂ ⁽¹⁾ (SF ₃ ⁽²⁾) ₂	(1) -88.2 (2) +6.6		20
CF ₈ Si ₂	SiF ₃ ⁽²⁾ ·CF ₂ ⁽¹⁾ ·SiF ₃ ⁽³⁾	(1) -146.6 (2) -150.5	(1-2) 2.97 (2-3) 1.66 (Si-2) 281.0	21
C ₂ F ₂ HBr ₂ Cl	CF ₂ ⁽¹⁾ Br·CH ⁽²⁾ BrCl	(1A) -52.2 (1B) -55.5	(1A-1B) 161 (1A-2) } (1B-2) } ^{6, 8}	23(a)
C ₂ F ₂ HCl ₃	CF ₂ ⁽¹⁾ Cl·CH ⁽²⁾ Cl ₂	-62.5	(1-2) 6	23(a)
C ₂ F ₂ HN	CF ₂ ⁽¹⁾ H ⁽²⁾ ·CN	-119.8	(1-2) 52.0	25
C ₂ F ₂ BrCl ₃	CF ₂ Cl·CBrCl ₂	-62.5		23(a)
C ₂ F ₂ Br ₂ Cl ₂	CF ₂ Cl·CBr ₂ Cl	-61.1		23(a)
C ₂ F ₂ Br ₂ Cl ₂	CF ₂ Br·CBrCl ₂	-55.6		23(a)
C ₂ F ₂ Br ₃ Cl	CF ₂ Br·CBr ₂ Cl	-54.1		23(a)

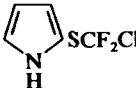
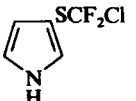
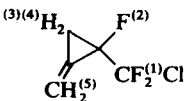
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₂ F ₂ Cl ₄	CF ₂ Cl·CCl ₃	-64.3		23(a)
C ₂ F ₃ Br ₂ Cl	CF ₂ ⁽¹⁾ Br·CF ⁽²⁾ BrCl	(1A) -60.0 (1B) -58.3 (2) -68.3	(1A-1B) 167 (1A-2) } (1B-2) } 14,13	23(a)
C ₂ F ₃ Cl ₃	CF ₂ ⁽¹⁾ Cl·CF ⁽²⁾ Cl ₂	(1) -67.4 (2) -71.4		23(a)
C ₂ F ₄ H ₂ BrNOS	CF ₂ ⁽¹⁾ Br·CH ₂ ·NSOF ₂ ⁽²⁾	(1) -69.8 (2) +50.7	(1-2) 10	33
C ₂ F ₄ H ₂ BrNOS	CH ₂ ⁽³⁾ Br·CF ₂ ⁽¹⁾ ·NSOF ₂ ⁽²⁾	(1) -55.8 (2) +46.8	(1-3) 4	33
C ₂ F ₄ H ₂ ClNOS	CF ₂ ⁽¹⁾ Cl·CH ₂ ⁽³⁾ ·NSOF ₂ ⁽²⁾	(1) -62.3 (2) +45.3	(1-2) 1.9 (1-3) 11	33
C ₂ F ₄ Br ₂	CF ₂ Br·CF ₂ Br	-62.8		23(a)
C ₂ F ₆ BrNOS	CF ₂ ⁽¹⁾ Br·CF ₂ ⁽²⁾ ·NSOF ₂ ⁽³⁾	(1) -68.9 (2) -83.6 (3) +50.6	(1-2) 5.5 (1-3) 1.5 (2-3) 9.2	42
C ₂ F ₆ Cl ₂ S ₂	CF ₂ ⁽¹⁾ Cl·SF ₂ ⁽³⁾ SCF ₂ ⁽²⁾ Cl	(1) -27.6 (2) -49.2 (3) +8.2	(1-3) 19.5 (2-3) 21	2
C ₂ F ₇ H ₂ ClO ₂ S	F ⁽¹⁾ SF ₄ ⁽²⁾ ·OO·CF ₂ ⁽³⁾ ·CH ₂ ⁽⁴⁾ Cl	(1)(2) +56.43 (3) -83.66	(1-3) 0.8 (2-3) 4 (3-4) 8.2	561
C ₂ F ₇ H ₂ ClS	F ⁽³⁾ SF ₄ ⁽²⁾ ·CH ₂ ⁽⁴⁾ ·CF ₂ ⁽¹⁾ Cl	(1) -55.7 (2) +68.7 (3) +77.4	(1-2) 11.6 (1-4) 12.4 (2-3) 139 (2-4) 7.2	52
C ₂ F ₇ ClSi	CF ₂ ⁽¹⁾ Cl·CF ₂ ⁽²⁾ ·SiF ₃ ⁽³⁾	(1) -71.0 (2) -126.5 (3) -147.3	(1-2) 8.7	49
C ₂ F ₇ Cl ₃ O ₂ S	F ⁽¹⁾ SF ₄ ⁽²⁾ ·OO·CF ₂ ⁽³⁾ CCl ₃	(1)(2) +57.47 (3) -89.85		561
C ₂ F ₇ Cl ₃ O ₂ S	F ⁽¹⁾ SF ₄ ⁽²⁾ ·OO·CCl ₂ ·CF ₂ ⁽³⁾ Cl	(1)(2) +58.74 (3) -62.83		561
C ₂ F ₈ Cl ₂ O ₂ S	F ⁽¹⁾ SF ₄ ⁽²⁾ ·OO·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ Cl ₂	(1)(2) +55.30 (3) -91.85 (4) -73.00		561
C ₂ F ₈ Cl ₂ O ₂ S	F ⁽¹⁾ SF ₄ ⁽²⁾ ·OO·CF ₂ ⁽³⁾ Cl·CF ₂ ⁽⁴⁾ Cl	(1)(2) +57.23 (3) -78.90 (4) -66.42	(2-3) ~4 (3-4) 6.75	561

$C_2F_9ClO_2S$	$F^{(1)}SF_4^{(2)} \cdot OO \cdot CF_2^{(3)} \cdot CF_2^{(4)}Cl$	(1)(2) +57·12 (3) -93·87 (4) -70·56	(2-3) ~4 (3-4) 2·7	561
$C_2F_{17}CoP_4$	$CF_2^{(1)}(CF_3)Co(PF_3^{(2)})_4$	(1) -47·76 (2) -12·50	(P-1) 51 (P-2) 1329	22(c)
$C_3F_2H_3BrCl_2$	$CH^{(2)}BrCl \cdot CF_2^{(1)} \cdot CH_2Cl$		(1-2) 10·0	24
$C_3F_2H_3Cl_3$	$CH^{(2)}Cl_2 \cdot CF_2^{(1)} \cdot CH_2Cl$		(1-2) 8·3	24
$C_3F_2H_3N$	$CH_3^{(2)} \cdot CF_2^{(1)} \cdot CN$	-85·4	(1-2) 18·1	25
$C_3F_2H_4ClI$	$CF_2^{(1)}I \cdot CH_2 \cdot CH_2Cl$	-37·9	(1-2) 14·2	57
$C_3F_2H_4Cl_2O$	$CH^{(2)}Cl_2 \cdot CF_2^{(1)} \cdot OCH_3$	-89·46	(1-2) 4·5	61(i)
$C_3F_3HCl_4$	$CF_2^{(1)(2)}Cl \cdot CH^{(4)}Cl \cdot CF^{(3)}Cl_2$	(1) -52·85 (2) -56·79 (3) -56·16	(1-2) 196·06 (1-3) 19·03 (1-4) 9·27 (2-3) 12·33 (2-4) 3·83 (3-4) 5·43	54
$C_3F_3H_2N$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot CN$	(1) -131·5 (2) -208·4	(1-2) 7·9 (1-3) 52·5 (1-4) 6·0 (2-3) 10·8 (2-4) 43·5	25
$C_3F_3H_2N$	$CF^{(2)}H_2^{(3)} \cdot CF_2^{(1)} \cdot CN$	(1) -102·9 (2) -236·5	(1-2) 18·9 (1-3) 11·7 (2-3) 45·4	25
$C_3F_3H_3ClI$	$CF_2^{(1)(2)}I \cdot CF^{(3)}H^{(4)} \cdot CH_2Cl$	(1)(2) -54·5 (3) -184·4	(1-2) 198·5 (1-3) 21·0 (2-3) 22·0 (1-4) 3·0 or 7·5 (2-4) 8·5 (3-4) 48·0	57
$C_3F_3H_3ClI$	$CF^{(3)}H^{(4)}I \cdot CF_2^{(1)(2)} \cdot CH_2Cl$	(1)(2) -109·3 (3) -166·3	(1-3) 21·5 (2-3) 21·5 (1-4) 11·0 (2-4) 8·5 (3-4) 47·0	57
$C_3F_3H_3I_2$	$CF^{(2)}H^{(3)}I \cdot CF_2^{(1)} \cdot CH_2I$		(1-3) 8·5 (2-3) 46	57
$C_3F_3H_4BrO$	$CF^{(2)}H^{(3)}Br \cdot CF_2^{(1)} \cdot CH_2^{(4)}OH$		(1-4) 12·0 (2-3) 48·0	64
C_3F_4HN	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CN$	(1) -135·1 (2) -109·6	(1-2) 5·6 (1-3) 53·6 (2-3) 1·9	25
$C_3F_4H_6Si$	$CF_2^{(1)}H \cdot SiF^{(2)}(CF^{(3)}H_2)CH_3^{(4)}$	(1) -141·8 (2) -179·6 (3) -282·39	(2-4) 6·9 (3-4) 5·5	45(a)

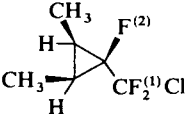
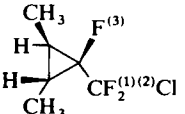
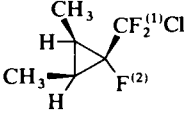
Molecular formula	Structure of compounds	Chemical shifts (pm)	Coupling constants (Hz)	Ref.
$C_3F_5H_2ClO$	$CF_2^{(1)}H^{(4)} \cdot OCF_2^{(2)} \cdot CF^{(3)}H^{(5)}Cl$	(neat) (1) -89.18 (2) -89.49 (3) -159.36 (shifts in several solvents reported)	(1-4) 70.7 (3-5) 48.0	61(i)
$C_3F_{14}CoOP$	$CF_3 \cdot CF_2^{(1)} \cdot Co(PF_3^{(2)})_3(CO)$	(1) -52.63 (2) -14.22	(P-1) 57 (P-2) 1342	22(c)
$C_3F_{19}CoP_4$	$CF_3 \cdot CF_2 \cdot CF_2^{(1)} \cdot Co(PF_3)_4$		(P-1) 52	22
$C_4F_2H_4Cl_2O_2$	$CH^{(2)}Cl_2 \cdot CF_2^{(1)} \cdot CO_2CH_3$		(1-2) 11.5	64
$C_4F_2H_5ClO$	$CF_2^{(1)}H^{(2)} \cdot OCH_2$ 		(1-2) 74.5 $^1J(1-C)$ 261 $^3J(1-C)$ 5	81
$C_4F_2H_5ClO$	$CF_2^{(1)}H^{(2)} \cdot O$ 		(1-2) 74.5 $^1J(1-C)$ 258	81
$C_4F_2H_6Cl_2O$	$CF_2^{(1)}H^{(2)} \cdot OCH(CH_2Cl)_2$		(1-2) 74.5	81
$C_4F_2H_6Cl_2O$	$CH^{(3)}Cl_2 \cdot CF_2^{(1)} \cdot CH_2^{(2)} \cdot OCH_3$		(1-2) 10.5 (1-3) 8.5	64
$C_4F_3H_6BrO$	$CF^{(2)}H^{(3)}Br \cdot CF_2^{(1)} \cdot CH_2^{(4)} \cdot OCH_3$		(1-4) 12.0 (2-3) 47.5	64
$C_4F_3H_6ClO$	$CF^{(2)}H^{(3)}Cl \cdot CF_2^{(1)} \cdot CH_2 \cdot OCH_3$		(1-4) 12.0 (2-3) 48.0	64
$C_4F_3H_6ClS_2$	$CH_3^{(3)}S \cdot CF_2^{(1)} \cdot CF^{(2)}Cl \cdot SCH_3^{(4)}$	(1) -83.9 (2) -91.2	(1-2) 17.9 (1-3) 1.2 (1-4) 1.1 (2-3) 1.2 (2-4) 1.3	69
$C_4F_3H_9Si$	$CF_2^{(1)}H^{(3)} \cdot Si(CH_2^{(4)}F^{(2)})(CH_3)_2$	(1) -139.67 (2) -275.11	(1-3) 45.6 (2-4) 46.4	45
$C_4F_4H_2Cl_4O$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH^{(4)}Cl \cdot OCCl_3$		(1-3) 53.5 (2-4) 8.0	64
$C_4F_4H_3Cl_3O$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH_2^{(4)} \cdot OCCl_3$		(1-3) 53.5 (2-4) 15.5	64
$C_4F_4H_3Cl_3O$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH^{(4)}Cl \cdot OCHCl_2$		(1-3) 53.5 (2-4) 8.0	64
$C_4F_4H_3Cl_3O$	$CF_2^{(1)}H^{(2)} \cdot CF_2 \cdot CCl_2 \cdot OCH_2Cl$		(1-2) 53.0	64
$C_4F_4H_4Cl_2O$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH_2^{(4)} \cdot OCH_3$		(1-3) 49.5 (2-4) 11.0	64

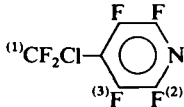
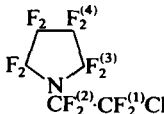
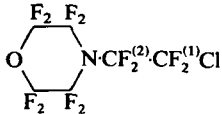
$C_4F_4H_6$	$CF_2^{(1)(2)}H^{(5)} \cdot CF^{(3)}(CH_3^{(6)}) \cdot CF^{(4)}H_2^{(7)}$	(1) -132.6 (3) -173.1	(2) -139.0 (4) -237.7	(1-2) 302.0 (1-6) ~1.0 (2-5) 54.0 (3-4) 11.7 (4-6) 2.2	(1-5) 54.0 (2-3) 9.0 (2-6) ~1.0 (3-6) 21.5 (4-7) 52.5	83
$C_4F_4H_6O$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH_2^{(4)} \cdot OCH_3$			(1-3) 53.5	(2-4) 12.0	64
$C_4F_4H_7O_3P$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot P(O)(OCH_3)_2$	(1) -139.9	(2) -130.7	(1-2) 5.6 (2-3) 5.7	(1-3) 52.8 (P-2) 91.3	84(a)
$C_4F_4H_8Ge$	$CF_2^{(1)}H^{(3)} \cdot Ge(CF^{(2)}H_2^{(4)})_2CH_3$	(1) -132.4	(2) -271.0	(1-2) 2.1 (2-4) 46.5	(1-3) 46.2	45(a)
$C_4F_4H_8Si$	$(CF_2^{(1)}H^{(2)})_2Si(CH_3)_2$	-137.72		(1-2) 46.0		45(a)
$C_4F_4H_8Si$	$CF_2^{(1)}H^{(3)} \cdot Si(CF^{(2)}H_2^{(4)})_2 \cdot CH_3^{(5)}$	(1) -139.58	(2) -278.10	(1-2) 1.5 (2-4) 47.0	(1-3) 45.8 (2-5) 1.0	45(a)
$C_4F_5H_5$	$(CF_2^{(1)}H^{(3)})_2CF^{(2)} \cdot CH_3^{(4)}$	(1) -138.0	(2) -184.0	(1-2) ~ (2-3) (1-3) ~50 (2-4) 21.7	~7.5 (1-4) 1.3	83
$C_4F_5H_5$	$CF_2^{(1)}H^{(4)} \cdot CF^{(2)}(CF^{(3)}H_2^{(5)})_2$	(1) -136.2 (3) -245.2	(2) -187.3	(1-2) 6.1 (2-4) 5.5 (2-5) 20.7 (3-5) 48.0	(1-4) 52.5 (2-3) 12.0 (3-4) 1.4	83
$C_4F_5H_7Ge$	$CF_2^{(1)}H^{(3)} \cdot Ge(CF^{(2)}H_2^{(4)})_3$	(1) -130.5	(2) -270.4	(1-2) 2.0 (2-4) 46.6	(1-3) 46.0	45(a)
$C_4F_5H_7Si$	$(CF_2^{(1)}H^{(3)})_2Si(CF^{(2)}H_2^{(4)})CH_3^{(5)}$	(1) -139.35	(2) -279.50	(1-2) 1.4 (2-4) 46.7	(1-3) 45.7 (2-5) 0.5	45(a)
$C_4F_5H_7Si$	$CF_2^{(1)}H^{(3)} \cdot Si(CF^{(2)}H_2^{(4)})_3$	(1) -138.53	(2) -279.92	(1-2) 1.7 (2-4) 46.6	(1-3) 45.4	45(a)
$C_4F_6H_2Cl_2O$	$CF_2^{(1)}H^{(3)} \cdot CF_2 \cdot CCl_2 \cdot OCF_2^{(2)}H^{(4)}$			(1-3) 52	(2-4) 68	64
$C_4F_6H_2Cl_2O$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH^{(4)}Cl \cdot OCF_2Cl$			(1-3) 53.5	(2-4) 8.0	64
$C_4F_6H_3ClO$	$CF_2^{(1)}H^{(4)} \cdot O \cdot CF^{(2)}H^{(5)} \cdot CF_2 \cdot CF^{(3)}H^{(6)}Cl$			(1-4) 70.5 (3-6) 48.0	(2-5) 56.0	64

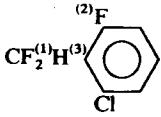
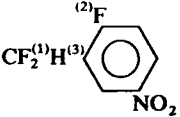
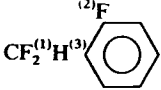
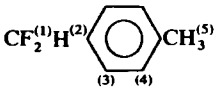
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_4F_6H_3ClO$	$CF_2Cl \cdot CF_2^{(1)} \cdot CH_2^{(2)} \cdot OCF_2H$		(1-2) 13.0	64
$C_4F_6H_4$	$(CF_2^{(1)}H^{(4)})_2CF^{(2)} \cdot CF^{(3)}H_2^{(5)}$	(1) -135.7 (2) -193.8 (3) -248.7	(1-2) 7.0 (1-3) 2.5 (1-4) 52.0 (1-5) 1.4 (2-3) 11.5 (2-4) 5.6 (2-5) 19.5 (3-5) 46.5	83
$C_4F_6H_6Ge$	$(CF_2^{(1)}H^{(3)})_2Ge(CF^{(2)}H_2^{(4)})_2$	(1) -129.9 (2) -271.0	(1-2) 2.5 (1-3) 46.0 (2-4) 46.6	45(a), 52
$C_4F_6H_6Si$	$(CF_2^{(1)}H^{(3)})_2Si(CF^{(2)}H_2^{(4)})_2$	(1) -137.38 (2) -281.37	(1-3) 45.4 (2-4) 46.4 (1-2) 1.9	45(a)
C_4F_6ClN	 $(-100^\circ C)$	(1) -54.0 (2) -88.0 (3) -84.0 (4) -37.0 (5) -52.5	(1-2) 34.4 (1-3) 6.2 (2-3) 24.6 (2-4) 5.5 (2-5) 3.4 (3-4) 23.0 (3-5) 1.6 (4-5) 69.0	85
$C_4F_7H_2ClO$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH^{(4)}Cl \cdot OCF_3$		(1-3) 53.5 (2-4) 6.6	64
$C_4F_7H_2ClO$	$CF_2^{(1)}H^{(3)} \cdot CF_2 \cdot CFCl \cdot OCF_2^{(2)}H^{(4)}$		(1-3) 49.5 (2-4) 61.0	64
$C_4F_7H_2ClO$	$CF_2Cl \cdot CF_2 \cdot CF^{(2)}H^{(4)} \cdot OCF_2^{(1)}H^{(3)}$		(1-3) 71.0 (2-4) 54.5	64
$C_4F_7H_3$	$(CF_2^{(1)}H^{(3)})_3CF^{(2)}$	(1) -135.1 (2) -202.1	(1-2) 7.9 (1-3) 53.0 (2-3) 6.5	83
$C_4F_7H_3O$	$CF_2^{(1)}H^{(4)} \cdot CF_2 \cdot CF^{(2)}H^{(5)} \cdot OCF_2^{(3)}H^{(6)}$		(1-4) 52.0 (2-5) 55.0 (3-6) 72.0	64
$C_4F_7H_3O$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH_2^{(4)} \cdot OCF_3$		(1-3) 53.0 (2-4) 12.0	64
$C_4F_7H_3Ge$	$(CF_2^{(1)}H^{(3)})_3GeCF^{(2)}H_2^{(4)}$	(1) -129.0 (2) -271.4	(1-2) 2.3 (1-3) 46.2 (2-4) 46.0	45(a), 92
$C_4F_7H_3Si$	$(CF_2^{(1)}H^{(3)})_3SiCF^{(2)}H_2^{(4)}$	(1) -138.07 (2) -282.89	(1-2) 2.1 (1-3) 45.0 (2-4) 46.6 (2-3) 0.4	45(a)
C_4F_8HHgI	$CF_2^{(1)}H^{(5)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot HgI$	(1) -137.7 (2) -132.4 (3) -119.9 (4) -101.4	(1-5) 59.0 (Hg-3) 150 (Hg-4) 1020	94

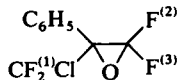
C ₄ F ₈ HI	CF ₂ ⁽¹⁾ H ⁽⁵⁾ ·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ I	(1) -137·7 (3) -121·0	(2) -130·8 (4) -63·9	(1-5) 59·0	94
C ₄ F ₈ Cl ₂ N ₂ S	(CF ₂ ⁽¹⁾ Cl·CF ₂ ⁽²⁾ ·N=) ₂ SF ₂ ⁽³⁾	(1) -73·7 (3) +63·2	(2) -84·7	(1-2) 4·1 (2-3) 9·6	42
C ₄ F ₈ I ₂	CF ₂ ⁽¹⁾ I·CF ₂ ⁽²⁾ ·CF ₂ ·CF ₂ I	(1) -59·9	(2) -112·7		94
C ₄ F ₉ Cl ₂ N	(CF ₂ ⁽¹⁾ Cl·CF ₂ ⁽²⁾) ₂ NF ⁽³⁾	(1) -70·9 (3) -92·0	(2) -107·5	(1-3) 20·0 (2-3) 22·6	99(a)
C ₄ F ₁₀ Br ₂ N ₂ S	(CF ₂ ⁽¹⁾ Br·CF ₂ ⁽²⁾ ·N=) ₂ SF ₂ ⁽³⁾	(1) -68·9 (3) +63·2	(2) -83·7	(1-2) 5·6 (2-3) 9·4	42
C ₄ F ₁₁ CoO ₂ P ₂	CF ₃ ·CF ₂ ⁽¹⁾ ·Co(PF ₃ ⁽²⁾) ₂ CO	(1) -54·79	(2) -14·87	(P-1) 63 (P-2) 1348	22(c)
C ₄ F ₁₆ CoOP ₃	CF ₃ ·CF ₂ ·CF ₂ ⁽¹⁾ ·Co(PF ₃) ₃ CO			(P-1) 56	22
C ₅ F ₂ H ₄ ClNS		-30·56			109
C ₅ F ₂ H ₄ ClNS		-30·53			109
C ₅ F ₂ H ₆ O _s	CF ₂ ⁽¹⁾ H ⁽²⁾ ·C(OH)=CH·COCH ₃ (Z)	-127·09		(1-2) 55	113
C ₅ F ₂ H ₁₂ P ₂	[(CH ₃) ₂ P] ₂ CF ₂	-116·8		(P-F) 58·3	115(a)
C ₅ F ₃ H ₄ Cl		(1) -63·5	(2) -191·5	(1-2) 13·5 (2-4) ~3	(2-3) 10·0 (2-5) 1·5 49
C ₅ F ₃ H ₄ Cl ₃ O ₂	CF ₂ ⁽²⁾ H ⁽⁴⁾ Cl·CF ₂ ⁽¹⁾ ·CH ₂ ⁽³⁾ ·O·CO·CHCl ₂			(1-3) 12·5 (2-4) 48·0	111
C ₅ F ₃ H ₇ Cl ₂	CF ₂ ⁽²⁾ H ⁽³⁾ Cl·CF ₂ ⁽¹⁾ ·CH(CH ₃)·CH ₂ Cl			(1-3) 7·0 (2-3) 48	116
C ₅ F ₃ H ₁₀ ClSi	CF ₂ ⁽¹⁾⁽²⁾ Cl·CF ₂ ⁽³⁾ H ⁽⁴⁾ ·Si(CH ₃) ₃	(1) -57·0 (3) -222·5	(2) -58·8	(1-3) 21·4 (1-4) 15·2 (3-4) 45·0	49

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_3F_4H_4Cl_2O_2$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot CH_2^{(4)} \cdot O \cdot CO \cdot CHCl_2$		(1-3) 52.5 (2-3) 3.6 (2-4) 12.5	111
$C_3F_4H_3ClO_2$		(1) -67.8 (2) -135.2	(1-2) 8.4 (2-3) 54	120
$C_3F_4H_6O_2$	As above with CF_2Cl replaced by CF_2H	-137.5	(2-3) 56.4	120
$C_3F_5H_3O$		(1) -133.8 (2) -119.05 (3A) -92.65 (3B) -84.68	(1-4) 55 (3A-3B) 160	121(a)
$C_3F_6H_4O$	$CF_2^{(3)}H^{(4)} \cdot CF_2^{(2)} \cdot CF_2^{(1)} \cdot COCH_3$	(1) -120.2 (2) -130.1 (3) -136.5	(2-4) 5.6 (3-4) 52.0	43
$C_3F_6H_4O_2$		(1) -136.0 (2) -127.8		120
$C_3F_8CoO_3P$	$CF_3 \cdot CF_2^{(1)} \cdot Co(PF_3^{(2)})(CO)_3$	(1) -55.46 (2) -14.24	(1-2) 6 (P-1) 63 (P-2) 1371	22(c)
$C_3F_{10}BrIO$	$CF_2^{(1)}Br \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot O \cdot CF_2^{(4)} \cdot CF_2^{(5)}I$	(1)(5) -63.5, -64.0 (3)(4) -81.8, -85 (2) -122		131
$C_3F_{10}Br_2O$	$CF_2^{(1)}Br \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot O \cdot CF_2^{(4)} \cdot CF_2^{(5)}Br$	(1)(5) 63.4, -69.1 (3)(4) -81.4, -86.0 (2) -122		131
$C_3F_{12}O_4S$	$F^{(1)}CO \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CO \cdot OOSF_5^{(5)}$	(1) +23.6 (2) -118.6 (3) -123.7 (4) -116.8 (5) +56.2	(1-4) 1.6 (1-3) 6.4 (1-2) 7.5	18
$C_3F_{16}O_6S_2$	$CF_2^{(1)}(CF_2^{(2)} \cdot CO \cdot OOSF_5^{(3)})_2$	(1) -122.6 (2) -116.1 (3) +56.8		18
$C_6F_2H_8O_2$	$CF_2^{(1)}H^{(2)} \cdot C(OH)=CH \cdot CO \cdot CH_2CH_3$ (Z)	-127.07	(1-2) 55	113

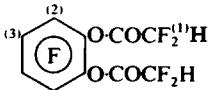
$C_6F_2H_{10}O$	$CF_2^{(1)}H^{(2)} \cdot CO \cdot C(CH_3)_3$	-126.8	(1-2) 53.5	(1-3) 0.75	134
$C_6F_2H_{11}Cl_2N$	$CH^{(2)}Cl_2 \cdot CF_2^{(1)} \cdot N(CH_2^{(3)}CH_3)_2$	-86.0	(1-2) 7.0	(1-3) 1.5	111
$C_6F_2H_{13}O_3P$	$CF_2^{(1)}H^{(2)} \cdot CH_2^{(3)} \cdot P(O)(OCH_2CH_3)_2$	-111.3	(1-2) 56.5 (P-1) 27.6	(1-3) 17.2	84(a)
$C_6F_2H_{15}GeP$	$(CH_3)_3Ge \cdot CF_2 \cdot P(CH_3)_2$	-119.3	(P-F) 59.7		115(a)
$C_6F_3H_8Cl$		(1) -64.3 (2) -222.5			49
$C_6F_3H_8Cl$		(1) -57.9 (2) -59.9 (3) -202.5	(1-3)(2-3) 12.0, 10.5		49
$C_6F_3H_8Cl$		(1) -56.3 (2) -201.5			49
$C_6F_3H_9O_2$	$CH_3 \cdot CH_2 \cdot CF_2^{(1)} \cdot CF^{(2)}H \cdot CO_2CH_3$	(1) -106.6 (2) -202.5	(1-2) 11		138
$C_6F_3H_{11}$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot C(CH_3)_3$	(1) -128.5 (2) -208.7	(1-3) 50.8 (2-4) 46.2		93(a)
$C_6F_3H_{11}$	$CF^{(2)}H_2^{(3)} \cdot CF_2^{(1)} \cdot C(CH_3)_3$	(1) -120.7 (2) -238.9	(2-3) 47.7		98(a)
$C_6F_3H_{11}ClO_3P$	$CF^{(2)}H^{(3)}Cl \cdot CF_2^{(1)} \cdot P(O)(OCH_2CH_3)_2$	(1) -124.5 (2) -154.7	(1-2) 16.9 (2-3) 47.1 (1-3) 5.6, 12.2 (P-1) 93.7		84(a)
$C_6F_4H_{10}$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot C(CH_3)_3$	(1) -132.7 (2) -124.9	(1-3) 53.4		98(a)
$C_6F_4H_{11}O_3P$	$CF_2^{(1)}H^{(3)} \cdot CF_2^{(2)} \cdot P(O)(OCH_2CH_3)_2$	(1) -139.3 (2) -131.1	(1-2) 6.5 (1-3) 52.5 (2-3) 5.9 (P-1) 2.0 (P-2) 92.0		84(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₆ F ₆ CIN		(1) -48.7 (2) -86.4 (3) -140.2	(1-3) 26.8	4
C ₆ F ₈ H ₆ S ₂	(CH ₃ S·CF ₂ ⁽¹⁾ ·CF ₂ ⁽²⁾) ₂	(1) -91.3 (2) -120.1		36
C ₆ F ₉ Cl ₃ O ₂	(CF ₂ ⁽¹⁾ Cl) ₂ CF ₂ ⁽²⁾ ·O·CF ₂ ⁽³⁾ ·CO·CF ₂ ⁽⁴⁾ Cl	(1)(4) -65.9 (2) -135.4 (3) -72.2	(2-3) 21.3 (1-3) ~6.5 (3-4) ~6.5 (1-2) 5.4	143
C ₆ F ₁₁ ClO	CF ₂ ⁽¹⁾ Cl·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽⁵⁾ ·COF ⁽⁶⁾	(1) -61.8 (2) -130.3 (3)(4) -123.2 (5) -116.4 (6) +21.0		97
C ₆ F ₁₂ Br ₂ O ₂	(CF ₂ ⁽¹⁾ Br·CF ₂ ⁽²⁾ ·OCF ₂ ⁽³⁾) ₂	(1) -69.5 (2)(3) -86.7, -89.3		131
C ₆ F ₁₂ CIN		(1) -73.1 (2) -61.8 (3) -92.5 (3) -135.2		135(a)
C ₆ F ₁₂ ClNO		(1) -73.9 (2) -94.0 (3) -93.7 (4) -89.2		135(a)
C ₆ F ₁₂ I ₂ O ₂	(CF ₂ ⁽¹⁾ I·CF ₂ ⁽²⁾ ·O·CF ₂ ⁽³⁾) ₂	(1) -64.4 (2)(3) -86.5, -101		131
C ₇ F ₂ H ₆ S	C ₆ H ₅ ·S·CF ₂ H	-121		161
C ₇ F ₂ H ₁₅ Br ₂ P	(CH ₃ CH ₂) ₃ P ⁺ CF ₂ Br Br ⁻	-52.2	(P-F) 59.1	192
C ₇ F ₂ H ₁₈ Br ₂ N ₃ P	[(CH ₃) ₂ N] ₃ P ⁺ CF ₂ Br Br ⁻	-49.0	(P-F) 87.1	192
C ₇ F ₂ H ₁₈ Cl ₂ N ₃ P	As above with Br replaced by Cl	-51.3	(P-F) 95.4	192
C ₇ F ₂ H ₁₈ I ₂ N ₃ P	As above with Br replaced by I	-58.1	(P-F) 86.6	192
C ₇ F ₂ H ₁₉ BrN ₃ P	[(CH ₃) ₂ N] ₃ P ⁺ CF ₂ ⁽¹⁾ H ⁽²⁾ Br ⁻	-129.3	(P-1) 84.6 (1-2) 47.2	192

$C_7F_3H_4Cl$		(1) -115.00 (2) -113.39	(1-2) 13.43 (1-3) 52.67	164
$C_7F_3H_4NO_2$		(1) -115.92 (2) -109.73	(1-2) 2.25 (1-3) 54.50	164
$C_7F_3H_5$		(1) -120.51 (2) -114.81	(1-2) 1.26 (1-3) 54.78	164
$C_7F_3H_{11}O_2$	$CH_3 \cdot (CH_2)_2 \cdot CF_2^{(1)} \cdot CF^{(2)}H^{(3)} \cdot CO_2CH_3$	(1) -108.5 (2) -202.5	(1-2) 12 (2-3) 46	138
$C_7F_4H_8Cl_2O_2$	$CF^{(1)}H^{(4)}Cl \cdot CF_2^{(2)} \cdot CH(CH_3) \cdot CH_2 \cdot O \cdot CO \cdot CF^{(3)}H^{(5)}Cl$		(1-4) 48 (2-4) 7 (3-5) 50	116
$C_8F_2H_6O$	$CF_2^{(1)}H^{(2)} \cdot CO \cdot C_6H_5$	-123.7	(1-2) 53	134
$C_8F_2H_7IO$	$CH_2^{(2)}I \cdot CF_2^{(1)} \cdot O \cdot C_6H_5$	-68.37	(1-2) 9.6	184(e)
$C_8F_2H_8$			(1-2) 56.622 (1-3) -1.169 (1-4) 0.891 (1-5) 1.46	186
$C_8F_2H_9P$	$CF_2^{(1)}H \cdot CH_2 \cdot PH \cdot C_6H_5$	-110.7 (AB system)	(1A-1B) 280	187(a)
$C_8F_2H_{12}O$	$CF_2^{(1)}H^{(2)} \cdot CO \cdot C_6H_{11}$	-128.6	(1-2) 53.6	134
$C_8F_2H_{12}O_2$	$CF_2^{(1)}H^{(2)} \cdot C(OH)=CH \cdot CO \cdot R (Z)$ $\left. \begin{matrix} R = nC_4H_9 \\ iC_4H_9 \end{matrix} \right\}$	-127.07	(1-2) 55	113
$C_8F_3H_{15}$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot CH(CH_3) \cdot C(CH_3)_3$	(1) -130.5 (2) -204.5	(1-2) ~13 (1-3) 56.0 (1-4) ~12 (2-4) 46.0 (2-3) ~10	49
$C_8F_3H_{15}$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot CH_2 \cdot CH_2 \cdot C(CH_3)_3$	(1) -133.0 (2) -206.5	(1-2) ~13 (1-3) 56.0 (1-4) ~12 (2-4) 46.0 (2-3) ~10	49

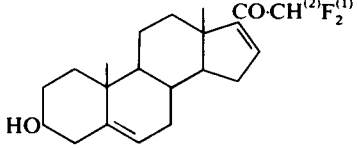
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_4H_7P$	$CF_2^{(2)}H^{(3)} \cdot CF_2^{(1)} \cdot PH^{(4)} \cdot C_6H_5$	(1) -110.9 (2) -134.9	(1-3) 4.6 (2-3) 53.8 (1-4) 18.6 (2-4) 3.0 (P-1) 82.6 (1A-1B) 298	187
$C_8F_8H_6O$	$CF_2^{(1)}H^{(5)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot C(OH)=CH \cdot CO \cdot CH_3$ (Z)	(1) -139.9 (2-4) -128.8, -139.9, -120.0	(1-5) 55	113
C_8F_8BrCl	$CF_2^{(1)}Br \cdot CF^{(2)}Cl$ 	(1) -62.9, -61.8 (2) -115.3 (3) -134.9 (4) -161.5 (5) -149.2	(1A-1B) 178	195(b)
$C_8F_{15}ClO$	$CF_2^{(1)}Cl \cdot (CF_2^{(2)})_4 \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CO \cdot F^{(5)}$	(1) -68.2 (2) -126.4 (3) -121.4 (4) -117.1 (5) +22.8		97
$C_8F_{16}H_2Hg$	$Hg(CF_2^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)}H^{(5)})_2$	(1) -108.0 (2) -121.9 (3) -132.4 (4) -137.7	(4-5) 59.0 (Hg-1) 746 (Hg-2) 148.0	94
$C_9F_2H_8Cl_2$	$C_6H_5CH_2^{(2)} \cdot CF_2^{(1)} \cdot CH^{(3)}Cl_2$	-104.8	(1-2) 16 (1-3) 8	203(a)
$C_9F_2H_8O$	$CF_2^{(1)}H^{(2)} \cdot CO \cdot CH_2^{(3)} \cdot C_6H_5$	-127.6	(1-2) 53.6 (1-3) 1.1	134
$C_9F_3H_5Cl_2O$	$CF_2^{(1)(2)}Cl \cdot CCl(C_6H_5) \cdot CO \cdot F^{(3)}$	(1) -60.0 (2) -59.8 (3) +32.9	(1-2) 169.7 (1-3) 9.1	4
$C_9F_3H_{10}OP$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot P(OCH_3)C_6H_5$ [two diastereoisomers (60:40); minor values given first]	(1) -127.3 (2) -218.5	(1-2) 17, 18 (2-3) 9, 10 (2-4) 46, 47 (P-2) 117, 103	187(a)
$C_9F_3H_{11}NP$	$CF_2^{(1)}H \cdot CF^{(2)}H \cdot P(NHCH_3)C_6H_5$	(1) -126.5 (2) -216.5		325(a)
$C_9F_3H_{17}$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot C(CH_3)_2C(CH_3)_2$	(1) -127.5 (2) -203.0	(1-2) 15.0 (1-3) 55.5 (1-4) ~13.5 (2-3) 10.5 (2-4) 45.6	49
$C_9F_4H_5ClO$		(1) -59.4 (2) -101.8 (3) -109.5	(1-2) 2.1 (1-3) 19.4 (2-3) 41.2	4

$C_9F_4H_5NOS$		(1) -76.2 (2) -148.7			204(a)
$C_9F_4H_7Cl_2NO_2S$	$(CF_2Cl)_2C(OH)NH \cdot S(O)C_6H_5$	-62.7 to -60.65			206(a)
$C_9F_5H_{14}Cl_2NO$	$CF_2^{(1)}H^{(4)} \cdot CF_2^{(2)} \cdot CH_2^{(5)} \cdot O \cdot CF^{(3)}(CHCl_2)N(CH_2CH_3)_2$	(1) -141 (2) -126 (3) -111 to -113.5	(1-4) 53.5 (2-5) 12.0 (1-5) 1.5		111
$C_9F_6H_7NO_2$		(1) -89.5 (2) -206.8 (3) -143.4 (4) -98.1 (5) -154.1			212
$C_9F_8H_8O_2$	$CF_2^{(1)}H^{(3)} \cdot (CF_2^{(2)})_3 \cdot C(OH)=CH \cdot CO \cdot CH_2CH_3$ (Z)	(1) -139.9 (2) -128.8, -139.9, -120.0	(1-3) 55		113
$C_9F_{18}Br_2O_2$	$(CF_2^{(1)}Br \cdot CF_2^{(2)} \cdot O \cdot CF_2^{(3)} \cdot CF_2^{(4)})_2CF_2^{(5)}$	(1) -70 (2) -87 (3) -84 (4)(5) -124, -128			131
$C_9F_{18}I_2O_2$	As above with Br replaced by I	(1) -65.5 (2) -87.6 (3) -83.5 (4) -126 (5) -123			131
$C_{10}F_2H_{10}Cl_2$	$C_6H_5 \cdot CH^{(3)}(CH_3) \cdot CF_2^{(1)(2)} \cdot CH^{(4)}Cl_2$	(1) -112.1 (2) -116.3	(1-2) 247 (1-3) 23 (1-4) 4 (2-3) 8 (2-4) 15		203(a)
$C_{10}F_2H_{21}Br_2P$	$(iC_3H_7)_3P^+CF_2Br Br^-$	-43.7	(P-F) 56.4		162
$C_{10}F_3H_{13}NP$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot P(C_6H_5)N(CH_3)_2$ (diastereoisomers)	(1) -126.7 (2) -216.1	(2-4) ~48 (1-3) ~55 [a] (1A-1B) 292 (1-2) 19 (2-4) 47.4 (2-3) 9.2 (P-2) 71 [b] (1-2) 16.5 (1-3) 54.6 (2-3) 10.6 (2-4) 47.6 (P-2) 125 (1-4)(P-1) 15.1, 19.8		325(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_3H_{19}$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot C(CH_3)_2 \cdot CH_2 \cdot C(CH_3)_3$	(1) -128.0 (2) -204.5	(1-3) 56.0 (2-4) 46.0 (1-4) 12.0, 9.0	49
$C_{10}F_3H_{19}$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot CH[CH(CH_3)_2] \cdot C(CH_3)_3$	(1) -131.5 (2) -207.5	(1-3) 56.0 (2-4) 46.0	49
$C_{10}F_4H_9Cl_2NO_2S$	$(CF_2Cl)_2C(OH)NH \cdot S(O)C_6H_4 \cdot CH_3p$	-62.5 to -60.8		206(a)
$C_{10}F_6H_4BrCl$	$CF_2^{(1)}Cl \cdot CF_2^{(2)} \begin{array}{c} F^{(3)} \\ \diagup \quad \diagdown \\ \text{C} = \text{C} \\ \diagdown \quad \diagup \\ F^{(4)} \end{array} \cdot mBrC_6H_4$	(1) -69.4 (2) -106.7 (3) -72.4 (4) -71.0	(1-2) 4.7 (1-3) 10.0 (2-3) 9.3 (2-4) 28.0 (3-4) 5.1	4
$C_{10}F_6H_4I_2O_2$	$(CH_2I \cdot CH_2 \cdot CF_2^{(1)} \cdot CF_2^{(2)} \cdot OCF_2^{(3)})_2$	(1) -119 (2)(3) -87.8, -89.3		131
$C_{10}F_6H_5Cl$	$CF_2^{(1)}Cl \cdot CF_2^{(2)} \begin{array}{c} F^{(3)} \\ \diagup \quad \diagdown \\ \text{C} = \text{C} \\ \diagdown \quad \diagup \\ C_6H_5 \end{array}$	(1) -69.4 (2) -106.7 (3) -73.8 (4) -72.4	(1-2) 4.6 (1-3) 9.4 (2-3) 27.4 (2-4) 8.6 (3-4) 8.3	4
$C_{10}F_6H_7Br$	$pBrC_6H_4 \cdot CH_2^{(6)} \cdot CF_2^{(3)(4)} \cdot CF^{(2)}H^{(5)} \cdot CF_3^{(1)}$	(2) -211.5 (3) -108.7 (4) -111.3	(3-4) 260 (2-3,4) 11.5 (1-5) 6.0 (2-5) 44.0 (3-5) 12.0 (4-5) 6.0 (2-6) 3.0 (3-6) 16.0 (4-6) 15.0	176
$C_{10}F_6H_8$	$CH_3^{(4)} \cdot CF_2^{(1)} \cdot CF^{(2)}(CF_3^{(3)}) \cdot C_6H_5$	(1) -130.1 (2) -180.5 (3) -75.2	(1-3) 14.0 (1-4) 19.0 (2-3) 8.0 (2-4) 2.5	176
$C_{10}F_6H_8$	$CF_2^{(1)}H^{(4)} \cdot CF^{(2)}(CF_3^{(3)}) \cdot CH_2^{(5)} \cdot C_6H_5$	(1) -136.5 (2) -187.0 (3) -77.0	(1-2) 21.0 (1-3) 8.0 (1-4) 54.0 (2-3) 8.0 (2-4) 4.0 (2-5) 24.5	176
$C_{10}F_8H_2O_4$		(1) -91.8 (2)(3) -136.6, -152.3		230
$C_{10}F_{12}H_6O_2$	$(CH_2=CH \cdot CF_2^{(1)} \cdot CF_2^{(2)} \cdot OCF_2^{(3)})_2$	(1) -117 (2)(3) -87.4, -88.3		131
$C_{10}F_{18}O_2$	$(COF^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CF_2^{(5)})_2$	(1) +25.4 (2) -116.7 (3) -119.6 (4) -119.6 (2) -120.8		151

$C_{11}F_4H_8Cl_2$	$C_6H_5 \cdot CH_2^{(3)} \cdot CF_2^{(1)} \cdot CCl_2 \cdot CF_2^{(2)} \cdot CH^{(4)}Cl_2$	(1)(2) -100.5 to -103.5	(1-3) 19	(2-4) 9	203(a)
$C_{11}F_8H_{12}O_2$	$CF_2^{(1)}H^{(3)} \cdot (CF_2^{(2)})_3 \cdot C(OH)=CH \cdot CO \cdot R \quad (Z)$ $R = \left. \begin{array}{l} nC_4H_9 \\ iC_4H_9 \\ tC_4H_9 \end{array} \right\}$	(1) -139.9 (2) -128.8, -139.9, -120.0	(1-3) 55		113
$C_{12}F_3H_{17}NP$	$CF_2^{(1)}H^{(3)} \cdot CF^{(2)}H^{(4)} \cdot P[N(CH_2CH_3)_2]C_6H_5$	(1) -126.3 (2) -215.5	(1-3) 55	(2-4) 47	325(a)
$C_{13}F_2H_{27}Br_2P$	$[nC_4H_9P^+CF_2Br] Br^-$	-53.0	(P-F) 66.8		162
$C_{13}F_2H_{27}Br_2P$	$[tC_4H_9P^+CF_2Br] Br^-$	-38.4	(P-F) 53.2		162
$C_{13}F_2H_{27}I_2P$	$[nC_4H_9P^+CF_2I] I^-$	-62.0	(P-F) 64.7		102
$C_{13}F_2H_{28}BrP$	$[nC_4H_9P^+CF_2^{(1)}H^{(2)}] Br^-$	-128.1	(P-1) 64.2	(1-2) 46.5	162
$C_{13}F_2H_{30}Br_2NP$	$[(CH_3CH_2)_2NP^+CF_2Br] Br^-$	-47.4	(P-F) 88.4		162
$C_{13}F_2H_{31}BrNP$	$[(CH_3CH_2)NP^+CF_2^{(1)}H^{(2)}] Br^-$	-127.8	(P-1) 88.6	(1-2) 48.5	162
$C_{13}F_8H_8O_2$	$CF_2^{(1)}H^{(3)} \cdot (CF_2^{(2)})_4 \cdot C(OH)=CH \cdot CO \cdot C_6H_5 \quad (Z)$	(1) -139.9 (2) -128.8, -139.9, -120.0	(1-3) 55		113
$C_{13}F_8H_8O_3$	$C_6H_5CH_2 \cdot CO \cdot CF_2^{(1)} \cdot CF_2^{(2)} \cdot CF_2^{(3)} \cdot CF_2^{(4)} \cdot CO_2H$	(1)(2) -123.7 (3) -121.3 (4) -120.9			241(a)
$C_{13}F_9H_{16}NO_3$	$\overline{O \cdot CH_2 \cdot CH_2 \cdot O \cdot CH(CF_2^{(1)})_4 \cdot CF^{(2)}H \cdot CO \cdot N(CH_2CH_3)_2}$	(1) -119, -121, -124 (2) -197			232
$C_{13}F_{11}H_7N_2O_2$	 $NH \cdot CO \cdot CF_2^{(3)} \cdot CF_2^{(2)} \cdot CF_2^{(1)} \cdot CO \cdot N(CH_3)_2$	(1)(3) -118.7 (2) -125.7 (4) -146.1 (5) -157.9 (6) -165.1			219(a)
$C_{13}F_{12}HN$	$CF_2^{(1)}H^{(5)} \cdot N \left(\text{2-fluorophenyl} \right)_2$	(1) -89.8 (2) -145.1 (3) -161.9 (4) -153.9	(1-5) 60.3	(2-5) 0.8	263(b)
$C_{13}F_{18}H_6O_2$	$(CH_2=CH \cdot CF_2^{(1)} \cdot CF_2^{(2)} \cdot O \cdot CF_2^{(3)} \cdot CF_2^{(4)})_2 CF_2^{(5)}$	(1) -122 (2)(3) -83.5, -87.8 (4) -130 (5) -127			131
$C_{14}F_3H_{11}$	$(C_6H_5)_2CF^{(2)} \cdot CF_2^{(1)}H^{(3)}$	(1) -141.7 (2) -172.5	(1-2) 11.7 (2-3) 5.5	(1-3) 54	185

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₁₄ F ₄ H ₁₂ P ₂	(C ₆ H ₅ PH·CF ₂) ₂	-100.5		187
C ₁₄ F ₆ H ₂₂ N ₂ O ₂	[(CH ₃ CH ₂) ₂ NCO·CF ⁽²⁾ H ⁽³⁾ ·CF ₂ ⁽¹⁾] ₂	(1) -122 to -123 (2) -197	(2-3) 45.8	232
C ₁₅ F ₉ H ₂₁ N ₂ O ₂	(CH ₃ CH ₂) ₂ N·CO·CF ₂ ⁽⁶⁾ ·CF ₂ ⁽⁵⁾ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽²⁾⁽³⁾ ·CF ⁽¹⁾ H ⁽⁷⁾ ·CO·N(CH ₂ CH ₃) ₂	(1) -197 (2)(3) -117, -122 (4-6) -119 to -121	(1-7) 46 (2-3) 310	232
C ₁₅ F ₂₈ O ₇	F ⁽¹⁾ CO·CF ₂ ⁽²⁾ ·CF ₂ ⁽³⁾ ·CF ₂ ⁽⁴⁾ CF ₂ ⁽⁵⁾ ·O·(CF ₂ ⁽⁶⁾ ·CF ₂ ⁽⁶⁾ ·O) ₄ ·CF ₂ ⁽⁷⁾ ·CO·F ⁽⁸⁾	(1) +21.2 (2) -120.6 (3) -124.9 (4) -127.6 (5) -85.4 (6) -90.8 (7) -79.2 (8) +10.45		275(a)
C ₁₆ F ₅ H ₂₀ NO ₂	C ₆ H ₅ CH ₂ ·O·CH ₂ ·CF ₂ ⁽¹⁾ ·CF ₂ ⁽²⁾⁽³⁾ ·CF ⁽⁴⁾ H ⁽⁵⁾ ·CO·N(CH ₂ CH ₃) ₂	(1) -121 (2) -122 (3) -127	(2-3) 294 (4-5) 47 (2-5) 15 (3-5) 7	232
C ₁₆ F ₆ H ₁₉ NO	C ₆ H ₅ ·CH ₂ ·O·CH ₂ ·CF ₂ ⁽⁴⁾ ·CF ₂ ⁽³⁾ ·CF ⁽²⁾ =CF ⁽¹⁾ ·N(CH ₂ CH ₃) ₂	(1)(2) -115, -120 (3)(4) -117, -122	(1-2) 12	232
C ₁₉ F ₂ H ₁₂ Br ₂ ClP	[(pClC ₆ H ₄) ₃ P ⁺ CF ₂ Br] Br ⁻	-52.8	(P-F) 86.0	162
C ₁₉ F ₂ H ₁₅ Br ₂ P	[(C ₆ H ₅) ₃ P ⁺ CF ₂ Br] Br ⁻	-50.5	(P-F) 84.5	162
C ₁₉ F ₂ H ₁₅ I ₂ P	As above with Br replaced by I	-58.1	(P-F) 86.6	162
C ₁₉ F ₂ H ₁₆ BrP	[(C ₆ H ₅) ₃ P ⁺ CF ₂ ⁽¹⁾ H ⁽²⁾] Br ⁻	-126.2	(P-1) 79.7 (1-2) 46.6	162
C ₁₉ F ₂ H ₃₃ BrP	[(C ₆ H ₁₁) ₃ P ⁺ CF ₂ Br] Br ⁻	-42.5	(P-F) 55.2	162
C ₂₀ F ₈ H ₁₀ D ₄ O ₆	[D ₂ O·C ₆ H ₅ ·CO·CO·CF ₂ ⁽¹⁾ ·CF ₂ ⁽²⁾] ₂	(1) -117.6 (2) -118.8		241(b)
C ₂₀ F ₈ H ₁₀ O ₄	[C ₆ H ₅ ·CO·CO·CF ₂ ⁽¹⁾ ·CF ₂ ⁽²⁾] ₂	(1) -118.4 (2) -121.3	(1-2) 13.5	241(b)
C ₂₀ F ₈ H ₁₄ O ₂	[C ₆ H ₅ CH ₂ ·CO·CF ₂ ⁽¹⁾ ·CF ₂ ⁽²⁾] ₂	(1) -121.4 (2) -123.4	(1-2) 12.5	241(b)
C ₂₀ F ₈ H ₁₄ O ₆	[H ₂ O·C ₆ H ₅ ·CO·CO·CF ₂ ⁽¹⁾ ·CF ₂ ⁽²⁾] ₂	(1) -119.5 (2) -121.6		241(b)

$C_{21}F_2H_{28}O_2$			(1-2) 53	283
$C_{22}F_2H_{21}Br_2OP$	$[(pCH_3OC_6H_4)_3P^+CF_2Br] Br^-$	-52.8	(P-F) 86.6	162
$C_{22}F_2H_{21}Br_2P$	$[(pCH_3C_6H_4)_3P^+CF_2Br] Br^-$	-50.8	(P-F) 83.5	162
$C_{23}F_4H_{34}Ge_2P_2$	$[(CH_3)_2GeF(CH_3)_2PC_6H_5]_2CF_2$		[a] (P-F) 69 [b] (P-F) 70	115
$C_{25}F_2H_{51}Br_2P$	$[(nC_8H_{17})_3P^+CF_2Br] Br^-$	-52.3	(P-F) 64	162
$C_{26}F_8H_{20}MoO_2PSb$	$[(\eta-C_5H_5)Mo(CO)_2(P(C_6H_5)_3)CF_2^{(1)}]SbF_6$	(1) +160.54	$^1J(1-C)$ 392	244(a)
$C_{39}F_4H_{30}Cl_2OPPt$	<i>cis</i> - $[PtCl(CF_2^{(1)} \cdot CO \cdot CF_2^{(2)}Cl)(P(C_6H_5)_3)_2]$	(1) -86.3 (2) -62.8	(1-2) 7.8 (Pt-1) 373 (Pt-2) 17 (P-1) <i>cis</i> 27 (P-1) <i>trans</i> 37	292(c)

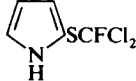
C. Fluorine nuclei in a CF group

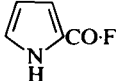
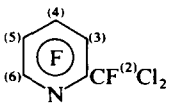
$CFBr_3$	$CFBr_3$		(Br-F) 30.5 (± 1.5)	1
$CFCl_3S$	$CFCl_2 \cdot SCl$	-29.3	(F-C) 329	2
CF_6H_3As	$CH_3^{(2)}F \rightarrow AsF_5$ (in SO_2ClF) (in SO_2F_2)	-198.4	(1-2) 41.4 (1-C) 133.7 (1-2) 41.3 (1-C) 134.1	17
CF_6H_3Sb	$CH_3^{(2)}F^{(1)} \rightarrow SbF_5$	-162.2	(1-2) 41.2 (1-C) 126.0	17
$CF_{12}O_5S_2$	$CF^{(1)}OF^{(2)} \cdot (OOSF_4^{(3)}F^{(4)})_2$	(1) -90.8 (2) +169.6 (3)(4) +57.1	(1-2) 20.0 (1-3) 4	18
$C_2FH_2BrCl_2$	$CF^{(1)}Cl_2 \cdot CH_2^{(2)}Br$		(1-2) 13.8	24
$C_2FH_2BrCl_2$	$CF^{(1)}H^{(2)}Cl \cdot CH^{(3)}BrCl$ (<i>erythro</i>) (<i>threo</i>)		(1-2) 53.6 (1-3) 16.2 (1-2) 51.2 (1-3) 12.5	24

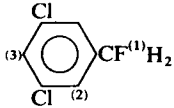
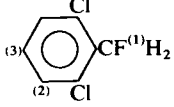
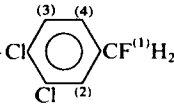
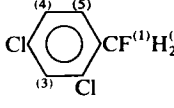
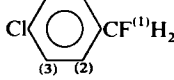
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_2FH_2Cl_3$	$CF^{(1)}Cl_2 \cdot CH_2^{(2)}Cl$		(1-2) 13.4	24
$C_2FH_2Cl_3$	$CF^{(1)}H^{(2)}Cl \cdot CH^{(3)}Cl_2$		(1-2) 52.6 (1-3) 13.0	24
C_2FH_2N	$CF^{(1)}H_2^{(2)} \cdot CN$	-232.3	(1-2) 45.0	25
$C_2FH_4Cl_3Si$	$CF^{(1)}H_2^{(2)} \cdot CH_2^{(3)} \cdot SiCl_3$	-210.51	(1-2) 46.92 (1-3) 20.42 (Si-1) 20.42 $^1J(F-C)$ 171.3 $^2J(F-C)$ 22.1	26
C_2FH_5	$CF^{(1)}H_2^{(2)} \cdot CH_3^{(3)}$		(1-2) +47.0 (1-3) +26.7 $^1J(F-C)$ -160.1 $^2J(F-C)$ +21.1	27
C_2FCl_5	$CFCl_2 \cdot CCl_3$	-62.4		23(a)
$C_2F_2Br_2Cl_2$	$CFBrCl \cdot CFBrCl$ (<i>meso</i>) (<i>dl</i>)	-63.8 -62.9		23(a)
$C_2F_2Cl_4$	$CFCl_2 \cdot CFCl_2$	-67.0		23(a)
$C_2F_3H_3ClNOS$	$CF^{(1)}H^{(3)}Cl \cdot CH_2 \cdot NSOF_2^{(2)}$	(1) -139.3 (2) +45.6	(1-2) 3.25 (1-3) 50	33
$C_2F_3H_3ClNOS$	$CH_2^{(3)}Cl \cdot CHF^{(1)} \cdot NSOF_2^{(2)}$	(1) -140.6 (2) +45.6	(1-3) 18.5 (2-3) 4.25	33
$C_2F_4N_2O_2S$	$CF^{(1)}(O)N=SF_2^{(2)}=NCF(O)$	(1) +17.9 (2) +51.2	(1-2) 9.5	37
$C_2F_5HCINOS$	$CF_2^{(3)}Cl \cdot CF^{(1)}H^{(4)} \cdot NSOF_2^{(2)}$	(1) -81.8 (2) +46.8	(3-4) 4.0 (1-4) 55 (1-2) 2.8	33
$C_2F_5HCINOS$	$CF^{(1)}H^{(4)}Cl \cdot CF_2^{(3)} \cdot NSOF_2^{(2)}$	(1) -73.0 (2) +47.0	(1-4) 51 (3-4) 4.0	33
$C_2F_7H_2BrS$	$CF^{(1)}H^{(5)}Br \cdot CH^{(6)}F^{(2)} \cdot SF_4^{(3)}F^{(4)}$ (<i>erythro</i>)	(1) -145.0 (2) -152.1 (3) +49.5 (4) +70.9	(1-2) 37 (1-3) 11 (1-4) 1 (2-4) 1 (3-4) 145 (1-5) 47 (1-6) 5 (2-6) 38 (3-6) 4.5 (4-6) 4.6 (2-5) 5	51

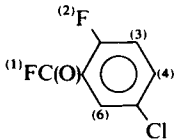
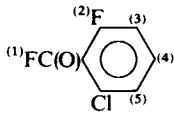
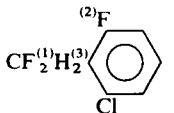
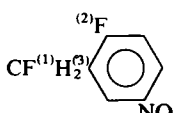
	(<i>threo</i>)	(1) -158.5 (2) -171.7 (3) +49.4 (4) +70.6	(1-2) 11 (1-3) 11 (3-4) 144 (1-5) 46 (1-6) 18 (2-6) 43 (3-6) 6 (2-5) 19	51
C ₂ F ₇ H ₂ ClO ₂ S	F ⁽¹⁾ SF ₄ ⁽²⁾ ·OO·CF ⁽³⁾ H ⁽⁵⁾ ·CF ⁽⁴⁾ H ⁽⁶⁾ Cl	(1)(2) +57.50 (3) -135.40 (4) -152.98	(1-3) 4.0 (2-3) 4 (3-4) 14.1 (3-5) 59.0 (3-6) 3.5 (4-5) 4.0 (4-6) 49.2	561
C ₃ FHCl ₆	CF ⁽¹⁾ Cl ₂ ·CH ⁽²⁾ Cl·CCl ₃	-50.08	(1-2) 2.67 Calculated: (1-2) <i>gauche</i> -1 ± 0.8 (1-2) <i>trans</i> +14.0 ± 1.6	54
C ₃ FH ₂ LiO ₃	CF ⁽¹⁾ H ₂ ⁽²⁾ ·C(O)CO ₂ ⁻ Li ⁺		(1-2) 46	55
C ₃ FH ₄ BrCl ₂	CF ⁽¹⁾ H ⁽²⁾ Cl·CH ⁽³⁾ Br·CH ₂ Cl (<i>erythro</i>) (<i>threo</i>)		(1-2) 48.7 (1-3) 9.2 (1-2) 49.5 (1-3) 14.8	24
C ₃ FH ₄ Cl ₃	CF ⁽¹⁾ H ⁽²⁾ Cl·CH ⁽³⁾ Cl·CH ₂ Cl (<i>erythro</i>) (<i>threo</i>)		(1-2) 48.7 (1-3) 9.4 (1-2) 49.3 (1-3) 13.4	24
C ₃ FH ₄ LiO ₄	CF ⁽¹⁾ H ₂ ⁽²⁾ ·C(OH) ₂ CO ₂ ⁻ Li ⁺		(1-2) 47	55
C ₃ FH ₄ N	CH ₃ ⁽³⁾ ·CF ⁽¹⁾ H ⁽²⁾ ·CN	-182.2	(1-2) 46.8 (1-3) 23.4	25
C ₃ FH ₅ Br ₂	(1 <i>S</i> , 2 <i>S</i>)-CF ⁽¹⁾ BrH ⁽²⁾ ·CH ⁽³⁾ Br·CH ₃ (1 <i>R</i> , 2 <i>S</i>)-isomer	-128.5 -145.5	(1-2) 49.8 (1-3) 9.1 (1-2) 50.3 (1-3) 21.6	56(a)
C ₃ FH ₅ ClI	CF ⁽¹⁾ H ⁽²⁾ I·CH ₂ ⁽³⁾⁽⁴⁾ ·CH ₂ Cl	-143.6	(1-2) 50.0 (1-3) 21.5 (1-4) 15.0	57
C ₃ FH ₅ ClI	CH ₂ ⁽³⁾ I·CF ⁽¹⁾ H ⁽²⁾ ·CH ₂ ⁽⁴⁾ Cl	-170.2	(1-2) 46.2 (1-3) 17.5 (1-4) 17.5	57
C ₃ FH ₅ DNO ₂	CF ⁽¹⁾ H ₂ ⁽²⁾ ·CD(NH ₂)·CO ₂ H	-227.3	(1-2) 46	58
C ₃ FH ₆ Br	CF ⁽¹⁾ BrH ⁽²⁾ ·CH ₂ ⁽³⁾ ·CH ₃	-132.0	(1-2) 50.5 (1-3) 18.0	56
C ₃ FH ₆ Br	CH ₃ ⁽³⁾ ·CF ⁽¹⁾ H ⁽²⁾ ·CH ₂ ⁽⁴⁾ Br	-170.4	(1-2) 47.8 (1-3) 23.0 (1-4) 18.2	59
C ₃ FH ₆ Cl	CF ⁽¹⁾ H ₂ ⁽²⁾ ·CH ⁽³⁾ Cl·CH ₃	-215.8	(1-2) 46.7 (1-3) 14.3	59

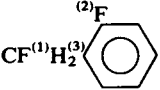
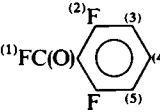
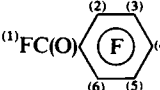
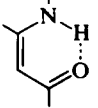
Molecular formula	Structure of compounds	Chemical shifts (ppm)			Coupling constants (Hz)			Ref.
C ₃ FH ₆ Cl ₃ Si	CF ⁽¹⁾ H ₂ ⁽²⁾ ·CH ₂ ·CH ₂ ·SiCl ₃	-219.85			(1-2) 46.5	¹ J(1-C) 169.9		26
					² J(1-C) 21.3	³ J(1-C) 5.1		
C ₃ FH ₆ I	CH ₃ ⁽³⁾ ·CF ⁽¹⁾ H ⁽²⁾ ·CH ₂ ⁽⁴⁾ I	-162.3			(1-2) 47.6	(1-3) 22.7		59
					(1-4) 17.1			
C ₃ FH ₆ NO ₂	CF ⁽¹⁾ H ₂ ⁽²⁾ ·CH ⁽³⁾ (NH ₂)·CO ₂ H	-233.2, -233.5			(1-2) 46	(1-3) 29.5		58
C ₃ FH ₆ NO ₂	CF ⁽¹⁾ H ₂ ⁽²⁾⁽³⁾ ·CH ⁽⁴⁾ (NH ₂)·CO ₂ H	(1-2)	(1-3)	(1-4)	¹ J(1-C)	² J(1-C)	³ J(1-C)	60
	(neutral)	45.98	47.40	29.55	169	20.0	6.3	
	(acid)	45.51	47.25	30.10	171	20.0	5.2	
	(alkaline)	46.81	47.49	29.94	166	19.4	7.6	
C ₃ FH ₆ NO ₂	CH ₂ ⁽³⁾⁽⁴⁾ NH ₂ ·CF ⁽¹⁾ H ⁽²⁾ ·CO ₂ H	(1-2)	(1-3)	(1-4)	¹ J(1-C2)	² J(1-C1)	² J(1-C3)	60
	(neutral)	-188.45	50.58	28.11	17.73	184	21.3	21.3
	(acid)	-189.83	49.92	27.77	18.54	184	22.5	21.1
	(alkaline)	-188.13	51.33	26.56	25.32	181	21.0	21.4
C ₃ FH ₇	CF ⁽¹⁾ H ₂ ⁽²⁾ ·CH ₂ ⁽³⁾ ·CH ₃ ⁽⁴⁾				(1-2) +48.0	(1-3) +25.2		27
					(1-4) 0.5	¹ J(1-C) -163.3		
					² J(1-C) +19.5			
					³ J(1-C) +6.7			
C ₃ FH ₇	CF ⁽¹⁾ H ⁽²⁾ ·(CH ₃ ⁽³⁾) ₂				(1-2) +48.4	(1-3) +23.2		27
					¹ J(1-C) -162.1			
					² J(1-C) +22.4			
C ₃ FH ₉ N ₂ O ₃	CF ⁽¹⁾ H ₂ ·C(OH)(NH ₂)CO ₂ ⁻ NH ₄ ⁺				¹ J(F-C) 172	² J(F-C) 19		55
					³ J(F-C) 3			
C ₃ FH ₁₀ N ₃ O ₂	CFH ₂ ·C(NH ₂) ₂ CO ₂ ⁻ NH ₄ ⁺				¹ J(F-C) 173	² J(F-C) 19		55
					³ J(F-C) 3			

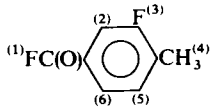
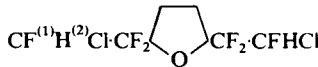
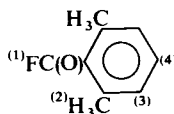
$C_3F_2HCl_5$	$CF^{(1)}Cl_2 \cdot CH^{(3)}Cl \cdot CF^{(2)}Cl_2$	-54.54	(1-2) 20.9 (1-3) 4.93 Calculated: (1-3) <i>gauche</i> 3.3 ± 0.3 in both conformers (1-3) <i>trans</i> 15.5 ± 0.6	54
$C_3F_2H_3N$	$CF^{(1)}H_2 \cdot CF^{(2)}H \cdot CN$	(1) -244.7 (2) -196.6		25
$C_3F_3H_7Si$	$(CH_2F^{(1)})_2SiF^{(2)} \cdot CH_3^{(3)}$	(1) -280.0 (2) -176.0	(1-3) 4.8 (2-3) 5.8	45(a)
C_4FH_6N	$CF^{(1)}H_2^{(2)} \cdot CH_2^{(3)} \cdot CH_2 \cdot CN$		(1-2) 47 (1-3) ~32	79
$C_4FH_7Br_2N_2O_4Si$	$F^{(1)}C(NO_2)_2 \cdot CH_2^{(2)} \cdot CH_2 \cdot SiBr_2 \cdot CH_3$	-103.5	(1-2) 17	80
$C_4FH_{11}Si$	$CF^{(1)}H_2^{(2)} \cdot Si(CH_3^{(3)})_3$	-270.37	(1-2) 47.53 (1-3) 0.64 (Si-1) 21.0 $^1J(1-C)$ 161.0 $^3J(1-C)$ 2.2	26
$C_4F_2H_{10}Si$	$(CF^{(1)}H_2^{(2)})_2Si(CH_3)_2$	-273.64	(1-2) 47.1	45(a)
$C_4F_3H_4BrO_2$	$CF^{(1)}H^{(2)}Br \cdot CF_2 \cdot CO_2CH_3$		(1-2) 47.5	64
$C_4F_3H_5BrNO$	$CF^{(1)}H^{(2)}Br \cdot CF_2 \cdot C(=NH)OCH_3$		(1-2) 45.5	64
$C_4F_3H_7N_2O_4Si$	$F^{(1)}C(NO_2)_2 \cdot CH_2^{(3)} \cdot CH_2 \cdot SiF_2^{(2)} \cdot CH_3^{(4)}$	(1) -104.5 (2) -132.3	(1-3) 16 (2-4) 6	80
$C_4F_3H_9Si$	$(CF^{(1)}H_2^{(2)})_3SiCH_3$	-276.23	(1-2) 46.8	45
$C_4F_4H_6$	$(CF^{(1)}H_2^{(3)})_3CF^{(2)}$	(1) -240.6 (2) -181.9	(1-2) 12.5 (1-3) 48.4 (2-3) 17.9	83
$C_4F_4H_8Ge$	$(CF^{(1)}H_2^{(2)})_4Ge$	-269.9	(1-2) 46.7	45(a)
$C_4F_4H_8Si$	$(CF^{(1)}H_2^{(2)})_4Si$	-279.63	(1-2) 46.9	45(a)
$C_4F_7H_2ClO$	$CF^{(1)}H_2^{(2)} \cdot O \cdot CF_2 \cdot CF_2 \cdot CF_2Cl$		(1-2) 51.0	64
$C_5FH_4Cl_2NS$		-22.51		109

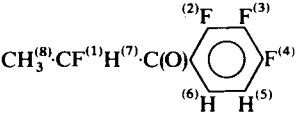
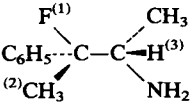
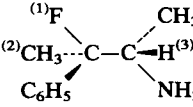
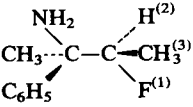
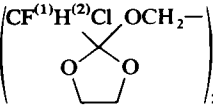
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₃ FH ₄ NOS		+41.11		109
C ₃ FH ₈ ClO	CH ₂ ⁽³⁾ Cl·CF ⁽¹⁾ H ⁽²⁾ ·CO·CH ₂ ·CH ₃	-196	(1-2) 48 (1-3) 22	110
C ₃ FH ₁₀ NO	(CH ₃ CH ₂) ₂ N·CO·F	-23.1		111
C ₃ FH ₁₀ NO ₂	(CH ₃ ⁽²⁾) ₂ CF ⁽¹⁾ ·CH ⁽³⁾ (NH ₂)·CO ₂ H	-143.5	(1-2) 22.5, 24 (1-3) 14	112
C ₃ FH ₁₁	(CH ₃) ₃ C·CH ₂ ⁽²⁾ F ⁽¹⁾	-222.23	(1-2) 48.0 ¹ J(1-C) 173.5 ² J(1-C) 17.6 ³ J(1-C) 4.4	26
C ₃ F ₂ H ₈ N ₂ O ₂ S	CF ⁽¹⁾ (O)·N=SF ⁽²⁾ · $\overbrace{\text{N} \cdot \text{CH}_2 \cdot \text{CH}_2 \cdot \text{O} \cdot \text{CH}_2 \cdot \text{CH}_2}^{\text{---}}$	(1)(2) +11.5, +27.9		114(c)
C ₃ F ₂ H ₁₀ N ₂ O ₅ Si	CF ⁽¹⁾ (NO ₂) ₂ CH ₂ ⁽³⁾ ·CH ₂ ·SiF ⁽²⁾ (CH ₃ ⁽⁴⁾)·OCH ₃	(1) -104.7 (2) -138.1	(1-3) 16 (2-4) 6	80
C ₃ F ₄ H ₇ Cl	CF ⁽¹⁾ H ⁽³⁾ Cl·CF ₂ ·CH(CH ₃)·CF ⁽²⁾ H ₂ ⁽⁴⁾		(1-3) 48 (2-4) 48	116
C ₆ FH ₁₀ ClO ₂	CF ⁽¹⁾ H ⁽²⁾ Cl·CO·O·(CH ₂) ₃ ·CH ₃		(1-2) 50	111
C ₆ FH ₁₃ N ₂ O ₄ Si	CF(NO ₂) ₂ ·CH ₂ ·CH ₂ ·Si(CH ₃) ₃	-106.0		80
C ₆ FH ₁₃ N ₂ O ₅ Si	CF(NO ₂) ₂ CH ₂ ·O·CH ₂ ·Si(CH ₃) ₃	-110.25		80
C ₆ FH ₁₅ Si	CF ⁽¹⁾ H ₂ ⁽²⁾ ·(CH ₂) ₂ ·Si(CH ₃) ₃	-215.09	(1-2) 48.0 (Si-1) 1.0 ¹ J(1-C) 169.9 ² J(1-C) 19.8 ³ J(1-C) 4.4	26
C ₆ F ₂ H ₁₀ N ₂ OS	CF(O)·N=SF·N(CH ₂) ₅	+11.4, +30.0		114(c)
C ₆ F ₅ Cl ₂ N		(2) -60.7 (3) -139.4 (4) -137.1 (5) -153.1 (6) -83.4	(2-3) 29.8 (3-4) 17.6 (3-5) 9.3 (3-6) 26.8 (4-5) 16.4 (4-6) 17.5 (5-6) 23.0	320(a)
C ₇ FH ₅ BrD	<i>p</i> BrC ₆ H ₄ ·CH ⁽²⁾ DF ⁽¹⁾	-208.42	(1-2) 47.9 (1-D) 7.2	158

C_7FH_5ClD	$pClC_6H_4 \cdot CH^{(2)}DF^{(1)}$	-207.90	(1-2) 47.7	(1-D) 7.3	158
$C_7FH_5Cl_2$			(1-2) -1.34	(1-3) 1.11	159
$C_7FH_5Cl_2$			(1-2) 1.25	(1-3) -2.43	159
$C_7FH_5Cl_2$			(1-2) -1.58 (1-4) -1.49	(1-3) 1.12	159
$C_7FH_5Cl_2$			(1-2) 47.17 (1-4) 0.71	(1-3) 1.35 (1-5) -0.70	159
$C_7FH_5DNO_2$	$pNO_2C_6H_4 \cdot CH^{(2)}DF^{(1)}$	-208.42	(1-2) 47.9	(1-D) 7.2	158
C_7FH_6Cl	$pClC_6H_4 \cdot CFH_2$	-208.34			160
C_7FH_6Cl			(1-2) -1.67	(1-3) 1.11	159
C_7FH_6D	$C_6H_5 \cdot CH^{(2)}DF^{(1)}$	-207.11	(1-2) 48.4	(1-D) 7.6	158
$C_7FH_6NO_2$	$pNO_2C_6H_4 \cdot CFH_2$	-216.46			160
C_7FH_7	$C_6H_5 \cdot CFH_2$	-207.60			160
C_7FH_7S	$C_6H_5 \cdot S \cdot CF^{(1)}H_2^{(2)}$	-185.3	(1-2) 60		161
$C_7FH_{12}ClO$	$CH_3 \cdot (CH_2)_3 \cdot CO \cdot CF^{(1)}H^{(2)} \cdot CH_2^{(2)}Cl$	-195.1	(1-2) 50	(1-3) 24	110

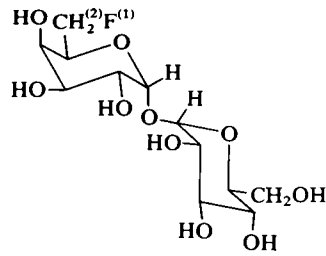
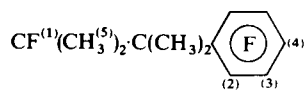
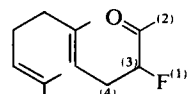
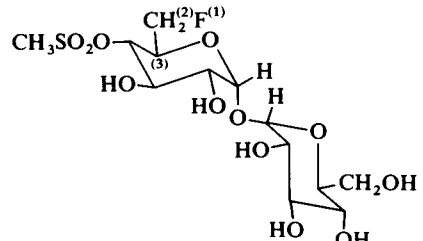
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_7FH_{15}	$CF^{(1)}H_2^{(2)} \cdot (CH_2)_5 \cdot CH_3$	-218.9	(1-2) 47.5 $^1J(1-C)$ 164.8 $^2J(1-C)$ 19.5 $^3J(1-C)$ 4.9	26
$C_7FH_{15}N_2O_4Si$	$CF^{(1)}(NO_2)_2 \cdot CH_2^{(2)} \cdot CH_2 \cdot CH_2 \cdot Si(CH_3)_3$	-102.8	(1-2) 19	80
$C_7FH_{18}Br_3NP$	$[(CH_3)_2N]_3P^+CFBr_2 Br^-$	-67.6	(P-F) 76.7	162
$C_7FH_{18}Cl_3NP$	As above with Br replaced by Cl	-58.8	(P-F) 84.2	162
$C_7FH_{19}Br_2NP$	$[(CH_3)_2N]_3P^+CF^{(1)}H^{(2)}Br Br^-$	-145.6	(P-1) 72.0 (1-2) 46.1	162
$C_7FH_{19}Cl_2NP$	As above with Br replaced by Cl	-153.6	(P-1) 78.0 (1-2) 43.0	162
$C_7FH_{20}BrNP$	$[(CH_3)_2N]_3P^+CF^{(1)}H_2^{(2)} Br^-$	-244.7	(P-1) 59.9 (1-2) 45.4	162
$C_7F_2H_3ClO$		(1) +31.14 (2) -109.15	(1-2) 43.56 (1-3) 0.89 (1-4) <0.03 (1-6) -0.68 (2-3) 9.76 (2-4) 4.11 (2-6) 5.95	163(n)
$C_7F_2H_3ClO$		(1) +51.15 (2) -107.28	(1-2) 30.60 (1-3) 0.76 (1-4) -0.29 (1-5) 0.82 (2-3) 9.30 (2-4) 5.83 (2-5) -1.43	163(n)
$C_7F_2H_5Cl$		(1) -211.71 (2) -113.81	(1-2) 2.86 (1-3) 47.47 (2-3) 2.07	164
$C_7F_2H_5NO_2$		(1) -219.67 (2) -109.19	(1-2) 1.10 (1-3) 46.65 (2-3) 1.28	164

$C_7F_2H_6$		(1) -216.72 (2) -119.28	(1-2) 0.82 (1-3) 46.90 (2-3) 1.13	164
$C_7F_2H_6$	$pFC_6H_4 \cdot CF^{(1)}H_2$	(1) -204.86		160
$C_7F_3H_3O$		(1) +46.98 (2) -106.0	(1-2) 37.53 (1-3) 1.00 (1-4) <0.03 (2-3) 9.63 (2-4) 6.03 (2-6) 0.51 (2-5) -1.32	163(n)
$C_7F_3H_8Cl_3O_2$	$CF^{(1)}H^{(2)}Cl \cdot CF_2 \cdot CH(CH_3) \cdot CH_2 \cdot O \cdot CO \cdot CHCl_2$		(1-2) 48	116
C_7F_6O		(1) +47.08 (2) -134.74 (3) -160.55 (4) -143.92	(1-2) 38.84 (1-3) 2.59 (1-4) 1.13 (2-3) -20.64 (2-4) 7.88 (2-6) -8.57 (3-4) -19.60 (3-5) -0.54 (2-5) 8.92	163
C_8FH_6N	$pCNC_6H_4 \cdot CFH_2$	-216.01		160
C_8FH_8D	$pCH_3C_6H_4 \cdot CF^{(1)}H^{(2)}D$	-204.23	(1-2) 48.4 (1-D) 7.6	158
C_8FH_9	$pCH_3C_6H_4 \cdot CFH_2$	-204.57		160
C_8FH_9O	$pCH_3OC_6H_4 \cdot CFH_2$	-200.14		160
$C_8FH_{11}KNO_3$	$CF^{(1)}H_2^{(2)}CH^{(3)}CO_2K^+$ 		(1-2) ~46 (1-3) ~31	182
$C_8FH_{13}O$	$CF^{(1)}H_2^{(2)} \cdot CO \cdot C_6H_{11}$	-277	(1-2) 48	183
$C_8FH_{14}ClO$	$CH_2^{(3)}Cl \cdot CF^{(1)}H^{(2)} \cdot CO \cdot (CH_2)_4 \cdot CH_3$	-195.4	(1-2) 49 (1-3) 23	110
$C_8FH_{19}Si$	$CF^{(1)}H_2^{(2)} \cdot (CH_2)_4 \cdot Si(CH_3)_3$	-218.06	(1-2) 48.0	26

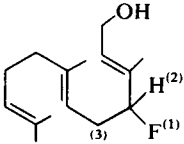
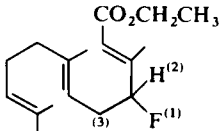
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_2H_6O$		(1) +17.16 (3) -114.87	(1-2) -0.481 (1-3) 4.516 (1-4) <0.03 (1-5) 1.563 (1-6) -0.592 (2-3) 9.516 (3-4) 2.119 (3-5) 7.310 (3-6) -0.531	163(n)
$C_8F_2H_8$	$CF^{(1)}H_2^{(3)} \cdot CF^{(2)}H^{(4)} \cdot C_6H_5$	(1) -247.5 (2) -208.5	(1-2) 16 (1-3) 48 (2-4) 50 (2-3) 22	185
$C_8F_4H_{14}N_4O_9Si_2$	$[CF^{(1)}(NO_2)_2 \cdot CH_2^{(3)} \cdot CH_2 \cdot SiF^{(2)}(CH_3^{(4)})]_2O$	(1) -105.5 (2) -132.0	(1-3) 20 (2-4) 6	80
$C_8F_6H_8Cl_2O$			(1-2) 48	116
C_9FH_8ClO	$CH_2^{(3)}Cl \cdot CF^{(1)}H^{(2)} \cdot CO \cdot C_6H_5$	-188.6	(1-2) 47 (1-3) 23	110
C_9FH_9O			(1-2) -3.091 (1-3) 1.087 (1-4) -0.370	202
C_9FH_9O	$CF^{(1)}H_2^{(2)} \cdot CO \cdot CH_2^{(3)} \cdot C_6H_5$	-228	(1-2) 47 (1-3) 3	183
$C_9FH_{16}N_3O_3S$	$F \cdot CO \cdot N = S \cdot \overbrace{(N \cdot CH_2 \cdot CH_2 \cdot O \cdot CH_2 \cdot CH_2)_2}$	+3.8		114(c)
$C_9F_2H_8Cl_2$	$CF^{(2)}H^{(3)}Cl \cdot CF^{(1)}Cl \cdot CH_2^{(4)} \cdot C_6H_5$	[a] (1) -116.8 (2) -143.9	(1-2) 20 (1-3) 6 (1-4) 21, 16 (2-3) 49	203(a)
		[b] (1) -117.4 (2) -143.0	(1-2) 20 (1-3) 5 (1-4) 20 (2-4) 49	
$C_9F_2H_{10}$	$CF^{(1)}H_2^{(3)} \cdot CF^{(2)}(CH_3^{(4)}) \cdot C_6H_5$	(1) -246 (2) -172.5	(1-2) 13.5 (1-3) 48 (1-4) 3 (2-3) 33 (2-4) 21	185

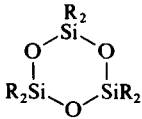
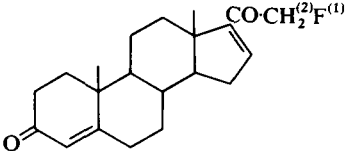
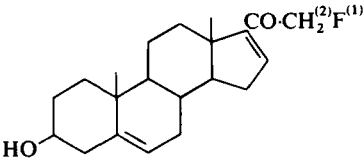
$C_9F_4H_6O$	$CH_3^{(8)} \cdot CF^{(1)}H^{(7)} \cdot C(O) \cdot$ 	$(1) -184.4$ $(2) -133.3$ $(3) -162.3$ $(4) -128.7$	$(1-2) 15.5$ $(1-7) 48.0$ $(1-8) 23.0$ $(2-3) 20.0$ $(2-4) 12.0$ $(3-4) 19.0$ $(3-5) 6.5$ $(3-6) 1.5$ $(4-5) 9.2$ $(4-6) 6.2$	424
$C_{10}FH_{10}ClO$	$CH_2^{(3)}Cl \cdot CF^{(1)}H^{(2)} \cdot CO \cdot CH_2 \cdot C_6H_5$	-193.9	$(1-2) 49$ $(1-3) 22$	110
$C_{10}FH_{11}O$	$(CH_3^{(2)})_2CF^{(1)} \cdot CO \cdot C_6H_5$	-144	$(1-2) 22$	223
$C_{10}FH_{14}N$	$C_6H_5 \cdot CF(CH_2CH_3) \cdot CH_2 \cdot NH_2$	-170.8		223
$C_{10}FH_{14}N$	$C_6H_5 \cdot CF^{(1)}H^{(2)} \cdot C(CH_3)_2 \cdot NH_2$	-186.4	$(1-2) 45$	223
$C_{10}FH_{14}N$	$C_6H_5 \cdot CH^{(2)}(NH_2) \cdot CF^{(1)}(CH_3^{(3)})_2$	-141.5	$(1-2) 12$ $(1-3) 21$	223
$C_{10}FH_{14}N$		-159.1	$(1-2) 23$ $(1-3) 16$	223
$C_{10}FH_{14}N$		-158.3	$(1-2) 23$ $(1-3) 16$	223
$C_{10}FH_{14}N$		-180.8	$(1-2) 48$ $(1-3) 24$	223
$C_{10}FH_{20}NO$	$CH_3 \cdot (CH_2)_4 \cdot CO \cdot CF^{(1)}H^{(2)} \cdot CH_2^{(3)} \cdot N(CH_3)_2$	-191.7	$(1-2) 50$ $(1-3) 26$	110
$C_{10}F_2H_{14}Cl_2O_6$		-149	$(1-2) 49.5$	116
$C_{10}F_3H_{17}ClNO_2$	$CF^{(1)}H^{(2)}Cl \cdot CF_2 \cdot CH(CH_3) \cdot CH_2 \cdot O \cdot CO \cdot N(CH_2CH_3)_2$		$(1-2) 48$	116
$C_{10}F_5H_{10}Cl_3O_4$	$CF^{(1)}H^{(4)}Cl \cdot CF_2 \cdot CH(OCO \cdot CF^{(2)}H^{(5)}Cl) \cdot (CH_2)_3 \cdot O \cdot CO \cdot CF^{(3)}H^{(6)}Cl$		$(1-4) 48$ $(2-5) 50$ $(3-6) 50$	116

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{11}FH_{12}BrO_2$	$oBrC_6H_4 \cdot CF^{(1)}H^{(2)} \cdot CH_2 \cdot CO_2CH_2CH_3$	-181.36	(1-2) 49.35	238(b)
	<i>m</i> -isomer	-175.92	45.05	
	<i>p</i> -isomer	-175.01	46.0	
$C_{11}FH_{12}ClO_2$	$oClC_6H_4 \cdot CF^{(1)}H^{(2)} \cdot CH_2 \cdot CO_2CH_2CH_3$	-182.01	(1-2) 48.90	238(b)
	<i>m</i> -isomer	-175.62	45.59	
	<i>p</i> -isomer	-173.05	45.80	
$C_{11}FH_{12}NO_3$	$oNO_2C_6H_4 \cdot CF^{(1)}H^{(2)} \cdot CH_2 \cdot CO_2CH_2CH_3$	-178.06	(1-2) 49.50	238(b)
	<i>m</i> -isomer	-177.62	47.0	
	<i>p</i> -isomer	-178.98	47.0	
$C_{11}FH_{13}O_2$	$C_6H_5 \cdot CF^{(1)}H^{(2)} \cdot CH_2 \cdot CO_2CH_2CH_3$	-174.14	(1-2) 46.50	238(b)
$C_{11}FH_{14}NO$	$C_6H_5 \cdot CO \cdot CF^{(1)}H^{(2)} \cdot CH_2^{(3)} \cdot N(CH_3)_2$	-189.2	(1-2) 49 (1-3) 26	110
$C_{11}FH_{16}N$	$C_6H_5 \cdot CF^{(1)}(CH_3^{(2)}) \cdot C(NH_2)(CH_3)_2$	-155.1	(1-2) 24	223
$C_{11}FH_{16}N$	$C_6H_5 \cdot C(NH_2)(CH_3) \cdot CF^{(1)}(CH_3^{(2)})_2$	-149.1	(1-2) 22	223
$C_{11}FH_{20}N_3OS$	$C(O)F \cdot N = S[N(CH_2)_5]_2$	+5.3		114(c)
$C_{11}F_2H_{12}O_2$	$oFC_6H_4 \cdot CF^{(1)}H^{(2)} \cdot CH_2 \cdot CO_2CH_2CH_3$	-181.16	(1-2) 47.94	238(b)
	<i>m</i> -isomer	-174.63	46.0	
	<i>p</i> -isomer	-172.23	46.0	
$C_{12}FH_{14}ClO$	$C_6H_5 \cdot (CH_2)_3 \cdot CO \cdot CF^{(1)}H^{(2)} \cdot CH_2^{(3)}Cl$	-195.8	(1-2) 50 (1-3) 23	110
$C_{12}FH_{15}O_2$	$mCH_3C_6H_4 \cdot CF^{(1)}H^{(2)} \cdot CH_2 \cdot CO_2CH_2CH_3$	-174.36	(1-2) 47.95	238(b)
$C_{12}FH_{15}O_3$	$mCH_3OC_6H_4 \cdot CF^{(1)}H^{(2)} \cdot CH_2 \cdot CO_2CH_2CH_3$	-173.45	(1-2) 47.48	238(b)
$C_{12}FH_{16}NO$	$C_6H_5 \cdot CH_2 \cdot CO \cdot CF^{(1)}H^{(2)} \cdot CH_2^{(3)} \cdot N(CH_3)_2$	-190.6	(1-2) 49 (1-3) 24	110

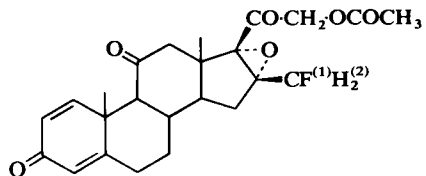
$C_{12}FH_{21}O_{10}$		-231.8	(1-2) ~40	(1-3) ~17	246(b)	
$C_{12}F_6H_{12}$	$CF^{(1)}(CH_3^{(5)})_2 \cdot C(CH_3)_2$ 	(1) -150 (3) -170	(2) -139 (4) -165	(1-2) 3.5 (2-3) 20	(1-5) 22 (3-4) 20	249
$C_{13}FH_{15}O$	$C_6H_5 \cdot CO \cdot CF^{(1)}(CH_2)_4 \cdot CH_2$	-165				223
$C_{13}FH_{17}O_2$	$C_6H_5 \cdot CF^{(1)}H^{(2)} \cdot C(CH_3)_2CO_2CH_2CH_3$	-174.65		(1-2) 45.75		238(b)
$C_{13}FH_{21}O$				(1-2) 4.5 (1-4) 26	(1-3) 50	257
$C_{13}FH_{23}O_{12}S$		-239.0	(1-2) 50	(1-3) 27		246(b)
$C_{14}FH_{11}Br_2$	$(C_6H_5)_2CF^{(1)} \cdot CH_2^{(2)}Br$	-158		(1-2) 22.5		264
$C_{14}FH_{11}O$	$C_6H_5 \cdot CO \cdot CF^{(1)}H^{(2)} \cdot C_6H_5$	-175.4		(1-2) 50		223

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}FH_{14}N$		-180.5	(1-2) 47 (1-3) 14	223
$C_{14}FH_{14}N$		-182.6	(1-2) 47 (1-3) 14	223
$C_{14}FH_{19}O_9$		-233.5	(1-2) 48.2 (1-3) 22.0	265
$C_{14}FH_{20}NO$	$C_6H_5 \cdot (CH_2)_3 \cdot CO \cdot CF^{(1)}H^{(2)} \cdot CH_2^{(3)} \cdot N(CH_3)_2$	-192	(1-2) 50 (1-3) 24	110
$C_{14}FH_{20}NO_8$		(α) -233.1 (β) -232.5	(1-2) 46.9 (1-3) 23.0 (1-2) 48.2 (1-3) 21.8	265
$C_{14}FH_{20}NO_8$		-230.8	(1-2) 47.5 (1-3) 18.5	265
$C_{14}FH_{20}NO_8$		-234.0, -233.3	(1-2) 49.5, 47.5 (1-3) 26.0, 23.5	265
$C_{14}F_2H_{11}Cl$	$(mClC_6H_4)C_6H_5CF^{(1)} \cdot CH_2^{(3)}F^{(2)}$	(1) -169.5 (2) -245	(1-2) 18 (1-3) 24 (2-3) 47	185
$C_{14}F_2H_{11}Cl$	As above but with <i>p</i> ClC ₆ H ₄	(1) -169.5 (2) -245	(1-2) 18 (1-3) 23 (2-3) 47	185

$C_{14}F_2H_{12}$	As above with Cl replaced by H	(1) -169.5 (2) -245	(1-2) 18 (2-3) 47	(1-3) 22	185
$C_{15}FH_{16}N$	$(C_6H_5)_2CF^{(1)} \cdot CH^{(2)}(CH_3) \cdot NH_2$	-174.5	(1-2) 25		223
$C_{15}FH_{25}O$			(1-2) ~48	(1-3) 26	257
$C_{15}F_2H_{14}$	$(pCH_3C_6H_4)C_6H_5 \cdot CF^{(2)} \cdot CH_2^{(3)}F^{(1)}$	(1) -254 (2) -169.5	(1-2) 18 (2-3) 22.5	(1-3) 47	185
$C_{15}F_2H_{14}$	As above but <i>m</i> CH ₃	(1) -245 (2) -170	(1-2) 18 (2-3) 24	(1-3) 47	185
$C_{15}F_2H_{14}$	$(C_6H_5)_2CF^{(1)} \cdot CF^{(2)}H^{(3)} \cdot CH_3$	(1) -173.3 (2) -194.3	(1-2) 17 (2-3) 45	(1-3) 18	185
$C_{15}F_2H_{14}O$	$(pCH_3OC_6H_4)C_6H_5 \cdot CF^{(2)} \cdot CH_2^{(3)}F^{(1)}$	(1) -254 (2) -167.5	(1-2) 18 (2-3) 22	(1-3) 47	185
$C_{16}FH_{17}N_2O_4Si$	$CF^{(1)}(NO_2)_2 \cdot CH_2^{(2)} \cdot CH_2 \cdot Si(CH_3)(C_6H_5)_2$	-104.4	(1-2) 22		80
$C_{16}F_2H_{22}O_2$	$C_6H_{11} \cdot C(OH)(CF^{(1)}H_2^{(3)}) \cdot CF^{(2)}H^{(4)} \cdot C_6H_{11}$	(1) -200.75, -204.1 (2) -228.25, -231.6	(1-3) 48, 48 (2-4) 48, 48		183
$C_{17}FH_{27}O_2$			(1-2) ~50	(1-3) 23	257
$C_{18}F_2H_{18}N_4O_8Si$	$[CF^{(1)}(NO_2)CH_2^{(2)}CH_2]_2Si(C_6H_5)_2$	-105.7	(1-2) 18		80
$C_{18}F_2H_{18}O_2$	$CF^{(1)}H_2^{(3)} \cdot C(OH)(CH_2C_6H_5) \cdot CF^{(2)}H^{(4)} \cdot CO \cdot CH_2 \cdot C_6H_5$	(1) -193 and -200.8 (2) -226.2 and -231.2	(1-3) 48 and 49 (2-4) 47 and 48		183

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{18}F_6H_{24}N_{12}O_{27}Si_3$	 $R = CF(NO_2)_2CH_2CH_2$	-106.0		80
$C_{19}FH_{15}Br_3P$	$[(C_6H_5)_3P^+CFBr_2] Br^-$	-74.2	(P-F) 77.9	162
$C_{19}FH_{16}Br_2P$	$[(C_6H_5)_3P^+CF^{(1)}H^{(2)}Br] Br^-$	-162.6	(P-1) 72.9 (1-2) 44.4	162
$C_{19}FH_{16}I_2P$	As above with Br replaced by I	-184.4	(P-1) 66.0 (1-2) 45.5	162
$C_{19}FH_{17}BrP$	$[(C_6H_5)_3P^+CF^{(1)}H_2^{(2)}] Br^-$	-242.9	(P-1) 57.7 (1-2) 44.1	162
$C_{19}FH_{19}$	$(C_6H_5)_2CH^{(3)} \cdot CF^{(1)}H^{(2)}$ 	-184.8	(1-2) 57 (1-3) 19.5	282
$C_{21}FH_{27}O_2$			(1-2) 47	283
$C_{21}FH_{29}O_2$			(1-2) 48	283
$C_{22}FH_{21}Br_3P$	$[(pCH_3C_6H_4)_3P^+CFBr_2] Br^-$	-78.2		162
$C_{22}FH_{22}Br_2P$	$[(pCH_3C_6H_4)_3P^+CF^{(1)}H^{(2)}Br] Br^-$	-162.2	(P-1) 72.5 (1-2) 44.9	162

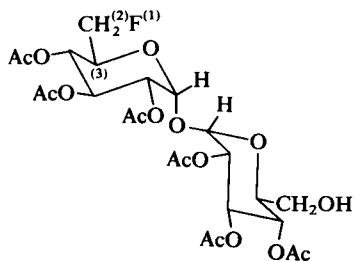
$C_{24}FH_{27}O_6$



(1-2) 47

285

$C_{24}FH_{33}O_{16}$



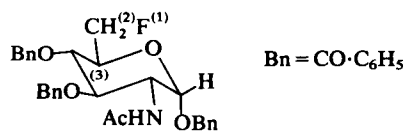
-231.5

(1-2) 45

(1-3) 21

246(b)

$C_{26}FH_{26}NO_8$



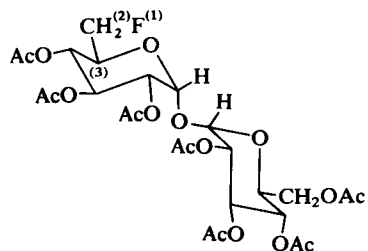
-233.7

(1-2) 47.5

(1-3) 22.5

265

$C_{26}FH_{35}O_{17}$

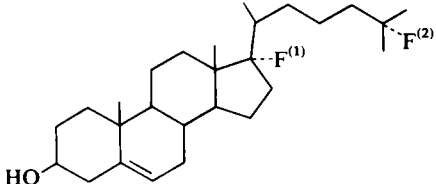
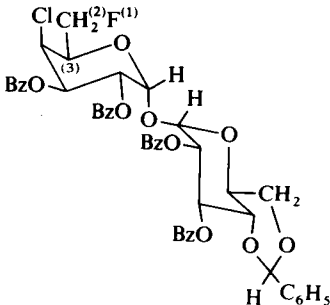


-237.9

(1-2) + (1-3) 92.5

(1-4) 21.5

287(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{27}F_2H_{44}O$		(1) -170 (2) -135.5		393
$C_{47}FH_{48}ClO_9$	 Bz = $CH_2C_6H_5$	-230.9		246(b)
$C_{48}FH_{51}O_{12}S$	As above with Cl replaced by equatorial SO_2CH_3	-237.9	(1-2) 50 (1-3) 24	246(b)
$C_{54}FH_{53}O_{11}$	As above with Cl replaced by $OCH_2C_6H_5$	-231.3		246(b)

D. Miscellaneous studies

The temperature and pressure dependence of the F-19 chemical shifts of CF_4 , CF_3H , CF_3Cl , CF_3Br , CH_3F , and $\text{CF}_3\cdot\text{CF}_3$ have been studied in the gas phase. (294–297) The angular momentum correlation time for CF_4 was deduced from the F-19 spin-lattice relaxation times. (298, 299) Other F-19 and H-1 relaxation studies have been made of *o*-, *m*-, and *p*-chlorobenzotrifluorides in which the individual contributions to the relaxation rate were evaluated. (300)

Several studies of the solvent effects upon the shifts of 1,1,1,10,10,10-hexafluorodecane and difluorobromomethane have been reported and empirical correlations with solvent refractive index and cohesive energy density have been made. (301–303, 484) Various association processes have been investigated and the relevant thermodynamic data evaluated. Thus the self association of sodium perfluorophosphinate and its association with *p*- $\text{R}\cdot\text{C}_6\text{H}_4\cdot\text{CF}_3$ ($\text{R} = \text{H}, \text{Cl}, \text{CF}_3$), (304) and the ionic association of lithium, sodium, potassium, and rubidium trifluoroacetates in propylene carbonate solutions (305) have been studied. Reversible proton transfer and ion-pair formation with no hydrogen bonding was observed in the $\text{CF}_3\text{CH}_2\text{NO}_2\text{--Bu}_3\text{N}$ system. (306) Multinuclear studies of $\text{Et}_3\text{N--CF}_3\text{CO}_2\text{H}$ have shown that the system is best described as a basic salt of 1:1 composition. (307) The effect of solvent and temperature upon the shift of the trifluoromethyl group of various substituted trifluoromethylphenyl isocyanates has been continued. (479) The shift to higher frequency of the signal with increasing temperature for the *ortho*-substituted compounds was attributed to intramolecular interactions, while the reverse situation for *meta*- and *para*-substituted compounds was attributed mainly to intermolecular interactions.

Variable-temperature studies of the reaction of various *o*-alkyl phenols and hexafluoroacetone have been reported and the importance of steric interactions has been enumerated. (308) The reaction of the benzyl ethers of phloroglucinol carboxylic acid, methyl ethers of phloroglucinol and trifluoroacetic anhydride has been monitored by F-19 NMR and shown to produce α,α,α -trifluoro-2,4,6-trimethoxyacetophenone. (309)

Dynamic polarization studies using F-19 NMR have been reported for $\text{CF}_3\text{CH}_2\text{OH}$, $(\text{CF}_3)_2\text{CHOH}$, and *m*-(CF_3) $_2\text{C}_6\text{H}_4$ doped with free radicals, (310) for the interaction of photoexcited α,α,α -trifluoroacetophenone with strong electron donors, (311–313) and for the anodic oxidation of 1H-perfluoroalkanes in $\text{HSO}_3\text{F--KSO}_3\text{F}$. (314).

Lanthanide shift reagent effects upon various alkyl fluorides have been measured (315) and their use in assigning the absolute configuration of the *R*-(+)- α -methoxy- α -trifluoromethylphenyl acetate esters of different stereoisomeric secondary alcohols has been evaluated. (316, 317)

This Page Intentionally Left Blank

III. FLUORINE BONDED TO A CARBON OF AN OLEFIN

This Page Intentionally Left Blank

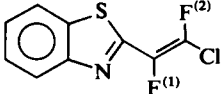
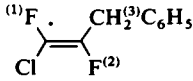
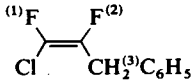
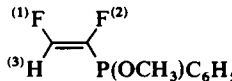
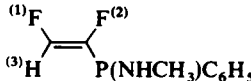
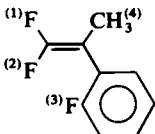
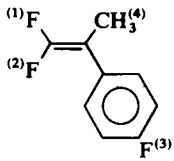
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₂ F ₅ OP	$ \begin{array}{c} {}^{(1)}\text{F} \quad \text{F}^{(3)} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ {}^{(2)}\text{F} \quad \text{P}(\text{O})\text{F}_2^{(4)} \end{array} $	(1) -71.6 (2) -92.9 (3) -199.8 (4) -65.2	(1-2) 16.1 (1-3) 29.5 (1-4) 11.9 (2-3) 121.3 (2-4) 4.2 (3-4) 8.1 (P-2) 20.1 (P-3) <1 (P-4) 1090	50
C ₂ F ₇ HS	$ \begin{array}{c} {}^{(1)}\text{F} \quad \text{F}^{(2)} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ {}^{(5)}\text{H} \quad \text{SF}_4^{(3)}\text{F}^{(4)} \end{array} $	(1) -158.7 (2) -128.2 (3) +59.5 (4) +72.1	(1-2) 12 (1-3) 12 (1-5) 69.3 (2-3) 4 (2-5) 13.5 (3-4) 145	51
C ₂ F ₇ HS	$ \begin{array}{c} {}^{(1)}\text{F} \quad \text{SF}_4^{(3)}\text{F}^{(4)} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ {}^{(5)}\text{H} \quad \text{F}^{(2)} \end{array} $	(1) -160.7 (2) -153.5 (3) +56.3 (4) +69.6	(1-2) 133 (1-3) 19 (1-5) 69.0 (2-5) 4.3 (3-5) 22 (3-4) 146	51
C ₂ F ₇ HS	$ \begin{array}{c} {}^{(1)}\text{F} \quad \text{H} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ {}^{(2)}\text{F} \quad \text{SF}_4^{(3)}\text{F}^{(4)} \end{array} $	(1) -85.8 (2) -67.0 (3) +72.0 (4) +78.0		52
C ₃ FH ₂ N	$ \begin{array}{c} {}^{(3)}\text{H} \quad \text{F}^{(1)} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ {}^{(2)}\text{H} \quad \text{CN} \end{array} $	-111.6	(1-2) 43.9 (1-3) 13.2	25
C ₃ FH ₄ Br	$ \begin{array}{c} {}^{(3)}\text{CH}_3 \quad \text{Br} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ {}^{(2)}\text{H} \quad \text{F}^{(1)} \end{array} $	-73.2	(1-2) 13.0 (1-3) 3.0	56(a)
C ₃ FH ₄ Br	$ \begin{array}{c} {}^{(3)}\text{CH}_3 \quad \text{F}^{(1)} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ {}^{(2)}\text{H} \quad \text{Br} \end{array} $	-77.5	(1-2) 31.0 (1-3) 7.0	56(a)
C ₃ FH ₄ Br	$ \begin{array}{c} {}^{(3)}\text{CH}_3 \quad \text{H}^{(2)} \\ \diagdown \quad \diagup \\ \text{C}=\text{C} \\ \diagup \quad \diagdown \\ \text{Br} \quad \text{F}^{(1)} \end{array} $	-120.4	(1-2) 82.3 (1-3) 4.8	56(a)

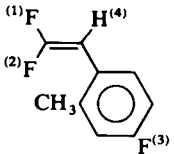
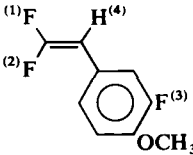
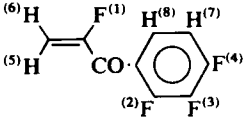
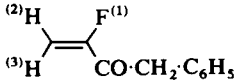
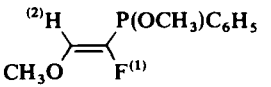
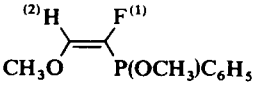
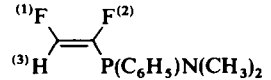
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_3FH_4Br		-119.8	(1-2) 83.3 (1-3) 4.1	56(a)
$C_3F_2H_3ClS$		(1) -108.9 (2) -128.1	(1-2) 139.4 (2-3) 2.5	69(a)
$C_3F_2H_3ClS$		(1) -91.7 (2) -116.9	(1-2) 16.9 (2-3) 2.1	69(a)
$C_4FH_4ClO_2$		-124.2	(1-2) 20	318
$C_4F_2H_4$		-87.63	(1-2) 4 (1-3) 4	319
C_5FH_7O		-117.9	(1-2) 14 (1-3) 46	110
C_5F_4ClN		(1) -86.9 (2) -147.0 (3) -174.6 (4) -141.7	(1-2) 11.2 (1-3) 3.4 (1-4) 13.6 (2-3) 38.4 (2-4) 3.2 (3-4) 10.6	320(a)
$C_6FH_{10}ClO_3$		-122.4	(1-2) 23	318

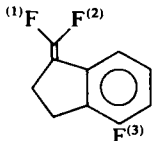
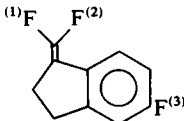
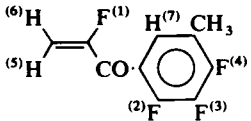
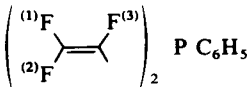
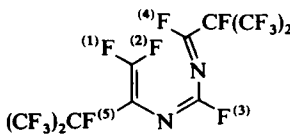
$C_6F_2H_{10}$		(1) -172.7 (2) -141.9	(1-2) 11.28 (1-3) 74.68 (2-3) 19.74	98(a)
$C_6F_3H_9$		(1) -106.3 (2) -121.7 (3) -178.3	(1-2) 85.0 (1-3) 45.0 (2-3) 110.0	98(a)
C_6F_9P			$^1J(1-C)$ 292 $^1J(2-C)$ 288.1 $^1J(3-C)$ 207.6 $^2J(1-C)$ 40 $^2J(2-C)$ 8.5 $^2J(3-C)$ 8.5	321
$C_6F_{11}N$		(1)(2) $-83.1, -93.0$ (3) $-42.6, -55.9$	(1-2) 29 (1-4) 59 (2-3) 15	322
C_7F_6ClN		(1) -4.70 (2) -153.9 (3) -165.6 (4) -162.3		320(a)
C_7F_7N		(1)(2)(3) $-93.8, -124.1, -175.6$ (4) -138.8 (5) -89.2		212
$C_7F_8H_3P$		(1) -153.6 (2) -121.9 (3) -173.1 (4) -104.9 (5) -82.2	(1-2) 140 (1-6) 24 (2-6) 6	323
C_8FH_6IO		-72.24	(1-2) 26.3	184(e)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_8FH_6IO		-69.43	(1-2) 1.9	184(e)
$C_8FH_{13}O$		-117.7	(1-2) 14 (1-3) 45	110
$C_8F_2H_4INO_2$		(1) -112.8 (2) -135.9	(1-2) 142	324(a)
$C_8F_5H_5Cl$		(1) -83.5 (2) -89.1	(1-2) 33.2 (1 or 2-3) 1.2	4
$C_8F_2H_5I$		(1) -117.3 (2) -132.4	(1-2) 142	324(a)
$C_8F_2H_8NP$		(1) -142.9 (2) -150.7	(1-2) 18.6 (1-3) 74.4 (2-3) 19 (P-1) 26.0 (P-2) 10.7	325(a)
$C_8F_2H_{14}$		(1) -91.02 (2) -93.38	(1-2) 49.76 (1-3) 3.03 (1-4) 1.56 (2-3) 25.13 (2-4) 2.02	326
$C_8F_3H_4Cl$		(1) -82.41 (2) -77.44 (3) -108.47	(1-2) 18.88 (1-3) 4.78 (1-4) 2.14 (2-3) 33.34 (2-4) 26.08	327

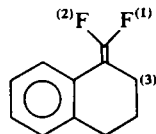
$C_8F_3H_4NO_2$	As above with Cl replaced by NO_2	(1) -78.24 (2) -77.77 (3) -107.61	(1-2) 18.52 (1-3) 4.10 (1-4) 3.13 (2-3) 5.51 (2-4) 25.09	327
$C_8F_3H_5$	As above with Cl replaced by H	(1) -81.39 (2) -81.08 (3) -117.50	(1-2) 27.85 (1-3) 4.87 (1-4) 3.43 (2-3) 4.28 (2-4) 26.23 (2- <i>Hortho</i>) 1.25	327
$C_8F_3H_5$		(1) -83.61 (2) -81.45 (3) -113.45	(1-2) 28.68 (1-3) 1.47 (1-4) 3.87 (2-3) 1.11 (2-4) 25.56	326
$C_8F_3H_5$		(1) -85.46 (2) -84.04 (3) -115.20	(1-2) 33.46 (1-3) 4.08 (1-4) 4.11 (2-3) 1.98 (2-4) 25.59	326
$C_8F_{11}P$		(1) -80.4 (2) -104.9 (3) -174.3 (4) -183.9 (5) -135.0 (6) -148.8 (7) -104.9 (8) -92.5		323
$C_8F_{13}N$		(1)(2)(3) -96.1, -113.8, -114.3, -136.7 (4) -32.9, -51.2	(2-3) 19 (3-4) 7.5	322
C_9FH_7O		-111.5	(1-2) 15 (1-3) 47	110

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9F_2H_4ClNS$		(1)(2) -106.3, -148.8	(1-2) 125	328(a)
$C_9F_2H_7Cl$		(1)(2) -142.1, -125.9	(1-2) 124 (2-3) 22	(1-3) 5 203(a)
$C_9F_2H_7Cl$		(1) -108.8 (2) -130.8	(1-2) 14 (2-3) 23	(1-3) 2 203(a)
$C_9F_2H_9OP$		(1) -137.9 (2) -148.4	(1-2) 17.5 (2-3) 17.4 (P-2) ~17	(1-3) 74.2 (P-1) 29.4 187
$C_9F_2H_{10}NP$		(1) -140.3 (2) -144.5	(1-2) 17.5 (2-3) 16.8 (P-2) ~17	(1-3) 75.6 (P-1) 28.2 325(a)
$C_9F_3H_7$		(1) -92.53 (2) -89.54 (3) -113.91	(1-2) 40.62 (1-4) 3.36 (2-4) 3.36	(1-3) 2.06 (2-3) 11.83 327
$C_9F_3H_7$		(1) -91.28 (2) -91.69 (3) -115.50	(1-2) 44.60 (1-4) 3.54 (2-4) 3.54	(1-3) 2.08 (2-3) 0.64 326

$C_9F_3H_7$		(1) -85.23 (2) -86.36 (3) -115.50	(1-2) 32.65 (1-3) 2.54 (1-4) 3.57 (2-3) 0.87 (2-4) 24.99	326
$C_9F_3H_7O$		(1) -85.73 (2) -83.94 (2) -135.33	(1-2) 33.82 (1-3) 1.10 (1-4) 3.98 (2-3) 1.10 (2-4) 25.37	326
$C_9F_4H_4O$		(1) -115.1 (2) -134.2 (3) -160.3 (4) -128.1	(1-2) 17.0 (1-5) 45.0 (1-6) 12.0 (2-3) 20.5 (2-4) 12.0 (2-7) 6.0 (2-8) 3.0 (3-4) 20.5 (3-7) 6.0 (3-8) 3.0 (4-7) 9.0 (4-8) 5.5	424
$C_{10}FH_9O$		-116.7	(1-2) 15 (1-3) 45	110
$C_{10}FH_{12}O_2P$		-150.9	(1-2) 22.7 (P-1) 10.2	187(a)
$C_{10}FH_{12}O_2P$		-162.3	(1-2) 24 (P-1) 55	187(a)
$C_{10}F_2H_{12}NP$		(1) -138.9 (2) -142.3	(1-2) 16.4 (1-3) 73.6 (2-3) 16.8 (P-1) 29.6 (P-2) ~16.5	325(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_3H_7$		(1)(2) -87.70 (3) -118.70	(1-2) + (2-3) 2.94	326
$C_{10}F_3H_7$		(1)(2) -88.03 (3) -117.12	(1-2) + (2-3) 3.68	326
$C_{10}F_3H_7$		(1) -89.91 (2) -90.32 (3) -115.68	(1-2) 51.77 (1-3) 6.10 (2-3) 2.84	326
$C_{10}F_4H_6O$		(1) -116.5 (2) -139.6 (3) -162.1 (4) -132.8	(1-2) 13.0 (1-5) 45.0 (1-6) 13.0 (2-3) 19.0 (2-7) 7.0 (3-4) 19.0 (3-7) 2.5 (4-7) 7.0	329
$C_{10}F_6H_5P$		(1) -84.04 (2) -107.00 (3) -176.48	(1-2) 44.06 (1-3) 33.97 (2-3) -124.31 (3-3') 12.21 (3-2') -1.02 (3-1') 4.35	321
$C_{10}F_{18}N_2$		(1)(2) -77.7, -91.3 (3)(4) -41.8	(1-2) 26 (1-5) 62	322

C₁₁F₂H₁₀

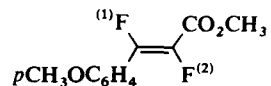


(1) -88.40 (2) -88.77

(1-2) 42.50 (1-3) 2.76
(2-H) 1.1

326

C₁₁F₂H₁₀O₃

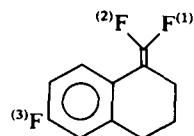


(1)(2) -164.2, -135.4

(1-2) 130

100

C₁₁F₃H₉

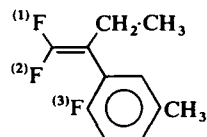


(1) -89.15 (2) -89.39
(3) -116.33

(1-2) 44.12 (1-3) 3.48
(2-3) 1.67

326

C₁₁F₃H₁₁

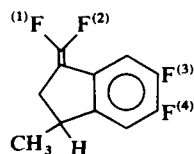


(1) -93.54 (2) -90.07
(3) -119.53

(1-2) 42.09 (1-3) 1.84
(2-3) 9.56 (2-H) 2.15

327

C₁₁F₄H₈

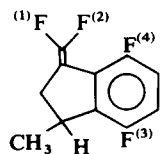


(1) -88.68 (2) -89.25
(2) -140.14 (4) -138.78

(1-2) 48.77 (1-3) 1.96
(1-4) 6.61 (2-3) 1.35
(2-4) 2.76 (3-4) -19.94

326

C₁₁F₄H₈



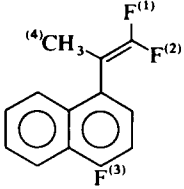
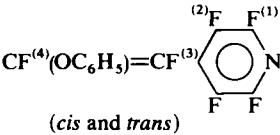
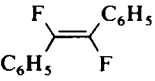
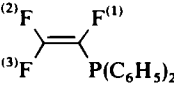
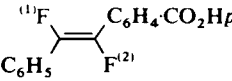
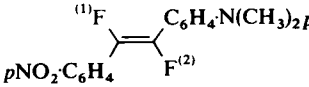
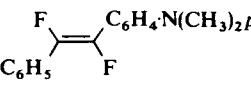
(1) -86.40 (2) -79.81
(3) -119.00 (4) -126.83

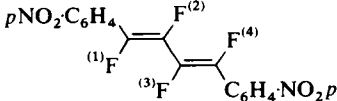
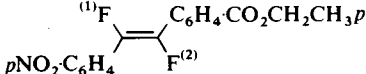
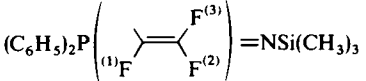
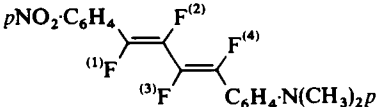
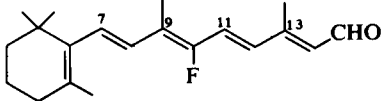
(1-2) 37.09 (1-3) 1.70
(1-4) 8.00 (2-3) 1.50
(2-4) 68.51 (3-4) 22.00

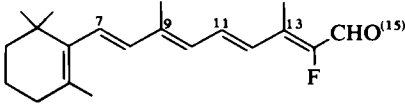
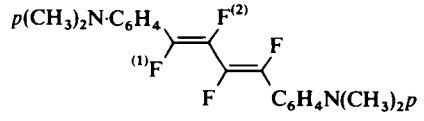
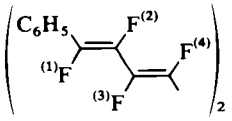
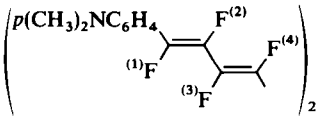
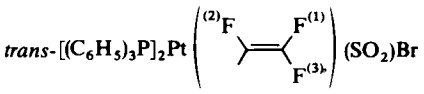
326

327

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₁₂ FH ₁₃ O	$\begin{array}{c} \text{(2)H} \quad \text{F}^{(1)} \\ \diagdown \quad / \\ \text{C} = \text{C} \\ / \quad \diagdown \\ \text{(3)H} \quad \text{CO} \cdot (\text{CH}_2)_3 \text{C}_6\text{H}_5 \end{array}$	-117.5	(1-2) 14 (1-3) 45	110
C ₁₂ FH ₁₆ O ₂ P	$\begin{array}{c} \text{(2)H} \quad \text{P(OCH}_2\text{CH}_3)_3 \\ \diagdown \quad / \\ \text{C} = \text{C} \\ / \quad \diagdown \\ \text{CH}_3\text{CH}_2\text{O} \quad \text{F}^{(1)} \end{array}$	-150.9	(1-2) 22.6 (P-1) 9.6	187(a)
C ₁₂ FH ₁₆ O ₂ P	$\begin{array}{c} \text{(2)H} \quad \text{F}^{(1)} \\ \diagdown \quad / \\ \text{C} = \text{C} \\ / \quad \diagdown \\ \text{CH}_3\text{CH}_2\text{O} \quad \text{P(OCH}_2\text{CH}_3)_3 \end{array}$	-164.1	(1-2) 24 (P-1) 64	187(a)
C ₁₂ F ₂ H ₁₆ NP	$\begin{array}{c} \text{(1)F} \quad \text{F}^{(2)} \\ \diagdown \quad / \\ \text{C} = \text{C} \\ / \quad \diagdown \\ \text{(3)H} \quad \text{P[N(CH}_2\text{CH}_3)_2]\text{C}_6\text{H}_5 \end{array}$	(1) -139.7 (2) -142.9	(1-2) 18.3 (1-3) 76.2 (2-3) 17.9 (P-1) 33.9 (P-2) 18	325(a)
C ₁₂ F ₃ H ₁₈ BrN ₂ Pt	$\text{trans-} [\text{PtBr} \left(\begin{array}{c} \text{F}^{(3)} \\ \diagdown \quad / \\ \text{C} = \text{C} \\ / \quad \diagdown \\ \text{(1)F} \quad \text{F}^{(2)} \end{array} \right) (\text{Bu}^t\text{NC})_2]$	(1) -146.5 (2) -98.8 (3) -124.4	(1-2) 40 (1-3) 109 (2-3) 92 (Pt-1) 598 (Pt-2) 73 (Pt-3) 39	330
C ₁₂ F ₃ H ₁₈ ClN ₂ Pt	As above with Br replaced by Cl	(1) -146.0 (2) -98.3 (3) -125.6	(1-2) 39 (1-3) 109 (2-3) 92.7 (Pt-1) 568 (Pt-2) 66 (Pt-3) 44	330
C ₁₂ F ₃ H ₁₈ ClN ₂ Pt	As above with Br replaced by Cl and <i>cis</i> -compound	(1) -162.3 (2) -96.8 (3) -125.7	(1-2) 35 (1-3) 107 (2-3) 91.6 (Pt-1) 396 (Pt-2) 84 (Pt-3) 49	330
C ₁₂ F ₃ H ₁₈ ClN ₂ Pt	$[\text{Pt} \left(\begin{array}{c} \text{(1)F} \quad \text{F}^{(3)} \\ \diagdown \quad / \\ \text{C} = \text{C} \\ / \quad \diagdown \\ \text{(2)F} \quad \text{Cl} \end{array} \right) (\text{Bu}^t\text{NC})_2]$	(1) -114.6 (2) -115.4 (3) -138.4	(1-2) 186.8 (1-3) <1 (2-3) 52.6 (Pt-1) 297 (Pt-2) 257 (Pt-3) 211	330

$C_{13}F_3H_9$		(1) -89.74 (2) -93.81 (3) -123.53	(1-2) 45.49 (1-4) 3.29 (2-4) 3.29 (1-H) 1.29	326
$C_{13}F_6H_5NO$	 (<i>cis</i> and <i>trans</i>)	(1) -90.9 (2) -137.3 (3)(4) -94.6, -161.4 and -114.4, -166.4		212
$C_{14}F_2H_{10}$		-151.76		331(a)
$C_{14}F_3H_{10}P$		(1) -178.70 (2) -85.45 (3) -108.46	(1-2) 29.92 (1-3) 125.05 (2-3) 46.40 (P-1) 11.8 (P-2) 9.19 (P-3) 61.36	321
$C_{15}F_2H_{10}O_2$		(1) -149.20 (2) -154.56	(1-2) 120	331(a)
$C_{15}F_2H_{12}$	As above with CO ₂ H replaced by CH ₃	(1) -152.40 (2) -151.46	(1-2) 120	331(a)
$C_{16}F_2H_{14}N_2O_2$		(1) -158.20 (2) -145.50	(1-2) 120	331(a)
$C_{16}F_2H_{15}N$		(1) -156.66 (2) -150.56	(1-2) 120	331(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{16}F_4H_8N_2O_4$		(1) -141.54 (2) -155.94	(1-2) 130	331(a), 324
$C_{16}F_4H_9NO_2$	As above with pNO_2 replaced by H	(1) -140.60 (2) -161.00 (3) -153.80 (4) -143.40	(1-2)(3-4) 129	331(a), 324
$C_{16}F_4H_{10}$	As above with each NO_2 replaced by H	(1) -141.71 (2) -158.70	(1-2) 120	331(a)
$C_{17}F_2H_{14}O_2$		(1) -148.26 (2) -152.61	(1-2) 120	331(a)
$C_{17}F_3H_{19}NPSi$		(1) -175.8 (2) -83.7 (3) -106.3		321
$C_{18}F_4H_4N_2O_2$		(1) -140.30 (2) -152.30 (3) -165.90 (4) -144.90	(1-2)(3-4) 120	331(a)
$C_{18}F_4H_{15}N$	As above with NO_2 replaced by H	(1) -142.50 (2) -157.70 (3) -164.50 (4) -143.70	(1-2)(3-4) 130	331(a)
$C_{20}FH_{27}O$	 7 <i>E</i> , 9 <i>Z</i> , 11 <i>E</i> , 13 <i>E</i> 7 <i>Z</i> , 9 <i>Z</i> , 11 <i>E</i> , 13 <i>E</i>		(F-11) 25.1 25.6	332

	7 <i>E</i> , 9 <i>Z</i> , 11 <i>Z</i> , 13 <i>E</i>		30·6	
	7 <i>E</i> , 9 <i>Z</i> , 11 <i>E</i> , 13 <i>Z</i>		27·5	
	7 <i>Z</i> , 9 <i>E</i> , 11 <i>E</i> , 13 <i>E</i>		27·5	
	7 <i>E</i> , 9 <i>E</i> , 11 <i>E</i> , 13 <i>Z</i>		26·2	
	7 <i>Z</i> , 9 <i>E</i> , 11 <i>E</i> , 13 <i>Z</i>		25·0	
	7 <i>E</i> , 9 <i>E</i> , 11 <i>E</i> , 13 <i>E</i>		27·3	
C ₂₀ FH ₂₇ O				332
	7 <i>E</i> , 9 <i>E</i> , 11 <i>E</i> , 13 <i>Z</i>		(F-15) 11·5	
	7 <i>E</i> , 9 <i>Z</i> , 11 <i>E</i> , 13 <i>Z</i>		11·0	
	7 <i>E</i> , 9 <i>E</i> , 11 <i>Z</i> , 13 <i>Z</i>		11·8	
	7 <i>E</i> , 9 <i>E</i> , 11 <i>E</i> , 13 <i>E</i>		11·0	
	7 <i>E</i> , 9 <i>E</i> , 11 <i>Z</i> , 13 <i>E</i>		11·7	
C ₂₀ F ₄ H ₂₀ N ₂		(1) -143·54 (2) -161·74	(1-2) 120	331(a)
C ₂₀ F ₈ H ₁₀		(1) -139·2 (2) -163·62 (3) -152·36 (4) -147·97	(1-2)(3-4) 130	333(a)
C ₂₄ F ₈ H ₂₀ N ₂		(1) -140·26 (2) -167·66 (3) -147·26 (4) -152·86	(1-2)(3-4) 130	333(a)
C ₃₈ F ₃ H ₃₀ BrO ₂ P ₂ PtS		(1) -100·2 (2) -128·6 (3) -149·0	(1-2) 93·2 (1-3) 30·8 (2-3) 109·7 (P-1) 6·0 (P-2) 3·8 (P-3) 1·5	334

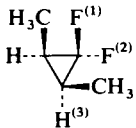
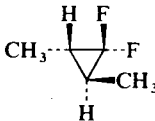
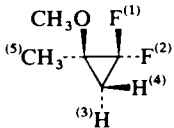
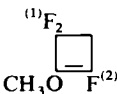
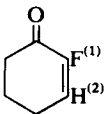
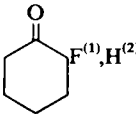
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{38}F_3H_{30}ClP_2$	$cis-[(C_6H_5)_3P]_2Pt \left(\begin{array}{c} ^{(2)}F \\ \diagup \quad \diagdown \\ C=C \\ \diagdown \quad \diagup \\ F^{(3)} \end{array} \right) F^{(1)}Cl$	(1) -100.1 (2) -127.4 (3) -152.3	(1-2) 95.2 (1-3) 29.7 (2-3) 101.5 (P-1) 18.0 (P-2) 0.5 (P-3) 26.0	334
$C_{39}F_3H_{30}ClO_5P_2Pt$	$[(C_6H_5)_3P]_2Pt \left(\begin{array}{c} ^{(2)}F \\ \diagup \quad \diagdown \\ C=C \\ \diagdown \quad \diagup \\ F^{(3)} \end{array} \right) F^{(1)}(CO)(ClO_4)$	(1) -93.1 (2) -120.8 (3) -162.3	(1-2) 92.0 (1-3) 29.0 (2-3) 107.5 (P-1) 6.5 (P-2) 5.5 (P-3) 2.8	334
$C_{39}F_3H_{30}NP_2Pt$	$trans-[(C_6H_5)_3P]_2Pt \left(\begin{array}{c} ^{(2)}F \\ \diagup \quad \diagdown \\ C=C \\ \diagdown \quad \diagup \\ F^{(3)} \end{array} \right) F^{(1)}(CN)$	(1) -96.3 (2) -126.0 (3) -154.3	(1-2) 96.0 (1-3) 30.8 (2-3) 107.4 (P-1) 5.9 (P-2) 4.6 (P-3) 2.0	334
$C_{39}F_3H_{30}NP_2PtS$	As above with (CN) replaced by (NCS)	(1) -101.0 (2) -128.7 (3) -147.0	(1-2) 98.5 (1-3) 31.0 (2-3) 105.5	334
$C_{42}F_6H_{33}P_4$	$\left[(C_6H_5)_2P \begin{array}{c} ^{(1)}F \\ \diagup \quad \diagdown \\ C=C \\ \diagdown \quad \diagup \\ ^{(2)}F \end{array} \right]_3 P$	(1)(2) -132.5		323
$C_{43}F_3H_{37}O_2P_2Pt$	$trans-[(C_6H_5)_3P]_2Pt \left(\begin{array}{c} ^{(2)}F \\ \diagup \quad \diagdown \\ C=C \\ \diagdown \quad \diagup \\ F^{(3)} \end{array} \right) F^{(1)}(CH_3COCHCOCH_3)$	(1) -99.8 (2) -128.0 (3) -149.2	(1-2) 98.0 (1-3) 32.2 (2-3) 106.8 (P-1) 6.1 (P-2) 4.0 (P-3) 1.8 (Pt-1) 70.2 (Pt-2) 65.0 (Pt-3) 549.0	334

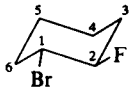
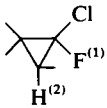
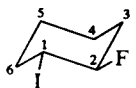
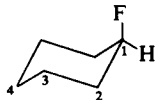
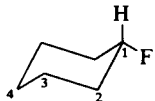
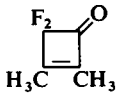
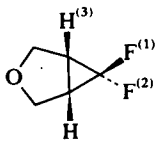
IV. FLUORINE BONDED TO A CARBON OF AN ALICYCLIC NON-AROMATIC SYSTEM

This Page Intentionally Left Blank

A. Data

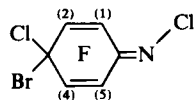
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_4FH_3D_4$			(1-2) 55.34 (1-3) 20.67 (1-4) 6.17	335
C_4FH_7			(1-2) 55.32 (1-3) 20.63 (1-4) 6.14 (1-5) 8.54 (1-6) -0.03 ¹ J(F-C) 214.55 ² J(F-C) 18.49 ³ J(F-C) 20.35	335
C_4FH_7	As above	-160.22	¹ J(F-C) 214.7 (1-2) 55 ² J(F-C) 19.1	336
$C_4F_2H_3ClO$		-99		318
$C_4F_2H_4$		-131.5		319(a)
C_4F_3IO		(1) -92.5 (2) -113.3	(1-2) 21.1	337(a)
$C_4F_6H_4O$		(1) -149 (2) -130	(1-2) 165 (1-3) 7 (1-4) 16 (1-5) 2.5 (2-3) 14.5 (2-4) 6 (2-5) 11	81

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_3FH_9	$F^{(1)}CH^{(2)}(CH_2)_3CH_2$	-170.54	(1-2) 53 $^1J(1-C)$ 173.5 $^2J(1-C)$ 22.1 $^3J(1-C)$ <1.5	336
$C_3F_2H_8$		(1) -158.3 (2) -128.7	(1-2) 158.7 (2-3) 12.0	338
$C_3F_2H_8$		-143.1		338
$C_5F_2H_8O$		(1) -144.9 (2) -135.8	(1-2) 159 (1-3) 6.1 (1-4) 16.1 (1-5) 2.9 (2-3) 14.2 (2-4) 5.5 (2-5) 1.8	81
$C_3F_3H_3O_2$		(1) -118.2 (2) -141.7		100
C_6FH_7O		-132.96	(1-2) 14	339
C_6FH_9O		-189.4	(1-2) 49.5	340(a)

$C_6FH_{10}Br$		-167.98	(F-C1) 20 (F-C3) 19 (F-C6) 5	(F-C2) 181 (F-C4) 10.3	336
$C_6FH_{10}Br$		-149.5			341
$C_6FH_{10}Cl$		-138.2	(1-2) 20.4		341
$C_6FH_{10}I$		-160.06	(F-C1) 19 (F-C3) 17.7 (F-C6) 5	(F-C2) 181 (F-C4) 11.8	336
C_6FH_{11}		-185.8	(F-C1) 167.9 (F-C3) <1.8	(F-C2) 21.1 (F-H1) 49	336
C_6FH_{11}		-165.4	(F-C1) 172.6 (F-C3) 11.0	(F-C2) 17.6 (F-H1) 49	336
$C_6F_2H_6O$		-110.8			100
$C_6F_2H_6O$		(1) -129.1 (2) -172.1	(1-2) 173.0 (2-3) 2.0	(1-3) 13.2	338

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_2H_8$		(1)(2) -137.1 (Centre AB)	(1-2) 156 (1-3) 7.6 (1-4) 10.2 (1-5) ~1.5 (2-3) 10.2 (2-4) 7.6 (2-5) 1.5 (2-6) ~2.4	81
$C_6F_2H_{10}$		(1) -157.2 (2) -126.6	(1-2) 156	81
$C_6F_2H_{10}$		(1)(2) -150.8, -139.0	(1-2) 150.4	338
$C_6F_2H_{10}O$		(1) -143.8 (2) -147.5	(1-2) 159 (2-3) 20.0	338
$C_6F_4HCl_2N$		(2) -114.5 (3) -116.7 (4) -148.9 (5) -153.3	(1-2) 10.1 (1-3) 0.9 (1-4) 0.9 (1-5) 5.1 (2-3) 27.2 (2-4) 1.6 (3-4) 25.5 (3-5) 6.8 (4-5) 5.9	175(a)
$C_6F_4H_6$		-115.6	(1-2) 1.25	100
$C_6F_4H_6O_2$	As above with CH_3 replaced by OCH_3	-111.4		100

C₆F₄BrCl₂N

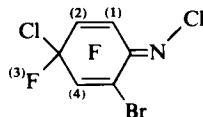


(1) -151.4 (2) -136.2
(4) -141.8 (5) -153.2

(1-2) 2.8 (2-4) 4.8
(4-5) 7.5

175(a)

C₆F₄BrCl₂N

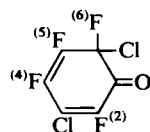


(1) -146.5 (2) -143.2
(3) -113.1 (4) -106.6

(1-2) 2.2 (1-3) 6.4
(2-3) 26.0 (3-4) 29.0

175(a)

C₆F₄Cl₂O

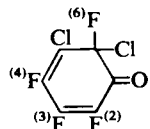


(2) -131.9 (4) -144.8
(5) -147.7 (6) -129.2

(2-4) 4.5 (2-5) 23
(2-6) 8.5 (4-5) 1.5
(4-6) 9 (5-6) 25.5

342(b)

C₆F₄Cl₂O

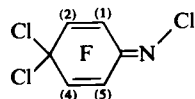


(2) -155.5 (3) -130.7
(4) -127.9 (6) -122.5

(2-3) 3 (2-6) = (4-6) 7.5
(3-4) 6.5 (3-6) 3

342(b)

C₆F₄Cl₃N

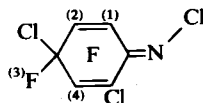


(1) -151.6 (2) -136.5
(4) -142.1 (5) -152.8

(1-2) ~1.5 (2-4) 6.0
(4-5) 6.8

175(a)

C₆F₄Cl₃N

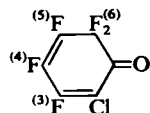


(1) -147.1 (2) -143.4
(3) -113.4 (4) -116.9

(1-3) 6.0 (2-3) 26.8
(3-4) 29.0

175(a)

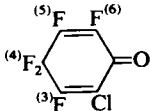
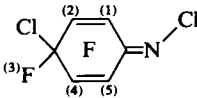
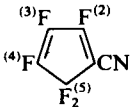
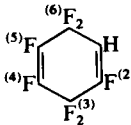
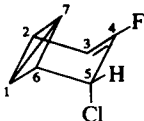
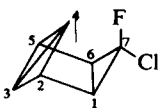
C₆F₅ClO

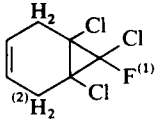
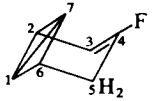
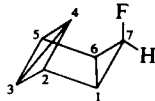
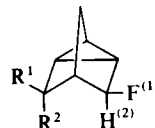
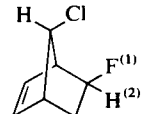


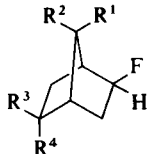
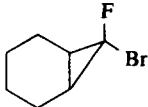
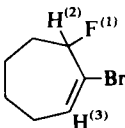
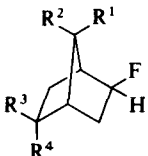
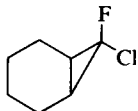
(3) -103.4 (4) -152.8
(5) -150.2 (6) -110.9

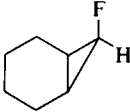
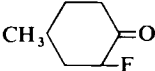
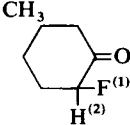
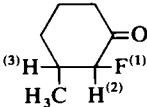
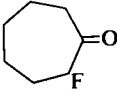
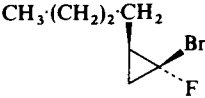
(3-4) 6.5 (3-5) 4.5
(3-6) 3.5 (4-5) 1.5
(4-6) 13.5 (5-6) 22.5

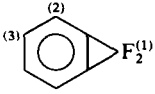
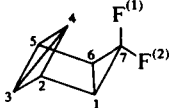
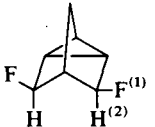
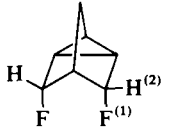
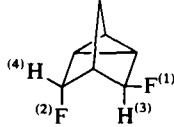
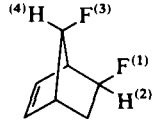
342(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₆ F ₅ ClO		(3) -117.4 (4) -114.2 (5) -149.1 (6) -148.1	(3-4) 23 (3-5) 4.5 (3-6) 2.5 (4-5) 21 (4-6) 10 (5-6) 4.5	342(b)
C ₆ F ₅ Cl ₂ N		(1) -150.1 (2) -143.1 (3) -113.6 (4) -148.7 (5) -151.3	(1-2) 1.2 (1-3) 6.0 (1-5) 2.6 (2-3) 26.2 (2-4) 4.2 (2-5) 1.2 (3-4) 25.9 (3-5) 6.0 (4-5) 6.3	175(a), 174
C ₆ F ₅ N		(2) -106.5 (3) -154.8 (4) -151.5 (5) -133.1	(2-3) 9.0 (2-4) 2.0 (2-5) 8.3 (3-4) 14.2 (3-5) 5.5 (4-5) 8.6	320(a)
C ₆ F ₇ H		(2) -129 (3) -116 (4) -156 (5) -161 (6) -102	(2-3) 21 (2-4) 1 (2-6) 10.5 (3-4) 21 (3-6) 5.25 (5-6) 21 (4-6) 10.5	343
C ₆ F ₇ Cl	As above with H replaced by Cl	(2) -131 (3) -115 (4) -156 (5) -161 (6) -108	(2-3) 24 (2-4) 3 (2-6) 10.5 (3-4) 21.5 (3-6) 5.25 (5-6) 21.5	343
C ₇ FH ₆ Cl			(F-C1) 2.2 (F-C2) 6.4 (F-C3) 9.6 (F-C4) 258.8 (F-C5) 20.6 (F-C6) 4.4 (F-C7) 1.5	344
C ₇ FH ₆ Cl			(F-C1) 13.2 (F-C2) 1.5 (F-C4) 5.9 (F-C7) 319.9	344

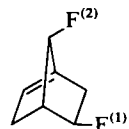
C_7FH_6Cl	As above with F and Cl interchanged		(F-C1) 7·4 (F-C3) 6·6 (F-C7) 280·0	344
$C_7FH_6Cl_3$		-65·4	(1-2) 2·0	345(b)
C_7FH_7			(F-C1) 2·2 (F-C2) 8·1 (F-C3) 8·1 (F-C4) 260 (F-C5) 19·1 (F-C6) 10·6 (F-C7) 2·2	344
C_7FH_7			(F-C1) 14·0 (F-C2) 2·9 (F-C3) 1·5 (F-C4) 5·1 (F-C7) 244·1	344
C_7FH_7	As above with F and H at C7 interchanged		(F-C1) 8·1 (F-C2) 2·2 (F-C3) 10·3 (F-C7) 222·1	344
C_7FH_8Br	 $R^1 = Br \quad R^2 = H$ $R^1 = H \quad R^2 = Br$	-194·25 -200·25	(1-2) 60 (1-2) 60	346
C_7FH_8Cl	As above with $R^1 = Cl \quad R^2 = H$ $R^1 = H \quad R^2 = Cl$	-195 -200·25	(1-2) 60 (1-2) 60	346
C_7FH_8Cl		-184·5		346

Molecular formula	Structure of compounds	Chemical shifts (pm)	Coupling constants (Hz)	Ref.																				
$C_7FH_{10}Br$	 <table data-bbox="689 272 867 401"><tr><th>R^1</th><th>R^2</th><th>R^3</th><th>R^4</th></tr><tr><td>Br</td><td>H</td><td>H</td><td>H</td></tr><tr><td>H</td><td>Br</td><td>H</td><td>H</td></tr><tr><td>H</td><td>H</td><td>Br</td><td>H</td></tr><tr><td>H</td><td>H</td><td>H</td><td>Br</td></tr></table>	R^1	R^2	R^3	R^4	Br	H	H	H	H	Br	H	H	H	H	Br	H	H	H	H	Br	-162 -162.75 -167.25 -163.5		347
R^1	R^2	R^3	R^4																					
Br	H	H	H																					
H	Br	H	H																					
H	H	Br	H																					
H	H	H	Br																					
$C_7FH_{10}Br$			$^1J(F-C)$ 305.6 $^2J(F-C)$ 10.2 $^3J(F-C)$ 3.1 $^4J(F-C)$ 1.7	348																				
$C_7FH_{10}Br$	As above with F and Br interchanged		$^1J(F-C)$ 295.6 $^3J(F-C)$ 10.5 $^3J(F-C)$ 2.6 $^4J(F-C)$ 0	348																				
$C_7FH_{10}Br$			(1-2) 47 (1-3) 2	349																				
$C_7FH_{10}Cl$	 <table data-bbox="689 796 867 931"><tr><th>R^1</th><th>R^2</th><th>R^3</th><th>R^4</th></tr><tr><td>Cl</td><td>H</td><td>H</td><td>H</td></tr><tr><td>H</td><td>Cl</td><td>H</td><td>H</td></tr><tr><td>H</td><td>H</td><td>Cl</td><td>H</td></tr><tr><td>H</td><td>H</td><td>H</td><td>Cl</td></tr></table>	R^1	R^2	R^3	R^4	Cl	H	H	H	H	Cl	H	H	H	H	Cl	H	H	H	H	Cl	-163.5 -162.75 -167.25 -162.75		347
R^1	R^2	R^3	R^4																					
Cl	H	H	H																					
H	Cl	H	H																					
H	H	Cl	H																					
H	H	H	Cl																					
$C_7FH_{10}Cl$			$^1J(F-C)$ 289.8 $^2J(F-C)$ 11.0 $^3J(F-C)$ 2.8 $^4J(F-C)$ 2.0	348																				
$C_7FH_{10}Cl$	As above with F and Cl interchanged		$^1J(F-C)$ 282.4 $^2J(F-C)$ 11.3 $^3J(F-C)$ 2.7 $^4J(F-C)$ 0	348																				

C_7FH_{11}			$^1J(F-C) 221.6$ $^2J(F-C) 10.6$ $^3J(F-C) 4.8$ $^4J(F-C) 1.1$	348
C_7FH_{11}	As above with F and H interchanged		$^1J(F-C) 221.6$ $^2J(F-C) 10.6$ $^3J(F-C) 3.4$ $^4J(F-C) 0$	348
$C_7FH_{11}O$		$-187.0(a), -188.5(e)$		340(a)
$C_7FH_{11}O$		$-186.6(a), -189.5(e)$	$(1-2) 47.9$	340(a)
$C_7FH_{11}O$		$-196.6(a), -194.3(e)$	$(1-2)_a 50.5$ $(1-2)_e 50.5$ $(1-3)_e 17.7$	340(a)
$C_7FH_{11}O$		-183.8		340(a)
$C_7FH_{12}Br$		-126.5		341
$C_7FH_{12}Br$	As above with Br and F interchanged	-149.4		341
C_7FH_{13}	$F^{(1)}\overline{CH^{(2)}(CH_2)_5}CH_2$	-164.92	$^1J(1-C) 167.7$ $^2J(1-C) 20.6$ $^3J(1-C) 8.8$ $(1-2) 47.5$	336

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_2H_4$			(1-2) 3.5 $^1J(1-C)$ 303 $^2J(1-C)$ 20 $^3J(1-C)$ 1 $^4J(1-C)$ 2.9	350
$C_7F_2H_6$			(1-C7) 314.0 (2-C7) 281.7 (1-C1) 13.2 (2-C1) 8.1 (1-C4) 4.4 (2-C3) 10.3	344
$C_7F_2H_8$		-201.4	(1-2) 60	346, 352
$C_7F_2H_8$		-201.8	(1-2) 60	352
$C_7F_2H_8$		(1) -201.8 (2) -213	(1-3) 60 (2-4) 60	346, 352
$C_7F_2H_8$		(1) -180	(1-2) 57 (3-4) 60.0 (3-H) 12	346

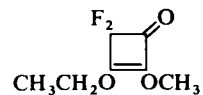
C₇F₂H₈



(1) -187.5 (2) -180

347

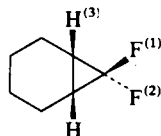
C₇F₂H₈O₂



-113.2

337(a)

C₇F₂H₁₀

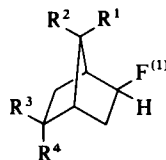


(1) -129.2 (2) -159.0

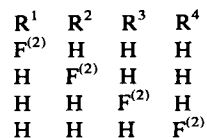
(1-2) 163.2 (1-3) 14.0

338

C₇F₂H₁₀



347



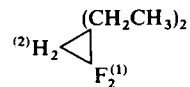
(1) -179.5 (2) -223.5

(1) -176.2 (2) -232.3

(1)(2) -181.5

(1)(2) -175.1

C₇F₂H₁₂

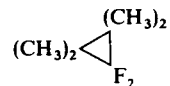


-145.2

(1-2) 8.4

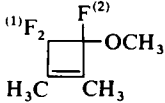
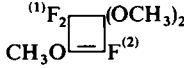
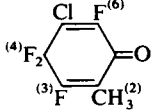
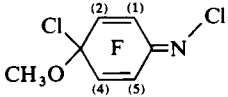
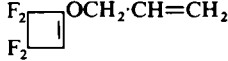
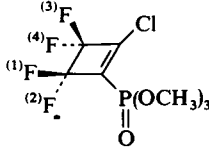
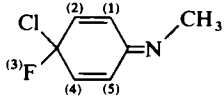
338

C₇F₂H₁₂

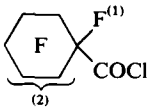
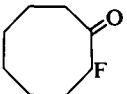
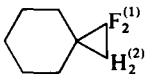
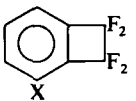
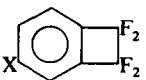
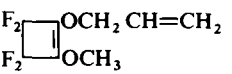


-148.9

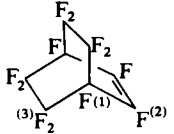
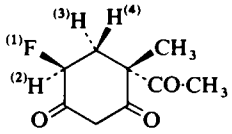
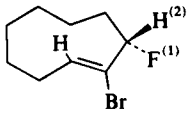
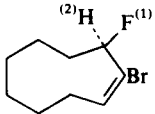
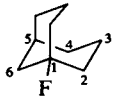
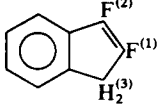
81, 338

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_3H_9O$		(1) -115.3 (2) -111.26		100
$C_7F_3H_9O_3$		(1) -117.0 (2) -135.2		100
$C_7F_4H_3ClO$		(3) -120.8 (4) -107.3 (6) -114.6	(2-3) 3 (2-4) 2 (3-4) 25 (3-6) 2.5 (4-6) 10	351(b)
$C_7F_4H_3Cl_2NO$		(1) -153.2 (2) -137.0 (4) -142.6 (5) -154.9	(1-2) ~2 (2-4) 5.4 (4-5) 7.4	175
$C_7F_4H_6O$		-109.1, -118.1		337(a)
$C_7F_4H_9ClO_4P$		(1)(2) or (3)(4) -112.090 (3)(4) or (1)(2) -116.982	(1-2) 197.04 (1-3) -12.57 (1-4) 27.88 (3-4) 196.41 (1,2-P) ± 4.86 (3,4-P) ± 12.24	353(b)
$C_7F_5H_3ClN$		(1) -150.6 (2) -153.3 (3) -110.2 (4) -155.4 (5) -152.5	(1-2) 3.1 (1-3) 6.4 (1-5) ~1.4 (2-3) 26.0 (2-4) 1.2 (3-4) 25.6 (3-5) 6.4 (4-5) 7.5	175

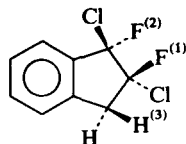
$C_7F_5H_5O$		(1) -140.7 (2)(3) -113.9, -117.3		337(a)
$C_7F_5H_9ClO_3P$		(1)(2) or (3)(4) -113.164 (3)(4) or (1)(2) -117.361 (5) -75.40	(1-2) 195.27 (1-3) -14.59 (1-4) 25.96 (1,2-5) 0.7 (3-4) 190.23 (5-6) 1.81 (1,2-P) $\mp$ 3.88 (3,4-P) $\pm$ 12.59 (5-P) -845	353(b)
$C_7F_6H_4O$		(1) -169.9 (2)(3)(4)(5) -139.5, -154.9 (6) -167.25	(1-7) 50 (1-6) 24 (6-7) 24	354
$C_7F_6H_4O_2$		(1)(2) -142.4, -151.8 (3)(4) -113.2, -124.0	(1-2) 4.5 (3-4) 2.4 (1-3 or 4) } (2-3 or 4) } 12.0	355
$C_7F_7H_5O_2$		(1) -162.3 (2)(3)(4) -116.2, -117.0, -130.6		355
$C_7F_9H_3O$		(1) -160.7 (2)(5) -117.2 (3)(4) -134.2		356
$C_7F_9H_3O$		(1)(2) -145.9, -157.8 (3) -130.0 (4)(5) -130.0, -131.1, 135.7 (2:1:1) (6A) -116.5 (6B) -122.0	(4A-4B) or (5A-5B) 265 (6A-6B) 285	356

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_{11}ClO$		(1) -167.0 (2) -119.0 and -131.0 , -123.5 and -138.9 , -122.9 and -141.6	$^2J(F-F)$ 294, 292, 291	237
$C_8FH_{13}O$		-189.9		340(a)
C_8FH_{15}	$F^{(1)}\overbrace{CH^{(2)}(CH_2)_6}CH_2$	-160.16	$^1J(1-C)$ 164.8 $^2J(1-C)$ 22.0 $^3J(1-C)$ 8.9 (1-2) 46.5	336
$C_8F_2H_{12}$		-140.1	(1-2) 8.5	338
$C_8F_4H_3Br$		X = Br -106.5 , -104.4		357(b)
$C_8F_4H_3NO_2$	As above with X = NO_2	-106.1 , -105.2		357(b)
$C_8F_4H_3NO_2$		X = NO_2 -107.1		357(b)
$C_8F_4H_4$	As above with X = H	X = H -104.4		357(b)
$C_8F_4H_8O_2$		-116.7 (centre of multiplet)		337(a)

$C_8F_4Br_4$		-107.7			357(b)
$C_8F_5H_3$		(1)(2) -106.4, -105.1 (3) -113.2			357(b)
$C_8F_6H_6$		(1) -139.2 (2)(3) -115.0, -110.0	(2-3) 235		358
$C_8F_7H_5$		(1) -205.0 (2)(3) -155.0, -140.0 (4)(5) -132.0, -130.0 (6)(7) -117.0, -115.0	(4-5) 225 (6-7) 225		358
$C_8F_8H_4O_2$		(3) -116.4 (4) -134.2			356
$C_8F_8H_4O_2$		(1)(2) -146.7, -159.8 (4)(5) -131.2, -132.4 (6) -117.8			356
$C_8F_9H_5O$		(2) -159.3 (3)(6) -117.0 (4)(5) -134.1			356
$C_8F_9H_5O$	As above	(2) -151.7 (3)-(6) -117.6, -126.7			355

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_8F_{12}		(1) -198.0 (2) -150.5 (3) -128.0		358
$C_9FH_{11}O_3$		(axial F) -189 (equatorial F) -188.5	(1-2) 49 (1-3) 30 (1-4) 15 (1-2) 49 (1-3) 10 (1-4) 10	359
$C_9FH_{14}Br$			(1-2) 47	349
$C_9FH_{14}Br$			(1-2) 47	349
C_9FH_{15}			(F-C1) 175.0 (F-C2) 22.1 (F-C3) 13.2 (F-C4) 2.9 (F-C5) 8.8 (F-C6) 16.2	336, 367
$C_9F_2H_6$		(1) -156.2 (2) -146.2	(1-2) 2 (1-3) 5 (2-3) 2	203(a)

C₉F₂H₆Cl₂



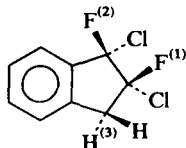
(1) -108.9 (2) -93.9

(1-2) 12
(2-3) <1

(1-3) 18

203(a)

C₉F₂H₆Cl₂



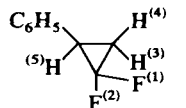
(1) -119.1 (2) -113.6

(1-2) 11
(2-3) <1

(1-3) 15, 16

203(a)

C₉F₂H₈

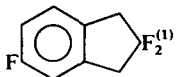


(1) -142.5 (2) -125.8

(1-2) 152.5 (1-3) 6.2
(1-4) 11.9 (1-5) 0
(2-3) 11.7 (2-4) 5.1
(2-5) 13.4

81

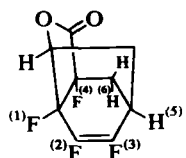
C₉F₃H₇



(1) +0.50 relative to
compound with aromatic F
replaced by H

360

C₉F₄H₆O₂

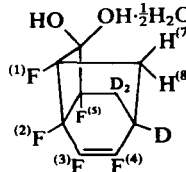


(1) -205.1 (2) -161.3
(3) -135.5 (4) -184.7

(3-5) 16.0 (4-6) 27.0

361

C₉F₅H₅D₃O_{2.5}



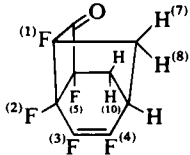
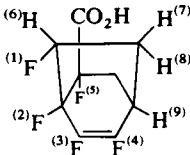
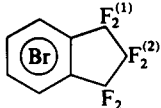
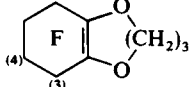
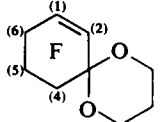
(1) -192.4 (2) -217.0
(3) -162.7 (4) -139.6
(5) -192.6

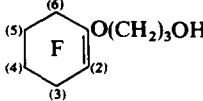
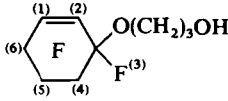
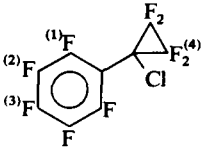
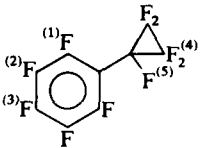
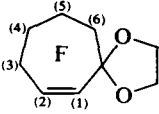
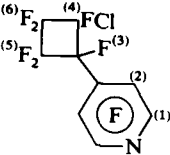
(1-7 or 8) 18.5

361

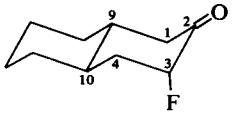
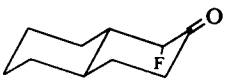
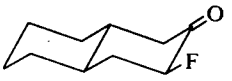
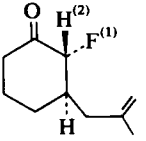
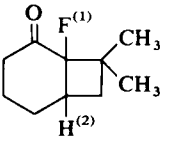
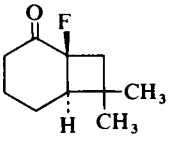
Data

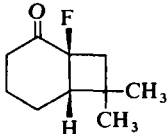
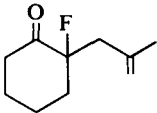
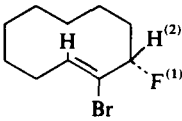
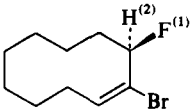
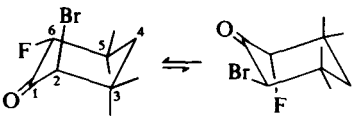
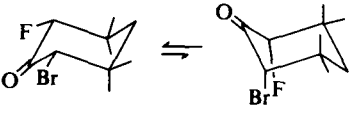
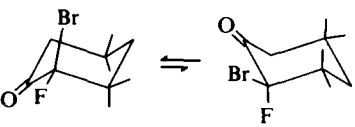
237

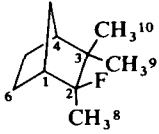
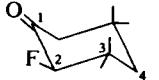
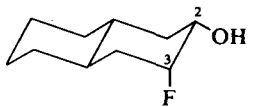
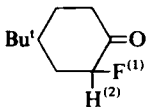
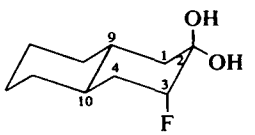
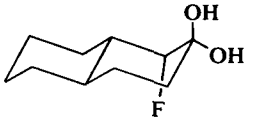
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9F_5H_5O$		(1)(5) -186.6 (2) -216.1 (3) -161.0 (4) -131.7	(1-8)(5-10) 15.0	361
$C_9F_5H_7O_2$		(1) -184.4 (2) -204.6 (3) -158.6 (4) -141.4 (5) -155.8	(1-5) 37.5 (1-6) 56.0 (1-7) 13.0 (1-8) 20.0 (4-9) 16.0	361
$C_9F_5H_8O_{2.5}$	As $C_9F_5H_5D_3O_{2.5}$ above with D replaced by H	(1)(5) -193.5 (2) -218.0 (3) -164.5 (4) -140.6	(1-2) 5 (1-7)(5-9) 5 (1-8)(5-10) 18 (2-5) 5 (4-6) 14.0	361
$C_9F_6Br_4$		(1) -107.7 (2) -126.9		357(b)
$C_9F_8H_6O_2$		(3) -117.0 (4) -134.6		356
$C_9F_8H_6O_2$		(1)(2) -143.0, -161.8 (4)(5) -125.2, -131.8 (6) -118.1		356

$C_9F_9H_7O_2$		(2) -158.8 (3)(6) -117.1 (4)(5) -134.3	356
$C_9F_9H_7O_2$		(1)(2) -144.6, -156.4 (3) -126.7 (4)(5) -130.7, -130.7, -134.3 (2:1:1) (6) -116.2, -120.4	(4A-4B) or (5A-5B) 268 (6A-6B) 283 356
C_9F_9Cl		(1) -136.5 (2) -159.1 (3) -142.6 (4) -145.0, -147.2	(4A-4B) 179 195(b)
C_9F_{10}		(1) -136.4 (2) -159.6 (3) -145.1 (4) -147.9, -156.5 (5) -145.9	(4A-4B) 205 195(b)
$C_9F_{10}H_4O_2$		(1)(2) -137.5, -150.3 (3)-(6) -114.7, -126.5, -126.5, -128.6	355
$C_9F_{10}ClN$		(1) -89.8 (2) -137.6 (3) -154.6, -160.6 (4) -141, -144 (5)(6) -117 to -136	212

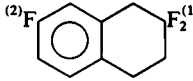
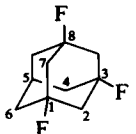
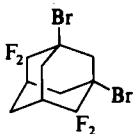
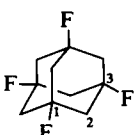
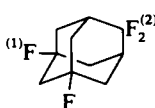
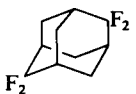
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9F_{11}H_5O_2$		(1) -139.8, (2)-(6) -112.0, -113.6, -128.7 (2:2:6)		355
$C_9F_{11}H_5O_2$		(1) -223.1	(1-2) 48	355
$C_{10}FH_9$		-123.75	(1-2) 15	362
$C_{10}FH_{10}I$		-165	(1-2) 52.5 (1-3) 7 (1-4) 26	363
$C_{10}FH_{10}I$		-139	(1-2) 54 (1-3) 9	363
$C_{10}FH_{11}$		-181.1	(1-2) 51	362

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}FH_{15}O$			(F-C1) 2 (F-C2) 21 (F-C3) 175 (F-C4) 21	365
$C_{10}FH_{15}O$			(F-C1) 177 (F-C2) 22 (F-C3) 2-5 (F-C9) 21	365
$C_{10}FH_{15}O$			(F-C2) 19 (F-C3) 172	365
$C_{10}FH_{15}O$		-196.5	(1-2) 47	339
$C_{10}FH_{15}O$		-152.52	(1-2) 22	339
$C_{10}FH_{15}O$		-173.2		339

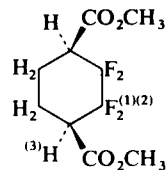
$C_{10}FH_{15}O$		-149.9	339
$C_{10}FH_{15}O$		-155.6	339
$C_{10}FH_{16}Br$		(1-2) 47	349
$C_{10}FH_{16}Br$		(1-2) 47	349
$C_{10}FH_{16}BrO$		(F-C1) 17.4 (F-C4) 4.4 (F-C5) 18.8 (F-C6) 193.0	366
$C_{10}FH_{16}BrO$		(F-C1) 15.9 (F-C4) 5.8 (F-C5) 18.1 (F-C6) 201.2	366
$C_{10}FH_{16}BrO$		(F-C1) 18.4 (F-C2) 274.8 (F-C3) 17.2 (F-C4) 4.2	366

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}FH_{17}$		-134.28	(F-C1) 24.8 (F-C2) 180.0 (F-C3) 17.4 (F-C6) 9.0 (F-C9) 1.3 (F-C10) 14.3	336
$C_{10}FH_{17}O$			(F-C1) 14.5 (F-C2) 194.7 (F-C3) 18.0 (F-C4) 5.3	366
$C_{10}FH_{17}O$			(F-C2) 19 (F-C3) 172	365
$C_{10}FH_{17}O$		-184.1(a), -186.9(e)	(1-2) 47.7	340(a)
$C_{10}FH_{17}O_2$			(F-C1) 0 (F-C2) 23 (F-C3) 174 (F-C4) 18	365
$C_{10}FH_{17}O_2$			(F-C1) 178.5 (F-C2) 22.5 (F-C9) 19.5	365

$C_{10}F_2H_{10}$		(1)(2) -180.4, -195	(1-2) 15 (2-3) 12	(1-4) 15	362
$C_{10}F_2H_{10}$		(1)(2) -183, -200.3	(1-2) 18 (2-3) 12	(1-4) 18	362
$C_{10}F_2H_{10}$		(1) -132.4 (2) -137.5	(1-2) 149 (1-4) 13 (2-4) 4.5	(1-3) 5 (2-3) 11.5	81
$C_{10}F_2H_{14}$		-132.5	(F-C1) 188 (F-C3) 13.3 (F-C6) 16.0	(F-C2) 19.0 (F-C5) 10.3	364
$C_{10}F_2H_{14}$		-99.75			239(a)
$C_{10}F_2H_{14}O$		(1) -109.3 (2) -115.5	(1-2) 280		134
$C_{10}F_3H_9$		(1) -0.32 relative to compound with aromatic F replaced by H (2) -116.81			360(d)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_3H_9$		(1) -0.39 relative to compound with atomic F replaced by H (2) -117.32		360(d)
$C_{10}F_3H_{13}$		-138.9	(F-C1) 191 (F-C2) 18.8 (F-C3) 14.8 (F-C6) 16.9 (F-C5) 12.0	364
$C_{10}F_4H_{10}Br_2$		$-105.73, -110.16$	(F-F) 234	239(a)
$C_{10}F_4H_{12}$		-148.5	(F-C1) 192 (F-C2) 19.4 (F-C3) 17.2	364
$C_{10}F_4H_{12}$		(1) -143.2 (2) -107.21		239(a)
$C_{10}F_4H_{12}$		-102.07		239(a)

C₁₀F₄H₁₂O₄



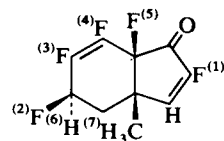
−124·0, −116·0

(1-2) 260

(1 or 2-3) 15

358

C₁₀F₅H₅O



(1) −139·7 (2) −193·4
(3) −143·4 (4) −153·8
(5) −177·5

(2-3) 24·0

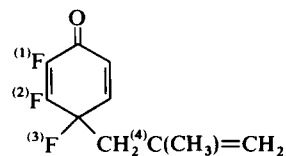
(2-6) 52·0

329

(4-5) 29·0

(5-7) 6·0

C₁₀F₅H₇O



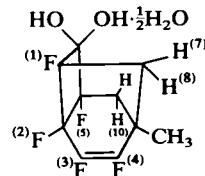
(1) −157·4 (2) −137·6
(3) −159·3

(2-3) 31·0

(3-4) 11·0

329

C₁₀F₅H₁₀O_{2·5}



(1)(5) −192·4 (2) −215·6
(3) −163·4 (4) −155·7

(1-2) 5

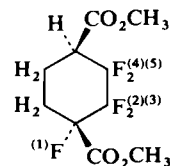
(1-8)(5-10) 18

361

(2-5) 5

(2-4) 8

C₁₀F₅H₁₁O₄



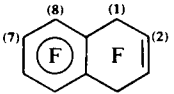
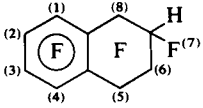
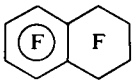
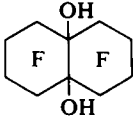
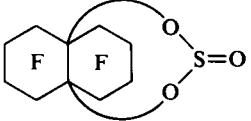
(1) −175·0
(2)(3) } −130·2
(4)(5) } −122·0, −114·0

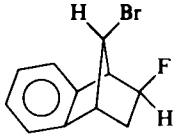
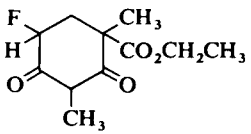
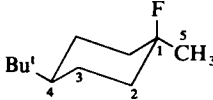
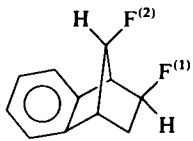
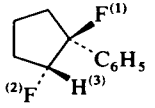
(2-3) or (4-5) 260

358

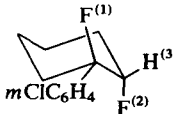
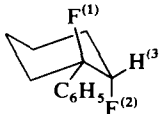
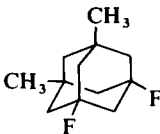
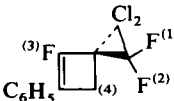
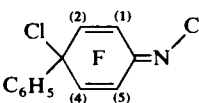
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_6H_{10}O_4$		(1) -177.0 (2) -129.0 (3) -130.0	(2-3) 235	358
$C_{10}F_6Cl_2O$		(3) -103.5 (4) -110.6 (5) -132.6 (6) -143.0 (7) -147.6 (8) -136.7	(3-4) 28 (3-6) 4 (4-5) 20 (5-6) 20 (5-7) 8.5 (5-8) 12.5 (6-7) 20 (6-8) 12.5 (7-8) 20	342(b)
$C_{10}F_7ClO$		(3) -114.1 (4) -99.7 (5) -137.7 (6) -143.9 (7) -147.5 (8) -137.7	(3-4) 25 (3-6) 4 (4-5) ~16 (5-6) ~19 (5-7) ~7 (6-7) ~19 (6-8) 12 (7-8) ~19	342(b)
$C_{10}F_7Cl_2N$		(1) -115.8, -116.0 (2)(3)(4) -133.5, -134.9, -149.5 to -151.5 (5) -140.0, -141.5 (6) -133.0, -136.2 (7) -152.3, -149.5 to -151.5	(1-2) 46 and 32 (5-6) 70 and 72	368(a)

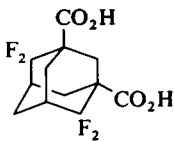
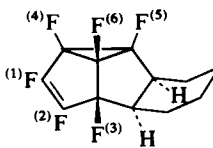
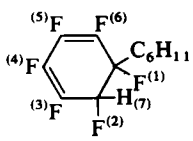
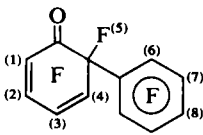
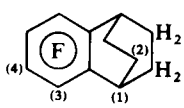
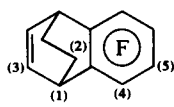
$C_{10}F_9H_7O_3$		(1) -156.5 (2)(5) -117.0 (3)(4) -133.9	356
$C_{10}F_9H_7O_3$		(1)(2) -146.0, -156.8 (4A-4B) or (5A-5B) 273 (3) -127.8 (6A-6B) 282 (4)(5) -131.3, -131.3 and -133.5 (6) -116.8, -121.0	356
$C_{10}F_9H_9O_2$		(1) -158.5 (2)(5) -116.8 (3)(4) -133.9	356
$C_{10}F_9H_9O_2$		(1)(2) -145.0, -157.3 (4A-4B) or (5A-5B) 274 (3) -126.2 (6A-6B) 282 (4)(5) -132.4, -132.4 and -134.9 (6) -116.5, -121.1	356
$C_{10}F_9H_{15}BN$		(1) -95.8 (2) -119.2 (3) -112.4 (4) -153.3	369
$C_{10}F_{10}$		(1)(2)(3)(4) -136.7, -147.2, -149.2, -158.8 (5)(6) } -118.1 (7)(8) } -126.6	230
$C_{10}F_{10}$	As above	(1)(10) -138.5, -140.6 (4-5,6) 32 (1-10) 62 (2)(3) -146.2, -148.3 (4) -135.7 (5)(6) -117.1 (7)(8) -125.5 (9) -158.6	233

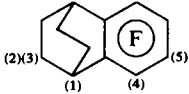
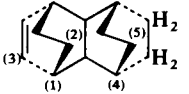
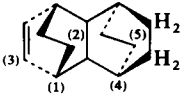
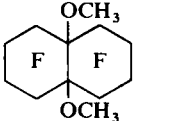
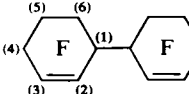
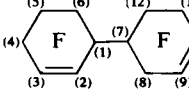
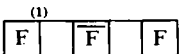
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_{10}$		(1) -99.5 (2) -155.5 (7) -147.7 (8) -137.2		233
$C_{10}F_{11}H$		(1)(2)(3)(4) -136.6, -146.6 (2:2) (5) } -92.4, -103.4 (6) } -105.7, -110.7 (8) } -127.4, -131.8 (7) -217.6	(5A-5B) } 296 (6A-6B) } 299 (8A-8B) } 274	230
$C_{10}F_{11}Br$	As above with H replaced by Br	(1)(2)(3)(4) -146.7, -136.8 (2:2) (5) } -81.7, -113.8 (6) } -95.0, -107.3 (8) } -117.3, -131.3 (7) -136.4	(5A-5B) } 299 (6A-6B) } 307 (8A-8B) } 279	230
$C_{10}F_{12}$		-106.9, -136.1, -145.2 (2:3:1)		370
$C_{10}F_{16}$		-111.3, -135.3 (1:1)		370
$C_{10}F_{16}H_2O_2$		-116.9 and -121.7 -128.3 and -130.6	(A-B) 285 (A-B) 282	370
$C_{10}F_{16}O_3S$		-110.0 and -120.6, -131.5 (1:1)	(A-B) 307	370

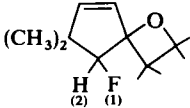
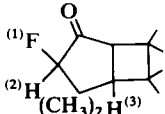
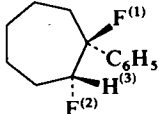
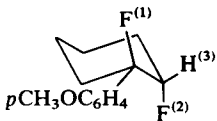
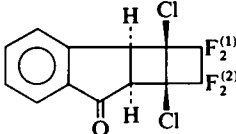
$C_{11}FH_{10}Br$		-177.75		347
$C_{11}FH_{10}Cl$	As above with Br replaced by Cl	-179.25		347
$C_{11}FH_{15}O_4$		(axial F) -185 (equatorial F) -182.5		359
$C_{11}FH_{21}$		-153.77	(F-C1) 167.7 (F-C2) 22.8 (F-C3) <0.8 (F-C5) 25.0	336
$C_{11}FH_{21}$	As above with F and CH ₃ interchanged	-127.6	(F-C1) 167.7 (F-C2) 20.6 (F-C3) 11.6 (F-C5) 25.7	336
$C_{11}F_2H_{10}$		(1) -182.25 (2) -183		347, 352
$C_{11}F_2H_{12}$		(1) -157.9 (2) -186.4	(1-3) 7.5 (2-3) 52	371
$C_{11}F_2H_{12}$	As above with F(2) and H(3) interchanged	(1) -172.5 (2) -207.4	(1-3) 18 (2-3) 54	371

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{11}F_2H_{16}$		(1)(2) -137.7, -140.2	(1-2) 158.7 (1-3) 10.5 (2-3) 7.5	81
$C_{11}F_4H_{15}NO$		-106.5, -114.2		337(a)
$C_{11}F_9H_3O$		(1)(2)(3) -132.4, -139.4, -143.7 (4)(7) -99.0 (5)(6) -155.4		230
$C_{12}FH_{13}BrCl$		-154.0	(1-2) 45	372
$C_{12}FH_{13}Br_2$	As above with Cl replaced by Br	-155.5	(1-2) 30	372
$C_{12}FH_{14}Br$	As above with Cl replaced by H	-157.3	(1-2) 45	372
$C_{12}FH_{17}O$		Isomer [a] [b]	(1-2) 52.0 (1-2) 52.0	373
$C_{12}FH_{19}$		-132.9		364

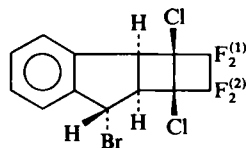
C ₁₂ FH ₂₃	$F^{(1)}\overbrace{CH^{(2)}(CH_2)_{10}CH_2}$			(1-2) 47 ² J(1-C) 20.6 ³ J(1-C) 5.9	¹ J(1-C) 166.2	336
C ₁₂ F ₂ H ₁₃ Cl		(1) -178.5 (2) -211	(1-3) <1 (2-3) 48			371
C ₁₂ F ₂ H ₁₃ Cl	As above with F(2) and H(3) interchanged	(1) -198 (2) -204	(1-3) 25 (2-3) 48			371
C ₁₂ F ₂ H ₁₄		(1) -178.4 (2) -212.0	(1-3) <1 (2-3) 49.5			371
C ₁₂ F ₂ H ₁₄	As above with F(2) and H(3) interchanged	(1) -197.5 (2) -205.5	(1-3) 25 (2-3) 48			371
C ₁₂ F ₂ H ₁₈		-139.8				364
C ₁₂ F ₃ H ₇ Cl ₂		(1) -129.29 (2) -133.70 (3) -95.43	(1-2) 228.4 (2-3) 2.93 (2-4) 1.50 (3-4) 12.52			374
C ₁₂ F ₄ H ₅ Cl ₂ N		(1) -151.1 (2) -131.7 (4) -137.5 (5) -152.0	(1-2) 5.4 (1-5) ~1 (2-4) 4.3 (2-5) ~1 (4-5) 9.8			175(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}F_4H_{12}O_4$		-107.87, -110.81	(F-F) 236	239(a)
$C_{12}F_6H_{10}$		(1)(2) -145, -147 (3) -170 (4) -198 (5)(6) -222.5, -239		249
$C_{12}F_6H_{12}$		(1) -164.25 (2) -170.25 (3)-(6) -152.25, -153.75, -146.25, -146.85	(1-2) 21 (2-7) 60	375
$C_{12}F_{10}O$		(1) -159.0 (2) -132.0 (3) -155.3 (4) -147.7 (5) -124.2 (6) -152.4 (7) -162.2 (8) -156.2		376
$C_{12}F_{10}H_4$		(1) -193.4 (2) -127.5, -129.6 (3) -146.6 (4) -153.1	(1-3) 46 (2A-2B) 232	377
$C_{12}F_{12}$		(1) -218.2 (2) -124.2, -125.0 (3) -158.0 (4) -146.0 (5) -154.8	(1-4) 36 (2A-2B) 215	377

$C_{12}F_{14}$		(1) -219.6 (2) -125.3 (3) -127.6 (4) -145.0 (5) -147.9	(1-4) 38 (2-3) 250	231
$C_{12}F_{14}H_4$		(1) -218.3 (2) -123.4, -124.8 (3) -153.1 (4) -198.6 (5) -131.6, -133.5	(2A-2B) 215 (1-4) 30 (5A-5B) 240	377
$C_{12}F_{14}H_4$		(1) -218.1 (2) -120.7, -122.8 (3) -152.9 (4) -197.6 (5) -125.3, -128.3	(2A-2B) 225 (1-4) 32 (5A-5B) 230	377
$C_{12}F_{16}H_6O_2$		-114, -129.5 (1:1)		370
$C_{12}F_{18}$		(1) -161.5 (2)(3) -139.7, -147.9 (4)-(6) -106 to -146		378
$C_{12}F_{18}$		(1) -163.1 (2)(3) -138.3 and -140.7, -146.9 and -148.3 (7) -173.5, -175.3 Rest -106 to -146		378
$C_{12}F_{18}$		(1) -180		379

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{13}FH_{21}O$		Isomer [a] [b]	(1-2) 50.0 (1-2) 50.0	373
$C_{13}FH_{21}O$			(1-2) 51 (1-3) 6.0	373
$C_{13}F_2H_{16}$		(1) -159.4 (2) -189.4	(1-3) 6 (2-3) 45	371
$C_{13}F_2H_{16}$	As above with F(2) and H(3) interchanged	(1)(2) -181.1	(1-3) 27 (2-3) 45	371
$C_{13}F_2H_{16}O$		(1) -178 (2) -211	(1-3) <1 (2-3) 48	371
$C_{13}F_2H_{16}O$	As above with F(2) and H(3) interchanged	(1) -197 (2) -205	(1-3) 24.5 (2-3) 48	371
$C_{13}F_4H_6Cl_2O$		(1) } -116.3 (2) } -115.5 and -118.6	(1A-1B) or (2A-2B) 203	380(a)

C₁₃F₄H₇BrCl₂



(1) } -114.9 and -119.7,
(2) } -116.5 and -118.0

(1A-1B) } 203
(2A-2B) } 183

380(a)

C₁₃F₄H₇Cl₃

As above with Br replaced by Cl

(1) } -115.0 and -119.3,
(2) } -116.2 and -118.0

(1A-1B) } 203
(2A-2B) } 172

380(a)

C₁₃F₄H₁₀

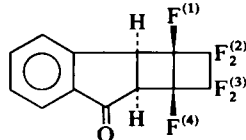
As above with Br and Cl replaced by H

(1) } -115.5,
(3) } -114.9 and -119.6

(1A-1B) or (2A-2B) 209

380(a)

C₁₃F₆H₆O

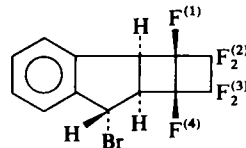


(1)(4) -176.4, -179.6
(2)(3) { -119.8 and -132.4,
 -120.9 and -131.6

(1-4) 26 (2A-2B) } 214
 (3A-3B) } 218

380(a)

C₁₃F₆H₇Br



(1)(4) -177.4, -190.1
(2)(3) { -121.6 and -134.4,
 -123.0 and -131.7

(1-4) 28 (2A-2B) } 218
 (3A-3B) } 223

380(a)

C₁₃F₆H₈

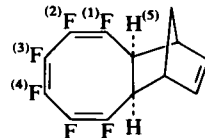
As above with Br replaced by H

(1)(4) -178.3, -190.4
(2)(3) { -122.5 and -132.4,
 -122.6 and -133.8

(1-4) 26 (2A-2B) } 219
 (3A-3B) } 219

380(a)

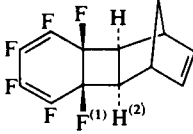
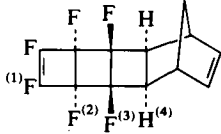
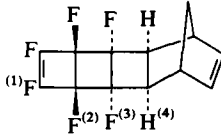
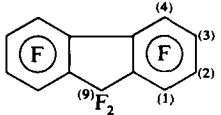
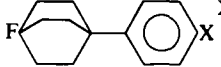
C₁₃F₆H₈



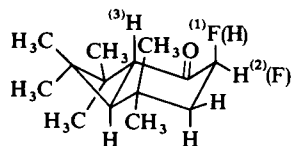
-159, -161.25, -168

(1-2)(3-4) 13 (1-5) 26

381

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{13}F_6H_8$		-162.3, -163.9, -188.62	(1-2) 7.5	381
$C_{13}F_6H_8$		(1) -123.75 (2) -180.75 (3) -194.25	(1-2) 9 (2-3) 15 (3-4) 7.5	381
$C_{13}F_6H_8$		(1) -125.25 (2) -180 (3) -164.25	(1-2) 4.5 (2-3) 15 (3-4) 24	381
$C_{13}F_{10}$		(1) -140.0 (2) -150.5 (3) -146.9 (4) -133.4 (9) -108.8		430
$C_{14}FH_{16}Br$	 X=Br	-153.11		382
$C_{14}FH_{16}NO_2$	As above with X=NO ₂	-153.42		382
$C_{14}FH_{17}$	As above with X=H	-152.77		382
$C_{14}FH_{18}N$	As above with X=NH ₂	-152.59		382

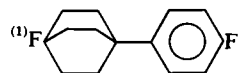
C₁₄FH₂₃O



Isomer [a] (1-2) 49.0 (1-3) 5.5
[b] (1-2) 48.0 (1-3) 0

373

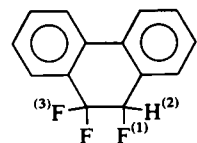
C₁₄F₂H₁₆



(1) -153.05

382

C₁₄F₃H₉

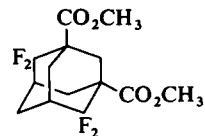


(1) -194.6

(1-2) 50.1 (1-3) 15.6

463

C₁₄F₄H₁₆O₄

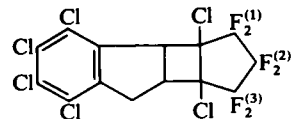


-106.90, -109.95

(F-F) 237

239(a)

C₁₄F₆H₄Cl₆

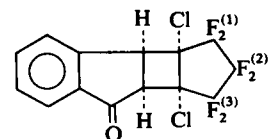


(1) -109.8 and -132.8
(2) -110.4 and -131.7
(3) -122.8 and -137.7

(1A-1B) 244
(2A-2B) 241
(3A-3B) 263

380(a)

C₁₄F₆H₆Cl₂O



(1) -100.6 and -127.1
(2) -124.6 and -135.2
(3) -103.6 and -124.6

(1A-1B) 254
(2A-2B) 255
(3A-3B) 253

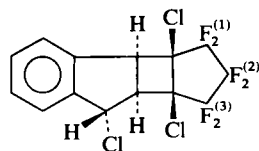
380(a)

Data

259

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_6H_6Cl_2O$		(1) -109.0 and -131.8, (2) -122.3 and -137.7, (3) -112.4 and -131.8	(1A-1B) 243 (2A-2B) 254 (3A-3B) 243	380(a)
$C_{14}F_6H_6O$		(1) -111.3, (2) -110.8 and -113.3, (3) -124.6	(1A-1B) } or (2A-2B) } 248 or (3A-3B) }	380(a)
$C_{14}F_6H_6O$		(1) -111.3, (2) -114.1, (3) -136.2		380(a)
$C_{14}F_6H_7BrCl_2$		(1) -107.6 and -132.6, (2) -111.9 and -130.1, (3) -122.0 and -137.1	(1A-1B) 247 (2A-2B) 245 (3A-3B) 263	380(a)
$C_{14}F_6H_7ClO$		(1) -109.6 and -133.0, (2) -111.2 and -127.3, (3) -123.9 and -138.1	(1A-1B) 248 (2A-2B) 256 (3A-3B) 261	380(a)

C₁₄F₆H₇Cl₃

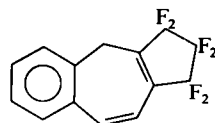


(1) } -107.9 and -132.8
(2) } -111.9 and -130.6
(3) } -122.2 and -137.4

(1A-1B) } 250
(2A-2B) } 246
(3A-3B) } 261

380(a)

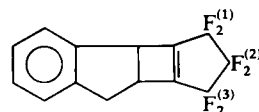
C₁₄F₆H₈



-110.1, -110.6, -128.8

380(a)

C₁₄F₆H₈

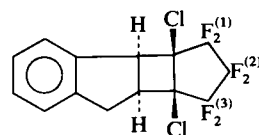


(1) } -110.5,
(2) } -110.3 and -112.3,
(3) } -124.7

(1A-1B) }
or (2A-2B) } 248
or (3A-3B) }

380(a)

C₁₄F₆H₈Cl₂

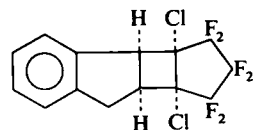


(1) } -109.0 and -133.2
(2) } -122.0 and -137.3
(3) } -110.8 and -130.6

(1A-1B) } 247
(2A-2B) } 260
(3A-3B) } 241

380(a)

C₁₄F₆H₈Cl₂

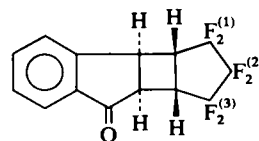


(1) } -98.3 and -126.9
(2) } -127.7 and -135.8
(3) } -100.3 and -127.0

(1A-1B) } 257
(2A-2B) } 253
(3A-3B) } 252

380(a)

C₁₄F₆H₈O



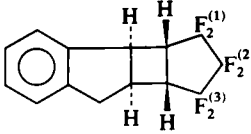
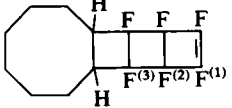
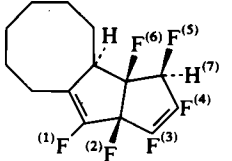
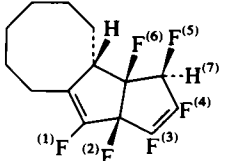
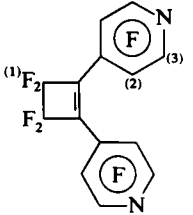
(1) } -111.2 and -128.0
(2) } -111.2 and -128.6
(3) } -124.8 and -142.1

(1A-1B) } 254
(2A-2B) } 254
(3A-3B) } 265

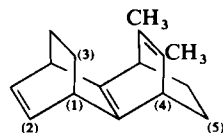
380a

Data

261

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_6H_{10}$		(1) -109.2 and -129.8 (2) -111.6 and -127.2 (3) -124.7 and -135.9	(1A-1B) 248 (2A-2B) 255 (3A-3B) 257	380(a)
$C_{14}F_6H_{14}$		Isomer [a] (1) -121.5 (2) -191 (3) -170 [b] (1) -123 (2) -186 (3) -160		249
$C_{14}F_6H_{14}$		(1)(3)(4) -141, -146, -148.5 (2) -164.5 (5) -190 (6) -195.5	(5-7) 53	249
$C_{14}F_6H_{14}$		(1)(3)(4) -139, -142, -147 (2) -164 (5) -197 (6) -178	(5-7) 55	249
$C_{14}F_{12}N_2$		(1) -113.0 (2) -138.6 (3) -91.1		212

C₁₄F₁₄H₆

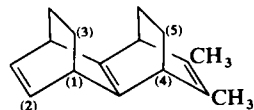


(1) -218.5 (2) -153.4
(3) -121.6, -122.9
(4) -208.9
(5) -120.9, -124.2

(1-4) 28 (3A-3B) 220
(5A-5B) 225

377

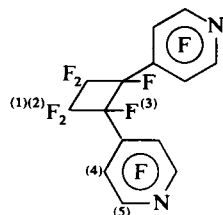
C₁₄F₁₄H₆



(1) -218.0 (2) -153.5
(3) -123.0 (4) -209.0
(5) -123.0

377

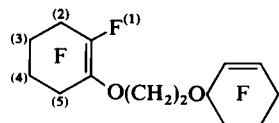
C₁₄F₁₄N₂



(1)(2) -121.1, -131.8
(3) -164.3 (4) -138.5
(5) -97.5

212

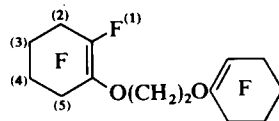
C₁₄F₁₈H₄O₂



(1) -153.4 (2)(5) -115.5
(3)(4) -132.1

356

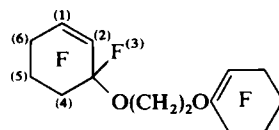
C₁₄F₁₈H₄O₂



(1) -155.8 (2)(5) -117.0
(3)(4) -134.1

356

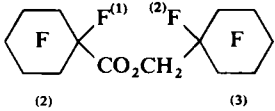
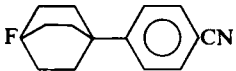
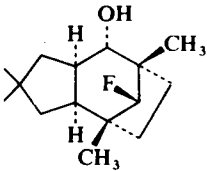
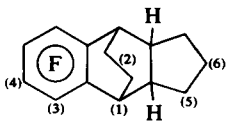
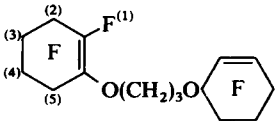
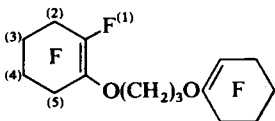
C₁₄F₁₈H₄O₂

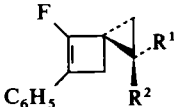
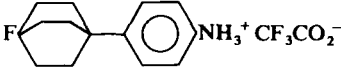
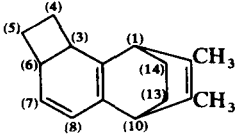
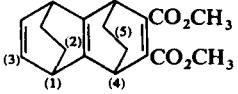
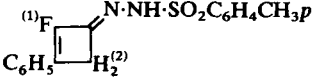
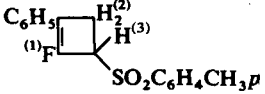


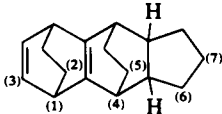
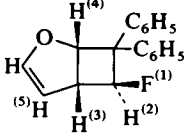
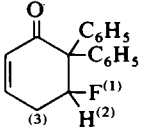
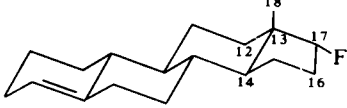
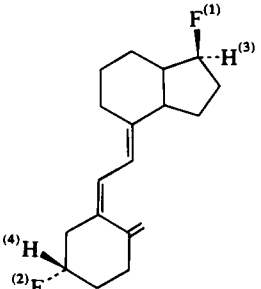
(1)(2) -145.5, -153.4
(3) -126.1
(4)(5) -127.8, -128.7 and -133.5
(6) -114.3 and -119.3

(4A-4B) or (5A-5B) 282
(6A-6B) 282

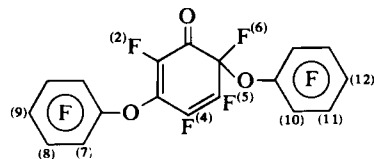
356

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_{22}H_2O_2$		(1) -180.2 (2) -189.2 (3) -115 to -150	$^2J(F-F)$ 290	237
$C_{15}FH_{16}N$		-153.35	$^2J(F-H)$ 53	382 383
$C_{15}FH_{25}O$				
$C_{15}F_{16}H_2$		(1) -199.1 (2) -126.8 (3) -145.2 (4) -150.1 (5) -116.3 and -121.4 (6) -133.9 and -142.9	(1-3) 40 (5A-5B) 254 (6A-6B) 252	377
$C_{15}F_{18}H_6O_2$		(1) -157.1 (2)(5) -117.2 (3)(4) -134.4		356
$C_{15}F_{18}H_6O_2$		(1) -157.1 (2)(5) -117.3 (3)(4) -134.3		356

$C_{16}FH_{15}$	 $R^1 = H$ $R^2 = C_6H_5$ $R^1 = C_6H_5$ $R^2 = H$	–91·02 –98·84	374	
$C_{16}F_4H_{19}NO_2$		–154·70	382	
$C_{16}F_{14}H_6$		(1) –205·5 (3)(5) –176·0, –189·7 (4)(5)(13)(14) –120·9, –123·8 (7) –151·1 (8) –124·0 (10) –205·3	233	
$C_{16}F_{16}H_6O_4$		Isomer [a] (1) –218·1 (2A-2B) 230 (2) –120·9, –122·5 (1-4) 30 (3) –152·1 (4) –208·2 (5) –120·7 [b] (1) –218·5 (2) –120·0 (1-4) 30 (3) –150·2 (4) –207·8 (5) –120·0	377	Data
$C_{17}FH_{15}N_2SO_2$		(1-2) 13·20	374	
$C_{17}FH_{15}O_2S$		–92·64 (1-2) 13·98 (1-3) 6·23	374	
$C_{17}F_6H_{22}ClP$		–105·2, –114, –129·6 (1:1:1)	384	265

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{17}F_{20}H_2$		(1) -216.8 (2) -121.3, -122.3 (3) -153.0 (4) -201.4 (5) -124.3 (6) -117.0, -119.7 (7) -132.6, -142.7	(2A-2B) 218 (6A-6B) 256 (7A-7B) 254	377
$C_{18}FH_{15}O$		-187.5	(1-2) 57 (1-3) 12 (1-4) 6 (1-5) 4.5	282
$C_{18}FH_{15}O$		-187.13	(1-2) 57 (1-3) 10	282
$C_{18}FH_{27}$			(F-C12) 8.6 (F-C13) 16.2 (F-C16) 15.8 (F-C17) 180.5 (F-C18) 6.9	336
$C_{18}F_2H_{24}$			(1-3) 50 (2-4) 50	385

C₁₈F₁₄O₃



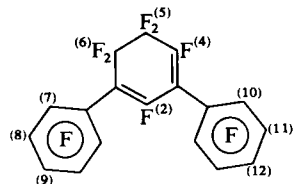
(2) -159.7 (4) -152.1
(5) -148.1 (6) -123.9
(7)(10) -155.8, -152.1
(8)(11) -162.3, -161.3
(9)(12) -157.0, -156.5

(2-5) 20

(5-6) 20

376

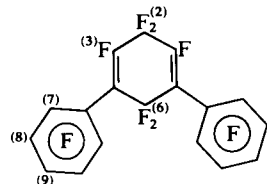
C₁₈F₁₆



(2) -100.3 (4) -113.7
(5) -130.5 (6) -119.0
(7)(10) -136.4, -137.4
(8)(11) -144.9, -160.3
(9)(12) -148.7, -150.0

386(b)

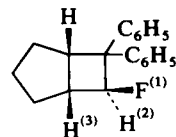
C₁₈F₁₆



(2) -120.5 (3) -113.6
(6) -92.9 (7) -137.3
(8) -160.2 (9) -148.5

386(b)

C₁₉FH₁₉



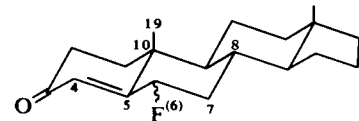
-232.5

(1-2) 60

(1-3) 18

282

C₁₉FH₂₅O₂



(axial F)

(6-4H) 5

(6-6H) 50

387

(6-19H) 2.2

(6-C3) <1

(6-C4) 9.2

(6-C5) 12.0

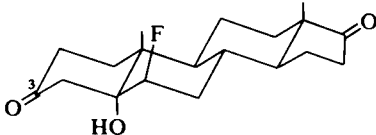
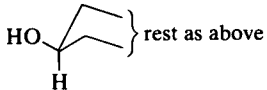
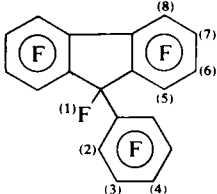
(6-C6) 167

(6-C7) 23

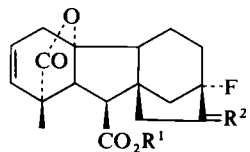
(6-C8) <1

(6-C10) 2.0

(6-C19) <1

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
		(equatorial F)	(6-4H) <1 (6-6H) 49 (6-19H) 0 (6-C3) <1 (6-C4) 14.7 (6-C5) 11.0 (6-C6) 185 (6-C7) 18.4 (6-C8) 11.0 (6-C10) 2.7 (6-C19) 0	
C ₁₉ FH ₂₇ O ₃	 (numbering as above)		(6-6H) 49 (6-19H) 3.6 (6-C4) <1 (6-C5) 23 (6-C6) 179 (6-C7) 27 (6-C8) <1 (6-C10) 1.8 (6-C19) 9.2	387
C ₁₉ FH ₂₉ O ₃	 rest as above		(6-6H) 50 (6-19H) 4.4 (6-C4) <1 (6-C5) 19.3 (6-C6) 178 (6-C7) 20 (6-C8) <1 (6-C10) 1.8 (6-C19) 9.2	387
C ₁₉ F ₁₄		(1) -158.4 (2) -141.8 (3) -160.7 (4) -151.5 (5) -140.2 (6) -151.5 (7) -148.4 (8) -132.9		430(b)

C₂₀FH₂₅O₄

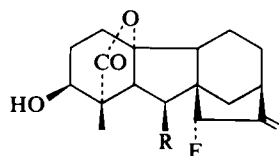


R¹ = CH₃ R² = αH, βCH₃
 R¹ = CH₃ R² = αCH₃, βH

-147.1
 -159.7

388

C₂₀FH₂₅O₅

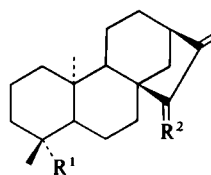


R = CO₂CH₃

²J(F-H) 55

389

C₂₀FH₂₉O₂



R¹ = CO₂H, R² = αF, βH

²J(F-H) 55

389

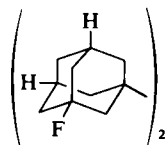
C₂₀FH₃₁O

As above with R¹ = CH₂OH, R² = αF, βH

²J(F-H) 55

389

C₂₀F₂H₂₈

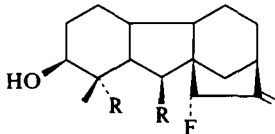
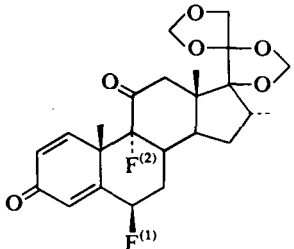
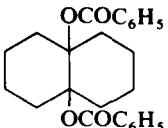
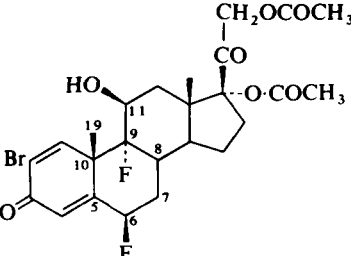


-127.8

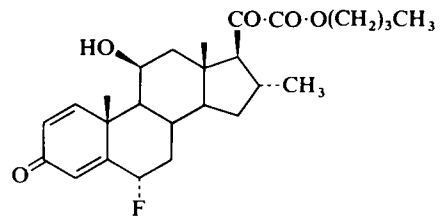
364

Data

269

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{21}FH_{29}O_5$	 $R = CO_2CH_3$		$^2J(F-H)$ 55	389
$C_{21}FH_{31}O_2$	As $C_{20}FH_{29}O_2$ above with $R^1 = CO_2CH_3$, $R^2 = \alpha F, \beta H$		$^2J(F-H)$ 55	389
$C_{24}F_2H_{28}O_6$		(1) -169 (2) -167.6	(1-6H) 50 (1-7H) 35, 17 (2-8H) 30	390
$C_{24}F_{16}H_{10}O_4$		-111.1 and -118.1, -131.9 (1:1)	(A-B) 296	370
$C_{25}F_2H_{29}BrO_7$		(6) -152.9 (9) -156.0	(6-6H) 50 (6-4H) 4 (9-19H) 2 (6-7H) 50, 14 (9-8H) 28 (9-11H) 8 (6-4C) 7.5 (6-5C) 12.6 (6-6C) 168.8 (6-7C) 20.2 (9-8C) 25.2 (9-9C) 173.9 (9-10C) 22.7 (9-11C) 35.3 (9-19C) 5.4	391(a)

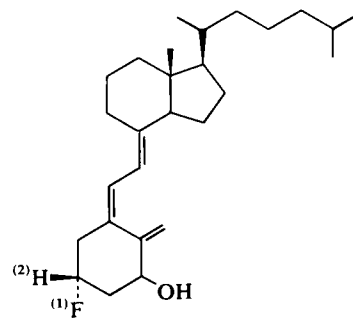
C₂₆FH₃₅O₅



²J(F-H) 49

392

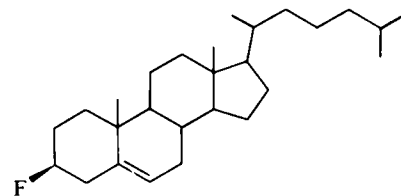
C₂₇FH₄₃O



(1-2) 50

385

C₂₇FH₄₅

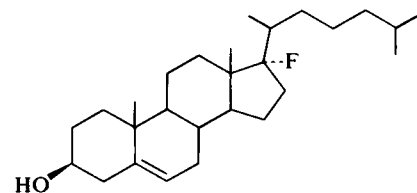


-168.4

(F-C1) 5.2 (F-C2) 18.1
(F-C3) 174.5 (F-C4) 18.9
(F-C5) 12.9 ²J(F-H) 48

336

C₂₇FH₄₅O



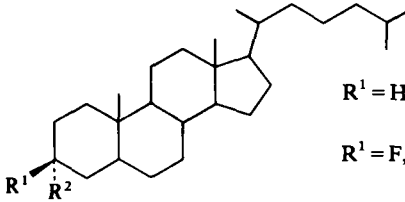
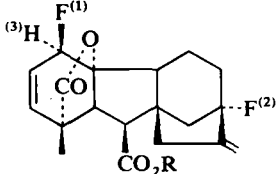
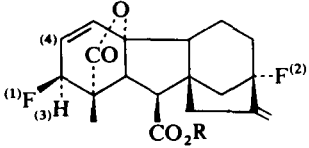
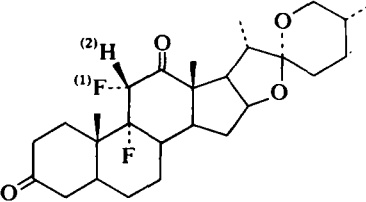
-170

(F-16 α H) = (F-16 β H) =
(F-20H) = 31

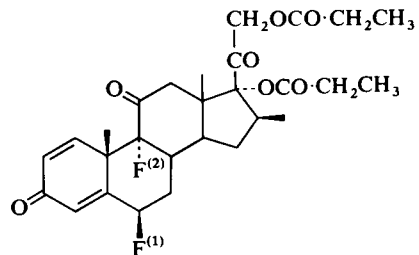
393

Data

271

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{27}FH_{47}$	 $R^1 = H, R^2 = F$ $R^1 = F, R^2 = H$	-181.4 -168.2	(F-C2) 21.3 (F-C3) 166.9 (F-C4) 22.1 $^2J(F-H)$ 49 (F-C2) 22.1 (F-C3) 172.2 (F-C4) 22.1 (F-C1) 8.8	336 336
$C_{27}F_2H_{25}BrO_5$	 $R = pBrC_6H_4CO \cdot CH_2$	(1) -176.0 (2) -147.3	(1-3) 47	388
$C_{27}F_2H_{25}BrO_5$	 $R = pBrC_6H_4CO \cdot CH_2$	(1) -179.3 (2) -147.3	(1-3) 47 (1-4) 9	388
$C_{27}F_2H_{38}O_4$			(1-2) 45	285

C₂₈F₂H₃₄O₇

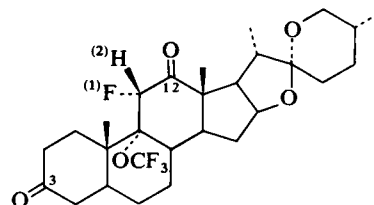


(1) -169.3 (2) -168.2

(1-6H) 50 (1-7H) 40
(2-8H) 30

390

C₂₈F₄H₃₈O₅



(1-2) 43

285

C₂₈F₄H₄₀O₅

As above with 12βOH, 12αH

(1-2) 46

285

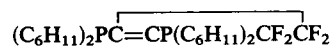
C₂₈F₄H₄₂O₅

As above with 12βOH, 12αH and 3βOH, 3αH

(1-2) 55

285

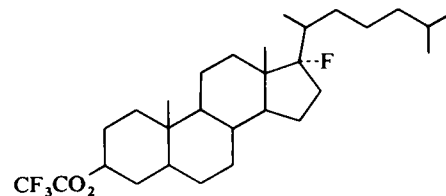
C₂₈F₄H₄₄P₂



-106.4

384

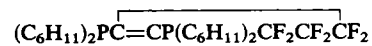
C₂₉F₄H₄₆O₂



-170

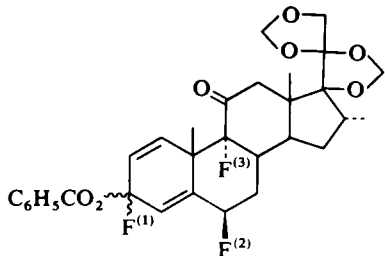
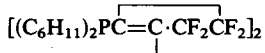
393

C₂₉F₆H₄₄P₂



-107.7, -134 (2:1)

384

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{31}F_3H_{33}O_7$		(1) -86 (2) -165.8 (3) -168.4	(2-6H) 50 (2-7H) 35 (3-8H) 30	390
$C_{32}F_8H_{44}P_2$		-105.8, -109.4 (1:1)		384

B. Miscellaneous studies

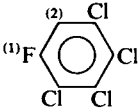
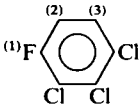
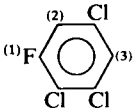
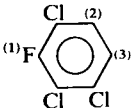
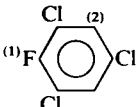
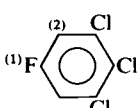
F-19 CIDNP studies have been made of the interaction of radicals with perfluorocycloalkanes, perfluorocycloalkenes, and their partially chlorinated analogues. (394) Dynamic F-19 NMR has been used to determine the rates of ring inversion of some isomeric octa- and deca-fluorocyclohexanes, and perfluorocyclohexane. (395)

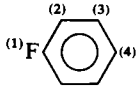
Association constants for the weak 1:1 complexes formed between 2,3,5,6-tetrafluoro-1,4-benzoquinone and benzene substituted and non-benzenoid macrocyclic ethers have been determined. (480)

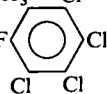
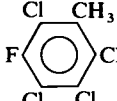
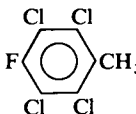
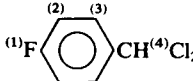
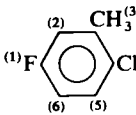
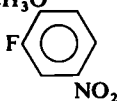
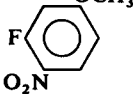
**V. FLUORINE BONDED TO A CARBON OF AN
AROMATIC SYSTEM, INCLUDING CONDENSED
ALICYCLIC AROMATIC SYSTEMS**

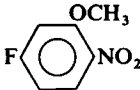
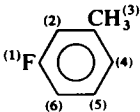
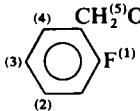
This Page Intentionally Left Blank

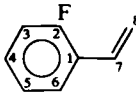
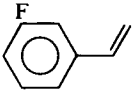
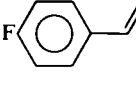
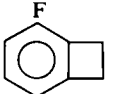
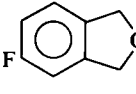
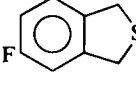
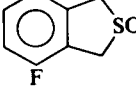
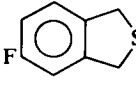
A. Fluorine nuclei in monofluorobenzene derivatives

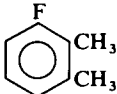
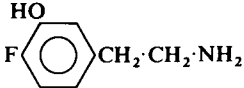
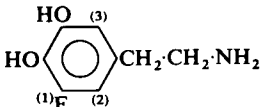
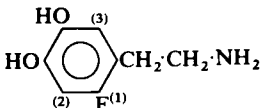
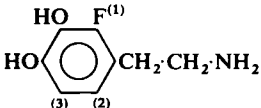
Molecular formula	Structure of compounds	Chemical shifts (pm)	Coupling constants (Hz)	Ref.
C_6FHCl_4		-109.5	(1-2) 9.6	397
C_6FHCl_4	<i>m</i> -isomer of above	-110.0	(1-2) 7.2	397
C_6FHCl_4	<i>p</i> -isomer of above	-103.7	(1-2) 2.4	397
$C_6FH_2Cl_3$		-107.2	(1-2) 7.8 (1-3) 4.2	398
$C_6FH_2Cl_2$		-106.1	(1-2) 8.2 (1-3) 1.9	398
$C_6FH_2Cl_3$		-106.0	(1-2) 4.2 (1-3) 1.4	398
$C_6FH_2Cl_3$		-116.6	(1-2) 6	398
$C_6FH_2Cl_3$		-110.6	(1-2) 8	398

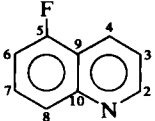
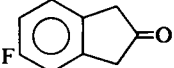
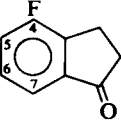
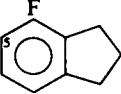
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6FH_4Cl_3Sn$	$mF \cdot C_6H_4 \cdot SnCl_3$	-118.36		399(d), 400
$C_6FH_4Cl_3Sn$	<i>p</i> -isomer of above	-122.92		399(d), 400
C_6FH_4NO	$mF \cdot C_6H_4 \cdot NO$	-111.24		401
C_6FH_4NO	<i>p</i> -isomer of above	-102.32		401
$C_6FH_4N_2^+$	$pF \cdot C_6H_4 \cdot N_2^+$	-82.8	(F-C4) 271.9 (F-C3) 25.2 (F-C2) 13.0	502
$C_6FH_4N_3$	$mF \cdot C_6H_4 \cdot N_3$	-111.81		401
$C_6FH_4N_3$	<i>p</i> -isomer of above	-118.56		401
C_6FH_5			(1-2) 9.176 (1-3) 5.759 (1-4) 0.348 (1-C1) -244.70 (1-C2) 20.98 (1-C3) 7.81 (1-C4) 3.18	402 403
C_6FH_5	As above		(1-C1) -245.07 (1-C2) 21.02 (1-C3) 7.79 (1-C4) 3.20	404
C_6FH_5S	$mF \cdot C_6H_4 \cdot SH$	-113.02		405(d)
C_6FH_5S	<i>p</i> -isomer of above	-118.16		405(d)
C_6FH_6NO	$mF \cdot C_6H_4 \cdot NH \cdot OH$	-113.56		401
C_6FH_6NO	<i>p</i> -isomer of above	-124.94		401
C_6FBr_5	C_6FBr_5	-80.3		396
$C_6F_3H_4NS$	$mF^{(1)} \cdot C_6H_4 \cdot SF_2^{(2)}N$	(1) -108.3 (2) +109.0		414
$C_6F_3H_4NS$	<i>p</i> -isomer of above	(1) -100.3 (2) +103.4		414

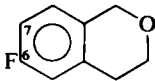
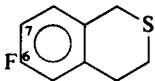
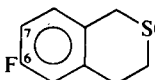
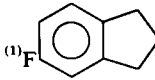
$C_7FH_3Cl_4$	⁽²⁾ CH ₃ Cl ⁽¹⁾ F 	-110	(1-2) 2·5	398
$C_7FH_3Cl_4$	Cl CH ₃ F 	-107		398
$C_7FH_3Cl_4$	Cl Cl F 	-108·3		398
$C_7FH_5Cl_2$	⁽²⁾ ⁽³⁾ ⁽¹⁾ F 		(1-4) 0·22	420
C_7FH_6Br	As above with CHCl ₂ replaced by CH ₂ Br		(1-4) 0·545	420
C_7FH_6Cl	As above with CHCl ₂ replaced by CH ₂ Cl		(1-2) 8·416 (1-3) 5·204 (1-4) 0·652	420
C_7FH_6Cl	⁽²⁾ CH ₃ ⁽³⁾ ⁽¹⁾ F 		(1-2) 8·858 (1-3) -0·226 (1-5) 5·169 (1-6) 7·735	421
$C_7FH_6NO_3$	CH ₃ O F 	-121·9		407
$C_7FH_6NO_3$	OCH ₃ F 	-113·4		407

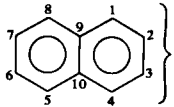
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7FH_6NO_3$		-101.0		407
C_7FH_7			(1-2) 9.732 (1-3) -0.253 (1-4) 0.144 (1-5) 6.008 (16) 8.529	421
C_7FH_7NO	$mF \cdot C_6H_4 \cdot NH \cdot CO \cdot NH_2$	-113.05		401
C_7FH_7NO	$pF \cdot C_6H_4 \cdot NH \cdot CO \cdot NH_2$	-122.04		401
$C_7FH_7N_4O_2$	$mF \cdot C_6H_4 \cdot N = C(NH_2) \cdot NH \cdot NO_2$	-112.01		422(d)
$C_7FH_7N_4O_2$	$pF \cdot C_6H_4 \cdot N = C(NH_2) \cdot NH \cdot NO_2$	-117.09		422(d)
C_7FH_7O		-119.39	(1-2) 5.301 (1-3) -0.203 (1-4) 7.477 (1-5) 1.233 (1-6) -0.78 (1-C5) 10.050	423(n)
$C_7FH_7O_2S$	$mF \cdot C_6H_4 \cdot SO_2CH_3$	-110.19		405
$C_7FH_7O_2S$	$pF \cdot C_6H_4 \cdot SO_2CH_3$	-105.35		405
C_7FH_7S	$mF \cdot C_6H_4 \cdot SCH_3$	-112.88		405
C_7FH_7S	$pF \cdot C_6H_4 \cdot SCH_3$	-118.01		405
$C_7FH_8N_3$	$mF \cdot C_6H_4 \cdot N = C(NH_2)_2$	-113.82		422(d)
$C_7FH_8N_3$	$pF \cdot C_6H_4 \cdot N = C(NH_2)_2$	-124.68		422(d)
$C_7FH_9N_4O_3$	$mF \cdot C_6H_4 \cdot N = C(NH_2)_2 \cdot HNO_3$	-111.34		422(d)
$C_7FH_9N_4O_3$	$pF \cdot C_6H_4 \cdot N = C(NH_2)_2 \cdot HNO_3$	-115.42		422(d)
C_8FH_6N	$pF^{(1)} \cdot C_6H_4 \cdot CH_2^{(2)}CN$		(1-2) 1.06	420

C_8FH_7			(F-C1) 12·4 (F-C2) 252·4 (F-C3) 21·3 (F-C4) 3·6 (F-C5) 3·5 (F-C6) 9·0 (F-C7) 4·7 (F-C8) 5·3	431
C_8FH_7			(F-C1) 8·3 (F-C2) 21·8 (F-C3) 248·2 (F-C4) 21·3 (F-C5) 8·3 (F-C6) 2·6 (F-C7) 2·4	431
C_8FH_7			(F-C1) 3·3 (F-C2) 8·1 (F-C3) 21·1 (F-C4) 249·7 (F-C8) 1·9	431
C_8FH_7		-123·32		360(d)
C_8FH_7O		-116·54		360(d)
$C_8FH_7O_2S$		-113·04		360(d)
$C_8FH_7O_2S$		-115·17		360(d)
C_8FH_7S		-117·60		360(d)
C_8FH_8N	As above with S replaced by NH	-117·50		360(d)
$C_8FH_8NO_2$	$pF \cdot C_6H_4 \cdot NH \cdot CO_2CH_3$	-121·1		401

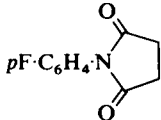
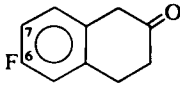
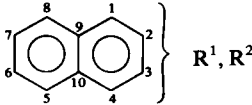
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8FH_8N_3O_2$	$mF \cdot C_6H_4 \cdot NH \cdot CO \cdot NH \cdot CO \cdot NH_2$	-111.68		401
$C_8FH_8N_3O_2$	<i>p</i> -isomer of above	-120.22		401
C_8FH_9		-118.16		360(d)
$C_8FH_9Cl_2Sn$	$pF \cdot C_6H_4 \cdot SnCl_2 \cdot CH_2 \cdot CH_3$	-119.44		400(d)
$C_8FH_9N_2O$	$mF \cdot C_6H_4 \cdot NH \cdot CO \cdot NH \cdot CH_3$	-112.76		422(d)
$C_8FH_9N_2O$	<i>p</i> -isomer of above	-122.51		422(d)
$C_8FH_{10}NO$		-133.2		432(b)
$C_8FH_{10}NO_2$		-131.9	(1-2) ± 10.8 (1-3) ∓ 1.6	432(b)
$C_8FH_{10}NO_2$		-125.9	(1-2) 11.0 (1-3) 7.6	432(b)
$C_8FH_{10}NO_2$		-136.6	(1-2) ± 8.2 (1-3) ∓ 1.8	432(b)
$C_8FH_{10}N_3$	$mF \cdot C_6H_4 \cdot CH_2 \cdot N = C(NH_2)_2$	-114.4		422(d)
$C_8FH_{10}N_3$	<i>p</i> -isomer of above	-117.33		422(d)

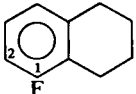
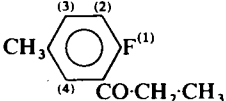
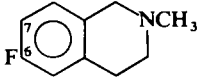
$C_8FH_{11}N_4O_3$	$mF \cdot C_6H_4 \cdot CH_2 \cdot N=C(NH_2)_2 \cdot HNO_3$	-113.15		422(d)
$C_8FH_{11}N_4O_3$	<i>p</i> -isomer of above	-115.14		422(d)
C_9FH_6N			(F-C3) 3.5 (F-C4) 4.7 (F-C5) 250 (F-C6) 19.2 (F-C7) 9.1 (F-C8) 2.4 (F-C9) 2.5 (F-C10) 16.4	431
C_9FH_6N	As above with F at C6		(F-C4) 5.3 (F-C5) 21.8 (F-C6) 250.1 (F-C7) 26.5 (F-C8) 9.1 (F-C9) 2.3 (F-C10) 10.0	431
C_9FH_6N	As above with F at C7		(F-C4) 2.7 (F-C5) 4.8 (F-C6) 7.5 (F-C7) 19.3 (F-C8) 259.9 (F-C9) 11.9 (F-C10) 2.3	431
C_9FH_7O		-115.38		360(d)
C_9FH_7O		-119.73		360(d)
C_9FH_7O	As above with F at C5	-104.66		360(d)
C_9FH_7O	As above with F at C6	-115.60		360(d)
C_9FH_9		-118.57		360(d)
C_9FH_9	As above with F at C5	-115.5		434

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₉ FH ₉ O		-116.87		360(d)
C ₉ FH ₉ O	As above with F at C7	-116.87		360(d)
C ₉ FH ₉ S		-116.54		360(d)
C ₉ FH ₉ S	As above with F at C7	-117.18		360(d)
C ₉ FH ₉ SO ₂		-114.59		360(d)
C ₉ FH ₉ SO ₂	As above with F at C7	-116.31		360(d)
C ₉ FH ₁₀ NO	<i>p</i> F·C ₆ H ₄ ·N=CH·O·CH ₂ CH ₃	-120.4		401
C ₉ FH ₁₁	<i>p</i> F ⁽¹⁾ ·C ₆ H ₄ ·CH ⁽⁴⁾ (CH ₃) ₂		(1-4) 0.56	420
C ₉ FH ₁₁ N ₂	<i>p</i> F·C ₆ H ₄ ·N=CH·N(CH ₃) ₂	-123.9		401
C ₉ FH ₁₁ O ₂ Si	<i>p</i> F·C ₆ H ₄ ·Si(CH ₃) ₂ O·CO·H	-116.81		435(d)
C ₉ FH ₁₃ IP	<i>m</i> F·C ₆ H ₄ ·P ⁺ (CH ₃) ₃ I ⁻	-110.95		405
C ₉ FH ₁₃ IP	<i>p</i> -isomer of above	-104.95		405
C ₉ FH ₁₃ Sn	<i>m</i> FC ₆ H ₄ ·Sn(CH ₃) ₃	-113.78		399(d)
C ₉ FH ₁₃ Sn	<i>p</i> -isomer of above	-113.38	(¹¹⁹ Sn-F) 9 (¹¹⁷ Sn-F) 9	399(d), 436
C ₉ F ₃ H ₇		(1) -115.26		360(d)

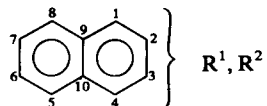
$C_9F_3H_9$	$mF \cdot C_6H_4 \cdot CH_2 \cdot CF_2 \cdot CH_3$	0.38 relative to <i>m</i> -fluoroethylbenzene	360(d)
$C_9F_3H_9$	<i>p</i> -isomer of above	2.40 relative to <i>p</i> -fluoroethylbenzene	360(d)
$C_{10}FH_6Br$	 R^1, R^2		405
	1-F, 5-Br	-121.12	(p)
	2-F, 5-Br	-116.09	(q)
$C_{10}FH_6Br$	As above with 2-F, 3-Br	-110.6	(F-H) 7.2, 9.3
$C_{10}FH_6Cl$	As above with 1-F, 4-Cl	-122.81	405(p)
$C_{10}FH_6Cl$	As above with 2-F, 6-Cl	-116.09	405(q)
$C_{10}FH_6Cl$	As above with 2-F, 7-Cl	-115.26	405(q)
$C_{10}FH_6Cl$	As above with 2-F, 8-Cl	-113.80	405(q)
$C_{10}FH_6I$	As above with 1-F, 4-I	-121.92	405(p)
$C_{10}FH_6I$	As above with 2-F, 6-I	-115.49	405(q)
$C_{10}FH_6I$	As above with 2-F, 7-I	-115.19	405(q)
$C_{10}FH_6I$	As above with 2-F, 8-I	-113.95	405(q)
$C_{10}FH_6NO_2$	As above with 1-F, 5-NO ₂	-119.53	405(p)
$C_{10}FH_6NO_2$	As above with 2-F, 5-NO ₂	-114.66	405(q)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}FH_7$	As above with 1-F, H		(1-2) +11.0 (1-3) +5.6 (1-4) 0.3 (1-5) +1.8 (1-7) 0.4 (1-8) 0.5 (1-C1) -251.18 (1-C2) +19.81 (1-C3) +8.36 (1-C4) +4.10 (1-C5) +3.18 (1-C6) 0.94 (1-C7) 1.82 (1-C8) +5.14 (1-C9) +16.40 (1-C10) +4.78	404
$C_{10}FH_7$	As above with 1-F, H		(1-2) 10.72 (1-3) 5.35 (1-4) 0 (1-C1) 251.5 (1-C2) 19.9 (1-C3) 8.3 (1-C4) 4.0 (1-C5) 3.3 (1-C6) ~0 (1-C7) 1.7 (1-C8) 5.2 (1-C9) 16.5 (1-C10) 4.9	440
$C_{10}FH_7$	As above with 1-F, H		Assignments wrong; see ref. 440	431
$C_{10}FH_7$	As above with 2-F, H		(2-C1) 20.2 (2-C2) 245.7 (2-C3) 25.3 (2-C4) 9.0 (2-C5) 1.2 (2-C6) 2.5 (2-C7) 0.9 (2-C8) 5.4 (2-C9) 9.1 (2-C10) 0.7	440
$C_{10}FH_7$	As above with 2-F, H		(1-2) +10.0 (2-3) +8.9 (2-4) +5.5 (2-C1) +20.24 (2-C2) -246.02 (2-C3) +25.26 (2-C4) +8.93	404

			(2-C5) 1·22 (2-C6) 2·63 (2-C7) 0·79 (2-C8) +5·39 (2-C9) +9·10 (2-C10) 0·98	
$C_{10}FH_7$	As above with 2-F, H		Assignments wrong; see ref. 440	431
$C_{10}FH_7S$	As above with 1-F, 4-SH	-127·30		405(p)
$C_{10}FH_7S$	As above with 2-F, 6-SH	-117·75		405(q)
$C_{10}FH_7S$	As above with 2-F, 7-SH	-116·05		405(q)
$C_{10}FH_8N$	As above with 1-F, 5-NH ₂	-121·91		405(p)
$C_{10}FH_8N$	As above with 2-F, 5-NH ₂	-118·78		405(q)
$C_{10}FH_8NO_2$		-114·2		401
$C_{10}FH_9O$		-116·88		360(d)
$C_{10}FH_9O$	As above with F at C7	-116·97		360(d)
$C_{10}FH_9O_2$	$pF \cdot C_6H_4 \cdot CO \cdot CH_2 \cdot CO \cdot CH_3$	-108·56		224(a)
$C_{10}FH_9Si$				441(q)
	2-F, 6-SiH ₃	-118·77		
	2-F, 7-SiH ₃	-117·84		

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}FH_{11}$		-119.27		360(d)
$C_{10}FH_{11}$	As above	-120.75		434
$C_{10}FH_{11}$	As above with F at C2	-121.8		434
$C_{10}FH_{11}O$			(1-2) 11.02 (1-3) 4.61 (1-4) 6.97	327
$C_{10}FH_{12}ClO_2Si$	$pF \cdot C_6H_4 \cdot Si(CH_3)_2 \cdot O \cdot CO \cdot CH_2Cl$	-117.04		435(d)
$C_{10}FH_{12}N$		-117.39		360(d)
$C_{10}FH_{12}N$	As above with F at C7	-117.91		360(d)
$C_{10}FH_{12}S^+$	As above with N replaced by S^+	-112.20		360(d)
$C_{10}FH_{12}S^+$	As above with N replaced by S^+ and F at C7	-115.81		360(d)
$C_{10}FH_{13}O_2Si$	$pF \cdot C_6H_4 \cdot Si(CH_3)_2O \cdot CO \cdot CH_3$	-116.23		435(d)
$C_{10}FH_{14}ClSn$	$pF \cdot C_6H_4 \cdot SnCl(CH_2CH_3)_2$	-116.14		400(d)
$C_{10}FH_{15}Pb$	$mF \cdot C_6H_4 \cdot CH_2 \cdot Pb(CH_3)_3$	-114.04		442(d)
$C_{10}FH_{15}Pb$	<i>p</i> -isomer of above	-120.70		442(d)
$C_{10}FH_{15}Sn$	$pF \cdot C_6H_4 \cdot CH_2 \cdot Sn(CH_3)_3$	-121.08	$(^{119}Sn-F)$ 24.5 $(^{117}Sn-F)$ 23.5	436

$C_{10}F_2H_6$



1-F, 5-F	(1) -120·85	405(p)
1-F, 6-F	(1) -121·78	
1-F, 7-F	(1) -122·40	
2-F, 5-F	(2) -115·26	405(q)
2-F, 6-F	(2) -118·04	
2-F, 7-F	(2) -115·69	
2-F, 8-F	(2) -114·81	

$C_{10}F_2H_6$

As above with 1-F, 8-F

-115·814	(1-8) 58·8 (1-C1) -255·4	408
	(1-C8) -1·6	

$C_{10}F_2H_6$

As above with 1-F, 5-F

1-F, 6-F

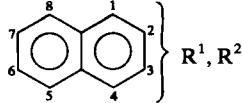
(1-C1) 254·5 (1-C3) 4·8	431
(1-C4) 4·8 (1-C5) 5·6	

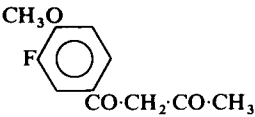
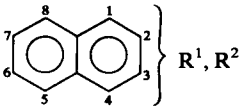
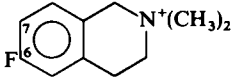
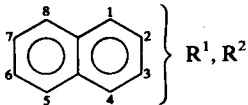
(1-C1) 256·9 (1-C2) 20·0	
(1-C3) 17·7 (1-C4) 8·3	
(1-C7) 2·3 (1-C9) 9·4	
(6-C2) 2·6 (6-C5) 26·2	
(6-C6) 252·3 (6-C7) 21·0	
(6-C9) 4·4	

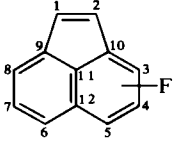
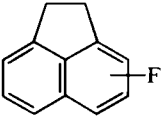
Assignments may be in error;
see ref. 440

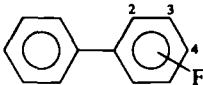
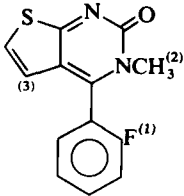
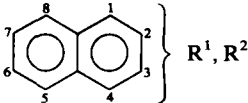
1-F, 7-F

(1-C1) 250·7 (1-C2) 19·6	
(1-C3) 9·1 (1-C4) 4·1	
(1-C7) 6·1 (1-C8) 6·0	
(1-C9) 8·8 (1-C10) 16·7	
(7-C3) 3·0 (7-C5) 4·9	
(7-C6) 25·6 (7-C7) 255·3	
(7-C8) 18·2 (7-C9) 3·0	
(7-C10) 9·1	

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
	2-F, 6-F		(2-6) 7 (2-C1) 25.4 (2-C2) 246.9 (2-C3) 19.9 (2-C4) 6.1 (2-C6) 3.1 (2-C8) 6.1 (2-C9) 4.2 (2-C10) 4.2 Assignments may be in error; see ref. 440	
	2-F, 7-F		(2-C1) 25.6 (2-C2) 258.8 (2-C3) 21.6 (2-C4) 9.7 (2-C6) 5.2 (2-C8) 2.4 (2-C10) 10 Assignments may be in error; see ref. 440	
C ₁₁ FH ₆ N				405
	1-F, 5-CN	-120.59		(p)
	2-F, 5-CN	-116.84		(q)
C ₁₁ FH ₇ O ₂	As above with 1-F, 5-CO ₂ H 2-F, 5-CO ₂ H	-121.20 -116.84		405(p) (q)
C ₁₁ FH ₉	As above with 1-F, 3-CH ₃ 1-F, 5-CH ₃ 1-F, 7-CH ₃ 2-F, 4-CH ₃ 2-F, 5-CH ₃	-123.96 -121.97 -123.60 -117.68 -118.09		405(p) (q)
C ₁₁ FH ₉ O ₂ S	As above with 1-F, 4-SO ₂ CH ₃ 2-F, 6-SO ₂ CH ₃ 2-F, 7-SO ₂ CH ₃	-111.88 -111.78 -114.18		405(p) (q)

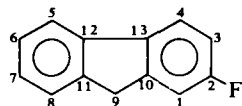
$C_{11}FH_9S$	As above with 1-F, 4-SCH ₃ 2-F, 6-SCH ₃ 2-F, 7-SCH ₃	-125·87 -118·01 -116·28	405(p) (q)
$C_{11}FH_{11}O_3$		-137·5	224(a)
$C_{11}FH_{11}Si$	 R^1, R^2		441(q)
	2-F, 6-SiH ₂ CH ₃ 2-F, 7-SiH ₂ CH ₃	-118·40 -117·60	
$C_{11}FH_{15}N^+$		-113·88	360(d)
$C_{11}FH_{15}N^+$	As above with F at C7	-115·33	360(d)
$C_{11}FH_{16}N_3$	$mF \cdot C_6H_4 \cdot NH \cdot C(NH_2) = N \cdot C(CH_3)_3$	-114·17	422(d)
$C_{11}FH_{16}N_3$	<i>p</i> -isomer of above	-125·27	422(d)
$C_{11}FH_{17}N_4O_3$	$mF \cdot C_6H_4 \cdot NH \cdot C(NH_2) = N \cdot C(CH_3)_3 \cdot HNO_3$	-111·87	422(d)
$C_{11}FH_{17}N_4O_3$	<i>p</i> -isomer of above	-115·35	422(d)
$C_{11}F_4H_6$	 R^1, R^2		405
	1-F, 3-CF ₃ 1-F, 4-CF ₃ 1-F, 5-CF ₃ 1-F, 7-CF ₃	(1) -119·30 (1) -114·72 (1) -119·97 (1) -121·29	(p)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
	2-F, 4-CF ₃	(2) -116.91		(q)
	2-F, 5-CF ₃	(2) -115.33		
	2-F, 6-CF ₃	(2) -113.01		
	2-F, 7-CF ₃	(2) -114.63		
	2-F, 8-CF ₃	(2) -112.28		
C ₁₂ FH ₇		F at C3	(3-C1) 2.44 (3-C2) 2.36 (3-C3) -263.34 (3-C4) +25.78 (3-C5) +8.05 (3-C6) 0.52 (3-C7) 2.02 (3-C8) 1.21 (3-C9) +4.20 (3-C10) +14.37 (3-C11) +10.14 (3-C12) 1.00	404
		F at C4	(4-C1) 0.54 (4-C2) 2.92 (4-C3) +27.71 (4-C4) -245.0 (4-C5) +22.75 (4-C6) +5.40 (4-C7) 0.61 (4-C8) 2.12 (4-C9) 1.01 (4-C10) +9.79 (4-C11) <0.15 (4-C12) +10.25	
		F at C5	(5-C1) 2.87 (5-C2) 0.44 (5-C3) +8.09 (5-C4) +21.54 (5-C5) -259.9 (5-C6) 1.56 (5-C7) 1.11 (5-C8) 0.92 (5-C9) 2.23 (5-C10) +3.95 (5-C11) +6.45 (5-C12) +19.34	
C ₁₂ FH ₉	 (Numbering as above)	F at C3	(3-C1) 0.57 (3-C2) 1.61 (3-C3) -245.48 (3-C4) +25.12 (3-C5) +6.93 (3-C6) 1.10 (3-C7) 2.76 (3-C8) 0.58 (3-C9) +4.98 (3-C10) +17.22 (3-C11) +8.26 (3-C12) 1.48	404

		F at C4	(4-C1) 0·39 (4-C3) +27·24 (4-C5) +22·70 (4-C7) 0·74 (4-C9) 1·33 (4-C11) <0·12	(4-C2) 2·53 (4-C4) -245·5 (4-C6) +5·70 (4-C8) 2·42 (4-C10) +10·18 (4-C12) +10·77	
		F at C5	(5-C1) <0·3 (5-C3) +7·43 (5-C5) -249·64 (5-C7) 1·47 (5-C9) +3·14 (5-C11) +6·94	(5-C2) 29·92 (5-C4) +20·46 (5-C6) 0·90 (5-C8) 1·46 (5-C10) +3·66 (5-C12) +19·18	
C ₁₂ FH ₉		F at C2 -118·2 F at C3 -113·3 F at C4 -116·1			448
C ₁₂ FH ₉ N ₂ OS				(1-2) 1·2 (1-3) 0·6	449
C ₁₂ FH ₁₂ N		-121·81			405
C ₁₂ FH ₁₂ S	As above with 1-F, 3-S(CH ₃) ₂ ⁺ 2-F, 6-S(CH ₃) ₂ ⁺ 2-F, 7-S(CH ₃) ₂ ⁺	-112·09 -111·00 -113·59			405(p) (q)
C ₁₂ FH ₁₃ Si	As above with 2-F, 6-SiH(CH ₃) ₂ 2-F, 7-SiH(CH ₃) ₂	-118·05 -117·38			441(q)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₁₂ FH ₁₉ Sn	<i>o</i> F·C ₆ H ₄ ·Sn(CH ₂ CH ₃) ₃	−93·38	(¹¹⁹ Sn-F) 34 (¹¹⁷ Sn-F) 33	436
C ₁₂ FH ₁₉ Sn	<i>m</i> -isomer of above	−112·41		400, 399(d)
C ₁₂ FH ₁₉ Sn	<i>p</i> -isomer of above	−112·62		400, 399(d)
C ₁₂ F ₂ H ₈ Cl ₂ Sn	(<i>m</i> F·C ₆ H ₄) ₂ SnCl ₂	−116·94		400, 399(d)
C ₁₂ F ₂ H ₈ Cl ₂ Sn	<i>p</i> -isomer of above	−116·14		400, 399(d)
C ₁₂ F ₂ H ₈ N ₂	<i>m</i> F·C ₆ H ₄ ·N=N·C ₆ H ₄ ·F <i>m</i> (<i>cis</i>) (<i>trans</i>)	−111·05 −112·04		401
C ₁₂ F ₂ H ₈ N ₂	<i>p</i> F·C ₆ H ₄ ·N=N·C ₆ H ₄ ·F <i>p</i> (<i>cis</i>) (<i>trans</i>)	−112·25 −109·71		401
C ₁₂ F ₂ H ₈ N ₂	<i>m</i> F ⁽¹⁾ ·C ₆ H ₄ ·N=N·C ₆ H ₄ ·F ⁽²⁾ <i>m</i> 	(1) −111·72 (2) −110·87		401
C ₁₂ F ₂ H ₈ N ₂ O	<i>p</i> F·C ₆ H ₄ ·N=N·C ₆ H ₄ ·F <i>p</i> 	−109·26, −108·93		401
C ₁₂ F ₂ H ₈ N ₂ O	(<i>m</i> F·C ₆ H ₄) ₂ N·NO	−111·42, −110·98		401
C ₁₂ F ₂ H ₈ N ₂ O	<i>p</i> -isomer of above	−115·79, −111·66		401
C ₁₂ F ₂ H ₉ N	(<i>m</i> F·C ₆ H ₄) ₂ NH	−111·77		401
C ₁₂ F ₂ H ₉ N	<i>p</i> -isomer of above	−122·45		401
C ₁₂ F ₂ H ₁₀ N ₂	(<i>p</i> F·C ₆ H ₄ ·NH) ₂	−126·9		401
C ₁₂ F ₂ H ₁₀ N ₂	(<i>p</i> F·C ₆ H ₄) ₂ N·NH ₂	−121·6		401
C ₁₂ F ₅ H ₆ N	<i>o</i> F·C ₆ H ₄ ·NH 	−130·0, −140·5, −157·1, −159·6, −168·5		451

C₁₃FH₉



(F-C1) +22.88 (F-C2) -245.0 404
 (F-C3) +23.00 (F-C4) +8.84
 (F-C5) 1.00 (F-C6) 0.50
 (F-C7) 1.01 (F-C8) 0.48
 (F-C9) +2.50 (F-C10) +8.68
 (F-C11) +1.99 (F-C12) 0.52
 (F-C13) +2.28

C₁₃FH₉BrCl



-113.32

454(b)

R = Br, X = Cl

C₁₃FH₉BrNO₂

As above with R = Br, X = NO₂

-112.35

454(b)

C₁₃FH₉ClNO₂

As above with R = Cl, X = NO₂

-112.41

454(b)

C₁₃FH₉Cl₂

As above with R = Cl, X = Cl

-113.45

454(b)

C₁₃FH₁₀Br

As above with R = Br, X = H

-113.96

454(b)

C₁₃FH₁₀Cl

As above with R = Cl, X = H

-114.09

454(b)

C₁₃FH₁₀Cl

As above with R = H, X = Cl

-116.52

454(b)

C₁₃FH₁₀ClO

As above with R = OH, X = Cl

-114.47

454(b)

C₁₃FH₁₀N

*p*F·C₆H₄·N=CH·C₆H₅

-117.5

401

C₁₃FH₁₀NO

*p*F·C₆H₄·NH·CO·C₆H₅

-119.1

401

C₁₃FH₁₀NO₂

As C₁₃FH₉BrCl above with
R = H, X = NO₂

-115.50

454(b)

C₁₃FH₁₀NO₃

As C₁₃FH₉BrCl above with
R = OH, X = NO₂

-113.12

454(b)

C₁₃FH₁₁

As C₁₃FH₉BrCl above with
R = H, X = H

-117.17

454(b)

C₁₃FH₁₁N₂O

*m*F·C₆H₄·NH·CO·NH·C₆H₅

-112.90

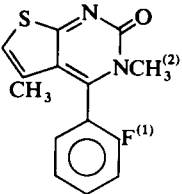
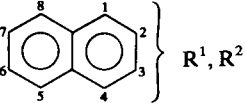
401

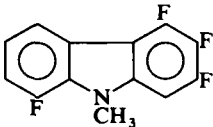
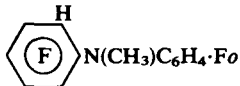
C₁₃FH₁₁N₂O

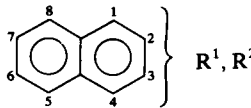
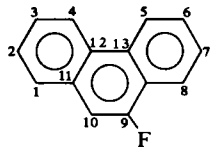
As above

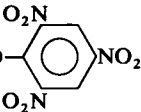
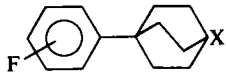
-112.33

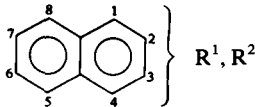
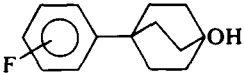
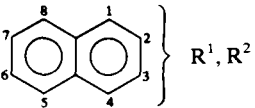
422(d)

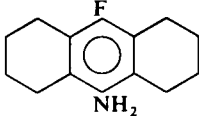
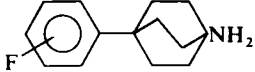
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{13}FH_{11}N_2O$	$pF \cdot C_6H_4 \cdot NH \cdot CO \cdot NH \cdot C_6H_5$	-121.95		401
$C_{13}FH_{11}N_2O$	As above	-121.48		422(d)
$C_{13}FH_{11}N_2OS$			(1-2) 0.5	449
$C_{13}FH_{11}O$	$pF \cdot C_6H_4 \cdot CH(OH) \cdot C_6H_5$	-115.30		454(d)
$C_{13}FH_{12}N_3$	$pF \cdot C_6H_4 \cdot N=N \cdot N(CH_3)C_6H_5$	-117.0		401
$C_{13}FH_{12}N_3$	$mF \cdot C_6H_4 \cdot N=C(NH_2) \cdot NH \cdot C_6H_5$	-113.15		422(d)
$C_{13}FH_{12}N_3$	p -isomer of above	-123.85		422(d)
$C_{13}FH_{13}N_4O_3$	$mF \cdot C_6H_4 \cdot N=C(NH_2)NH \cdot C_6H_5 \cdot HNO_3$	-111.59		422(d)
$C_{13}FH_{13}N_4O_3$	p -isomer of above	-115.90		422(d)
$C_{13}FH_{15}N^+$	 1-F, 4-N(CH ₃) ₃ ⁺ 1-F, 5-N(CH ₃) ₃ ⁺ 2-F, 6-N(CH ₃) ₃ ⁺ 2-F, 7-N(CH ₃) ₃ ⁺	-116.56 -116.98 -113.60 -114.31		(p) (q)
$C_{13}FH_{15}P^+$	As above with 1-F, 4-P(CH ₃) ₃ ⁺ 2-F, 6-P(CH ₃) ₃ ⁺ 2-F, 7-P(CH ₃) ₃ ⁺	-111.75 -111.45 -114.11		405(p) (q)

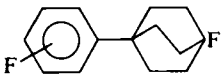
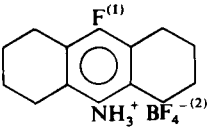
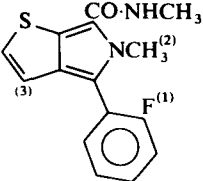
C ₁₃ FH ₁₅ Si	As above with 1-F, 5-Si(CH ₃) ₃	-121.47	405(p)
	2-F, 5-Si(CH ₃) ₃	-117.92	(q)
	1-F, 7-Si(CH ₃) ₃	-123.23	(p)
	2-F, 8-Si(CH ₃) ₃	-115.78	(q)
C ₁₃ FH ₁₅ Si	As above with 2-F, 6-Si(CH ₃) ₃	-117.77	441(q)
	2-F, 7-Si(CH ₃) ₃	-117.17	
C ₁₃ F ₂ H ₈ N ₂	(<i>m</i> F·C ₆ H ₄ ·N=) ₂ C	-111.51	401
C ₁₃ F ₂ H ₈ N ₂	<i>p</i> -isomer of above	-117.12	401
C ₁₃ F ₂ H ₉ Br	(<i>p</i> F·C ₆ H ₄) ₂ CHBr	-113.53	454(b)
C ₁₃ F ₂ H ₉ Cl	As above with Br replaced by Cl	-113.65	454(b)
C ₁₃ F ₂ H ₁₀	As above with Br replaced by H	-116.73	454(b)
C ₁₃ F ₂ H ₁₀ N ₂ O	(<i>m</i> F·C ₆ H ₄ ·NH) ₂ CO	-112.87	401
C ₁₃ F ₂ H ₁₀ N ₂ O	<i>p</i> -isomer of above	-120.98	401
C ₁₃ F ₂ H ₁₀ O	(<i>p</i> F·C ₆ H ₄) ₂ CH·OH	-114.75	454(b)
C ₁₃ F ₄ H ₇ N		-136.6, -143.3, -173.8 (1:1:2)	451
C ₁₃ F ₄ H ₈	<i>p</i> F·C ₆ H ₄ ·C ₆ H ₄ ·CF ₃ <i>m</i>	+1.35 relative to compound with CF ₃ replaced by H	405
C ₁₃ F ₄ H ₈	<i>p</i> F·C ₆ H ₄ ·C ₆ H ₄ ·CF ₃ <i>p</i>	+1.65 relative to compound with CF ₃ replaced by H	405
C ₁₃ F ₅ H ₈ N		-118.7, -137.3, -146.2. -157.1, -166.5	451

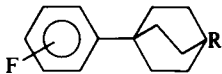
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}FH_6N_3$	 1-F, 4-C(CN) ₃ 2-F, 6-C(CN) ₃ 2-F, 7-C(CN) ₃	-114.26 -111.78 -113.11		457 (p) (q)
$C_{14}FH_8Br$	$pF \cdot C_6H_4 \cdot C \equiv C \cdot C_6H_4 \cdot Xp$ X=Br	-110.79		454(b)
$C_{14}FH_8Cl$	As above with X = Cl	-110.85		454(b)
$C_{14}FH_8NO_2$	As above with X = NO ₂	-109.25		454(b)
$C_{14}FH_9$	As above with X = H	-111.49		454(b)
$C_{14}FH_9$			(9-C1) +5.50 (9-C2) 0.71 (9-C3) 2.42 (9-C4) 1.41 (9-C5) +3.19 (9-C6) 0.52 (9-C7) 1.40 (9-C8) +6.29 (9-C9) -251.63 (9-C10) +19.81 (9-C11) +9.88 (9-C12) 1.09 (9-C13) +5.68 (9-C14) +18.79	404
$C_{14}FH_{10}Br$	$pF \cdot C_6H_4 \cdot CH=CH \cdot C_6H_4 \cdot Brp$	<i>cis</i> -113.77 <i>trans</i> -113.54		454(b)
$C_{14}FH_{10}Br_2Cl$	$pF \cdot C_6H_4 \cdot CHBr \cdot CHBr \cdot C_6H_4 \cdot Clp$	<i>erythro</i> -111.94 <i>threo</i> -111.99		454(b)
$C_{14}FH_{10}Br_2NO_2$	$pF \cdot C_6H_4 \cdot CHBr \cdot CHBr \cdot C_6H_4 \cdot NO_2p$	<i>erythro</i> -111.44 <i>threo</i> -111.08		454(b)
$C_{14}FH_{10}Br_3$	$pF \cdot C_6H_4 \cdot CHBr \cdot CHBr \cdot C_6H_4 \cdot Brp$	<i>erythro</i> -111.92 <i>threo</i> -111.90		454(b)
$C_{14}FH_{10}Cl$	$pF \cdot C_6H_4 \cdot CH=CH \cdot C_6H_4 \cdot Clp$	<i>cis</i> -113.83 <i>trans</i> -113.62		454(b)

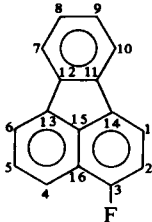
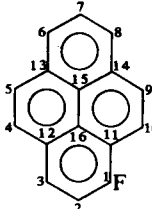
C ₁₄ FH ₁₀ NO ₂	<i>p</i> F·C ₆ H ₄ ·CH=CH·C ₆ H ₄ ·NO ₂ <i>p</i>	<i>cis</i> -112·55 <i>trans</i> -111·87	454(b)
C ₁₄ FH ₁₁	<i>p</i> F·C ₆ H ₄ ·CH=CH·C ₆ H ₅	<i>cis</i> -114·51 <i>trans</i> -114·28	454(b)
C ₁₄ FH ₁₁ Br ₂	<i>p</i> F·C ₆ H ₄ ·CHBr·CHBr·C ₆ H ₅	<i>erythro</i> -112·36 <i>threo</i> -112·55	454(b)
C ₁₄ FH ₁₂ Br	<i>p</i> F·C ₆ H ₄ ·CHBr·C ₆ H ₄ ·CH ₃ <i>p</i>	-114·18	454(b)
C ₁₄ FH ₁₂ BrO	<i>p</i> F·C ₆ H ₄ ·CHBr·C ₆ H ₄ ·OCH ₃ <i>p</i>	-114·27	454(b)
C ₁₄ FH ₁₂ Cl	<i>p</i> F·C ₆ H ₄ ·CHCl·C ₆ H ₄ ·CH ₃ <i>p</i>	-114·33	454(b)
C ₁₄ FH ₁₂ Cl	<i>p</i> F·C ₆ H ₄ ·CH ₂ ·CH ₂ ·C ₆ H ₄ ·Cl <i>p</i>	-116·86	454(b)
C ₁₄ FH ₁₂ ClO	<i>p</i> F·C ₆ H ₄ ·CHCl·C ₆ H ₄ ·OCH ₃ <i>p</i>	-114·42	454(b)
C ₁₄ FH ₁₂ N	<i>p</i> F·C ₆ H ₄ ·CH=CH·C ₆ H ₄ ·NH ₂ <i>p</i>	<i>cis</i> -115·35 <i>trans</i> -115·63	454(b)
C ₁₄ FH ₁₂ N ₃ O ₇ S	<i>m</i> F·C ₆ H ₄ ·S(CH ₃) ₂ ⁺ ·O- 	-108·18	422(d)
C ₁₄ FH ₁₂ N ₃ O ₇ S	<i>p</i> -isomer of above	-104·56	422(d)
C ₁₄ FH ₁₃	<i>p</i> F·C ₆ H ₄ ·CH ₂ ·CH ₂ ·C ₆ H ₅	-117·32	454(b)
C ₁₄ FH ₁₃	<i>p</i> F·C ₆ H ₄ ·CH ₂ ·C ₆ H ₄ ·CH ₃ <i>p</i>	-117·39	454(b)
C ₁₄ FH ₁₃ O	<i>p</i> F·C ₆ H ₄ ·CH ₂ ·C ₆ H ₄ ·OCH ₃ <i>p</i>	-117·43	454(b)
C ₁₄ FH ₁₃ O	<i>p</i> F·C ₆ H ₄ ·CH(OH)·C ₆ H ₄ ·CH ₃ <i>p</i>	-115·61	454(b)
C ₁₄ FH ₁₃ O ₂	<i>p</i> F·C ₆ H ₄ ·CH(OH)·C ₆ H ₄ ·OCH ₃ <i>p</i>	-115·73	454(b)
C ₁₄ FH ₁₄ N	<i>p</i> F·C ₆ H ₄ ·CH ₂ ·CH ₂ ·C ₆ H ₄ ·NH ₂ <i>p</i>	-117·71	454(b)
C ₁₄ FH ₁₆ Br		X = Br <i>o</i> -109·94 <i>m</i> -113·91 <i>p</i> -119·70	459(f) 458 460

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}FH_{16}Cl$	As above with $X = Cl$	<i>o</i> -109.95 <i>m</i> -113.92 <i>p</i> -119.01		459(f) 458 460
$C_{14}FH_{16}I$	As above with $X = I$	<i>o</i> -109.97 <i>m</i> -113.92 <i>p</i> -119.05		459(f) 458 460
$C_{14}FH_{16}NO_2$	As above with $X = NO_2$	<i>o</i> -109.89 <i>m</i> -113.81 <i>p</i> -118.73		459(f) 458
$C_{14}FH_{17}$	As above with $X = H$	<i>o</i> -109.42 <i>m</i> -114.21 <i>p</i> -119.71		459(f)
$C_{14}FH_{17}Ge$	 1-F, 4-CH ₂ Ge(CH ₃) ₃ 2-F, 6-CH ₂ Ge(CH ₃) ₃ 2-F, 7-CH ₂ Ge(CH ₃) ₃	-127.73 -119.23 -117.31		442 (p) (q)
$C_{14}FH_{17}O$		<i>o</i> -109.83 <i>m</i> -114.17 <i>p</i> -119.48		459(f) 460
$C_{14}FH_{17}Pb$	 1-F, 4-CH ₂ Pb(CH ₃) ₃ 2-F, 6-CH ₂ Pb(CH ₃) ₃ 2-F, 7-CH ₂ Pb(CH ₃) ₃	-128.21 -119.62 -117.25		442 (p) (q)

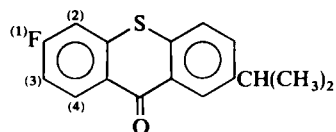
$C_{14}FH_{17}Si$	As above with 1-F, 4- $CH_2Si(CH_3)_3$ 2-F, 6- $CH_2Si(CH_3)_3$ 2-F, 7- $CH_2Si(CH_3)_3$	-127·73 -119·17 -117·33	442(p) (q)
$C_{14}FH_{17}Sn$	As above with 1-F, 4- $CH_2Sn(CH_3)_3$ 2-F, 6- $CH_2Sn(CH_3)_3$ 2-F, 7- $CH_2Sn(CH_3)_3$	-128·49 -119·63 -117·28	442(p) (q)
$C_{14}FH_{18}N$		-131·1	461
$C_{14}FH_{18}N$		<i>o</i> -109·76 <i>m</i> -114·20 <i>p</i> -119·57	459(f) 458 460
$C_{14}FH_{19}N^+$	As above with NH_2 replaced by NH_3^+	<i>o</i> -110·50 <i>m</i> -113·17 <i>p</i> -117·53	459(f) 458 460
$C_{14}F_2H_8$	$pF \cdot C_6H_4 \cdot C \equiv C \cdot C_6H_4 \cdot Fp$	-111·17	454(b)
$C_{14}F_2H_8N_2O_2$	$(mF \cdot C_6H_4 \cdot NCO)_2$	-110·82	401
$C_{14}F_2H_8N_2O_2$	<i>p</i> -isomer of above	-117·85	401
$C_{14}F_2H_{10}$	$pF \cdot C_6H_4 \cdot CH=CH \cdot C_6H_4 \cdot Fp$	<i>cis</i> -114·10 <i>trans</i> -114·01	454(b)
$C_{14}F_2H_{10}Br_2$	$pF \cdot C_6H_4 \cdot CHBr \cdot CHBr \cdot C_6H_4 \cdot p$	<i>erythro</i> -112·05 <i>threo</i> -112·10	454(b)
$C_{14}F_2H_{11}N_3O_2$	$(mF \cdot C_6H_4 \cdot NH \cdot CO)_2NH$	-112·47	401
$C_{14}F_2H_{11}N_3O_2$	<i>p</i> -isomer of above	-119·69	401
$C_{14}F_2H_{12}$	$pF \cdot C_6H_4 \cdot CH_2 \cdot CH_2 \cdot C_6H_4 \cdot Fp$	-117·00	454(b)
$C_{14}F_2H_{14}Sn$	$(pF \cdot C_6H_4)_2Sn(CH_3)_2$	-112·65	399(d)
$C_{14}F_2H_{14}Sn$	<i>m</i> -isomer of above	-113·22	399(d)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_2H_{16}$		<i>o</i> -109.97 <i>m</i> -113.98 <i>p</i> -119.05		459(f) 458 460
$C_{14}F_5H_{19}BN$		(1) -120.4 (2) -146.2		461
$C_{15}FH_8N$	$pF \cdot C_6H_4 \cdot C \equiv C \cdot C_6H_4 \cdot CN_p$	-109.54		454(b)
$C_{15}FH_{10}Br_2N$	$pF \cdot C_6H_4 \cdot CHBr \cdot CHBr \cdot C_6H_4 \cdot CN_p$	<i>erythro</i> -111.54 <i>threo</i> -111.25		454(b)
$C_{15}FH_{10}N$	$pF \cdot C_6H_4 \cdot CH=CH \cdot C_6H_4 \cdot CN_p$	<i>cis</i> -112.82 <i>trans</i> -112.20		454(b)
$C_{15}FH_{11}$	$pF \cdot C_6H_4 \cdot C \equiv C \cdot C_6H_4 \cdot CH_3_p$	-111.86		454(b)
$C_{15}FH_{11}O$	$pF \cdot C_6H_4 \cdot C \equiv C \cdot C_6H_4 \cdot OCH_3_p$	-112.17		454(b)
$C_{15}FH_{13}$	$pF \cdot C_6H_4 \cdot CH=CH \cdot C_6H_4 \cdot CH_3_p$	<i>cis</i> -114.77 <i>trans</i> -114.68		454(b)
$C_{15}FH_{13}Br_2$	$pF \cdot C_6H_4 \cdot CHBr \cdot CHBr \cdot C_6H_4 \cdot CH_3_p$	<i>erythro</i> -112.50 <i>threo</i> -112.81		454(b)
$C_{15}FH_{13}Br_2O$	$pF \cdot C_6H_4 \cdot CHBr \cdot CHBr \cdot C_6H_4 \cdot OCH_3_p$	<i>erythro</i> -112.57 <i>threo</i> -112.90		454(b)
$C_{15}FH_{13}N_2OS$			(1-2) 1.8 (1-3) 0.8	559

$C_{15}FH_{13}O$	$pF \cdot C_6H_4 \cdot CH=CH \cdot C_6H_4 \cdot OCH_3p$	<i>cis</i> -114·93 <i>trans</i> -115·04	454(b)
$C_{15}FH_{15}$	$pF \cdot C_6H_4 \cdot CH_2 \cdot CH_2 \cdot C_6H_4 \cdot CH_3p$	-117·45	454(b)
$C_{15}FH_{15}O$	$pF \cdot C_6H_4 \cdot CH_2 \cdot CH_2 \cdot C_6H_4 \cdot OCH_3p$	-117·47	454(b)
$C_{15}FH_{16}N$		R = CN <i>m</i> -113·84 <i>p</i> -118·94	459(f) 458
$C_{15}FH_{16}N$	As above with R = NC	<i>m</i> -113·86	459(f)
$C_{15}FH_{17}O$	As above with R = CHO	<i>p</i> -119·25	459(f)
$C_{15}FH_{17}O_2$	As above with R = CO ₂ H	<i>p</i> -119·37	459(f)
$C_{15}FH_{18}Cl$	As above with R = CH ₂ Cl	<i>p</i> -119·35	459(f)
$C_{15}FH_{19}$	As above with R = CH ₃	<i>o</i> -109·59 <i>m</i> -114·24 <i>p</i> -119·63	459(f) 464
$C_{15}FH_{19}O$	As above with R = OCH ₃	<i>o</i> -109·86 <i>m</i> -114·10 <i>p</i> -119·33	459(f) 458 460
$C_{15}FH_{19}O$	As above with R = CH ₂ OH	<i>p</i> -119·62	459(f)
$C_{15}F_4H_{10}$	$pF \cdot C_6H_4 \cdot CH=CH \cdot C_6H_4 \cdot CF_3p$	<i>cis</i> -113·27 <i>trans</i> -112·78	454(b)
$C_{15}F_4H_{10}Br_2$	$pF \cdot C_6H_4 \cdot CHBr \cdot CHBr \cdot C_6H_4 \cdot CF_3p$	<i>erythro</i> -111·64 <i>threo</i> -111·42	454(b)
$C_{15}F_4H_{12}$	$pF \cdot C_6H_4 \cdot CH_2 \cdot CH_2 \cdot C_6H_4 \cdot CF_3p$	-116·57	454(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{16}FH_9$			(3-C1) +8.00 (3-C2) +21.40 (3-C3) -257.4 (3-C4) 1.00 (3-C5) 1.44 (3-C6) 1.50 (3-C7) 0.25 (3-C8) 1.07 (3-C9) 0.59 (3-C10) 1.05 (3-C11) 0.85 (3-C12) 2.14 (3-C13) 3.96 (3-C14) +2.81 (3-C15) +6.73 (3-C16) +19.41	404
$C_{16}FH_9$			(1-C1) -251.1 (1-C2) +22.17 (1-C3) +8.21 (1-C4) 1.75 (1-C5) 2.99 (1-C6) 2.06 (1-C7) 1.08 (1-C8) 2.44 (1-C9) 2.45 (1-C10) +4.59 (1-C11) +15.36 (1-C12) +2.97 (1-C13) 0.60 (1-C14) 1.11 (1-C15) +3.70 (1-C16) +5.65	404
$C_{16}FH_9$	As above with F at C2		(2-C1) +22.75 (2-C2) -244.46 (2-C4) +3.91 (2-C5) 0.70 (2-C6) 1.03 (2-C7) 1.11 (2-C11) +9.58 (2-C13) 1.37 (2-C15) 0.92 (2-C16) 0.79	404
$C_{16}FH_9$	As above with F at C4		(4-C1) 1.20 (4-C2) 1.12 (4-C3) +4.42 (4-C4) -253.0 (4-C5) +19.96 (4-C6) +6.06 (4-C7) <0.2 (4-C8) 2.58 (4-C9) 0.75 (4-C10) 0.44 (4-C11) +3.15 (4-C12) +20.07 (4-C13) +9.68 (4-C14) 1.3 (4-C15) 0.63	404

C₁₆FH₁₃OS

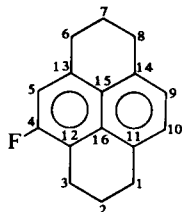


-116.8

(1-2) 8.5 (1-3) 8.5
(1-3) 6.0

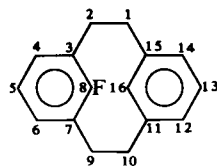
465

C₁₆FH₁₅



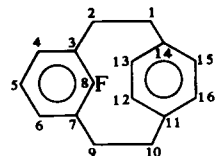
(4-C1) <0.3 (4-C2) 0.42 404
(4-C3) +3.56 (4-C4) -242.10
(4-C5) +25.68 (4-C6) +1.81
(4-C7) <0.3 (4-C8) <0.3
(4-C9) 2.41 (4-C10) 0.86
(4-C11) +6.75 (4-C12) +16.55
(4-C13) +8.74 (4-C14) 1.91
(4-C15) 1.11 (4-C16) +6.75

C₁₆FH₁₅



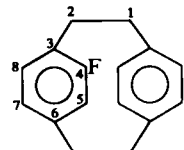
(8-C3) 18.3 (8-C4) 4.9 466
(8-C5) 4.9 (8-C8) 248.4
(8-C16) 3.7

C₁₆FH₁₅



(8-C3) 19.5 (8-C4) 4.9 466
(8-C5) 3.7 (8-C8) 246.6
(8-C15) 4.9

C₁₆FH₁₅



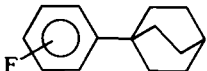
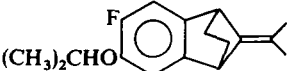
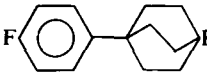
(4-C3) 18.3 (4-C4) 245.5 466
(4-C5) 22.0 (4-C6) 7.3
(4-C7) 2.4 (4-C8) 6.1

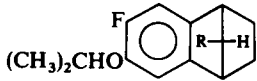
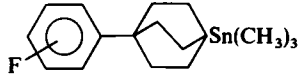
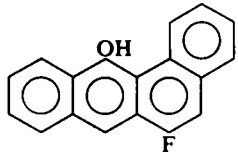
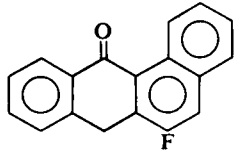
C₁₆FH₁₆N

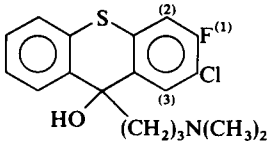
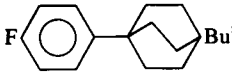
*p*F·C₆H₄·CH=CH·C₆H₄·N(CH₃)₂*p*

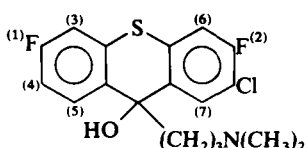
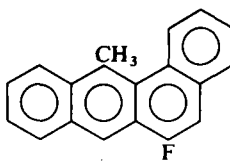
cis -115.57
trans -115.91

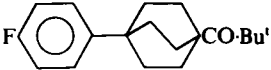
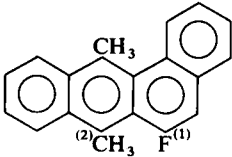
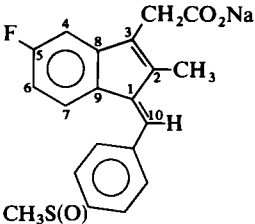
454(b)

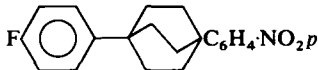
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{16}FH_{18}N$	$pF \cdot C_6H_4 \cdot CH_2 \cdot CH_2 \cdot C_6H_4 \cdot N(CH_3)_2$	-117.79		454(b)
$C_{16}FH_{18}N$	 $R = CH_2CN$	$p \ -119.30$		459(f)
$C_{16}FH_{19}O$	As above with $R = CO \cdot CH_3$	$p \ -119.28$		459(f)
$C_{16}FH_{19}O_2$	As above with $R = O \cdot CO \cdot CH_3$	$o \ -109.97$ $m \ -114.01$ $p \ -119.16$		459(f) 458 460
$C_{16}FH_{20}NO$	As above with $R = NH \cdot CO \cdot CH_3$	$o \ -109.83$ $m \ -114.12$ $p \ -119.43$		459(f) 460
$C_{16}FH_{21}$	As above with $R = CH_2 \cdot CH_3$	$m \ -114.24$ $p \ -119.64$		459(f) 464
$C_{16}F_2H_{18}Sn$	$(oF \cdot C_6H_4)_2Sn(CH_2CH_3)_2$	-93.18	$(^{119}Sn-F) \ 34 \ (^{117}Sn-F) \ 33$	436, 399, 400
$C_{16}F_2H_{18}Sn$	m -isomer of above	-112.87	$(^{119}Sn-F) \ 21.5$ $(^{117}Sn-F) \ 20.5$	436, 399, 400
$C_{16}F_2H_{18}Sn$	p -isomer of above	-112.49	$(^{119}Sn-F) \ 8.5 \ (^{117}Sn-F) \ 8.5$	436, 399, 400
$C_{17}FH_{21}O$		-137.3		462
$C_{17}FH_{21}O_2$	 $R = CO_2 \cdot CH_2 \cdot CH_3$	-119.27		459(f)

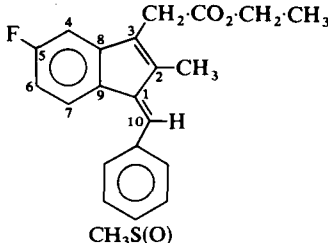
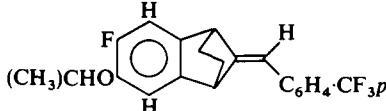
$C_{17}FH_{22}Cl$	As above with $R = CCl(CH_3)_2$	-119·41		459(f)
$C_{17}FH_{22}NO$	As above with $R = CO \cdot N(CH_3)_2$	-119·41		459(f)
$C_{17}FH_{23}$	As above with $R = (CH_2)_2 \cdot CH_3$	-119·65		459(f), 464
$C_{17}FH_{23}O$	As above with $R = C(OH)(CH_3)_2$	-119·70		459(f)
$C_{17}FH_{23}O$	 $R = CH(CH_3)_2$	-137·6		462
$C_{17}FH_{25}$		<i>o</i> -109·35 <i>p</i> -119·71		459(f)
$C_{18}FH_8BiCl_6$	$pF \cdot C_6H_4 \cdot Bi(3,4,5-Cl_3C_6H_2)_2$	-110·49		467(d)
$C_{18}FH_8Cl_6P$	As above with Bi replaced by P	-108·75		468(d)
$C_{18}FH_{10}BiCl_4$	$pF \cdot C_6H_4 \cdot Bi(3,4-Cl_2C_6H_3)_2$	-111·55		467(d)
$C_{18}FH_{10}Cl_4N$	As above with Bi replaced by N	-116·43		467(d)
$C_{18}FH_{10}Cl_4P$	As above with Bi replaced by P	-110·05		468(d)
$C_{18}FH_{11}O$		-124·0	(F-H) 13·3	439
$C_{18}FH_{11}O$		-121·0	(F-H) 7,12	439(b)
$C_{18}FH_{12}BiCl_2$	$pF \cdot C_6H_4 \cdot Bi(C_6H_4 \cdot Clm)_2$	-114·2		467(d)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{18}FH_{12}BiCl_2$	$pF \cdot C_6H_4 \cdot Bi(C_6H_4 \cdot Clp)_2$	-113.78		467(d)
$C_{18}FH_{12}Cl_2N$	$pF \cdot C_6H_4 \cdot N(C_6H_4 \cdot Clp)_2$	-118.29		467(d)
$C_{18}FH_{12}Cl_2N$	$pF \cdot C_6H_4 \cdot N(C_6H_4 \cdot Clm)_2$	-117.37		467(d)
$C_{18}FH_{12}Cl_2P$	$pF \cdot C_6H_4 \cdot P(C_6H_4 \cdot Clp)_2$	-111.45		468(d)
$C_{18}FH_{12}Cl_2P$	$pF \cdot C_6H_4 \cdot P(C_6H_4 \cdot Clm)_2$	-110.75		468(d)
$C_{18}FH_{12}Cl_2Sb$	$pF \cdot C_6H_4 \cdot Sb(C_6H_4 \cdot Clp)_2$	-111.53		467(d)
$C_{18}FH_{12}Cl_2Sb$	$pF \cdot C_6H_4 \cdot Sb(C_6H_4 \cdot Clm)_2$	-111.10		467(d)
$C_{18}FH_{14}Bi$	$pF \cdot C_6H_4 \cdot Bi(C_6H_5)_2$	-113.26		467(d)
$C_{18}FH_{14}N$	As above with Bi replaced by N	-119.75		467(d)
$C_{18}FH_{14}P$	As above with Bi replaced by P	-112.15		468(d)
$C_{18}FH_{14}Sb$	As above with Bi replaced by Sb	-112.57		467(d)
$C_{18}FH_{19}ClNOS$		-120.0	(1-2) 8.5 (1-3) 6.5	465
$C_{18}FH_{25}$		-119.72		459(f) 464
$C_{18}FH_{25}$	As above with Bu ⁱ replaced by Bu ⁱ	-119.65		459(f) 464
$C_{18}FH_{31}Sn$	$mF \cdot C_6H_4 \cdot Sn[(CH_2)_3CH_3]_3$	-113.58		399(d)
$C_{18}FH_{31}Sn$	<i>p</i> -isomer of above	-113.52		399(d)
$C_{18}FH_{34}NO_2P_2Pt$	<i>trans</i> -[P(CH ₂ CH ₃) ₃] ₂ Pt(C ₆ H ₄ F)NO ₂	<i>m</i> -115.07 <i>p</i> -122.13		469(d)

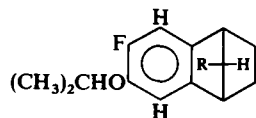
$C_{18}FH_{34}N_3P_2Pt$	As above with NO_2 replaced by N_3	$m - 115.33$ $p - 122.96$			469(d)
$C_{18}F_2H_{18}ClNOS$		(1) -117.0 (2) -119.6	(1-3) 8.0 (1-5) 5.5 (2-7) 6.5	(1-4) 8.0 (2-6) 8.5	465
$C_{18}F_3H_{12}Bi$	$(pF \cdot C_6H_4)_3Bi$	-112.83			467(d)
$C_{18}F_3H_{12}ClSn$	$(mF \cdot C_6H_4)_3SnCl$	-111.92			399(d), 400
$C_{18}F_3H_{12}ClSn$	<i>p</i> -isomer of above	-110.15			399(d), 400
$C_{18}F_3H_{12}N$	$(mF \cdot C_6H_4)_3N$	-111.13			401
$C_{18}F_3H_{12}N$	<i>p</i> -isomer of above	-119.83			401
$C_{18}F_3H_{12}N$	As above	-120.05			467(d)
$C_{18}F_3H_{12}P$	As above with N replaced by P	-111.85			468(d)
$C_{18}F_3H_{12}P$	$pF^{(1)} \cdot C_6H_4 \cdot P(C_6H_4 \cdot Fm)$	(1) -111.25			468(d)
$C_{18}F_3H_{12}Sb$	$(pF \cdot C_6H_4)_3Sb$	-112.02			468(d)
$C_{19}FH_9Cl_6$	$pF \cdot C_6H_4 \cdot CH(3,4,5-Cl_3C_6H_2)_2$	-113.88			467(d)
$C_{19}FH_{11}Cl_4$	$pF \cdot C_6H_4 \cdot CH(3,4-Cl_2C_6H_3)_2$	-114.96			467(d)
$C_{19}FH_{13}$		-123.6	(F-H) 10.3		439
$C_{19}FH_{13}Cl_2$	$pF \cdot C_6H_4 \cdot CH(C_6H_4 \cdot Clm)_2$	-115.90			467(d)
$C_{19}FH_{13}Cl_2$	$pF \cdot C_6H_4 \cdot CH(C_6H_4 \cdot Clp)_2$	-115.97			467(d)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{19}FH_{15}$	As above with Cl replaced by H	-116.90		467(d)
$C_{19}FH_{16}N_3O$	$pF \cdot C_6H_4 \cdot NH \cdot CO \cdot N(C_6H_5)NH \cdot C_6H_5$	-120.8		401
$C_{19}FH_{25}O$		-119.40		459(f)
$C_{19}F_3H_{13}$	$(pF \cdot C_6H_4)_3CH$	-116.32		467(d)
$C_{19}F_3H_{13}N_2O$	$(pF \cdot C_6H_4)_2N \cdot CO \cdot NH \cdot C_6H_4 \cdot Fp$	-120.5, -116.0		401
$C_{19}F_3H_{15}Sn$	$(mF \cdot C_6H_4)_3SnCH_3$	-112.58		399(d)
$C_{20}FH_{15}$		-110.6	(1-2) 6.3 (F-H) 15.4	439
$C_{20}FH_{16}NaO_3S$			(5-C3) 2 (5-C4) 23 (5-C5) 243 (5-C6) 23 (5-C7) 8 (5-C8) 9 (5-C9) 3	475
$C_{20}FH_{17}O_2S$	As above with Na replaced by H and S(O)CH ₃ replaced by SCH ₃	(Z)-isomer	(5-C3) 3 (5-C4) 24 (5-C5) 243 (5-C6) 23 (5-C7) 9 (5-C8) 9 (5-C9) 3	475

			(<i>E</i>)-isomer	(5-C3) 2 (5-C5) 241 (5-C7) 9 (5-C9) 3	(5-C4) 24 (5-C6) 23 (5-C8) 9	475
			(<i>Z</i>)-isomer	(5-C3) 2 (5-C5) 245 (5-C7) 9 (5-C9) 3	(5-C4) 24 (5-C6) 23 (5-C8) 9	475
			(<i>E</i>)-isomer	(5-C3) 2 (5-C5) 244 (5-C7) 9 (5-C9) 2	(5-C4) 24 (5-C6) 24 (5-C8) 7	475
$C_{20}FH_{17}O_3S$	As above with Na replaced by H					
$C_{20}FH_{18}Bi$	$pF \cdot C_6H_4 \cdot Bi(C_6H_4 \cdot CH_3p)_2$	-113·65				467(d)
$C_{20}FH_{18}BiO_2$	$pF \cdot C_6H_4 \cdot Bi(C_6H_4 \cdot OCH_3p)_2$	-113·56				467(d)
$C_{20}FH_{18}N$	$pF \cdot C_6H_4 \cdot N(C_6H_4 \cdot CH_3p)_2$	-120·92				467(d)
$C_{20}FH_{18}P$	As above with N replaced by P	-112·65				468(d)
$C_{20}FH_{18}P$	$pF \cdot C_6H_4 \cdot P(C_6H_4 \cdot CH_3m)_2$	-112·45				468(d)
$C_{20}FH_{18}Sb$	$pF \cdot C_6H_4 \cdot Sb(C_6H_4 \cdot CH_3p)_2$	-112·57				467(d)
$C_{20}FH_{20}NO_2$		-119·23				459(f)
$C_{20}FH_{21}$	As above with NO_2 replaced by H	-119·42				459(d)
$C_{20}F_2H_{26}Sn$	$(mF \cdot C_6H_4)_2Sn(CH_2 \cdot CH_2 \cdot CH_2 \cdot CH_3)_2$	-112·99				399(d)
$C_{20}F_2H_{26}Sn$	<i>p</i> -isomer of above	-112·70				399(d)
$C_{20}F_3H_{14}N_3O_2$	$mF^{(1)} \cdot C_6H_4 \cdot N(CO \cdot NH \cdot C_6H_4 \cdot Fm)_2$	(1) -111·60				401
$C_{20}F_3H_{14}N_3O_2$	$pF \cdot C_6H_4 \cdot N(CO \cdot NH \cdot C_6H_4 \cdot Fp)_2$	-112·90, -119·0				401
$C_{20}F_3H_{17}Sn$	$(mF \cdot C_6H_4)_3Sn \cdot CH_2 \cdot CH_3$	-112·45				399(d)
$C_{20}F_3H_{17}Sn$	<i>p</i> -isomer of above	-111·84				400, 399(d)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.	
C ₂₁ FH ₁₉	<i>p</i> F·C ₆ H ₄ ·CH(C ₆ H ₄ ·CH ₃ <i>m</i>) ₂	-117.09		400, 467(d)	
C ₂₁ FH ₁₉	<i>p</i> F·C ₆ H ₄ ·CH(C ₆ H ₄ ·CH ₃ <i>p</i>) ₂	-117.13		467(d)	
C ₂₁ FH ₁₉ O ₂	<i>p</i> F·C ₆ H ₄ ·CH(C ₆ H ₄ ·OCH ₃ <i>m</i>) ₂	-116.83		467(d)	
C ₂₁ FH ₁₉ O ₂	<i>p</i> F·C ₆ H ₄ ·CH(C ₆ H ₄ ·OCH ₃ <i>p</i>) ₂	-117.28		467(d)	
C ₂₁ F ₃ N ₃ O ₃	(<i>m</i> F·C ₆ H ₄ ·NCO) ₃	-112.30		401	
C ₂₁ F ₃ N ₃ O ₃	<i>p</i> -isomer of above	-112.76		401	
C ₂₁ F ₇ H ₁₃	<i>p</i> F·C ₆ H ₄ ·CH(C ₆ H ₄ ·CF ₃ <i>m</i>) ₂	-113.56		467(d)	
C ₂₂ FH ₂₁ O ₃ S			(5-C3) 3 (5-C5) 247 (5-C7) ~9 (5-C9) 3	(5-C4) 24 (5-C6) 23 (5-C8) 9 (5-C10) 1.5	475
C ₂₂ FH ₂₄ BiN ₂	<i>p</i> F·C ₆ H ₄ ·Bi[C ₆ H ₄ ·N(CH ₃) ₂ <i>p</i>] ₂	-114.35		467(d)	
C ₂₂ FH ₂₄ N ₂ P	As above with Bi replaced by P	-114.35		468(d)	
C ₂₂ FH ₂₄ N ₃	As above with Bi replaced by N	-124.45		467(d)	
C ₂₂ F ₃ H ₂₁ Sn	(<i>m</i> F·C ₆ H ₄) ₃ Sn(CH ₂) ₃ CH ₃	-112.40		399(d)	
C ₂₂ F ₃ H ₂₁ Sn	<i>p</i> -isomer of above	-111.86		399(d)	
C ₂₂ F ₄ H ₂₀ O		-136.1		462	

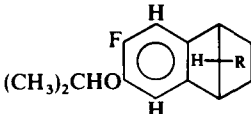
$C_{23}FH_{25}N_2$	$pF \cdot C_6H_4 \cdot CH[C_6H_4 \cdot N(CH_3)_3 p]_2$	-117.90	467(d)
$C_{24}FH_{10}Cl_9Si$	$pF \cdot C_6H_4 \cdot Si(3,4,5-Cl_3C_6H_2)_3$	-106.75 (-6.4)	476(d)
$C_{24}FH_{13}Cl_6Si$	$pF \cdot C_6H_4 \cdot Si(3,4-Cl_2C_6H_3)_3$	-108.15 (-5.0)	476(d)
$C_{24}FH_{16}ClNp$	$mF \cdot C_6H_4 \cdot N=P(C_6H_4 \cdot Clp)_3$	-114.20	477(d)
$C_{24}FH_{16}ClNP$	p -isomer of above	-127.01	477(d)
$C_{24}FH_{16}Cl_3Si$	$pF \cdot C_6H_4 \cdot Si(C_6H_4 \cdot Clm)_3$	-109.15 (-4.0)	476(d)
$C_{24}FH_{16}N_4O_6P$	$mF \cdot C_6H_4(C_6H_5)_2P=N[2,4,6-(NO_2)_3C_6H_2]$	-109.60	477(d)
$C_{24}FH_{16}N_4O_6P$	p -isomer of above	-104.90	477(d)
$C_{24}FH_{18}ClNP$	$mF \cdot C_6H_4(C_6H_5)_2P=N \cdot C_6H_4 \cdot Clp$	-110.62	477(d)
$C_{24}FH_{18}ClNP$	p -isomer of above	-106.83	477(d)
$C_{24}FH_{18}ClNP$	$mF \cdot C_6H_4 \cdot N=P(C_6H_5)_2 \cdot C_6H_4 \cdot Clp$	-114.66	477(d)
$C_{24}FH_{18}ClNP$	p -isomer of above	-127.76	477(d)
$C_{24}FH_{18}N_2O_2P$	$mF \cdot C_6H_4(C_6H_5)_2P=N \cdot C_6H_4 \cdot NO_2p$	-109.89	477(d)
$C_{24}FH_{18}N_2O_2P$	p -isomer of above	-105.57	477(d)
$C_{24}FH_{19}Np$	$mF \cdot C_6H_4(C_6H_5)_2P=N \cdot C_6H_5$	-110.88	477(d)
$C_{24}FH_{19}NP$	$mF \cdot C_6H_4 \cdot N=P(C_6H_5)_3$	-114.84	477(d)
$C_{24}FH_{19}NP$	p -isomer of above	-128.20	477(d)
$C_{24}FH_{19}Si$	$pF \cdot C_6H_4 \cdot Si(C_6H_5)_3$	-110.85 (-2.3)	476(d)
$C_{24}FH_{19}Sn$	$mF \cdot C_6H_4 \cdot Sn(C_6H_5)_3$	-113.58	399(d)
$C_{24}FH_{19}Sn$	p -isomer of above	-111.46	400, 399(d)

 $C_{24}FH_{27}O_3$  $R = C_6H_4 \cdot CO_2Pr^i p$ -137.0

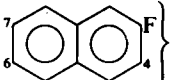
462

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₂₄ F ₂ H ₁₈ NP	<i>m</i> F ⁽¹⁾ ·C ₆ H ₄ ·N=P(C ₆ H ₅)C ₆ H ₄ ·F ⁽²⁾ <i>m</i>	(1) -114.66 (2) -110.63		477(d)
C ₂₄ F ₂ H ₁₈ NP	<i>m</i> F ⁽¹⁾ ·C ₆ H ₄ ·N=P(C ₆ H ₅)C ₆ H ₄ ·F ⁽²⁾ <i>p</i>	(1) -114.68 (2) -106.85		477(d)
C ₂₄ F ₂ H ₁₈ NP	<i>p</i> F ⁽¹⁾ ·C ₆ H ₄ ·N=P(C ₆ H ₅)C ₆ H ₄ ·F ⁽²⁾ <i>m</i>	(1) -127.76 (2) -110.77		477(d)
C ₂₄ F ₂ H ₁₈ NP	<i>p</i> F ⁽¹⁾ ·C ₆ H ₄ ·N=P(C ₆ H ₅)C ₆ H ₄ ·F ⁽²⁾ <i>p</i>	(1) -127.82 (2) -107.06		477(d)
C ₂₄ F ₂ H ₁₈ Sn	(<i>m</i> F·C ₆ H ₄) ₂ Sn(C ₆ H ₅) ₂	-112.16		399(d), 400
C ₂₄ F ₂ H ₁₈ Sn	<i>p</i> -isomer of above	-111.42		399(d)
C ₂₄ F ₃ H ₁₄ N ₄ O ₆ P	(<i>m</i> F·C ₆ H ₄) ₃ P=N[2,4,6-(NO ₂) ₃ C ₆ H ₂]	-108.81		400, 477(d)
C ₂₄ F ₃ H ₁₄ N ₄ O ₆ P	<i>p</i> -isomer of above	-104.11		477(d)
C ₂₄ F ₃ H ₁₆ ClNP	(<i>m</i> F·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·Cl <i>p</i>	-109.89		477(d)
C ₂₄ F ₃ H ₁₆ ClNP	<i>p</i> -isomer of above	-106.12		477(d)
C ₂₄ F ₃ H ₁₆ N ₂ O ₂ P	(<i>m</i> F·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·NO ₂ <i>p</i>	-109.17		477(d)
C ₂₄ F ₃ H ₁₆ N ₂ O ₂ P	<i>p</i> -isomer of above	-104.86		477(d)
C ₂₄ F ₃ H ₁₇ NP	(<i>m</i> F·C ₆ H ₄) ₃ P=N·C ₆ H ₅	-110.15		477(d)
C ₂₄ F ₃ H ₁₇ NP	<i>p</i> -isomer of above	-106.58		477(d)
C ₂₄ F ₃ H ₁₇ Sn	(<i>m</i> F·C ₆ H ₄) ₃ SnC ₆ H ₅	-114.32		399(d), 400
C ₂₄ F ₃ H ₁₇ Sn	<i>p</i> -isomer of above	-111.30		399(d), 400
C ₂₄ F ₄ H ₁₆ NP	(<i>m</i> F ⁽¹⁾ ·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·F ⁽²⁾ <i>m</i>	(1) -109.89 (2) -114.31		477(d)
C ₂₄ F ₄ H ₁₆ NP	(<i>m</i> F ⁽¹⁾ ·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·F ⁽²⁾ <i>p</i>	(1) -110.00 (2) —		477(d)
C ₂₄ F ₄ H ₁₆ NP	(<i>p</i> F ⁽¹⁾ ·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·F ⁽²⁾ <i>m</i>	(1) -106.10 (2) -114.32		477(d)
C ₂₄ F ₄ H ₁₆ NP	(<i>p</i> F ⁽¹⁾ ·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·F ⁽²⁾ <i>p</i>	(1) -106.36 (2) -127.32		477(d)

C ₂₄ F ₄ H ₁₆ Si	(<i>p</i> F·C ₆ H ₄) ₄ Si	−110.05 (−3.1)	476(d)
C ₂₄ F ₄ H ₁₆ Si	<i>p</i> F ⁽¹⁾ ·C ₆ H ₄ ·Si(C ₆ H ₄ ·F <i>m</i>) ₃	(1) −109.45 (−3.7)	476(d)
C ₂₄ F ₄ H ₁₆ Sn	(<i>m</i> F·C ₆ H ₄) ₄ Sn	−111.89	399(d), 400
C ₂₄ F ₄ H ₁₆ Sn	(<i>m</i> F ⁽¹⁾ ·C ₆ H ₄) ₃ SnC ₆ H ₄ ·F ⁽²⁾ <i>p</i>	(1) −111.90 (2) −110.62	399(d), 400
C ₂₄ F ₄ H ₁₆ Sn	(<i>m</i> F ⁽¹⁾ ·C ₆ H ₄) ₂ Sn(C ₆ H ₄ ·F ⁽²⁾ <i>p</i>) ₂	(1) −111.96 (2) −110.77	399(d), 400
C ₂₄ F ₄ H ₁₆ Sn	<i>m</i> F ⁽¹⁾ ·C ₆ H ₄ ·Sn(C ₆ H ₄ ·F ⁽²⁾ <i>p</i>) ₃	(1) −111.95 (2) −110.89	399(d), 400
C ₂₄ F ₄ H ₁₆ Sn	(<i>p</i> F·C ₆ H ₄) ₄ Sn	−111.07	399(d) 400
C ₂₅ FH ₁₈ N ₂ P	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅) ₂ P=N·C ₆ H ₄ ·CN <i>p</i>	−110.12	477(d)
C ₂₅ FH ₁₈ N ₂ P	<i>p</i> -isomer of above	−106.83	477(d)
C ₂₅ FH ₁₉ Cl ₃ INP	<i>m</i> F·C ₆ H ₄ ·N(CH ₃)P ⁺ (C ₆ H ₄ ·Cl <i>p</i>) ₃ I [−]	−108.79	477(d)
C ₂₅ FH ₁₉ Cl ₃ INP	<i>p</i> -isomer of above	−110.92	477(d)
C ₂₅ FH ₂₁ ClINP	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ N(CH ₃)C ₆ H ₄ ·Cl <i>p</i> I [−]	−106.68	477(d)
C ₂₅ FH ₂₁ ClINP	<i>p</i> -isomer of above	−98.67	477(d)
C ₂₅ FH ₂₁ NOP	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅) ₂ P=N·C ₆ H ₄ ·OCH ₃ <i>p</i>	−110.95	477(d)
C ₂₅ FH ₂₁ NOP	<i>p</i> -isomer of above	−107.38	477(d)
C ₂₅ FH ₂₁ NP	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅) ₂ P=N·C ₆ H ₄ ·CH ₃ <i>p</i>	−110.96	477(d)
C ₂₅ FH ₂₁ NP	<i>p</i> -isomer of above	−107.39	477(d)
C ₂₅ FH ₂₁ N ₂ O ₂ P	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ NCH ₃ (C ₆ H ₄ ·NO ₂ <i>p</i>) I [−]	−106.17	477(d)
C ₂₅ FH ₂₁ N ₂ O ₂ P	<i>p</i> -isomer of above	−98.00	477(d)
C ₂₅ FH ₂₂ INP	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ N(CH ₃)C ₆ H ₅ I [−]	−106.96	477(d)
C ₂₅ FH ₂₂ INP	<i>p</i> -isomer of above	−99.00	477(d)

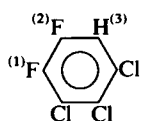
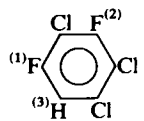
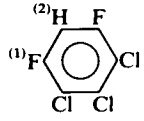
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₂₅ FH ₂₂ INP	(C ₆ H ₅) ₃ P ⁺ N(CH ₃)C ₆ H ₄ ·F <i>m</i> I [−]	−109.43		477(d)
C ₂₅ FH ₂₂ INP	<i>p</i> -isomer of above	−111.71		477(d)
C ₂₅ FH ₂₉ O ₃	 (CH ₃) ₂ CHO R = CH₂C₆H₄CO₂Prⁱ<i>p</i>	−137.4		462
C ₂₅ F ₂ H ₂₁ INP	<i>m</i> F ⁽¹⁾ ·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ N(CH ₃)C ₆ H ₄ ·F ⁽²⁾ <i>m</i> I [−]	(1) −106.69 (2) −109.17		477(d)
C ₂₅ F ₂ H ₂₁ INP	<i>m</i> F ⁽¹⁾ ·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ N(CH ₃)C ₆ H ₄ ·F ⁽²⁾ <i>p</i> I [−]	(1) −106.80 (2) −111.34		477(d)
C ₂₅ F ₂ H ₂₁ INP	<i>p</i> F ⁽¹⁾ ·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ N(CH ₃)C ₆ H ₄ ·F ⁽²⁾ <i>m</i> I [−]	(1) −98.62 (2) −109.33		477(d)
C ₂₅ F ₂ H ₂₁ INP	<i>p</i> F ⁽¹⁾ ·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ N(CH ₃)C ₆ H ₄ ·F ⁽²⁾ <i>p</i> I [−]	(1) −98.85 (2) −111.59		477(d)
C ₂₅ F ₃ H ₁₆ N ₂ P	(<i>m</i> F·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·CN <i>p</i>	−109.39		477(d)
C ₂₅ F ₃ H ₁₆ N ₂ P	<i>p</i> -isomer of above	−105.26		477(d)
C ₂₅ F ₃ H ₁₉ ClINP	(<i>m</i> F·C ₆ H ₄) ₃ P ⁺ N(CH ₃)C ₆ H ₄ ·Cl <i>p</i> I [−]	−105.89		477(d)
C ₂₅ F ₃ H ₁₉ ClINP	<i>p</i> -isomer of above	−98.06		477(d)
C ₂₅ F ₃ H ₁₉ IN ₂ O ₂ P	(<i>m</i> F·C ₆ H ₄) ₃ P ⁺ N(CH ₃)C ₆ H ₄ ·NO ₂ <i>p</i> I [−]	−105.47		477(d)
C ₂₅ F ₃ H ₁₉ IN ₂ O ₂ P	<i>p</i> -isomer of above	−97.36		477(d)
C ₂₅ F ₃ H ₁₉ NOP	(<i>m</i> F·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·OCH ₃ <i>p</i>	−110.25		477(d)
C ₂₅ F ₃ H ₁₉ NOP	<i>p</i> -isomer of above	−106.75		477(d)
C ₂₅ F ₃ H ₁₉ NP	(<i>m</i> F·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·CH ₃ <i>p</i>	−110.24		477(d)
C ₂₅ F ₃ H ₁₉ NP	<i>p</i> -isomer of above	−106.73		477(d)
C ₂₅ F ₃ H ₂₀ INP	(<i>m</i> F·C ₆ H ₄) ₃ P ⁺ N(CH ₃)C ₆ H ₅ I [−]	−106.25		477(d)
C ₂₅ F ₃ H ₂₀ INP	<i>p</i> -isomer of above	−98.44		477(d)
C ₂₅ F ₄ H ₁₉ INP	(<i>m</i> F ⁽²⁾ ·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·F <i>m</i>) I [−]	(1) −108.69 (2) −105.97		477(d)

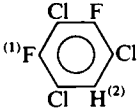
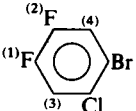
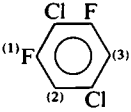
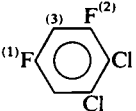
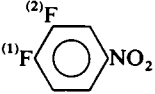
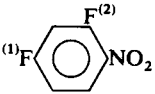
C ₂₅ F ₄ H ₁₉ INP	(<i>p</i> F ⁽²⁾ ·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·F ⁽¹⁾ <i>m</i>) I [−]	(1) −109·02 (2) −98·09	477(d)
C ₂₅ F ₄ H ₁₉ INP	(<i>m</i> F ⁽²⁾ ·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·F ⁽¹⁾ <i>p</i>) I [−]	(1) −110·61 (2) −105·99	477(d)
C ₂₅ F ₄ H ₁₉ INP	(<i>p</i> F ⁽²⁾ ·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·F ⁽¹⁾ <i>p</i>) I [−]	(1) −111·18 (2) −98·20	477(d)
C ₂₅ F ₆ H ₁₆ NP	(<i>m</i> F ⁽¹⁾ ·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·CF ₃ <i>p</i>	(1) −109·67	477(d)
C ₂₅ F ₆ H ₁₆ NP	(<i>p</i> F ⁽¹⁾ ·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·CF ₃ <i>p</i>	(1) −105·80	477(d)
C ₂₆ FH ₂₁ IN ₂ P	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅)P ⁺ NCH ₃ (C ₆ H ₄ ·CN <i>p</i>) I [−]	−106·22	477(d)
C ₂₆ FH ₂₁ IN ₂ P	<i>p</i> -isomer of above	−98·11	477(d)
C ₂₆ FH ₂₄ INOP	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ NCH ₃ (C ₆ H ₄ ·OCH ₃ <i>p</i>) I [−]	−106·99	477(d)
C ₂₆ FH ₂₄ INOP	<i>p</i> -isomer of above	−99·10	477(d)
C ₂₆ FH ₂₄ INP	<i>m</i> F·C ₆ H ₄ (C ₆ H ₅) ₂ P ⁺ NCH ₃ (C ₆ H ₄ ·CH ₃ <i>p</i>) I [−]	−107·01	477(d)
C ₂₆ FH ₂₄ INP	<i>p</i> -isomer of above	−99·05	477(d)
C ₂₆ F ₃ H ₁₉ IN ₂ P	(<i>m</i> F·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·CN <i>p</i>) I [−]	−105·53	477(d)
C ₂₆ F ₃ H ₁₉ IN ₂ P	<i>p</i> -isomer of above	−97·50	477(d)
C ₂₆ F ₃ H ₂₂ INOP	(<i>m</i> F·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·OCH ₃ <i>p</i>) I [−]	−106·26	477(d)
C ₂₆ F ₃ H ₂₂ INOP	<i>p</i> -isomer of above	−98·61	477(d)
C ₂₆ F ₃ H ₂₂ INP	(<i>m</i> F·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·CH ₃ <i>p</i>) I [−]	−106·25	477(d)
C ₂₆ F ₃ H ₂₂ INP	<i>p</i> -isomer of above	−98·59	477(d)
C ₂₆ F ₄ H ₁₇ N ₃ O ₂	(<i>p</i> F·C ₆ H ₄) ₂ N·CO·N(<i>p</i> F·C ₆ H ₄)CO·NH·C ₆ H ₄ ·F <i>p</i>	−119·8, −115·6, −115·3	401
C ₂₆ F ₆ H ₁₉ INP	(<i>m</i> F ⁽¹⁾ ·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·CF ₃ <i>p</i>) I [−]	(1) −105·78	477(d)
C ₂₆ F ₆ H ₁₉ INP	<i>p</i> -isomer of above	(1) −97·82	477(d)
C ₂₇ FH ₂₅ NO ₃ P	(<i>p</i> CH ₃ O·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·F <i>m</i>	−114·92	477(d)
C ₂₇ FH ₂₅ NO ₃ P	<i>p</i> -isomer of above	−128·68	477(d)
C ₂₇ FH ₂₅ NP	(<i>p</i> CH ₃ ·C ₆ H ₄) ₃ P=N·C ₆ H ₄ ·F <i>m</i>	−115·00	477(d)
C ₂₇ FH ₂₅ NP	<i>p</i> -isomer of above	−128·67	477(d)

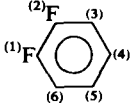
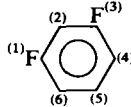
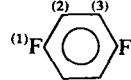
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₂₇ FH ₂₅ O ₃ Si	(<i>p</i> CH ₃ O·C ₆ H ₄) ₃ Si·C ₆ H ₄ ·F <i>p</i>	−111·35 (−1·8)		476(d)
C ₂₇ FH ₂₅ Si	As above with OCH ₃ replaced by CH ₃	−111·45 (−1·7)		476(d)
C ₂₇ FH ₂₅ Si	As above with <i>p</i> CH ₃ O replaced by <i>m</i> CH ₃	−111·35 (−1·8)		476(d)
C ₂₈ FH ₂₈ INOP	(<i>p</i> CH ₃ O·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·F <i>m</i>) I [−]	−109·74		477(d)
C ₂₈ FH ₂₈ INOP	<i>p</i> -isomer of above	−112·31		477(d)
C ₂₈ FH ₂₈ INP	(<i>p</i> CH ₃ ·C ₆ H ₄) ₃ P ⁺ NCH ₃ (C ₆ H ₄ ·F <i>m</i>) I [−]	−109·69		477(d)
C ₂₈ FH ₂₈ INP	<i>p</i> -isomer of above	−112·09		477(d)
C ₂₉ FH ₂₃ Ge				442(q)
	4-CH ₂ Ge(C ₆ H ₅) ₃	−121·03		
	6-CH ₂ Ge(C ₆ H ₅) ₃	−118·89		
	7-CH ₂ Ge(C ₆ H ₅) ₃	−117·25		
C ₂₉ FH ₂₃ Pb	As above with Ge replaced by Pb			442(q)
	4-CH ₂ Pb(C ₆ H ₅) ₃	−121·11		
	6-CH ₂ Pb(C ₆ H ₅) ₃	−119·01		
	7-CH ₂ Pb(C ₆ H ₅) ₃	−117·11		
C ₂₉ FH ₂₃ Si	As above with Ge replaced by Si			442(q)
	4-CH ₂ Si(C ₆ H ₅) ₃	−121·09		
	6-CH ₂ Si(C ₆ H ₅) ₃	−118·89		
	7-CH ₂ Si(C ₆ H ₅) ₃	−117·30		
C ₂₉ FH ₂₃ Sn	As above with Ge replaced by Sn			442(q)
	4-CH ₂ Sn(C ₆ H ₅) ₃	−121·73		
	6-CH ₂ Sn(C ₆ H ₅) ₃	−119·17		
	7-CH ₂ Sn(C ₆ H ₅) ₃	−117·13		
C ₃₀ FH ₃₄ N ₃ Si	[<i>p</i> (CH ₃) ₂ N·C ₆ H ₄] ₃ Si·C ₆ H ₄ ·F <i>p</i>	−112·35 (−0·8)		476(d)

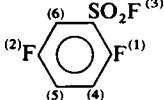
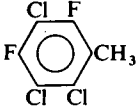
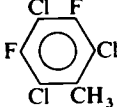
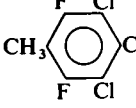
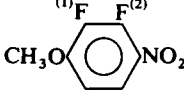
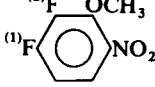
$C_{34}F_3H_{22}N_5O_4$		$\left. \begin{matrix} (6)(10) \\ (8) \end{matrix} \right\} -111.6, -112.7$	401
$C_{34}F_3H_{22}N_5O_4$	<i>p</i> -isomer of above	$\left. \begin{matrix} (6)(10) \\ (8) \end{matrix} \right\} -111.3, -109.4$	401
$C_{34}F_5H_{20}N_5O_4$	As above with $R^1 = R^3 = R^6 = R^8 = R^{10} = mF \cdot C_6H_4$	$\begin{matrix} (1) -109.10 & (3) -110.70 \\ (6)(10) -110.40 & (8) -112.50 \end{matrix}$	401
$C_{34}F_5H_{20}N_5O_4$	As above with $R^1 = R^3 = R^6 = R^8 = R^{10} = pF \cdot C_6H_4$	$\begin{matrix} (1) -112.40 & (3) -111.40 \\ (6)(10) -110.50 & (8) -113.30 \end{matrix}$	401

B. Fluorine nuclei in difluorobenzene derivatives

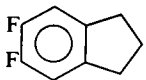
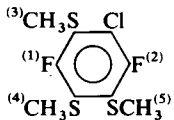
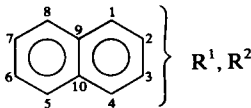
$C_6F_2HCl_3$		$\begin{matrix} (1) -132.2 & (2) -133.9 \end{matrix}$	$\begin{matrix} (1-2) 21.6 \\ (2-3) 10.0 \end{matrix}$	$\begin{matrix} (1-3) 7.8 \end{matrix}$	397
$C_6F_2HCl_3$		$\begin{matrix} (1) -113.0 & (2) -106.7 \end{matrix}$	$\begin{matrix} (1-2) 2.0 \\ (2-3) 2.0 \end{matrix}$	$\begin{matrix} (1-3) 8.4 \end{matrix}$	397
$C_6F_2HCl_3$		-107.8	$\begin{matrix} (1-2) 9.6 \end{matrix}$		397

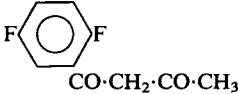
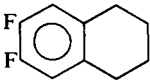
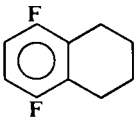
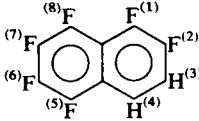
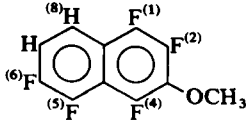
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_2HCl_3$		(1) -135.5 (2) -128.8	(1-2) 19.0 (1-3) 7.0 (2-3) 2.0	397
$C_6F_2HCl_3$		-112.6	(1-2) 7.2	397
$C_6F_2H_2BrCl$			(1-2) -20.633 (1-3) +10.228 (1-4) +7.907 (2-3) +7.588 (2-4) +9.904	406
$C_6F_2H_2Cl_2$		-109.7	(1-2) 8 (1-3) 1	398
$C_6F_2H_2Cl_2$		(1) -104.6 (2) -108.5	(1-3) 6 (2-3) 8	398
$C_6F_2H_3NO_2$		(1) -128.7 (2) -135.0		407
$C_6F_2H_3NO_2$		(1) -98.2 (2) -111.9		407

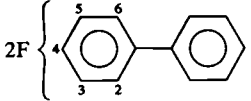
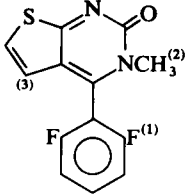
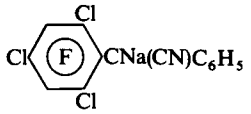
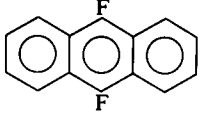
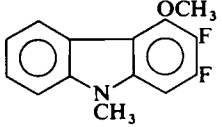
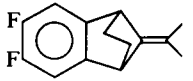
$C_6F_2H_4$		-138.619	(1-2) -20.5 (1-C1) -247.1 (1-C2) 12.5	408
$C_6F_2H_4$	As above		(1-2) -20.821 (1-3) 8.060 (1-4) -1.395 (1-5) 4.527 (1-6) 10.846 (1-C1) -246.64 (1-C2) 12.50 (1-C3) 0.53 $\frac{1}{2}[(1-C4)+(1-C5)]$ 5.19 (1-C6) 17.02	409 402
$C_6F_2H_4$		-119.848	(1-3) 6.6 (1-C1) -246.8 (1-C3) 11.8	408
$C_6F_2H_4$	As above		(1-2) 9.419 (1-3) 6.515 (1-4) -0.807 (1-5) 6.627 (1-6) 8.433 (1-C1) -247.55 (1-C2) 25.36 (1-C3) 11.88 (1-C4) 3.54 (1-C5) 9.80 (1-C6) 21.26	402
$C_6F_2H_4$		-110.879	(1-4) 17.8 (1-C1) -241.2 (1-C4) 2.5	408
$C_6F_2H_4$	As above		(1-2) 8.092 (1-3) 4.157 (1-4) 17.646 (1-C1) -240.89 (1-C2) 24.36 (1-C3) 8.52 (1-C4) 2.42	402
$C_6F_2Br_4$		-118.0		396

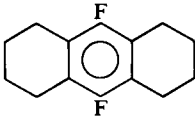
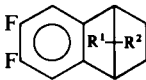
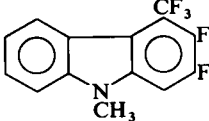
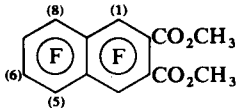
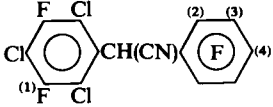
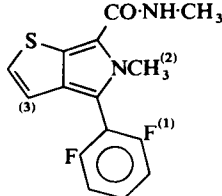
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_6F_2Br		-88.4		396
$C_6F_3H_3O_2S$		(1) -113.008 (2) -114.679 (3) 63.681	(1-2) 17.569 (1-3) 11.594 (1-4) 8.746 (1-5) 3.699 (1-6) 5.018 (2-3) 1.690 (2-4) 3.892 (2-5) 7.198 (2-6) 6.988 (3-4) 0.781 (3-5) -0.583 (3-6) -0.225	413(n)
$C_7F_2H_3Cl_3$		-114, -111.9		398
$C_7F_2H_3Cl_3$		-111.3		398
$C_7F_2H_3Cl_3$		-112.1		398
$C_7F_2H_5NO_3$		(1) -155.0 (2) -140.6		407
$C_7F_2H_5NO_3$		(1) -126.8 (2) -150.1		407

$C_7F_2H_5NO_3$		(1) -102.0 (2) 119.7			407
$C_7F_2H_5NO_3$		-117.0			407
$C_7F_2H_5NO_3$		(1) -117.6 (2) -136.8			407
$C_7F_2H_5NO_3$		(1) -145.4 (2) -125.4			407
$C_8F_2H_9NO$		-129.4	(1-2) 12.6 (1-4) ±2.1	(1-3) ±11.3	432(b)
$C_8F_2H_9NS_2$		(1) -112.3 (2) -137.8	(1-2) 14.2 (2-3) 5.9	(1-2) 9.1	425
$C_8F_2H_{10}O_3S_2$	$mF \cdot C_6H_4 \cdot S^+(CH_3)_2 FSO_3^-$	-108.52			405
$C_8F_2H_{10}O_3S_2$	<i>p</i> -isomer of above	-104.69			405

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_2N_8$		-118.4		410
$C_9F_2H_8$		-139.5		434
$C_9F_2H_9ClS_3$		(1) -99.4 (2) -107.6	(1-2) 14.8 (1-3) 1.55 (1-4) 1.55 (2-5) 2.3	427
$C_9F_2H_9NO_2S_3$	As above with Cl replaced by NO_2	(1) -100.3 (2) -122.0	(1-2) 15.5 (1-3) 1.0 (1-4) 2.65 (2-5) 2.65	427
$C_9F_2H_{10}OS_3$	As above with Cl replaced by OH	(1) -100.4 (2) -129.9	(1-2) 14.5 (1-4) 0.7 (2-5) 2.5	427
$C_{10}F_2H_6$	 R^1, R^2			405
	1-F, 3-F	(1) -118.10		(p)
	1-F, 4-F	(1) -119.07		
	2-F, 4-F	(2) -114.48		(q)
$C_{10}F_2H_6$	As above with 1-F, 4-F	-127.9		443
$C_{10}F_2H_6$	As above with 1-F, 3-F		(1-C1) 248.1 (1-C2) 24.2 (1-C3) 13.2 (1-C4) 5.1 (1-C6) 1.7 (1-C9) 10.9	431

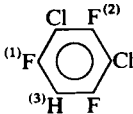
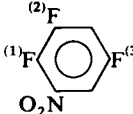
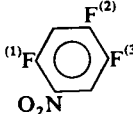
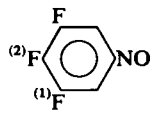
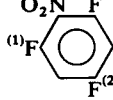
				(1-C10) 6·7 (3-C1) 12·6 (3-C2) 29·5 (3-C3) 256·7 (3-C4) 20·9 (3-C9) 5·8 (3-C10) 4·8	
	1-F, 4-F			(1-C1) 254 (1-C2) 16·0 (1-C4) 5·9 (1-C5) 3·0 (1-C8) 3·0 (1-C10) 11·7	
	2-F, 3-F			(2-C1) 7·0 (2-C2) 255·2 (2-C3) 17·7 (2-C10) 3·8	
C ₁₀ F ₂ H ₈ O ₂		-119·12, -119·71			224(a)
C ₁₀ F ₂ H ₁₀		-127			434
C ₁₀ F ₂ H ₁₀		-101·25			434
C ₁₀ F ₆ H ₂		(1) -137·2 (2) -144·3 (5) -148·5 (6) -158·0 (7) -155·8 (8) -146·8	(1-8) 53 (5-6) 17 (5-7) 3 (5-8) 16 (6-7) 17 (6-8) 3·5 (7-8) 17 (4-5) 17		433
C ₁₁ F ₅ H ₅ O		(1) -151·0 (2) -153·2 (4) -141·4 (5) -145·4 (6) -139·0	(1-2) 17 (1-8) 17 (3-4) 17 (4-5) 56 (5-6) 16		433

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}F_2H_8$		2-F, 3-F (2) -144.2 (3) -138.6 2-F, 4-F (2) -113.5 (4) -112.2 2-F, 5-F (2) -124.2 (5) -120.1 2-F, 6-F -114.8 3-F, 4-F (3) -138.1 (4) -139.9 3-F, 5-F -109.4		448
$C_{12}F_2H_8N_2OS$			(1-2) 0.8 (1-3) 1.0	449
$C_{14}F_2H_5Cl_3NNa$		-113.6		444(b)
$C_{14}F_2H_8$		-131.9		461
$C_{14}F_2H_{11}NO$		-138.5, -169.6		451
$C_{14}F_2H_{14}$		-143.5		462

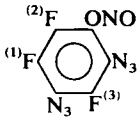
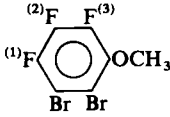
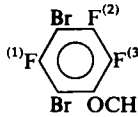
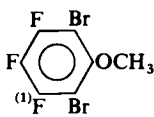
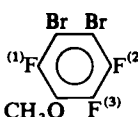
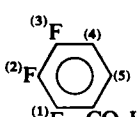
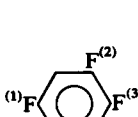
$C_{14}F_2H_{16}$		-126.2				461
$C_{14}F_2H_{16}$	 $R^1 = (CH_3)_2CH, R^2 = H$ $R^1 = H, R^2 = (CH_3)_2CH$	-143.4 -143.9				462
$C_{14}F_5H_8NO$		-138.0, -150.8				451
$C_{14}F_6H_4O_4$		(1) -119.5 (5) -141.3 (6) -151.2	(1-8) 70			377
$C_{14}F_7HCl_3N$		(1) -108.7 (2) -138.1 (3) -161.7 (4) -153.0	(2-4) 3.5 (3-4) 21			444(b)
$C_{14}F_7Cl_3LiN$	As above with H replaced by Li	(1) -114.0 (2) -145.2 (3) -167.5 (4) -172.8	(3-4) 22 (2-4) 6.1			444(b)
$C_{14}F_7Cl_3NNa$	As above with H replaced by Na	(1) -116.6 (2) -145.5 (3) -168.0 (4) -174.1				444(b)
$C_{15}F_2H_{12}N_2OS$			(1-2) 1.3 (1-3) 1.0			449

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{18}F_4H_{18}S_7$		(1) -101.0 (2) -100.9	(1-2) 14.63 (1-3) 1.21 (1-4) 1.97 (2-5) 1.99	427
$C_{19}F_5H_{13}$		-142.0		462
$C_{19}F_5H_{15}$		$R = pCF_3 \cdot C_6H_4$ -142.5		462
$C_{21}F_2H_{20}O_2$		$R = C_6H_4 \cdot CO_2Pr^i p$ -142.8		462
$C_{22}F_2H_{22}O_2$	As above with R and H interchanged and $R = CH_2 \cdot C_6H_4 \cdot CO_2Pr^i p$	-143.5		462
$C_{54}F_2H_{40}N_4$		-123.8		471(b)

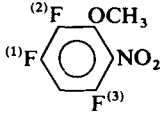
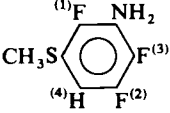
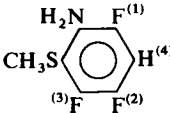
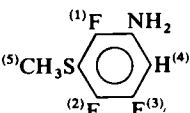
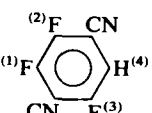
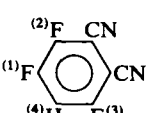
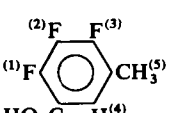
C. Fluorine nuclei in trifluorobenzene derivatives

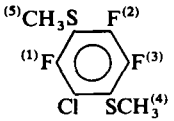
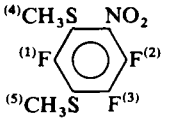
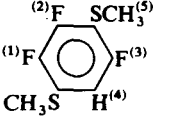
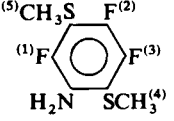
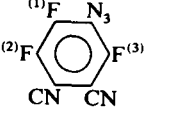
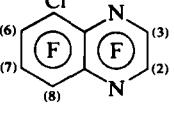
$C_6F_3HCl_2$		(1) -111.8	(2) -110.4	(1-2) 2.0	(1-3) 7.2	397
$C_6F_3H_2NO_2$		(1) -137.5	(2) -155.9			407
$C_6F_3H_2NO_2$		(1) -147.8	(2) -129.8			407
$C_6F_3H_2NO_2$		(1) -117.6	(2) -121.8			407
$C_6F_3H_2NO_2$		(1) -130.6	(2) -150.1			407
$C_6F_3H_2NO_2$		(1) -115.3	(2) -98.8			407

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_3H_3$			(1-2) -19.640 (1-3) 6.135 (1-4) -2.033 (1-5) 5.730 (1-6) 9.989 (1-C1) -248.66 (1-C2) 15.22 (1-C3) 3.25 (1-C4) 3.51 (1-C5) 8.40 (1-C6) 17.60 (2-C1) 9.76 (2-C2) -249.68 (2-C4) -0.73 (2-C5) 5.12	411
$C_6F_3H_3$			(1-2) -20.424 (1-3) 6.352 (1-4) 15.094 (1-5) 3.310 (1-6) 10.086 (2-3) 10.652 (2-4) 3.190 (2-5) -2.012 (2-6) 8.975 (3-4) 8.384 (4-5) 7.864 (4-6) 5.090 (1-C1) -243.08 (1-C2) 14.45 (1-C3) 0.53 (1-C4) 2.71 (1-C5) 6.42 (1-C6) 19.69 (2-C1) 12.51 (2-C2) -249.95 (2-C3) 21.00 (2-C4) 9.59 (2-C5) 4.00 (2-C6) 1.39 (4-C1) 3.91 (4-C2) 12.53 (4-C3) 27.83 (4-C4) -244.41 (4-C5) 23.95 (4-C6) 9.88	402
$C_6F_3H_3$			(1-2) 9.122 (1-3) 5.828 (1-4) -1.682 (1-C1) -248.66 (1-C2) 25.91 (1-C3) 15.11 (1-C4) 4.34	402, 408
$C_6F_3Br_3$		-95.4		396

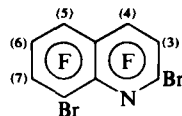
$C_6F_3N_7O_2$		(1) -138.4 (3) -139.9	(2) -149.5	(1-2) 24.0 (2-3) 9.8	(1-3) 2.6	410
$C_7F_3H_3Br_2O$		(1) -125.8 (3) -149.2	(2) -156.2			396
$C_7F_3H_3Br_2O$		(1) NA (3) -156.0	(2) -128.3			396
$C_7F_3H_3Br_2O$		(1) -125.4				396
$C_7F_3H_3Br_2O$		(1) -119.5 (3) -150.1	(2) -126.9			396
$C_7F_3H_3O_2$		(1) -132.1 (3) -130.4	(2) -163.0	(1-2) 19.0 (1-4) 2.5 (2-3) 19.0 (2-5) 2.5 (3-5) 6.0	(1-3) 12.5 (1-5) 7.0 (2-4) 6.5 (3-4) 8.5	424
$C_7F_3H_4BrO$		(1) -109.0 (3) -158.2	(2) -135.2			396

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_3H_4BrO$		(1) -161.8 (2) -129.5 (3) -111.5		396
$C_7F_3H_4BrO$		(1) -130.9 (2)(3) -140.0		396
$C_7F_3H_4BrO$		(1) -130.9 (2) -132.7 (3) -139.1	(1-2) 11.2 (1-3) 0.5 (1-4) 11.3 (1-5) 1.2 (2-3) 22.6 (2-4) 7.3 (3-4) 9.8	396
$C_7F_3H_4NO_3$		(1) -139.5 (2) -141.5 (3) -158.4		407
$C_7F_3H_4NO_3$		(1) -132.5 (2) -150.3 (3) -145.7		407
$C_7F_3H_4NO_3$		(1) -118.9 (2) -142.8 (3) -161.5		407

$C_7F_3H_4NO_3$		(1) -128.3 (2) -157.1 (3) -126.0		407
$C_7F_3H_6NS$		(1) -135.9 (2) -143.2 (3) -159.0	(1-4)(3-4) 6.8, 8.3 (1-2) 12.0 (1-3) 9.6 (2-3) 19.8 (2-4) 10.1	425
$C_7F_3H_6NS$		(1)(2)(3) -136.5, -149.6	(1-4)(2-4) 10.6, 9.8 (3-4) 7.1	425
$C_7F_3H_6NS$		(1) -133.9 (2) -145.5 (3) -142.1	(1-4)(2-4) 7.8, 8.3 (1-2) 4.3 (1-3) 12.7 (1-5) 0.6 (2-3) 23.6 (2-5) 1.0 (3-4) 11.1	425
$C_8F_3HN_2$		(1) -118.0 (2) -129.8 (3) -105.3	(1-2) 20.0 (1-3) 9.3 (1-4) 9.3 (2-3) 12.9 (2-4) 6.5 (3-4) 8.3	410
$C_8F_3HN_2$		(1) -122.7 (2) -134.6 (3) -107.6	(1-2) 21.7 (1-3) 10 (1-4) 10 (2-3) 13.3 (2-4) 6.8 (3-4) 9	410
$C_8F_3H_5O_2$		(1) -136.2 (2) -162.6 (3) -132.4	(1-2) 19.0 (1-3) 12.0 (1-4) 8.0 (2-3) 19.0 (2-4) 3.0 (3-4) 8.0 (3-5) 2.0	329

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_3H_6ClS_2$		(1) -108.5 (2) -130.3 (3) -132.9	(1-2) 2.65 (1-3) 12.6 (1-5) 0.9 (2-3) 25.3 (2-5) 0.9 (3-4) 1.85	427
$C_8F_3H_6NO_2S$		(1) -100.1 (2) -126.8 (3) -149.8	(1-2) 5.3 (1-3) 13.2 (1-4) 0.8 (1-5) 1.3 (2-3) 24.3 (2-5) 1.3	427
$C_8F_3H_7S_2$		(1) -139.9 (2) -128.9 (3) -110.0	(1-2) 22.8 (1-3) 12.9 (1-4) 5.8 (1-5) 0.6 (2-3) ~2 (2-4) 2.4 (3-4) 9.2	425
$C_8F_3H_8NS_2$		(1) -132.8 (2) -146.2 (3) -135.7	(1-2) 5.0 (1-3) 13.0 (1-5) 1.0 (2-3) 25.5 (2-5) 0.96	427
$C_8F_3N_5$		(1) -127.7 (2) -128.7 (3) -116.6	(1-2) 18.8 (1-3) 11.3 (2-3) 11.0	410
$C_8F_3ClN_2$		(2) -80.74 (3) -78.82 (6) -128.38 (7) -153.38 (6) -146.20	(3-6) -1.13 (3-7) +8.20 (3-8) -4.10 (6-7) -20.85 (6-8) +5.48 (7-8) -17.53 (2-3) -29.7 (2-6) +8.45 (2-7) -1.15 (2-8) +7.70	496

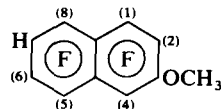
C₉F₃Br₂N



(3) -137.7 (4) -128.6 (3-4) 15.8 (3-5) 6.7
 (5) -141.2 (6) -155.0 (3-6) 2.8 (3-7) 8.4
 (7) -117.2 (4-5) 49.5 (4-6) 5.3
 (4-7) 3.6 (5-6) 16.8
 (5-7) 7.0 (6-7) 22.2

499

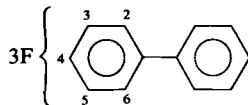
C₁₁F₆H₄O



(1) -147.3 (2) -151.2 (1-8) 65 (4-5) 63.7
 (4) -141.0 (5) -151.2
 (6) -135.8 (8) -117.5

433

C₁₂F₃H₇

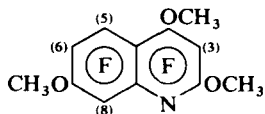


2-F, 3-F, 4-F
 2-F, 3-F, 5-F
 2-F, 3-F, 6-F
 2-F, 4-F, 5-F
 2-F, 4-F, 6-F

(2) -139.3 (3) -160.7 (4) -136.5
 (2) -149.0 (3) -133.5 (5) -116.2
 (2) -138.1 (3) -142.8 (6) -120.1
 (2) -120.1 (4) -135.7 (5) -143.5
 (2)(6) -111.5 (4) -109.8

448

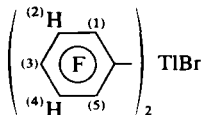
C₁₂F₄H₉NO₃



(3) -159.8 (5) -147.9 (3-5) 6.3 (3-6) 1.2
 (6) -157.0 (8) -146.6 (3-8) 3.0 (5-6) 17.8
 (5-8) 14.7 (6-8) 2.8

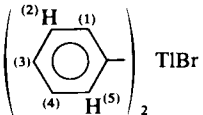
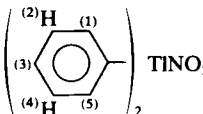
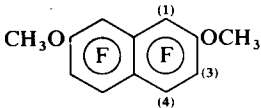
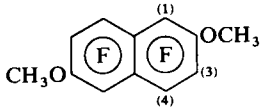
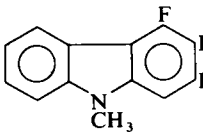
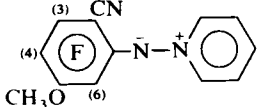
499

C₁₂F₆H₄BrTi



(1) -88.7 (3) -106.8 (1-2) ±7.0 (1-3) 8.2
 (1-4) ±1.5 (1-5)(2-4) 0, 3.6
 (2-3) 9.1
 (²⁰³Tl-1) +785.8
 (²⁰³Tl-1) +778.7
 (Tl-2) -22.5

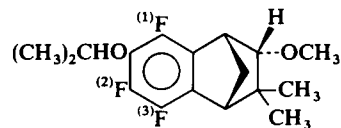
453

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}F_6H_4BrTi$		(1) -94.7 (3) -131.9 (4) -142.6	(1-2) 5.4 (1-3) 4.7 (1-4) 15.5 (1-5) 2.4 (2-3) 10.4 (2-4) 5.8 (3-4) 20.1 (3-5) 9.7 (4-5) 9.3 (Ti-1) +1087 (Ti-3) -48.2 (Ti-4) +141.5	453
$C_{12}F_6H_4NO_2Ti$		(1) -89.4 (3) -104.8	(^{205}Ti -1) 950.5 (^{203}Ti -1) 938.5 (Ti-2) 26.3	453
$C_{12}F_6H_6O_2$		(1) -140.5 (3) -151.0 (4) -148.8		433
$C_{12}F_6H_6O_2$		(1) -141.2 (3) -151.0 (4) -148.0	(1-8) 68	433
$C_{13}F_3H_8N$		-137.5, -143.1, -175.4		451
$C_{13}F_3H_8N_3O$		(3) -132.8 (4) -152.2 (6) -143.7		447(a)

$C_{14}F_3H_{10}NO$		-138.0, -144.2, -175.2			451
$C_{14}F_3H_{12}NO$		-147.4, -156.8 (2:1)			451
$C_{14}F_3H_{13}O$		(1) -139.4 (2) -154.5 (3) -147.6			446
$C_{14}F_3H_{15}O$	As above with double bond reduced	(1) -140.0 (2) -153.7 (3) -148.6			446
$C_{14}F_8HCl_2N$		(1) -135.3 (2) -134.1 (3) -114.4 (4) -139.1 (5) -161.2 (6) -152.4	(1-2) 20 (4-6) 3.3 (4-7) 11 (5-6) 21		444(b)
$C_{14}F_8Cl_2LiN$	As above with H replaced by Li	(1) -140.1 (2) -142.1 (3) -119.2 (4) -144.3 (5) -167.6 (6) -172.1	(1-2) 22 (1-3) 8.5 (1-4) 0 (4-6) 5.5 (5-6) 21.4		444(b)
$C_{14}F_8Cl_2NNa$	As above with H replaced by Na	(1) -140.4 (2) -142.5 (3) -119.3 (4) -144.3 (5) -167.9 (6) -173.1	(1-2) 23 (1-3) 8.5 (1-4) 0 (4-6) 6.1 (5-6) 22.2		444(b)
$C_{15}F_3H_{13}O$		(1) -143.9 (2) -157.0 (3) -152.3			446

Molecular formula	Structure of compounds	Chemical shifts (pm)	Coupling constants (Hz)	Ref.
$C_{15}F_3H_{17}O$		(1) -143.9 (2) -155.7 (3) -152.7		446
$C_{16}F_3H_{17}O$		(1) -136.2 (2) -150.4 (3) -146.7		446
$C_{16}F_3H_{17}O$		(1) -130.3 (2) -150.7 (3) -144.7		446
$C_{16}F_3H_{19}O$		(1) -139.1 (2) -153.0 (3) -148.6		446
$C_{16}F_3H_{19}O$		(1) -140.0 (2) -152.8 (3) -147.6		446
$C_{17}F_3H_{21}O_2$		(1) -140.6 (2) -153.4 (3) -147.8		446

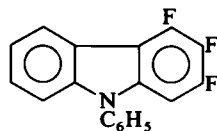
C₁₇F₃H₂₁O₂



(1) -139.2 (2) -153.8
(3) -149.2

446

C₁₈F₃H₁₀N



-136.9, -142.8, -172.2

451

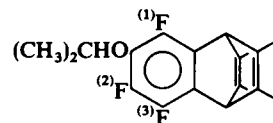
C₁₉F₃H₁₂N

As above with C₆H₅ replaced by CH₂·C₆H₅

-136.8, -144.6, -173.8

451

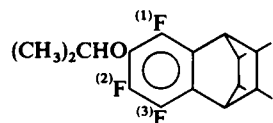
C₁₉F₃H₂₁O



(1) -145.1 (2) -157.1
(3) -153.3

446

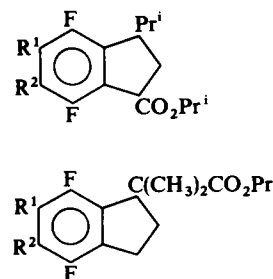
C₁₉F₃H₂₅O



(1) -144.6 (2) -155.2
(3) -153.5

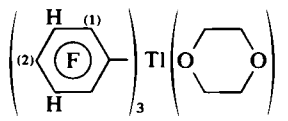
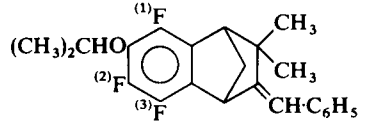
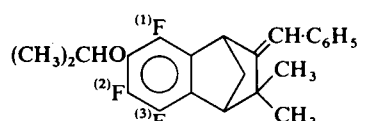
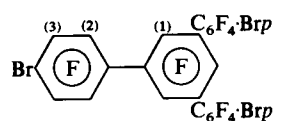
446

C₁₉F₃H₂₅O₃

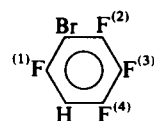


R¹ = PrⁱO, R² = F -130.2, -145.2, -151.9
R¹ = F, R² = PrⁱO and -136.4, -139.7, -152.8

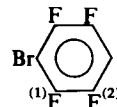
446

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{22}F_9H_{14}O_2Tl$		(1) -86.8 (2) -108.5	(Tl-1) 497 (Tl-2) 14.3	453
$C_{23}F_3H_{23}O$		(1) -139.5 (2) -152.3 (3) -145.4		446
$C_{23}F_3H_{23}O$		(1) -137.2 (2) -153.1 (3) -148.0		446
$C_{24}F_{15}Br_3$		(1) -101.3 (2) -136.6 (3) -132.2		386(b)
$C_{24}F_{18}$	As above with Br replaced by F(4)	(1) -102.3 (2) -137.5 (3) -161.3 (4) -150.9		386(b)

D. Fluorine nuclei in tetrafluorobenzene derivatives

C_6F_4HBr		(1) -109.7 (2) -125.5 (3) -163.2 (4) -133.2	396
-------------	---	--	-----

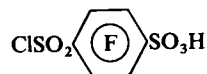
C₆F₄HBr



(1) -133.4 (2) -137.9

396

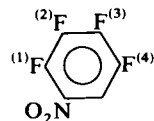
C₆F₄HClO₅S₂



-139.98, -140.92

415(a)

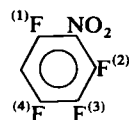
C₆F₄HNO₂



(1) -145.1 (2) -154.2
(3) -148.8 (4) -138.8

407

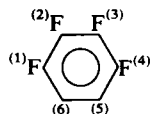
C₆F₄HNO₂



(1) -128.8 (2) -141.0
(3) -163.5 (4) -125.6

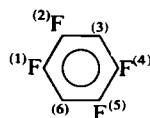
407

C₆F₄H₂



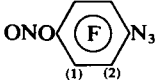
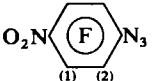
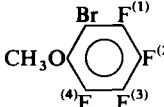
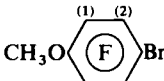
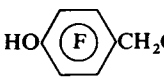
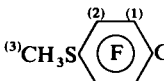
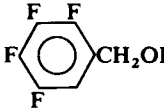
(1-2) -19.517 (1-3) 2.547 402
(1-4) 12.706 (1-5) 4.459
(1-6) 9.807 (2-3) -18.487
(2-5) -2.698 (2-6) 8.149
(1-C1) -245.87 (1-C2) 16.85
(1-C3) 3.61 (1-C4) 3.67
(1-C5) 7.81 (1-C6) 19.73
(2-C1) 10.16 (2-C2) -252.11
(2-C3) 12.32 (2-C4) 1.52
(2-C5) 4.06 (2-C6) 0.15

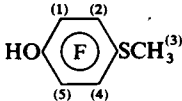
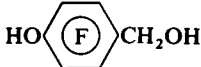
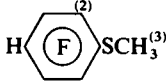
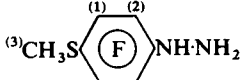
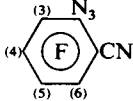
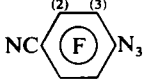
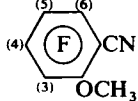
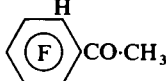
C₆F₄H₂

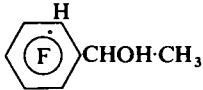
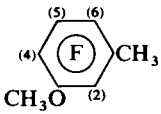
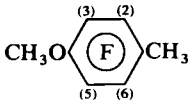
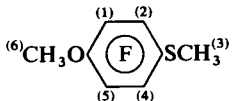
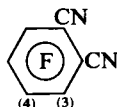
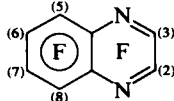
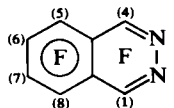


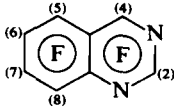
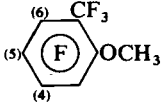
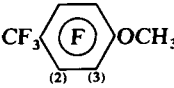
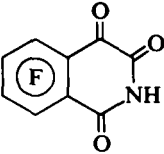
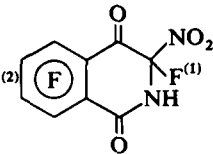
(1-2) -20.736 (1-3) 7.449 402
(1-4) 13.334 (1-5) -0.224
(1-6) 10.063 (1-C1) -247.00
(1-C2) 14.10 (1-C3) 1.05
(1-C4) 3.89 (1-C5) 9.57
(1-C6) 22.98

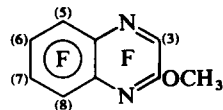
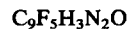
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_4H_2$			(1-2) -19.735 (1-3) 5.557 402 (1-4) -2.439 (1-5) 1.780 (1-6) 10.419 (2-4) 5.599 (2-5) 10.901 (1-C1) -250.38 (1-C2) 15.34 (1-C3) 5.34 (1-C4) 3.96 (1-C5) 13.08 (1-C6) 21.93 (2-C1) 11.09 (2-C2) -246.89 (2-C4) -1.12 (2-C5) 3.82 (5-C1) 14.99 (5-C2) 5.72 (5-C4) 27.83 (5-C5) -246.35	
$C_6F_4H_2O_3S_2$		-141.75, -147.02		415(a)
$C_6F_4H_2O_6S_2$		-145.15		415(a)
$C_6F_4H_3N$		(HSO_3F solvent) (2) -146.8 (3) -134.1		416(a)
$C_6F_4Br_2$		(1) -125.0 (2) -154.3		396
$C_6F_4Br_2$		(1) -102.3 (2) -125.1 (3) -159.4		396

$C_6F_4K_2O_3S_2$		$-65.77, -74.88$ relative to CF_3CO_2K	415
$C_6F_4N_4O_2$		(1) -161.1 (2) -152.4	410
$C_6F_4N_4O_2$		(1) -149.6 (2) -147.0	410
$C_7F_4H_3BrO$		(1) -132.1 (2) -161.4 (3) -161.1 (4) -156.0	396
$C_7F_4H_3BrO$		(1) -156.7 (2) -135.2	396
$C_7F_4H_3ClO$		$-162.0, -144.9$	426(b)
$C_7F_4H_3ClS$		(1) -146.6 (2) -134.3 (2-3) 1.0	427
$C_7F_4H_3NO_2S$	As above with Cl replaced by NO_2	(1) -148.1 (2) -133.3 (2-3) 1.6	427
$C_7F_4H_4O$		$-139.8, -146.3, -157.0, -157.9$	428

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_4H_4OS$		(1) -161.8 (2) -134.7	(1-2) 23.7 (1-4) 8.5 (1-5) 2.7 (2-3) 0.4 (2-4) 2.7	427
$C_7F_4H_4O_2$		-164.0, -148.6		426(b)
$C_7F_4H_4S$			(2-3) 0.95	427
$C_7F_4H_6N_2S$		(1) -158.5 (2) -135.9	(1-3) 0.4	425
$C_7F_4N_4$		(3) -155.0 (4) -145.6 (5) -155.0 (6) -134.0		410
$C_7F_4N_4$		(2) -133.6 (3) -149.6		410
$C_8F_4H_3NO$		(3) -156.2 (4) -146.4 (5) -162.8 (6) -134.0	(3-4) 18 (3-5) 1 (3-6) 8.7 (4-5) 19.7 (4-6) 4.8 (5-6) 20.3	410
$C_8F_4H_4O$		-136.9, -138.3, -148.8, -155.0		428

$C_8F_4H_6O$		-139.4, -146.7, -157.0 -158.0			428
$C_8F_4H_6O$		(2) -137.8 (4) -155.0 (5) -166.0 (6) -145.8	(2-5) 8.5 (4-5) 19.6 (5-6) 19.6		433
$C_8F_4H_6O$		(2) -145.8 (3) -159.5	(2-3) 20 (2-5) 9 (2-6) 1		433
$C_8F_4H_6OS$		(1) -156.9 (2) -135.2	(1-2) 22.8 (1-4) 8.4 (1-5) 3.0 (1-6) 1.45 (2-3) 0.6 (2-4) 0.9		427
$C_8F_4N_2$		(3) -126.3 (4) -141.6			410
$C_8F_4N_2$		-129.2			410
$C_8F_6N_2$		(2) -79.09 (5) -151.03 (6) -153.83	(5-6) -17.8 (5-7) +1.3 (5-8) +14.0 (6-7) -19.6		496
$C_8F_6N_2$		(1) -79.67 (5) -138.83 (6) -143.57	(5-6) -18.45 (5-7) +8.45 (5-8) +17.00 (6-7) -19.7 (6-8) +8.45 (7-8) -18.45		496

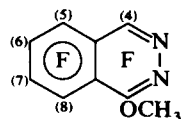
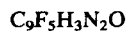
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_6N_2$		(2) -43.31 (4) -47.04 (5) -139.68 (6) -155.77 (7) -143.20 (8) -150.62	(2-5) -4.00 (2-6) 6.76 (2-7) -1.82 (2-8) 4.92 (4-5) 50.75 (4-6) 2.55 (4-7) 0.48 (4-8) 0.80 (5-6) -18.80 (5-7) 6.77 (5-8) 16.00 (6-7) -19.17 (6-8) 2.08 (7-8) -17.10	497
$C_8F_7H_3O$		(5) -162.0	(4-5) 22 (5-6) 22	433
$C_8F_7H_3O$		(2) -142.4 (3) -157.0		433
$C_9F_4HNO_3$		-140.2, -141.7, -147.8, -150.7		437
$C_9F_5HN_2O_4$		(1) -119.5 (2) -138.1, -140.2, -147.3, -149.4		437



(3) -77.25 (5) -152.70
 (6) -159.52 (7) -156.77
 (8) -153.49

(3-5) +7.75 (3-6) -0.50
 (3-7) +7.90 (3-8) -3.50
 (5-6) -18.45 (5-7) +0.55
 (5-8) +13.45 (6-7) -19.90
 (6-8) -0.60 (7-8) -18.30

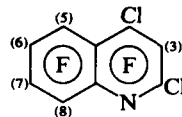
496



(4) -86.05 (5) -140.47
 (6) -147.02 (7) -146.38
 (8) -136.36

(4-5) +55.84 (4-6) +0.90
 (4-7) -2.83 (4-8) +3.15
 (5-6) -19.28 (5-7) +6.81
 (5-8) +16.51 (6-7) -19.47
 (6-8) +8.24 (7-8) -18.57

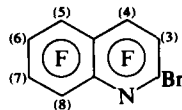
496



(3) -116.62 (5) -145.41
 (6) -153.98 (7) -153.19
 (8) -148.57

(3-5) 8.9 (3-6) -2.1
 (3-7) 7.0 (3-8) -4.2
 (5-6) -16.8 (5-7) 1.8
 (5-8) 15.0 (6-7) -19.0
 (6-8) 2.1 (7-8) -17.4

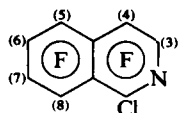
499



(3) -136.87 (4) -129.32
 (5) -146.89 (6) -154.73
 (7) -152.76 (8) -149.53

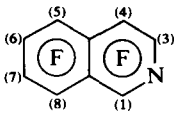
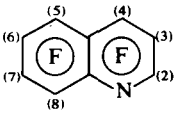
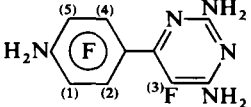
(3-4) -15.4 (3-5) 6.8
 (3-6) -2.8 (3-7) 7.6
 (3-8) -4.3 (4-5) 45.8
 (4-6) 5.0 (4-7) -3.5
 (4-8) 1.4 (5-6) -17.4
 (5-7) 1.4 (5-8) 15.4
 (6-7) -19.0 (6-8) 3.0
 (7-8) -16.6

499

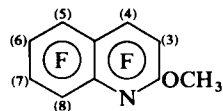


(3) -96.65 (4) -153.0
 (5) -146.31 (6) -147.67
 (7) -153.5 (8) -138.68

500

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_9F_7N		(1) -62.94 (3) -98.57 (4) -155.93 (5) -145.82 (6) -146.78 (7) -153.45 (8) -140.70	(1-3) 12 (1-4) 32.2 (1-5) 1.7 (1-6) -2.5 (1-7) 3.3 (1-8) 61.5 (3-4) -16.6 (3-5) 4.3 (3-6) -3.5 (3-7) 6.4 (3-8) -4.7 (4-5) 46.4 (4-6) 5.0 (4-7) -4.6 (4-8) 1.5 (5-6) -19.3 (5-7) 1.8 (5-8) 16.4 (6-7) -20.0 (6-8) 7.9 (7-8) -19.0	500
C_9F_7N		(2) -73.98 (3) -161.93 (4) -125.73 (5) -147.06 (6) -156.25 (7) -152.60 (8) -149.77	(2-3) -27.6 (2-4) 26.5 (2-5) -3.1 (2-6) 8.3 (2-7) -1.5 (2-8) 6.1 (3-4) -14.5 (3-5) 7.0 (3-6) -2.5 (3-7) 8.15 (3-8) -4.05 (4-5) 48.0 (4-6) 4.55 (4-7) -3.4 (4-8) 1.5 (5-6) -17.9 (5-7) 1.9 (5-8) 14.95 (6-7) -19.15 (6-8) 2.45 (7-8) -17.1	499
$C_{10}F_4H_3N_2Na$		(1) -146.6 (2) -148.0		444(b)
$C_{10}F_4H_4N_2$	As above with Na replaced by H	(1) -141.3 (2) -141.3		444(b)
$C_{10}F_5H_6N_5$		(1) -164.7 (2) -146.4 (3) -167.9	(1-2) 21.2 (1-4) 8.6 (1-5) 8.1 (2-3) 9.3 (2-4) 2.3	445(a)

$C_{10}F_6H_3NO$

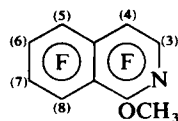


(3) -161.5 (4) -134.6
(5) -148.5 (6) -160.8
(7) -155.1 (8) -151.9

(3-4) 12.6 (3-5) 7.4
(3-6) 1.5 (3-7) 7.6
(3-8) 3.6 (4-5) 47.8
(4-6) 3.4 (4-7) 3.4
(4-8) 1.6 (5-6) 18.4
(5-7) 1.4 (5-8) 14.8
(6-7) 18.5 (6-8) 1.5
(7-8) 17.0

499

$C_{10}F_6H_3NO$

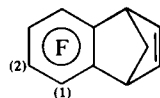


(3) -99.37 (4) -166.37
(5) -148.60 (6) -150.17
(7) -157.86 (8) -137.69

(3-4) -18.5 (3-5) 4.8
(3-6) -3.2 (3-7) 6.4
(3-8) -4.8 (4-5) 50.2
(4-6) 4.0 (4-7) -4.0
(4-8) 1.5 (5-6) -17.9
(5-7) ±0.5 (5-8) 16.0
(6-7) -19.3 (6-8) 8.0
(7-8) -17.9

500

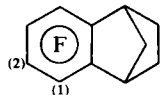
$C_{11}F_4H_6$



(1) -146.2 (2) -160.9

446

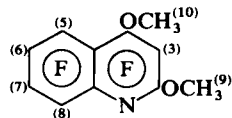
$C_{11}F_4H_8$



(1) -147.3 (2) -160.4

446

$C_{11}F_5H_6NO_2$



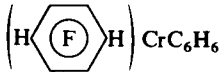
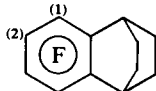
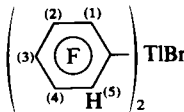
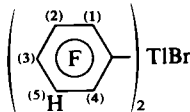
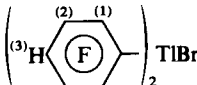
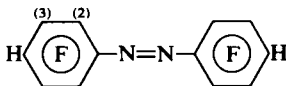
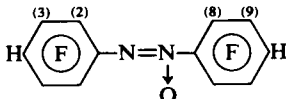
(3) -159.06 (5) -146.38
(7) -157.69 (8) -152.86
(6) -163.28

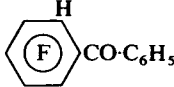
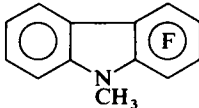
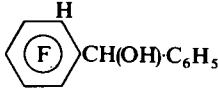
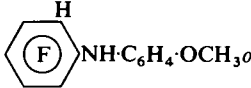
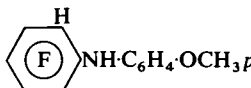
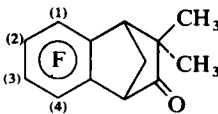
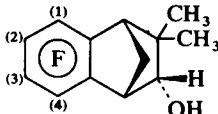
(3-5) +6.45 (3-6) -1.62
(3-7) +6.55 (3-8) -3.20
(5-6) -18.58 (5-7) +1.74
(5-8) +13.96 (6-7) -19.84
(6-8) <0.2 (7-8) -18.53
(3-9) <0.1 (3-10) 4.58
(5-10) 0.38

496

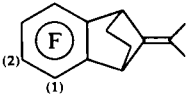
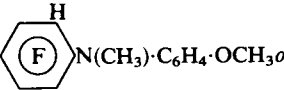
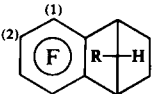
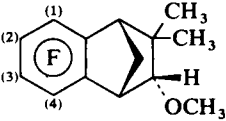
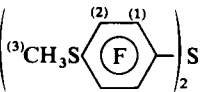
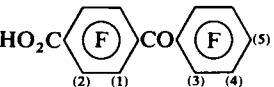
499

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{11}F_5H_6NO_2$		(3) -151.04 (5) -143.39 (6) -165.23 (7) -154.54 (8) -149.12	(3-5) +5.70 (3-6) -1.10 (3-7) +7.19 (3-8) 2.70 (5-6) -20.59 (5-7) +4.23 (5-8) +10.70 (6-7) -21.30 (6-8) -1.15 (7-8) -19.15 (3-9) 0.50 (3-10) 4.40 (5-10) 0.37 (8-9) 9.82	496
$C_{12}F_{14}H_5N_2NaO_2$		(1) -136.6 (2) -138.8		444(b)
$C_{12}F_4H_5N_3$		(1) -152.5 (2) -132.3		447(a)
$C_{12}F_4H_6$		(1) -150.9 (2) -163.5		446
$C_{12}F_4H_6$		2-F, 3-F, 4-F, 5-F (2) -143.8 (3) -156.2 (4) -158.5 (5) -140.5 2-F, 3-F, 4-F, 6-F (2) -135.8 (3) -165.5 (4) -134.1 (6) -118.5 2-F, 3-F, 5-F, 6-F (2) -144.5 (3) -140.0		448
$C_{12}F_4H_6N_2O_2$		(1) -137.6 (2) -132.5		444(b)

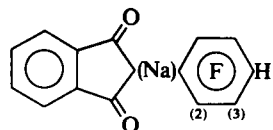
$C_{12}F_4H_8Cr$		-178.0			450
$C_{12}F_4H_{10}$		(1) -150.9 (2) -162.2			446
$C_{12}F_8H_2BrTi$		(1) -117.8 (2) -154.9 (3) -154.3 (4) -138.2	(1-2) 26.8 (1-3) 2.4 (1-4) 13.6 (2-3) 19.3 (2-4) ~1.5 (2-5) ~2 (3-4) 19.7 (3-5) 8.2 (4-5) 8.6 (Ti-1) +1188 (Ti-2) +525 (Ti-3) -93.5 (Ti-4) +205.4	453	
$C_{12}F_8H_2BrTi$		(1) -114.2 (2) -165.5 (3) -131.2 (4) -94.4	(1-2) 25 (1-3) 7 (1-5) ~2 (2-3) 19 (2-5) 7 (2-4) 11 (3-5) 11 (3-4) 4 (4-5) 7 (Ti-1) +792 (Ti-2) +282 (Ti-3) -67 (Ti-4) +709	453	
$C_{12}F_8H_2BrTi$		(1) -120.7 (2) -137.8	(1-3) 7.1 (2-3) 9.8 (Ti-1) 783 (Ti-2) 369	453	
$C_{12}F_8H_2N_2$		(HSO ₃ F solvent) (2) -131.7 (3) -135.9		416(a)	
$C_{12}F_8H_2N_2O$		(HSO ₃ F solvent) (8) -130.1 (9) -134.0 (2) -133.6 (3) -140.4		416(a)	

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{13}F_4H_6O$		-137.1, -138.4, -151.5, -154.7		428
$C_{13}F_4H_7N$		-151.7, -162.5, -165.3, -171.7		451
$C_{13}F_4H_8O$		-139.4, -145.1, -156.6, -157.7		428
$C_{13}F_4H_9NO$		-141.0, -158.0, -160.3, -170.8		451
$C_{13}F_4H_9NO$		-140.8, -158.4, -163.7, -172.4		451
$C_{13}F_4H_{10}O$		(1) -145.0 (2)(3) -156.7 (4) -143.0		446
$C_{13}F_4H_{12}O$		(1) -147.0 (2)(3) -159.3 (4) -146.1		446

$C_{13}F_5H_9N_2O_3$		(2) -142.3 (3) -158.6 (4) -160.4	(2-3) 21.3 (2-5) 9.1 (2-6) 4.1 (3-5) 0.9 (2-7) 11.4 (3-4) 1.6	445(a)
$C_{13}F_8H_2O$		-139.3, -140.6, -149.7, -156.4		428
$C_{13}F_8H_4O$		-138.5, -144.6, -155.6, -155.6		428
$C_{13}F_8H_4O_2$		-162.6, -145.7		426(b)
$C_{13}F_9H_3O$		-164.2, -162.9, -156.3, -145.3, -143.1 (2:2:1:2:2)		426(b)
$C_{13}F_9H_4N$		-140.1, -148.8, -153.4, -156.1, -158.7, -163.0, -166.0 (1:2:1:1:1:2:1)		451
$C_{14}F_4H_6BrNaO$		(2) -140.4 (3) -138.3		272(b)
$C_{14}F_4H_6ClNaO$	As above with Br replaced by Cl	(2) -140.8 (3) -146.0		272(b)
$C_{14}F_4H_7BrO$	As above with Na replaced by H	(2) -141.1 (3) -135.1		272(b)
$C_{14}F_4H_7ClO$	As above with Na replaced by H and Br by Cl	(2) -141.3 (3) -142.6		272(b)
$C_{14}F_4H_7NaO$	As above with Br replaced by H	(2) -142.4 (3) -142.4		272(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_4H_8O$	As above with Br and Na replaced by H	(2) -142.8 (3) -139.9		272(b)
$C_{14}F_4H_{10}$		(1) -147.7 (2) -160.8		462
$C_{14}F_4H_{11}NO$	 $N(CH_3) \cdot C_6H_4 \cdot OCH_3o$	-141.7, -152.5, -158.0, -169.9		451
$C_{14}F_4H_{11}NO$	<i>p</i> - OCH_3 isomer of above	-140.8, -148.1, -158.1, -165.8		451
$C_{14}F_4H_{14}$	 $R = CH(CH_3)_2$	(1) -147.7 (2) -160.3		462
$C_{14}F_4H_{14}O$		(1) -147.5 (2)(3) -160.0 (4) -146.1		446
$C_{14}F_8H_6S_3$		(1)(2) -132.5	(2-3) 1.01	427
$C_{14}F_9HO_3$		(1)(3) -142.1 (2) -139.5 (4) -161.0 (5) -146.8	(3-5) 5.2 (4-5) 18.8	269

C₁₅F₄H₅NaO₂



(2) -137.2 (3) -142.1

272(b)

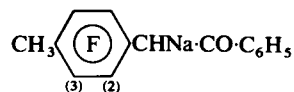
C₁₅F₄H₆O₂

As above with Na replaced by H

(2) -141.4 (3) -138.9

272(b)

C₁₅F₄H₉NaO



(2) -143.4 (3) -147.4

272(b)

C₁₅F₄H₁₀O

As above with Na replaced by H

(2) -144.7 (3) -144.7

272(b)

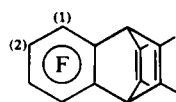
C₁₅F₈H₂O₅



(2) -142.4 (3) -139.9

269

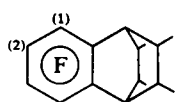
C₁₆F₄H₁₄



(1) -152.2 (2) -163.9

446

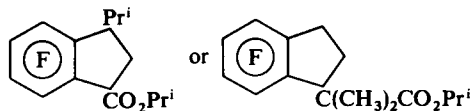
C₁₆F₄H₁₈



(1) -151.8 (2) -161.8

446

C₁₆F₄H₁₈O₂



-137.9, -143.4, -158.2,
-159.0

446

C₁₈F₄H₈Br₂O₂



-155.2

470

C₁₈F₄H₈N₂O₆

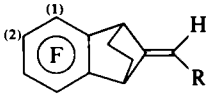
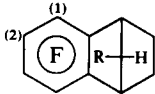
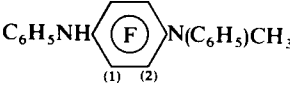
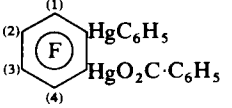
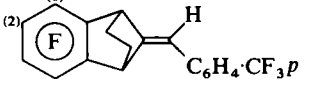
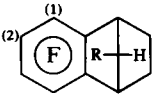
As above with Br replaced by NO₂

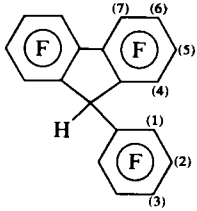
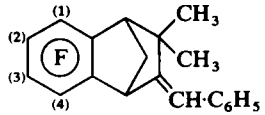
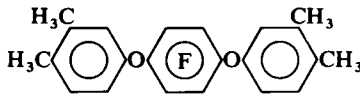
-154.5

470

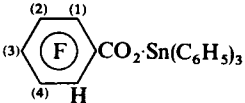
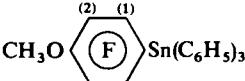
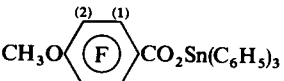
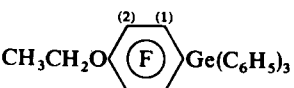
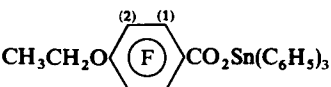
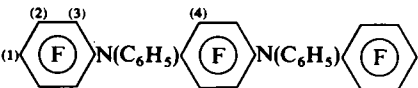
Fluorine nuclei in tetrafluorobenzene derivatives

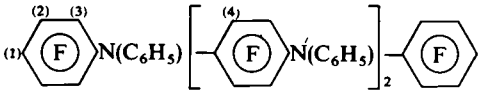
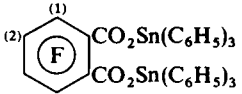
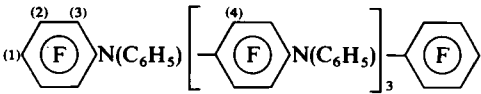
355

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{18}F_4H_{11}NO_2$		$R = pNO_2 \cdot C_6H_4$ (1) -147.5 (2) -158.8		462
$C_{18}F_4H_{12}$	As above with $R = C_6H_5$	(1) -146.8 (2) -159.6		462
$C_{18}F_4H_{13}N$	As above with $R = pNH_2 \cdot C_6H_4$	(1) -147.0 (2) -159.7		462
$C_{18}F_4H_{14}$		$R = CH_2C_6H_5$ (1) -147.1 (2) -159.4		462
$C_{19}F_4H_9N_2$		(1) -151.2 (2) -147.9		452
$C_{19}F_4H_{10}Hg_2O_2$		(1) -131.3 (2)(3) -150.2, -155.4 (4) -115.0		473
$C_{19}F_7H_{11}$		(1) -147.0 (2) -159.3		462
$C_{19}F_7H_{13}$		$R = pCF_3 \cdot C_6H_4$ (1) -146.9 (2) -159.1		462

$C_{19}F_{13}H$		(1) -142.0 (2) -161.4 (3) -153.8 (4) -143.2 (5) -153.8 (6) -153.8 (7) -134.6	474(b)
$C_{20}F_4H_{10}O_6$	$mHO_2C \cdot C_6H_4 \cdot O \cdot \text{C}_6\text{H}_4(F) \cdot O \cdot C_6H_4 \cdot CO_2Hm$	-151.3	470
$C_{20}F_4H_{14}N_2O$	$C_6H_5NH \cdot \text{C}_6\text{H}_4(F) \cdot N(C_6H_5)CO \cdot CH_3$ (1) (2)	(1) -151.45 (2) -147.55	452(b)
$C_{20}F_4H_{14}O_2$	$mCH_3 \cdot C_6H_4 \cdot O \cdot \text{C}_6\text{H}_4(F) \cdot O \cdot C_6H_4 \cdot CH_3m$	-149.6	470
$C_{20}F_4H_{16}$		(1) -146.3 (2) -159.5 (3) -158.5 (4) -144.0	446
$C_{20}F_8H_7NaO$	$\left(H \cdot \text{C}_6\text{H}_4(F) \right)_2 \text{CNa} \cdot CO \cdot C_6H_5$ (3) (2)	(2) -137.1 (3) -142.1	272(b)
$C_{20}F_8H_8O$	As above with Na replaced by H	(2) -138.5 (3) -140.1	272(b)
$C_{22}F_4H_{18}O_2$		-150.5	470

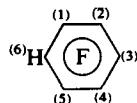
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{22}F_{12}H_{11}O_2Tl$		(1) -119.3 (2) -137.6	(Tl-1) 529 (Tl-2) 232	453
$C_{22}F_{12}H_{11}O_2Tl$		(1) -112.1 (2) -165.2 (3) -131.8 (4) -93.4	(Tl-1) 530 (Tl-2) 472 (Tl-3) 46 (Tl-4) 175	453
$C_{24}F_4H_{16}Ge$		(1) -123.6 (2) -138.4		473
$C_{24}F_4H_{16}N_2$		(1) -151.10 (2) -147.75		452(b)
$C_{24}F_4H_{16}Pb$		(1) -119.2 (2) -137.8		473
$C_{24}F_4H_{16}Sn$	As above with Pb replaced by Sn	(1) -120.0 (2) -137.8		473
$C_{24}F_6H_{10}Fe_2N_2O_4$	 R = Fe(CO) ₂ C ₅ H ₅	(2) -145.7 (3) -108.6 (7) -114.5 (8) -55.0	(2-7) 11.5 (7-8) 30.6	445(a)
$C_{24}F_9H_{13}Cr_2$		-185, -195 to -205 (ratio 2:7)		450

$C_{25}F_4H_{16}O_2Sn$		(1)(4) -137.4 , -139.4 (2)(3) -152.3 , -155.2	473
$C_{25}F_4H_{18}OSn$		(1) -120.3 (2) -156.2	473
$C_{26}F_4H_{18}O_3Sn$		(1) -139.7 (2) -158.5	473
$C_{26}F_4H_{20}GeO$		(1) -124.5 (2) -156.3	473
$C_{26}F_4H_{20}OSn$	As above with Ge replaced by Sn	(1) -120.5 (2) -155.4	473
$C_{27}F_4H_{20}O_3Sn$		(1) -140.0 (2) -157.9	473
$C_{30}F_4H_{20}N_2$		-146.80	471(b)
$C_{30}F_{14}H_{10}N_2$		(1) -157.40 (2) -162.35 (3)(4) -146.55 , -146.75	471(b)
$C_{42}F_4H_{30}Ge_2$		-122.6	473
$C_{42}F_4H_{30}Pb_2$	As above with Ge replaced by Pb	-117.7	473

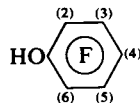
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{42}F_4H_{30}Sn_2$	As above with Ge replaced by Sn	-118.6		473
$C_{42}F_{18}H_{15}N_3$		(1) -157.35 (2) -162.30 (3) -146.65 (4) -147.00		471(b)
$C_{44}F_4H_{30}O_4Sn_2$		(1) -137.7 (2) -151.9		473
$C_{63}F_{22}H_{20}N_4$		(1) -157.35 (2) -162.30 (3) -146.60 (4) -146.95, -147.05		471(b)

E. Fluorine nuclei in pentafluorobenzene derivatives

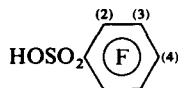
C_6F_5H



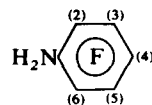
(1-2) -20.566	(1-3) 1.213	402
(1-4) 8.782	(1-5) -2.041	417
(1-6) 10.416	(2-3) -18.747	
(2-4) -1.230	(2-6) 7.004	
(3-6) -2.656	(1-C1) -248.09	
(1-C2) 16.67	(1-C3) 5.17	
(1-C4) 5.21	(1-C5) 12.32	
(1-C6) 23.44	(2-C1) 11.22	
(2-C2) -249.76	(2-C3) 13.39	
(2-C4) 1.41	(2-C5) 4.25	
(2-C6) 0.50	(3-C1) 3.78	
(3-C2) 12.86	(3-C3) -252.89	
(3-C6) 3.56		



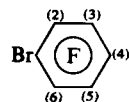
(2-3) -20.65	(2-4) -5.73	417
(2-5) 4.40	(2-6) 2.24	
(3-4) -20.75	(3-5) -3.19	
(2-C1) 14.27	(2-C2) -242.76	
(2-C3) 14.57	(2-C4) 3.82	
(2-C5) 4.96	(2-C6) 5.25	
(3-C1) 2.38	(3-C2) 12.78	
(3-C3) -247.48	(3-C4) 13.33	
(3-C5) 3.72	(3-C6) 4.57	
(4-C1) 4.08	(4-C2) 3.24	
(4-C3) 13.44	(4-C4) -245.10	



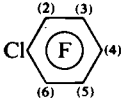
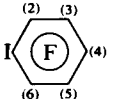
(2)(3) -147.70, -169.97	415(a)
(4) -158.88	

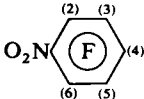
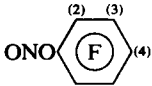
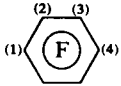
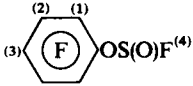
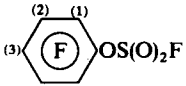


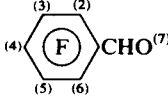
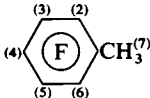
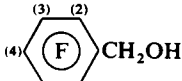
(2-3) -20.52	(2-4) -7.33	417
(2-5) 5.13	(2-6) 4.64	
(3-4) -20.75	(3-5) -2.40	
(2-C1) 14.52	(2-C2) -238.90	
(2-C3) 14.65	(2-C4) 4.30	
(2-C5) 4.71	(2-C6) 7.21	
(3-C1) 2.10	(3-C2) 13.23	
(3-C3) -245.42	(3-C4) 13.94	
(3-C5) 3.58	(3-C6) 4.21	
(4-C1) 3.29	(4-C2) 3.29	
(4-C3) 13.38	(4-C4) -241.64	



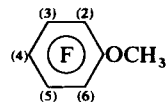
(2) -132.7	(3) -160.9	396
(4) -154.9		

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_6F_5Br	As above		(2-3) -21.47 (2-4) 1.22 (2-5) 6.28 (2-6) -5.30 (3-4) -19.62 (3-5) -1.65 (2-C1) 23.02 (2-C2) -247.71 (2-C3) 17.29 (2-C4) 4.40 (2-C5) 5.24 (2-C6) 4.07 (3-C1) 0.00 (3-C2) 12.54 (3-C3) -253.53 (3-C4) 13.56 (3-C5) 2.17 (3-C6) 4.68 (4-C1) 5.10 (4-C2) 3.47 (4-C3) 13.36 (4-C4) -254.61	417
C_6F_5Cl			(2-3) -20.63 (2-4) 0.47 (2-5) 5.99 (2-6) -5.29 (3-4) -19.77 (3-5) -1.96 (2-C1) 19.44 (2-C2) -249.91 (2-C3) 16.24 (2-C4) 4.18 (2-C5) 5.24 (2-C6) 3.18 (3-C1) 1.07 (3-C2) 12.85 (3-C3) -252.51 (3-C4) 13.63 (3-C5) 2.48 (3-C6) 5.14 (4-C1) 5.14 (4-C2) 3.31 (4-C3) 13.43 (4-C4) -254.34	417
C_6F_5I			(2-3) -22.91 (2-4) 1.97 (2-5) 7.28 (2-6) -4.72 (3-4) -19.46 (3-5) -1.25 (2-C1) 28.44 (2-C2) -244.58 (2-C3) 18.83 (2-C4) 4.79 (2-C5) 5.21 (2-C6) 6.15 (3-C1) 1.44 (3-C2) 11.96 (3-C3) -254.66 (3-C4) 13.57 (3-C5) 1.69 (3-C6) 4.48 (4-C1) 4.75 (4-C2) 3.59 (4-C3) 13.24 (4-C4) -254.84	417

$C_6F_5NO_2$		(2) -146.4 (3) -158.6 (4) -147.6		410
$C_6F_5NO_2$	As above		(2-3) -21.24 (2-4) 5.40 (2-5) 6.68 (2-6) -9.98 (3-4) -19.94 (3-5) -0.37 (2-C2) -262.12 (2-C3) 15.39 (2-C4) 4.26 (2-C5) 5.31 (2-C6) 2.30 (3-C2) 14.06 (3-C3) -255.29 (3-C4) 13.42 (3-C5) 1.82 (3-C6) 2.30 (4-C2) 3.79 (4-C3) 13.41 (4-C4) -261.95	417
$C_6F_5NO_2$		(2) -160.2 (3) -161.0 (4) -143.3		410
C_6F_6			(1-2) -20.282 (1-3) -3.003 (1-4) 3.869 (1-C1) -251.13 (1-C2) 13.73 (1-C3) 3.19 (1-C4) 5.21	402 417
C_6F_6	As above		(1-2) -21.65 (1-3) -2.84 (1-4) 3.98	418
$C_6F_6O_2S$		(1) -151.0 (2) -162.0 (3) -155.7 (4) +69.6	(1-2) 16.8 (1-4) 9.0 (2-3) 20.8	419
$C_6F_6O_3S$		(1) -151.8 (2) -160.6 (3) -153.3 (4) +42.0	(1-2) 16.2 (1-4) 11.2 (2-3) 20.5	419

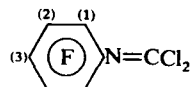
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C_7F_5HO			(2-3) -20.18 (2-4) 6.46 (2-5) 9.49 (2-6) -4.82 (3-4) -19.13 (3-5) -1.11 (2-7) 1.05 (3-7) -1.34 (4-7) 0.21 (2-C1) 10.21 (2-C2) -261.33 (2-C3) 16.16 (2-C4) 5.29 (2-C5) 5.28 (2-C6) 6.69 (3-C1) 1.86 (3-C2) 11.47 (3-C3) -252.90 (3-C4) 13.46 (3-C5) 1.33 (3-C6) 4.32 (4-C1) 3.83 (4-C2) 4.12 (4-C3) 12.98 (4-C4) -261.22	417
$C_7F_5H_3$			(2-3) -20.89 (2-4) -0.06 (2-5) 8.55 (2-6) -0.63 (3-4) -19.43 (3-5) -2.27 (2-7) 2.28 (3-7) 0.25 (4-7) 1.38 (2-C1) 19.49 (2-C2) -244.66 (2-C3) 17.47 (2-C4) 5.19 (2-C5) 5.09 (2-C6) 8.76 (3-C1) 0.86 (3-C2) 11.02 (3-C3) -248.70 (3-C4) 13.54 (3-C5) 2.06 (3-C6) 4.26 (4-C1) 3.94 (4-C2) 3.61 (4-C3) 12.76 (4-C4) -250.03	417
$C_7F_5H_3O$		(2)(3) -150.0, -168.0 (4) -159.75		354

C₇F₅H₃O



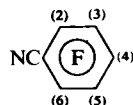
(2-3) -20.44	(2-4) -3.20	417
(2-5) 4.65	(2-6) -0.44	
(3-4) -20.62	(3-5) -3.36	
(2-C1) 12.51	(2-C2) -247.70	
(2-C3) 14.97	(2-C4) 3.95	
(2-C5) 5.03	(2-C6) 5.22	
(3-C1) 1.75	(3-C2) 12.00	
(3-C3) -248.67	(3-C4) 13.84	
(3-C5) 3.66	(3-C6) 4.48	
(4-C1) 4.49	(4-C2) 3.36	
(4-C3) 13.21	(4-C4) -248.21	

C₇F₅Cl₂N



(1) -150.1	(2) -162.7	271(b)
(3) -159.1		

C₇F₅N



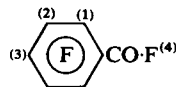
(2) -131.9	(3) -163.5	410
(4) -143.4		

C₇F₅N

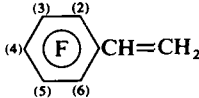
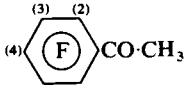
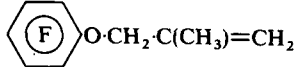
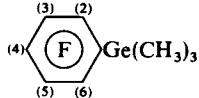
As above

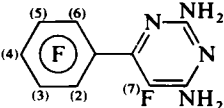
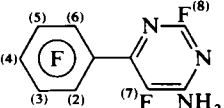
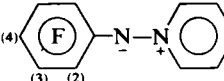
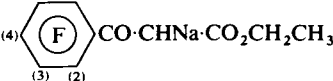
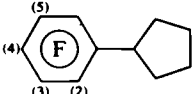
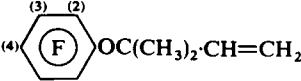
(2-3) -20.12	(2-4) 5.88	417
(2-5) 8.05	(2-6) -7.85	
(3-4) -19.47	(3-5) 0.28	
(2-C2) -260.00	(2-C3) 15.01	
(2-C4) 4.94	(2-C5) 5.11	
(2-C6) 4.98	(3-C2) 13.04	
(3-C3) -253.59	(3-C4) 13.33	
(3-C5) 1.97	(3-C6) 4.03	
(4-C2) 4.29	(4-C3) 13.32	
(4-C4) -262.85		

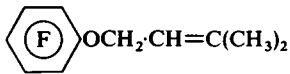
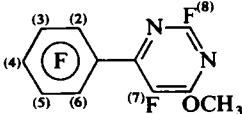
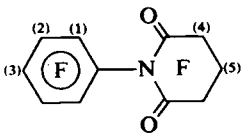
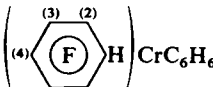
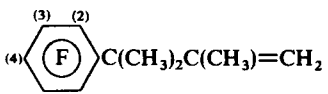
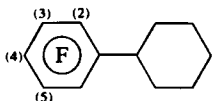
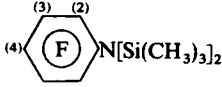
C₇F₆O

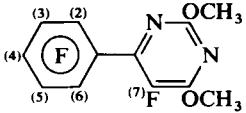
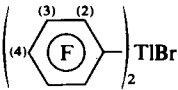
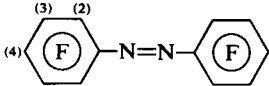
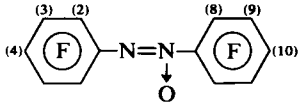
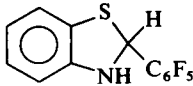
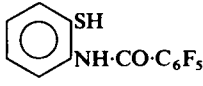
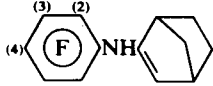
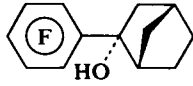


(1) -154.1	(2) -161.8	(1-2) 15.6	(2-3) 20.6	429
(3) -156.1	(4) -22.6			

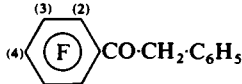
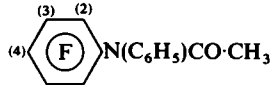
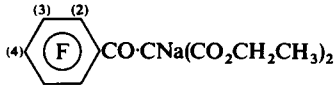
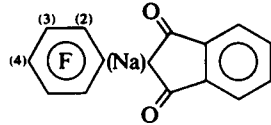
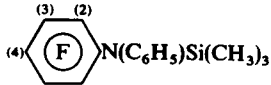
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_8F_5H_3$			(2-3) -20.47 (2-4) 1.37 (2-5) 8.22 (2-6) -2.20 (3-4) -19.66 (3-5) -2.00 (2-C1) 13.95 (2-C2) -250.41 (2-C3) 17.30 (2-C4) 13.67 (2-C5) 5.13 (2-C6) 7.89 (3-C1) 0.79 (3-C2) 11.49 (3-C3) -249.37 (3-C4) 13.67 (3-C5) 2.22 (3-C6) 4.32 (4-C1) 4.26 (4-C2) 3.35 (4-C3) 12.88 (4-C4) -253.75	417
$C_8F_5H_3O$		(2) -154.0 (3) -164.1 (4) -159.6	(2-3) 19.0 (3-4) 22.0	429
$C_9F_5H_7O$		-159.7 and -167.8 (2:3)		329
$C_9F_5H_9Ge$		(2) -127.9 (3) -161.51 (4) -153.4	(2-3) -23.9 (2-4) 2.5 (2-5) 10.7 (2-6) -4.2 (3-4) -20.2 (3-5) -1.2	438(b)
$C_9F_5H_9Si$	As above with Ge replaced by Si	(2) -127.8 (3) -161.75 (4) -152.5	(2-3) -23.9 (2-4) 3.3 (2-5) 10.7 (2-6) -4.2 (3-4) -19.2 (3-5) -1.2	438(b)
$C_9F_5H_9Sn$	As above with Ge replaced by Sn	(2) -121.9 (3) -160.66 (4) -152.96	(2-3) -27.25 (2-4) 1.97 (2-5) 11.58 (2-6) -4.38 (3-4) -19.7 (3-5) -1.37	438(b)

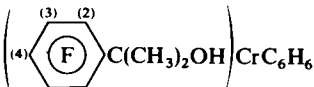
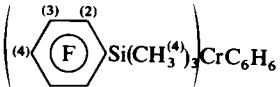
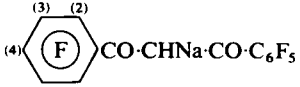
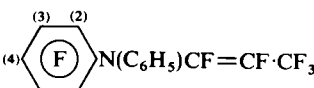
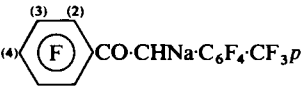
$C_{10}F_6H_4N_4$		(2) -142.8 (3) -165.0 (4) -156.3 (7) -168.4	(2-3) 21.7 (2-4) 2.1 (2-5) 8.1 (2-6) 3.4 (3-5) 0.8 (3-4) 20.5 (2-7) 7.9	445(a)
$C_{10}F_7H_2N_3$		(2) -142.3 (3) -164.4 (4) -154.4 (7) -156.5 (8) -51.7	(2-3) 21.5 (2-4) 3.2 (2-5) 7.9 (2-6) 4.4 (2-7) 10.9 (3-4) 19.9 (3-5) 1.2 (7-8) 27.6	445(a)
$C_{10}F_8N_2$	As above with NH_2 replaced by F(9)	(2) -142.5 (3) -164.1 (4) -152.0 (7) -154.4 (8) -48.7 (9) -75.4	(2-3) 20.4 (2-4) 4.2 (2-5) 8.2 (2-6) 5.8 (2-7) 13.0 (3-4) 19.0 (7-8) 28.0 (7-9) 20.3 (8-9) 2.4	445(a)
$C_{11}F_5H_5N_2$		(2) -153.5 (3) -167.0 (4) -177.4		447(a)
$C_{11}F_5H_6NaO_3$		(2) -142.9 (3) -163.8 (4) -160.0	(3-4) 20.5	272(b)
$C_{11}F_5H_7O_3$	As above with Na replaced by H (50% enol)	(2) -139.4, -140.4 (3) -161.7 (4) -152.0, -149.9	(2-4) 3.5 (3-4) 21	272(b)
$C_{11}F_5H_9$		(2) -145.5 (3) -165.75 (4) -152.25	(2-3) 24 (2-5) 9	375
$C_{11}F_5H_9O$		(2) -153.4 (3) -165.7 (4) -167.8	(2-3) 20.0 (3-4) 20.0 (2-4) 3.0	329

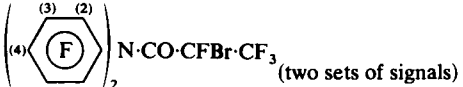
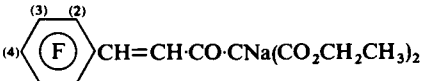
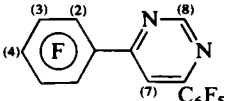
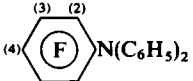
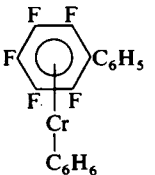
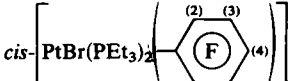
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{11}F_5H_9O$		-158.4, -167.0 (2:3)		329
$C_{11}F_7H_3O$		(2) -140.3 (3) -162.1 (4) -151.3 (7) -153.6 (8) -48.0	(2-3) 21.8 (2-4) 3.5 (2-5) 8.2 (2-6) 4.5 (3-4) 19.7 (3-5) 0.5 (2-7) 11.7 (7-8) 27.4	445(a)
$C_{11}F_{11}NO_2$		(1) -143.4 (2) -147.3 (3) -159.4 (4) -121.3 (5) -137.1		219(a)
$C_{12}F_5H_7Cr$		(2) -185.0 (3) -194.9 (4) -196.8		450
$C_{12}F_5H_{11}$		(2) -152 (3) -177 (4) -172	(2-3) 20 (3-4) 20	249
$C_{12}F_5H_{11}$		(2) -148.5 (3) -169.5 (4) -165	(2-3) 24 (2-5) 9 (3-4) 24	375
$C_{12}F_5H_{18}NSi_2$		(2) -147.9 (3) -162.75 (4) -165.4		452

$C_{12}F_6H_6N_2O_2$		(2) -140.5 (3) -162.6 (4) -152.7 (7) -160.4	(2-3) 21.3 (2-4) 2.6 (2-5) 8.1 (2-6) 4.3 (3-4) 20.6 (3-5) 0.8 (2-7) 10.7	445(a)
$C_{12}F_6H_6Cr$	$C_6F_5 \cdot Cr \cdot C_6H_5$	-193.3	$^1J(F-C)$ 303	624
$C_{12}F_{10}BrTi$		(2) -119.7 (3) -160.4 (4) -152.5	(2-4) ~0 (3-4) 19.1 (TI-2) 780 (TI-3) 339 (TI-4) 82	453
$C_{12}F_{10}N_2$		(HSO ₃ F solvent) (2) -133.9 (3) -154.4 (4) -123.7		416(a)
$C_{12}F_{10}N_2O$		(HSO ₃ F solvent) (2) -140.0 (3) -157.4 (4) -141.6 (8) -133.1 (9) -154.4 (10) -134.1		416(a)
$C_{13}F_5H_4NS$		-139.0, -151.0, -161.0		252
$C_{13}F_5H_6NOS$		-131.0, -143.0, -152.0		252
$C_{13}F_5H_{10}N$		(2) -161.5 (3) -166.5 (4) -175.0		455(a) (?)
$C_{13}F_5H_{11}O$		Centre of signals -138.9		456

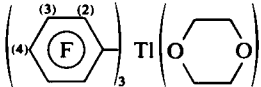
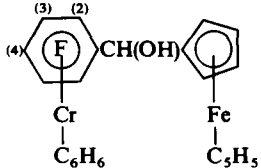
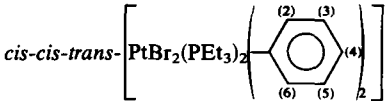
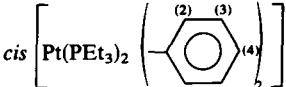
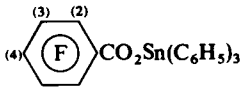
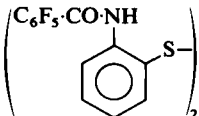
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{13}F_5H_{12}NO$		(2) -163.0 (3) -166.5 (4) -175.0		455(a)
$C_{13}F_9HNNaO$		(2) -142.8 (3) -163.7 (4) -160.0 (8) -145.2 (9) -98.4		272(b)
$C_{13}F_9HN_2$		(2) -139.2 (3) -160.7 (4) -151.6 (8) -139.6 (9) -89.7	(2-4) 3.2 (3-4) 21	444(b)
$C_{13}F_9H_2NO$		(2) -139.9 (3) -160.8 (4) -148.7 (8) -142.9 (9) -91.8	(3-4) 20.5	272(b)
$C_{13}F_9N_2Na$		(2) -140.5 (3) -166.6 (4) -165.1 (8) -153.1 (9) -100.6	(3-4) 20	444(b)
$C_{13}F_{10}HNO$		(two sets of signals) [a] (2) -146.8 (3) -161.8 (4) -152.7 [b] (2) -144.0 (3) -160.5 (4) -152.0	(2-4) 2.6	263(b)
$C_{13}F_{10}O_2$		(2) -153.9 (3) -162.2 (4) -156.7	(2-3) 16.5 (3-4) 20.7	429
$C_{14}F_5H_6NaO$		(2) -142.0 (3) -166.4 (4) -170.3	(2-4) 3.5 (3-4) 20.5	272(b)

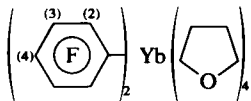
$C_{14}F_5H_7O$	As above with Na replaced by H	(2) -142.3 (4) -157.0	(3) -163.7	(3-4) 20.5	272(b)
$C_{14}F_5H_7O$		(2) -141.1 (4) -151.0	(3) -161.1	(2-4) 3.4 (3-4) 20	272(b)
$C_{14}F_5H_8NO$		(2) -145.5 (4) -163.1	(3) -155.85		452(b)
$C_{14}F_5H_{10}NaO_5$		(2) -143.9 (4) -161.6	(3) -165.3		272(b)
$C_{14}F_5H_{11}O_5$	As above with Na replaced by H	(2) -140.0 (4) -151.8	(3) -162.0	(2-4) 2.5 (3-4) 22	272(b)
$C_{14}F_{10}HNaO$		(2) -140.6 (4) -169.2 (7) -164.1	(3) -166.4 (6) -142.8 (8) -160.9	(2-4) 3.5 (3-4) 20.4 (6-8) 0 (7-8) 20	272(b)
$C_{14}F_{10}H_2O$	As above with Na replaced by H	(2) -141.9 (4) -155.6 (7) -160.9	(3) -163.1 (6) -140.3 (8) -149.5	(2-4) 0 (3-4) 20 (6-8) 4.5 (7-8) 20.4	272(b)
$C_{15}F_5H_4NaO_2$		(2) -136.1 (4) -161.9	(3) -165.7		272(b)
$C_{15}F_5H_5O_2$	As above with Na replaced by H	(2) -140.6 (4) -154.3	(3) -162.3	(2-4) 1.5 (3-4) 20.7	272(b)
$C_{15}F_5H_{11}NSi$		(2) -147.1 (4) -163.85	(3) -158.9		452(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{15}F_5H_{13}CrO$		(2) -182.3 (2) -194.5 (3) -198.8		450
$C_{15}F_5H_{15}CrSi$		(2) -175.2 (2) -193.7 (3) -200.7	(2-4) 1.5	450
$C_{15}F_{10}HNaO_2$		(2) -142.7 (3) -163.1 (4) -158.1	(2-4) 0 (3-4) 19.8	272(b)
$C_{15}F_{10}H_2O_2$	As above with Na replaced by H	(2) -139.1 (3) -161.1 (4) -149.5	(2-4) 4.4 (3-4) 20.5	272(b)
$C_{15}F_{10}H_3NaO$		(2) -142.6 (3) -166.8 (4) -171.7	(2-4) 4.5 (3-4) 21.7	272(b)
$C_{15}F_{10}H_4O$	As above with Na replaced by H	(2) -142.2 (3) -163.3 (4) -156.5	(2-4) 0 (3-4) 21	272(b)
$C_{15}F_{10}H_5N$		(2) -146.7 (3) -161.7 (4) -155.0		263(b)
$C_{15}F_{12}HNaO$		(2) -142.9 (3) -164.0 (4) -160.4	(2-4) 0 (3-4) 21	272(b)
$C_{15}F_{12}H_2O$	As above with Na replaced by H	(2) -139.9 (3) -160.8 (4) -148.9	(2-4) 4.7 (3-4) 21	272(b)

$C_{15}F_{14}BrNO$		[a] (2) -143.3 (3) -160.9 (4) -151.3 [b] (2) -141.3 (3) -160.5 (4) -149.5			263(b)
$C_{16}F_5H_{12}NaO_5$		(2) -141.6 (3) -163.7 (4) -158.0	(2-4) 0 (3-4) 21.5		272(b)
$C_{16}F_5H_{13}O_5$	As above with Na replaced by H (enol form)	(2) -140.8 (3) -162.4 (4) -153.1	(2-4) 2.3 (3-4) 21.0		272(b)
$C_{16}F_{12}N_2$		(2) -142.1 (3) -163.9 (4) -151.7 (7) -134.5 (8) -51.3	(2-3) 22.4 (2-4) 3.6 (2-5) 8.3 (2-6) 5.8 (3-4) 19.8 (2-7) 15.0 (7-8) 28.6		445(a)
$C_{17}F_7H_5FeN_2O_2$	As above with C_6F_5 replaced by $Fe(CO)_2C_5H_5$	(2) -142.5 (3) -164.6 (4) -154.6 (7) -114.3 (8) -54.7	(2-3) 20.9 (2-4) 2.7 (2-5) 7.7 (2-6) 3.7 (3-4) 20.3 (2-7) 11.8 (3-7) 0.8 (7-8) 30.1		445(a)
$C_{18}F_5H_{10}N$		(2) -145.30 (3) -163.05 (4) -159.10			471(b)
$C_{18}F_5H_{11}Cr$		-181.9, -194.5, -196.6 (ratio 2:2:1)			624
$C_{18}F_5H_{30}BrP_2Pt$		(2) -119.0 (3) -164.3 (4) -162.8	(2-3) 21 (3-4) 20 (Pt-2) 309 (Pt-3) 62		472

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{18}F_5H_{30}Br_3P_2Pt$	<i>mer-trans</i> - $\left[PtBr_3(PEt_3)_2\right]\left(\begin{array}{c} (2) \quad (3) \\ \text{F} \\ (6) \quad (5) \end{array} \begin{array}{c} (4) \end{array}\right)$	(2) -106.5 (3) -161.4 (4) -157.3	(2-3) 22 (3-4) 21 (Pt-2) 93 (Pt-3) 38	472
$C_{18}F_5H_{30}Br_3P_2Pt$	As above but <i>mer-cis</i> -compound (at 180 K)	(2) -108.4 (3) -162.1 (4) -157.9 (5) -161.4 (6) -105.6	(2-3) 29 (3-4) 22 (5-6) 29 (Pt-2) ~85 (Pt-6) ~60	472
$C_{18}F_5H_{30}ClP_2Pt$	<i>cis</i> - $\left[PtCl(PEt_3)_2\right]\left(\begin{array}{c} (2) \quad (3) \\ \text{F} \\ (6) \quad (5) \end{array} \begin{array}{c} (4) \end{array}\right)$	(2) -119.6 (3) -163.7 (4) -162.2	(2-3) 20 (3-4) 20 (Pt-2) 307 (Pt-3) 70	472
$C_{18}F_5H_{30}Cl_3P_2Pt$	<i>mer-trans</i> - $\left[PtCl_3(PEt_3)_2\right]\left(\begin{array}{c} (2) \quad (3) \\ \text{F} \\ (6) \quad (5) \end{array} \begin{array}{c} (4) \end{array}\right)$	(2) -111.5 (3) -162.8 (4) -159.0	(2-3) 22 (3-4) 21 (Pt-2) 85 (Pt-3) 39 (Pt-4) 20	472
$C_{18}F_5H_{30}Cl_3P_2Pt$	As above but <i>mer-cis</i> -compound (at 180 K)	(2) -115.3 (3) -162.0 (4) -157.9 (5) -161.3 (6) -113.9	(2-3) 27 (3-4) 20 (5-6) 23 (4-5) 20 (Pt-2) ~65 (Pt-6) ~50	472
$C_{18}F_5H_{30}IP_2Pt$	<i>cis</i> - $\left[PtI(PEt_3)_2\right]\left(\begin{array}{c} (2) \quad (3) \\ \text{F} \\ (6) \quad (5) \end{array} \begin{array}{c} (4) \end{array}\right)$	(2) -116.5 (3) -165.0 (4) -163.9	(2-3) 23 (3-4) 20 (Pt-2) 311 (Pt-3) 61	472
$C_{18}F_{10}H_5N$	$\left(\begin{array}{c} (3) \quad (2) \\ \text{F} \\ (4) \end{array} \begin{array}{c} \text{NC}_6\text{H}_5 \end{array}\right)_2$	(2) -146.50 (3) -162.4 (4) -157.35		471(b)
$C_{20}F_{10}H_5NaO$	$\left(\begin{array}{c} (3) \quad (2) \\ \text{F} \\ (4) \end{array} \begin{array}{c} \text{CNa}\cdot\text{CO}\cdot\text{C}_6\text{H}_5 \end{array}\right)_2$	(2) -139.2 (3) -166.3 (4) -164.4	(3-4) 20.7	272(b)
$C_{20}F_{10}H_6O$	As above with Na replaced by H	(2) -139.2 (3) -162.0 (4) -153.9	(2-4) 1.5 (3-4) 20.5	272(b)

$C_{22}F_{15}H_8O_2Ti$		(2) -118.1 (4) -151.5	(3) -159.3	(Ti-2) 546 (Ti-4) 58	(Ti-3) 217	453
$C_{23}F_5H_{17}CrFeO$		(2) -187.3 (4) -200.6	(3) -197.3			450
$C_{24}F_{10}H_{30}Br_2P_2Pt$	$cis-cis-trans-\left[PtBr_2(PEt_3)_2\left(\text{C}_6\text{H}_4\right)_2\right]$ 	(2) -115.9 (4) -157.8 (6) -100.8	(3) -160.9 (5) -160.7	(2-3) 25 (2-5) = (3-5) = (3-6) ~5 (5-6) 25 (Pt-3) 45 (Pt-5) 40	(3-4) ≈ (4-5) 21 (Pt-2) 136 (Pt-4) 19 (Pt-6) 102	472
$C_{24}F_{10}H_{30}ClP_2Pt$	As above with Br replaced by Cl	(2) -116.3 (4) -157.9 (6) -106.7	(3) -163.2 (5) -160.9	(2-3) 24 (2-5) = (3-5) = (3-6) ~5 (5-6) 26 (Pt-3) 45 (Pt-5) 40	(3-4) ≈ (4-5) 20 (Pt-2) 145 (Pt-4) 18 (Pt-6) 95	472
$C_{24}F_{10}H_{30}P_2Pt$	$cis\left[Pt(PEt_3)_2\left(\text{C}_6\text{H}_4\right)_2\right]$ 	(2) -117.6 (4) -163.0	(3) -163.7	(3-4) 20 (Pt-3) 68	(Pt-2) 354 (Pt-4) 13	472
$C_{25}F_5H_{15}O_2Sn$		(2) -138.3 (4) -150.9	(3) -161.8			473
$C_{26}F_{10}N_2O_2S_2$		-141.0, -153.0, -162.0				252

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{28}F_{10}H_{32}O_4Yb$		(2) -108.3 (3)(4) -161.4		478

F. Miscellaneous studies

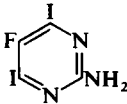
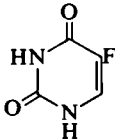
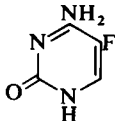
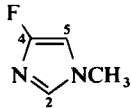
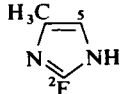
A review of fluorine coupling constant data (481) has enabled the additivity of substituent effects $J(F-H)$ in polysubstituted fluorobenzenes to be demonstrated (482) and those upon $J(F-F)$ to be updated. (483)

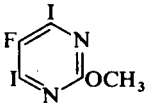
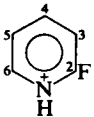
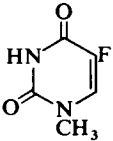
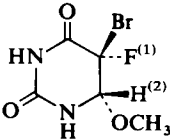
Solvent effects upon the shifts of hexafluorobenzene, (484) fluoroaniline, (479) fluoronitrobenzene, (479) and 2,3,4,5,6-pentafluorodiphenylamine (485) have been reported. A dynamic polarization study of C_6F_6 , doped with free radicals, has been performed. (310)

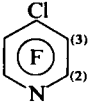
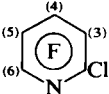
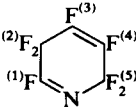
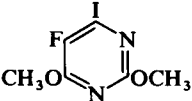
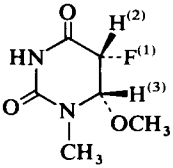
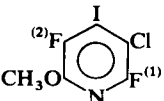
VI. FLUORINE BONDED TO A CARBON OF A HETEROCYCLIC SYSTEM

This Page Intentionally Left Blank

A. Nitrogen heterocycles

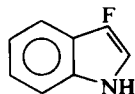
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_4FH_2I_2N_3$		-117.0		486(a)
$C_4FH_3N_2O_2$		-171		487
$C_4FH_4N_3O$		-171		487
$C_4FH_5N_2$			(2-4) 1.8 (4-5) 8.0	488
$C_4FH_5N_2$	As above with F at C2		(2-4) 1.6 (2-5) ~0	488
$C_4FH_5N_2$	As above with F at C5		(4-5) 7.5 (2-5) ~0	488
$C_4FH_5N_2$			(2-5) 1.3	488

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_5FH_3I_2N_2O$		-109.5		486(a)
C_5FH_4N			(F-C2) 238.3 (F-C3) 37.8 (F-C4) 8.0 (F-C5) 3.8 (F-C6) 15.5	431
C_5FH_4N	As above with F at C3		(F-C2) 23.3 (F-C3) 258.8 (F-C4) 18.2 (F-C5) 3.8 (F-C6) 3.6	431
$C_5FH_5N^+$			(F-C2) 250.8 (F-C3) 32.6 (F-C4) 8.9 (F-C5) 4.4 (F-C6) 10.0	431
$C_5FH_5N^+$	As above with F at C3		(F-C2) 34.4 (F-C3) 257.2 (F-C4) 18.2 (F-C5) 7.1 (F-C6) 3.6	431
$C_5FH_5N_2O_2$		-170		487
$C_5FH_6BrN_2O_3$		-139	(1-2) ~0.7	487

$C_5FH_7N_2O_3$	As above with Br replaced by H(3)	-208		(1-2) 2·2	(1-3) 45·5	487
$C_5F_2H_6N_2O_3$	As above with Br replaced by F(3)	(1) -131	(3) -112	(1-2) 2 (2-3) 6	(1-3) 274	487
C_5F_4ClN		(2) -87·5	(3) -141·5			4
C_5F_4ClN		(3) -141·6 (5) -160·8	(4) -138·3 (6) -86·2	(3-4) 17·6 (3-6) 25·6 (4-6) 15·7	(3-5) 4·2 (4-5) 17·0 (5-6) 22·2	320(a)
C_5F_7N		(1) -57·8 (3) -159·1 (5) -85·9	(2) -113·2 (4) -144·5			489
$C_6FH_6IN_2O_2$		-145·0				486(a)
$C_6FH_9N_2O_3$		-209		(1-2) 46	(1-3) ~2	487
$C_6F_2H_3ClIN$		(1) -74·2	(2) -119·1	(1-2) 27		490(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_8HN_3$		(1) -106.9 (2)(3) -132.5, -134.8 (4) -92.4	(1-5) 2	142
$C_6F_9H_2N$		(1) -164.3 (2) -118.8 (3) -134.77, -129.63 (4) -141.31, -126.70 (5) -93.11, -88.89	(3A-3B) 288 (4A-4B) 300 (5A-5B) 228	142
$C_7FH_5N_2$			(F-C3) 244 (F-C4) 4 (F-C9) 25.5	491
$C_7F_2H_6Cl_2N_2OS$		-74.6		492(a)
$C_7F_5H_4N$		(1) -64.6 (2)(3)(4) -149.5, -151.7, -159.8 (5) -140.7		493(a) 494
$C_7F_8H_6N_2$		(3) -112.5 (4)(5) -133.5 (6) -81.5		179(a)

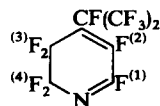
C₈FH₆N



-175

495

C₈F₁₃N

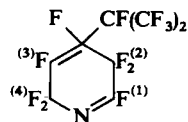


(1)(2) -53, -108
(3)(4) -103, -115

(1-2) 20

498

C₈F₁₃N

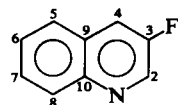


(1)(3) -58, -110
(2)(4) -88, -103

(1-2) 26

498

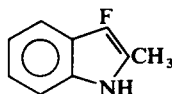
C₉FH₆N



(F-C2) 27·0 (F-C3) 257·0
(F-C4) 16·1 (F-C5) 3·6
(F-C9) 1·7 (F-C10) 5·3

431

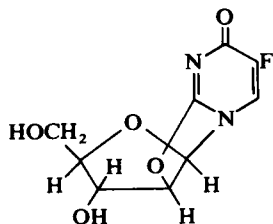
C₉FH₈N



-180

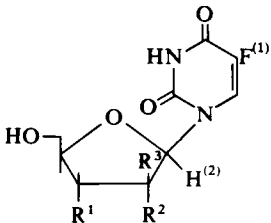
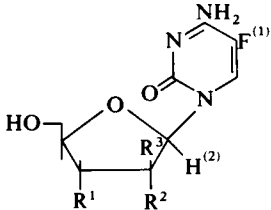
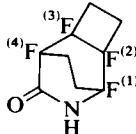
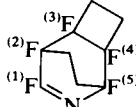
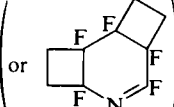
495

C₉FH₁₀N₂O₅

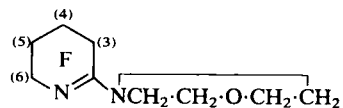


-131

487

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9FH_{11}N_2O_5$	 $R^1 = OH, R^2 = H, R^3 = H$	-166	(1-2) 1·6	487
$C_9FH_{11}N_2O_6$	As above with $R^1 = R^2 = OH, R^3 = H$	-166	(1-2) 1·6	487
$C_9FH_{11}N_2O_6$	As above with $R^1 = R^3 = OH, R^2 = H$	-169	(1-2) 1·8	487
$C_9FH_{12}N_3O_4$	 $R^1 = OH, R^2 = H, R^3 = H$	-165	(1-2) 1·8	487
$C_9FH_{12}N_3O_5$	As above with $R^1 = R^2 = OH, R^3 = H$	-166	(1-2) 1·6	487
$C_9FH_{12}N_3O_5$	As above with $R^1 = R^3 = OH, R^2 = H$	-167	(1-2) 1·6	487
$C_9F_4H_9NO$		(1) -166·0 (2) -175·5 (3) -185·3 (4) -195·2		494(a)
$C_9F_5H_8N$	 	(1) -56·0 (2) -198·8 (3) -186·3 (4) -173·9 (5) -153·6		494(a)

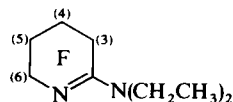
C₉F₈H₈N₂O



(3) -112.5 (4)(5) -134.5
(6) -82.5

179(a)

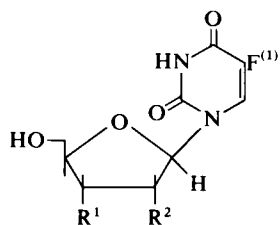
C₉F₈H₁₀N₂



(3) -113.5 (4)(5) -135.0
(6) -80.7

179(a)

C₁₀FH₁₃N₂O₆

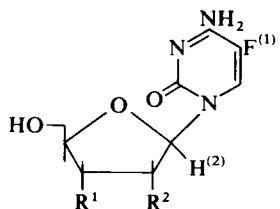


R¹ = OH, R² = OCH₃ -168
R¹ = OCH₃, R² = OH -168

(1-2) 1.8
(1-2) 2.0

487

C₁₀FH₁₄N₃O₅

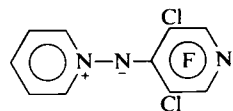


R¹ = OH, R² = OCH₃ -165
R¹ = OCH₃, R² = OH -169

(1-2) 1.6
(1-2) 1.6

487

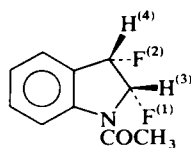
C₁₀F₂H₅Cl₂N₃



-77.2

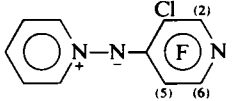
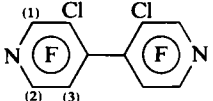
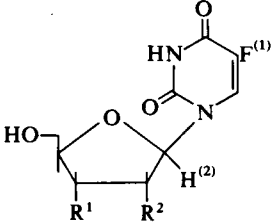
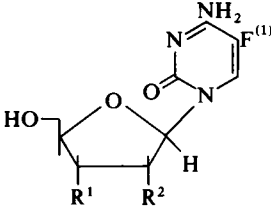
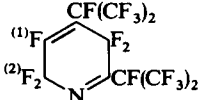
447(a)

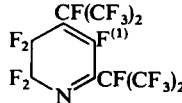
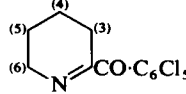
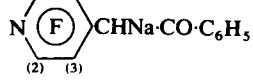
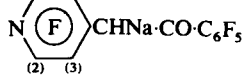
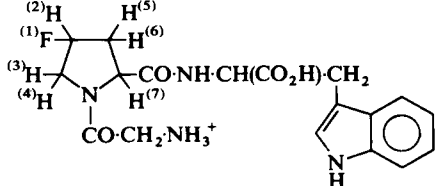
C₁₀F₂H₉NO

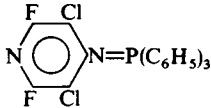
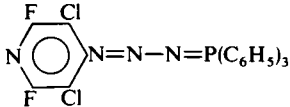


(1) -158 (2) -203 (1-3) 70 (2-4) 70

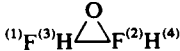
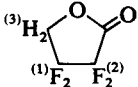
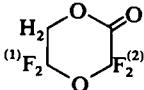
495

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_3H_5ClN_3$		(2) -80.5 (5) -167.0 (6) -98.0		447(a)
$C_{10}F_4H_5N_3$	As above with Cl replaced by F(3)	(2) -98.9 (3) -163.3		447(a)
$C_{10}F_6Cl_2N_2$		(1) -74.8 (2) -89.4 (3) -142.5		490(a)
$C_{11}FH_{15}N_2O_6$	 $R^1 = R^2 = OCH_3$	-166	(1-2) 1.5	487
$C_{11}FH_{16}N_3O_5$	 $R^1 = R^2 = OCH_3$	-168	(1-2) ~1.6	487
$C_{11}F_{19}N$		(1) -102	(1-2) 27	498

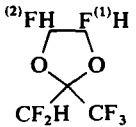
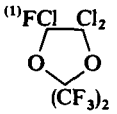
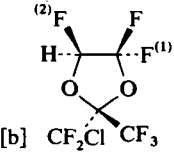
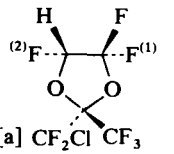
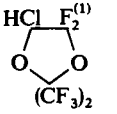
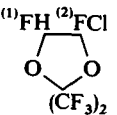
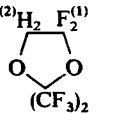
$C_{11}F_{19}N$		(1) -101			498		
$C_{12}F_8HN_3$		(1) -140.1	(2) -89.7		444(a)		
$C_{12}F_8Cl_5NO$		(2) -119.8	(4) -134.9	(5) -133.5	(6) -89.5	179(a)	
$C_{12}F_8N_3Na$		(1) -146.8	(2) -98.5			444(a)	
$C_{13}F_4H_6NNaO$		(2) -99.4	(3) -147.3			272(b)	
$C_{13}F_4H_7NO$	As above with Na replaced by H	(2) -92.9	(3) -143.6			272(b)	
$C_{13}F_9HNNaO$		(2) -98.4	(3) -145.2			272(b)	
$C_{13}F_9H_2NO$	As above with Na replaced by H	(3) -91.8	(3) -142.9			272(b)	
$C_{18}FH_{22}N_4O_4^+$		Two forms, arising probably from restricted rotation about glycol-fluoropropyl bond $\delta_{(minor-major)} = -0.1063$ ppm			Major (1-2) 51.8 (1-3) 38.0 (1-4) 20.9 (1-5) 41.5 (1-6) 19.2 (1-7) -0.4	Minor 51.8 37.3 21.4 40.8 20.4 -0.4	501

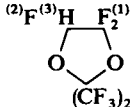
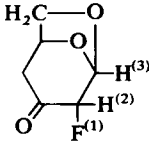
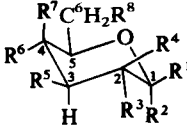
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{23}F_2H_{15}Cl_2N_2P$		-77.3		492(a)
$C_{23}F_2H_{15}Cl_2N_4P$		-71.3		492(a)

B. Oxygen heterocycles

$C_2F_2H_2O$		<i>cis</i> -165	(1-2) 40.3 (1-3) 85.9	506
		<i>trans</i> -83.7	(1-4) -3.8 (1-2) 15.3 (1-3) 82.0 (1-4) -0.5	
$C_4F_4H_2O_2$		(1) -127.1 (2) -131.1	(1-2) 6.2 (1-3) 11.6	237
$C_4F_4H_2O_3$		(1) -75.8 (2) -76.3		237

$C_4F_5HO_3$		(1) -69.62 (2) -70.54 (3) -78.26 (4) -89.51 (5) -135.58	(1-2) 177.6 (1-3) 4.34 (1-4) 2.16 (1-5) 3.23 (1-6) 0.50 (2-3) 8.88 (2-4) 2.15 (2-5) 8.02 (2-6) 0.15 (3-4) 165.1 (3-5) 9.34 (3-6) 1.44 (4-5) 12.94 (4-6) 1.64 (5-6) 51.25	237
C_4F_6O		(1) -77.7 (2) -155.3		508
$C_5F_6H_2O_2$		(1) -125.0 (2) -135.7 (3) -121.2	(1-4) 11.0	237
$C_3F_7HCl_2O_2$		(1) -57.28	(1-2) 7.0	120
$C_5F_7HO_3$		(1) -83.7, -84.3 (2) -126.4, -134.8 (3) -126.9, -131.5 (4) -120.8	(1A-1B) 150 (2A-2B) 250 (3A-3B) 250	508
$C_5F_7H_2ClO_2$		(1a) -127.4 (1e) -128.2		120
$C_5F_7H_3O_2$		(1a) -75.3 (1e) -89.9 (2) -128.1	(1a-2) 7.0 (1e-2) 11.5	120

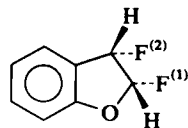
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_5F_7H_3O_2$		(1)(2) -128.1, -128.8		120
$C_5F_7Cl_3O_2$		(1) -44.2		120
$C_5F_8HClO_2$	 	[a] (1a) -85.2 (1e) -86.1 (2) -129.4 [b] (1a) -75.1		120
$C_5F_8HClO_2$		(1a) -72.07 (1e) -77.11		120
$C_5F_8HClO_2$		(1a) -118.2 (2a) -58.9	(1-2) 1.6	120
$C_5F_8H_2O_2$		(1) -75.4	(1-2) 9.0	120

$C_5F_9HO_2$		(1a) -76.5 (1e) -89.3 (2a) -129.6	(1a-2a) 6.7 (1e-2a) 9.0	120																																																							
$C_6FH_7O_3$			(1-2) 49 (1-3) 2.0	509																																																							
$C_6FH_{11}O_5$				510																																																							
	<table><tr><th>R^1</th><th>R^2</th><th>R^3</th><th>R^4</th><th>R^5</th><th>R^6</th><th>R^7</th><th>R^8</th></tr><tr><td>F</td><td>H</td><td>OH</td><td>H</td><td>OH</td><td>OH</td><td>H</td><td>OH</td></tr><tr><td>H</td><td>OH</td><td>F</td><td>H</td><td>OH</td><td>OH</td><td>H</td><td>OH</td></tr><tr><td>OH</td><td>H</td><td>F</td><td>H</td><td>OH</td><td>OH</td><td>H</td><td>OH</td></tr><tr><td>H</td><td>OH</td><td>OH</td><td>H</td><td>F</td><td>OH</td><td>H</td><td>OH</td></tr><tr><td>OH</td><td>H</td><td>OH</td><td>H</td><td>F</td><td>OH</td><td>H</td><td>OH</td></tr><tr><td>H</td><td>OH</td><td>OH</td><td>H</td><td>OH</td><td>F</td><td>H</td><td>OH</td></tr></table>	R^1	R^2	R^3	R^4	R^5	R^6	R^7	R^8	F	H	OH	H	OH	OH	H	OH	H	OH	F	H	OH	OH	H	OH	OH	H	F	H	OH	OH	H	OH	H	OH	OH	H	F	OH	H	OH	OH	H	OH	H	F	OH	H	OH	H	OH	OH	H	OH	F	H	OH	(F-C1) 210.6 (F-C2) 21.6 (F-C3) 11.7 (F-C5) 4.7 (F-C1) 21.4 (F-C2) 185.9 (F-C3) 17.5 (F-C4) 8.0 (F-C5) 1.3 (F-C1) 23.0 (F-C2) 183.1 (F-C3) 17.3 (F-C4) 8.2 (F-C5) 1.2 (F-C1) 10.9 (F-C2) 16.8 (F-C3) 178.6 (F-C4) 17.6 (F-C5) 7.3 (F-C6) 1.5 (F-C1) 12.4 (F-C2) 17.2 (F-C3) 180.5 (F-C4) 17.6 (F-C5) 8.2 (F-C6) 1.9 *interchangeable (F-C1) 1.4 (F-C2) 8.3 (F-C3) 17.6* (F-C4) 179.9 (F-C5) 23.9*	
R^1	R^2	R^3	R^4	R^5	R^6	R^7	R^8																																																				
F	H	OH	H	OH	OH	H	OH																																																				
H	OH	F	H	OH	OH	H	OH																																																				
OH	H	F	H	OH	OH	H	OH																																																				
H	OH	OH	H	F	OH	H	OH																																																				
OH	H	OH	H	F	OH	H	OH																																																				
H	OH	OH	H	OH	F	H	OH																																																				

*interchangeable

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
		interchangeable	(F-C1) 1·4 (F-C2) 8·8 (F-C3) 18·0 (F-C4) 180·0 (F-C5) 24·5*	
			(F-C3) 0·7 (F-C4) 6·6 (F-C5) 17·2 (F-C6) 167·2	
			(F-C4) 6·9 (F-C5) 18·7 (F-C6) 167·6	
			(F-C1) 29·6 (F-C2) 172·2 (F-C3) 17·5 (F-C4) 1·4	
			(F-C1) 15·8 (F-C2) 180·0 (F-C3) 17·6 (F-C4) 1·5	
			(F-C2) 2·2 (F-C3) 17·6 (F-C4) 177·0 (F-C5) 17·6 (F-C6) 5·5	
			(F-C2) 1·1 (F-C3) 18·0 (F-C4) 177·7 (F-C5) 17·3 (F-C6) 5·5	
C ₆ F ₂ H ₁₀ O		(1) -155·5 (2) -163·0	(1-2) 36·0 (1-3) 2·5 (1-4) 0·8 (2-3) 85·0	134
C ₇ F ₆ H ₄ O ₄		(1) -205·4 (2)(3) -126·5, -126·7	(1-4) 45	511
C ₈ FH ₅ O		-177	(1-2) 4·5 (1-3) 1·5	495

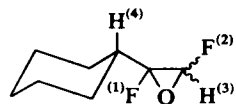
C₈F₂H₆O



(1) -145 (2) -198

495

C₈F₂H₁₂O

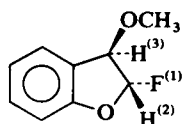


(1) -152.0 (2) -163.5

(1-2) 35.3 (1-4) 12.0
(1-3) 2.2 (2-3) 84.0

134

C₉FH₉O₂

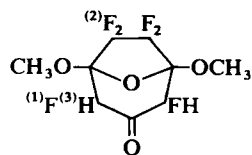


-127

(1-2) 60 (1-3) 15

495

C₉F₆H₈O₄

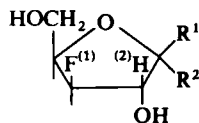


(1) -206.3
(2A) -124.5 (2B) -127.7

(1-3) 45 (2A-3) 22.5
(2A-2B) 250

511

C₁₀FH₁₃N₂O₆

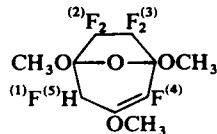


R¹ = H, R² = uracil
R¹ = uracil, R² = H

(1-2) 24
(1-2) 16

512

C₁₀F₆H₁₀O₄

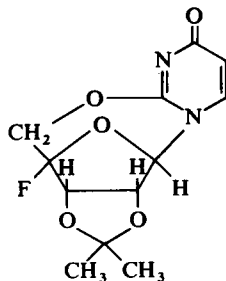


(1) 206.4
(2) -110.9 and -127.1
(3) -131.1 and -131.2
(4) -156.0

(1-5) 51
(2A-2B) 254
(3A-3B) 250

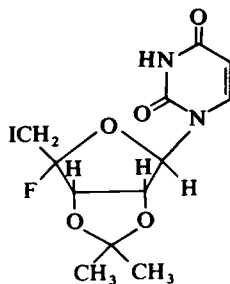
511

Molecular formula	Structure of compounds	Chemical shifts (pm)	Coupling constants (Hz)	Ref.
$C_{10}F_{18}O$		(1) -117		197
$C_{11}FH_{12}N_5O_5$			(1-2) 1.5	513
$C_{11}FH_{15}O_7$		$R = CO \cdot CH_3$ (α) -122 (β) -128.7	(1-2) 63.0 (1-3) 10.0 (1-2) 67.0 (1-3) 22.5 (1-5) 6.5	504
$C_{11}FH_{15}O_7$		$R = CO \cdot CH_3$ -134.8	(1-2) 61.8 (1-3) 17.2 (1-5) 1.4	504
$C_{11}FH_{15}O_7$		(α) -133.1 (β) -116.6	(1-2) 67.7 (1-3) 20.7 (1-2) 61.8 (1-3) 4.2 (1-4) 2.3 (1-5) 6.9	504

$C_{12}FH_{13}N_2O_5$ 

-141.76

513

 $C_{12}FH_{14}IN_2O_5$ 

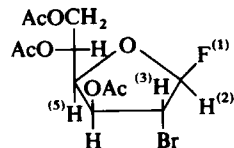
-101.8

513

 $C_{12}FH_{14}N_3O_5$ As above with I replaced by N_3

-109.5

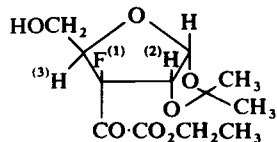
513

 $C_{12}FH_{16}BrO_7$ Ac = $CO \cdot CH_3$

-99.7

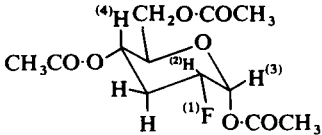
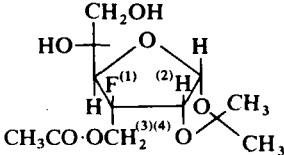
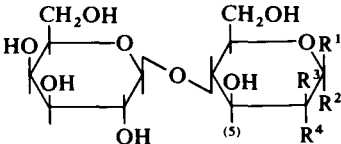
(1-2) 64.5 (1-3) 6.0
(1-5) 6.5

504

 $C_{12}FH_{17}O_7$ 

(1-2) 12 (1-3) 27

514

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}FH_{17}O_7$			(1-2) 47 (1-4) 1·4	509
$C_{12}FH_{19}O_7$			(1-2) 11 (1-4) 34	512
$C_{12}F_2H_{20}O_9$				260
	$R^1 = R^3 = F, R^2 = R^4 = H$	(1) -164·9 (3) -239·0	(1-2) 49 (1-4) 5 (3-4) 52	(1-3) 13·5 (2-3) 18 (3-5) 32·5
	$R^2 = R^4 = F, R^1 = R^3 = H$	(2) -166·3 (4) -219·8	(1-2) 52 (2-3) 22 (3-4) 48	(1-4) 13 (2-4) 18

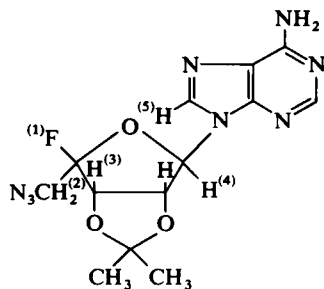
$C_{13}FH_{15}N_8O_3$

-105.01

(1-2) 17
(1-4) 2.5

(1-3) 6

515



$C_{13}FH_{17}N_6O_3$

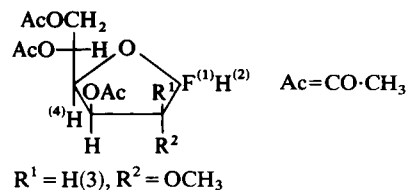
As above but with N_3 replaced by NH_2

(1-4) 2

(1-5) 2

515

$C_{13}FH_{19}O_8$



(α) -138.8

(1-2) 62.5
(1-4) ~1

(1-3) 14.5

504

(β) -119.7

(1-2) 62.0
(1-4) 6.0

(1-3) 5.0

$R^1 = OCH_3, R^2 = H(3)$

(α) -122.3

(1-2) 64.7
(1-4) 2.0

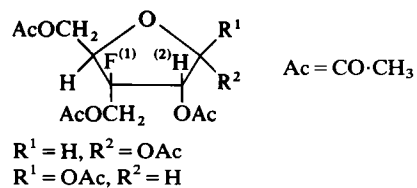
(1-3) 16.1

(β) -128.0

(1-2) 65.8
(1-4) 7.0

(1-3) 22.6

$C_{14}FH_{19}O_9$



(α) -138.8

(1-2) 62.5
(1-4) ~1

(1-3) 14.5

512

(β) -119.7

(1-2) 62.0
(1-4) 6.0

(1-3) 5.0

$R^1 = H, R^2 = OAc$

(α) -122.3

(1-2) 64.7

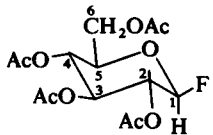
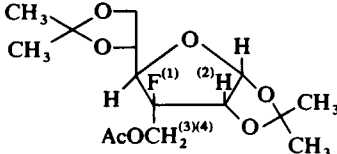
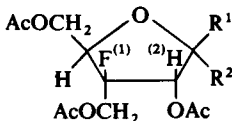
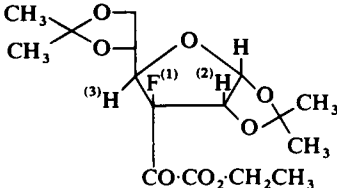
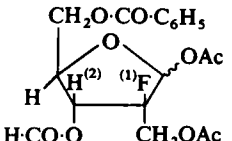
(1-3) 16.1

$R^1 = OAc, R^2 = H$

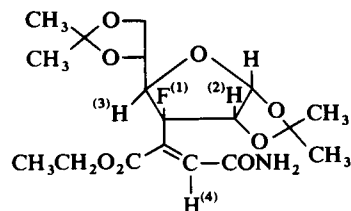
(β) -128.0

(1-2) 65.8

(1-3) 22.6

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}FH_{19}O_9$	 Ac = CO·CH₃		(F-C1) 219.4 (F-C2) 29.3 (F-C3) 8.8 (F-C5) 4.4	510
$C_{14}FH_{20}O_7$	 Ac = CO·CH₃		(1-2) 11 (1-3) 13 (1-4) 34	512
$C_{16}FH_{19}N_2O_9$	 Ac = CO·CH₃ R¹ = H, R² = uracil R¹ = uracil, R² = H		(1-2) 24 (1-2) 18	512
$C_{16}FH_{23}O_8$	 CO·CO₂CH₂CH₃		(1-2) 13.5 (1-3) 27	514
$C_{18}FH_{19}O_9$	 Ac = CO·CH₃ (α) (β)		(1-2) 16 (1-2) 21	512

C₁₈FH₂₆NO₈

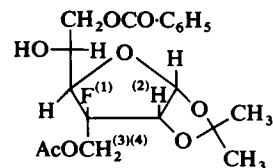


(1-2) 10
(1-4) 1

(1-3) 31

514

C₁₉FH₂₃O₈



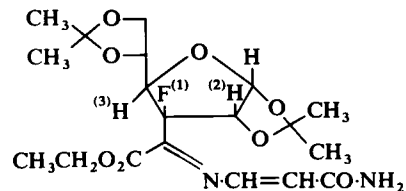
Ac = CO·CH₃

(1-2) 10
(1-4) 34

(1-3) 14
(1-5) 25

512

C₁₉FH₂₇N₂O₈

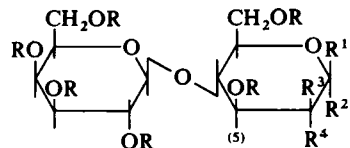


(1-2) 14·5

(1-3) 29

514

C₂₄F₂H₃₂O₁₅



R = CO·CH₃

260

R¹ = R³ = F, R² = R⁴ = H

(1) -147·8 (3) -219·5

(1-2) 50

(1-3) 15

(1-4) 9

(2-3) 12

(3-4) 49

(3-5) 21

R² = R⁴ = F, R¹ = R³ = H

(2) -151·2 (4) -203·9

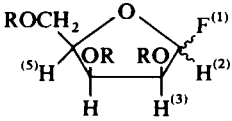
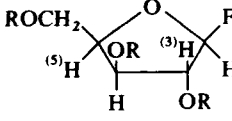
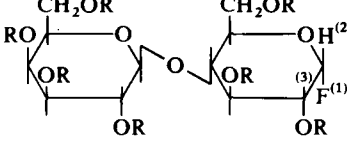
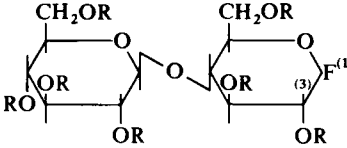
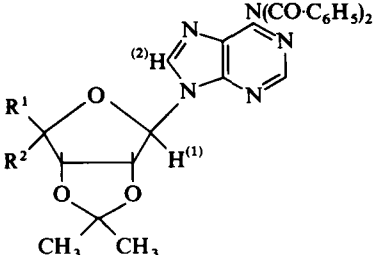
(1-2) 53

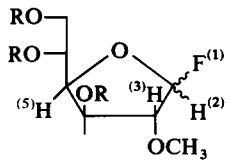
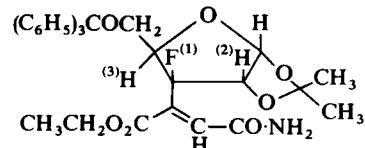
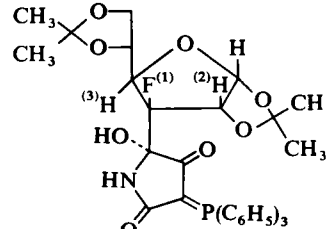
(1-4) 12·2

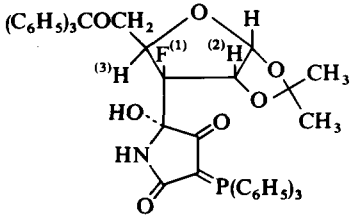
(2-3) 23·7

(2-4) 19

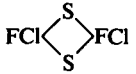
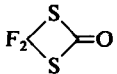
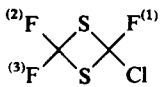
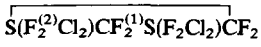
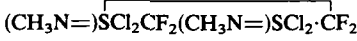
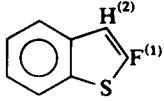
(3-4) 48

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{26}FH_{21}O_7$	 $R = CO \cdot C_6H_5$	$(\alpha) -121.0$ $(\beta) -133.8$	$(1-2) 61.0$ $(1-3) 8.5$ $(1-5) 2.2$ $(1-2) 67.0$ $(1-3) 21.5$ $(1-5) 6.0$	504
$C_{26}FH_{21}O_7$		-118.5	$(1-2) 61.5$ $(1-3) 4.5$ $(1-5) 6.0$	504
$C_{26}FH_{35}O_{17}$	 $R = CO \cdot CH_3$	-149.3	$(1-2) 53.5$ $(1-3) 24.5$	260
$C_{26}FH_{35}O_{17}$		$(\alpha) -147.6$ $(\beta) -136.4$	$(1-2) 53.8$ $(1-3) 24.5$ $(1-2) 53.8$ $(1-3) 9.7$	516
$C_{27}FH_{22}I_2N_5O_5$	 $R^1 = F(3) \quad R^2 = CHI_2$		$(1-3) 2.5$ $(2-3) \sim 0.5$	515

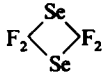
$C_{27}FH_{23}ClN_5O_5$	As above with $R^1 = F$, $R^2 = CH_2Cl$		(1-3) 2·5	(2-3) ~0·5	515
$C_{27}FH_{23}IN_5O_5$	As above with $R^1 = CH_2I$, $R^2 = F$	-103·6			515
$C_{27}FH_{23}IN_5O_5$	As above with $R^1 = F$, $R^2 = CH_2I$	-97·3	(1-3) 2·5	(2-3) ~0·5	515
$C_{27}FH_{24}N_5O_6$	As above with $R^1 = F$, $R^2 = CH_2OH$		(1-3) 2·5	(2-3) 1·5	515
$C_{27}FH_{25}N_5O_8S$	As above with $R^1 = F$, $R^2 = CH_2 \cdot OSO_2 \cdot NH_2$		(1-3) 2·5	(2-3) 1·5	515
$C_{28}FH_{22}O_8$	 $R = CO \cdot C_6H_5$	$(\alpha) -137 \cdot 5$ $(\beta) -119 \cdot 5$	(1-2) 62·5 (1-5) ~1 (1-2) 62·5 (1-5) 6·5	(1-3) 16·5 (1-3) 5·3	504
$C_{29}FH_{26}N_5O_7$	As $C_{27}FH_{22}I_2N_5O_5$ above but with $R^1 = F$, $R^2 = CH_2O \cdot CO \cdot CH_3$		(1-3) 2	(2-3) 1	515
$C_{33}FH_{34}NO_7$			(1-2) 11	(1-3) 28	514
$C_{34}FH_{35}NO_8P$			(1-2) 10	(1-3) 7·5	514

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{49}FH_{43}NO_7P$			(1-2) 10.5 (1-3) 26	514

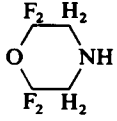
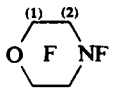
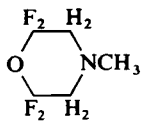
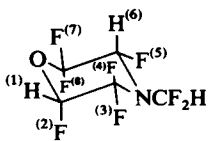
C. Sulphur heterocycles

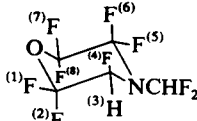
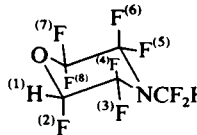
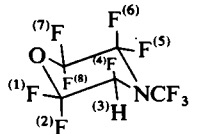
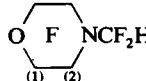
$C_2F_2Cl_2S_2$		<i>cis</i> -34.4 <i>trans</i> -43.8		89
$C_2F_2OS_2$		-64.3		89
$C_2F_3ClS_2$		(1) -42.7 (2) -54.9 (3) -44.9	(1-2) 31.5 (1-3) 4.7 (2-3) 135.5	89
$C_2F_8Cl_4S_2$		(1) -102.2 (2) +148.3		201
$C_4F_4H_6Cl_4N_2S_2$		-97.6		201
C_8FH_5S		-130.2	(1-2) 2.5	340(a)

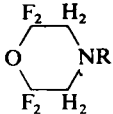
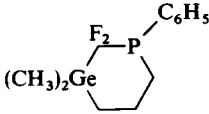
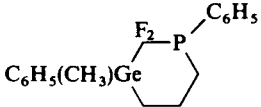
D. Seleneium heterocycles

$C_2F_4Se_2$		-45.7	(F-Se) 67.9	507
--------------	---	-------	-------------	-----

E. Mixed systems

$C_4F_4H_5NO$		-73.7		517
C_4F_9NO		(1) -85.9 (2) -110.0		518
$C_3F_4H_7NO$		-72.4		517
$C_5F_8H_3NO$		(2) -134.4 (3) -90.15 (4) -82.15 (5) -145.1 (7) -73.3 (8) -78.9	(1-2) 52 (3-4) 202 (5-6) 52 (7-8) 162.8	518
$C_5F_8H_3NO$	As above with H(6) and F(5) interchanged	(2) -139.0 (3) -109.8 (4) -73.8 (5) -152.5 (7) -87.15 (8) -73.55	(1-2) 50.3 (3-4) 208.0 (5-6) 54 (7-8) 163.3	518

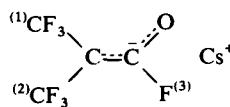
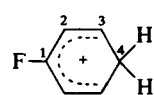
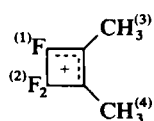
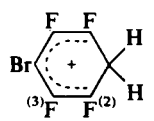
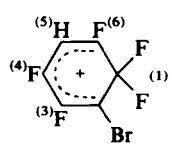
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_5F_9H_2NO$		(1) -88.6 (2) -96.9		518
$C_5F_9H_2NO$		(1) -84.6 (2) -76.3 (4) -151.6 (5) -106.3 (6) -84.7 (7) -88.7 (8) -84.9	(1-2) 161.0 (3-4) 52.3 (5-6) 197 (7-8) 146	518
$C_5F_9H_2NO$	As above with F(5) replaced by H(5) and CHF ₂ by CF ₃	(1) -85.3 (2) -75.4 (4) -152.8	(1-2) 166.0 (3-4) 53.8	518
$C_5F_9H_2NO$		(2) -139.4 (3) -101.1 (4) -78.5 (5) -100.5 (6) -89.7 (7) -91.3 (8) -82.1	(1-2) 50.4 (3-4) 209 (5-6) 198 (7-8) 151	518
$C_5F_{10}HNO$		(1) -83.5 (2) -74.8 (4) -150.6 (5) -104.35 (6) -88.95 (7) -88.0 (8) -86.0	(1-2) 162.35 (3-4) 53.0 (5-6) 190 (7-8) 146	518
$C_5F_{10}HNO$		(1) -87.75 (2) -95.45		518
$C_5F_{11}NO$	As above but with H replaced by F	(1) -86.1 (2) -93.9		518

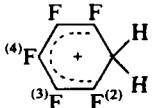
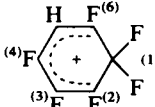
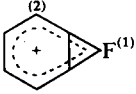
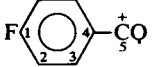
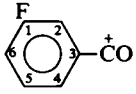
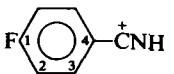
$C_6F_4H_9NO$		$R = CH_2 \cdot CH_3$	-71.8	517
$C_7F_4H_{11}NO$	As above with $R = CH(CH_3)_2$		-72.5	517
$C_8F_4H_{13}NO$	As above $R = C(CH_3)_3$		-72.8	517
$C_{10}F_4H_9NO$	As above with $R = C_6H_5$		-72.7	517
$C_{10}F_4H_{15}NO$	As above with $R = C_6H_{11}$		-72.7	517
$C_{11}F_4H_{11}NO$	As above with $R = CH_2 \cdot C_6H_5$		-72.7	517
$C_{12}F_2H_{17}GeP$			(P-F) 69	115
$C_{17}F_2H_{19}GeP$		(diastereoisomers)	[a] (P-F) 69 [b] (P-F) 72	115

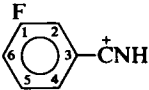
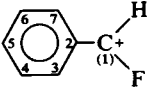
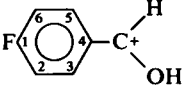
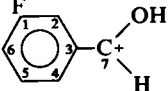
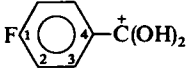
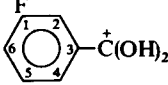
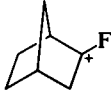
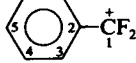
This Page Intentionally Left Blank

**VII. FLUORINE BONDED TO A CARBON OF A CARBONIUM
IONIC OR CARBANIONIC SYSTEM**

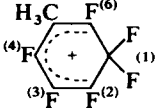
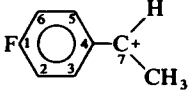
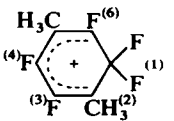
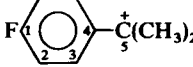
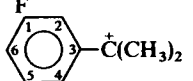
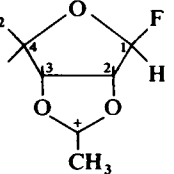
This Page Intentionally Left Blank

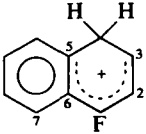
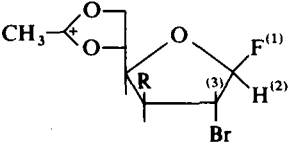
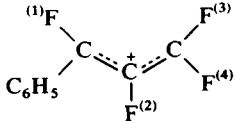
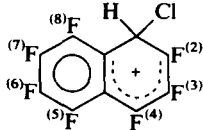
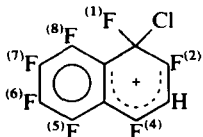
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_4F_4H_3O^+$	$CH_3O \cdot CF^{(1)} \equiv CF^{(2)} \equiv CF_2^{(3)}$	(1)(2) -76.4, -73.7 (3) -74.2		100
C_4F_7CsO		(1) -50.8 (2) -49.5 (3) +0.5	(1-2) 10.7 (1,2-3) 19.4, 12.1	145(a)
$C_6FH_6^+$		-11.0	(F-C1) 324.8 (F-C2) 19.7 (F-C3) 26.7	502
$C_6F_3H_6^+$		(1) 18.2 (2) -98.0	(1-2) 10 (1-3) 4.5 (2-3) 3.9	100
$C_6F_3H_6O_2^+$	As above with CH_3 replaced by OCH_3	(1) -41.1 and -31.6 (2) -102.3 and -86.5	(1-2) 10.5 and 10.5	100
$C_6F_4H_2Br^+$		(2) -56.3 (3) -119.1		503(b)
$C_6F_4H_2Cl^+$	As above with Br replaced by Cl	(2) -53.6 (3) -126.4		503(b)
$C_6F_4H_3^+$	As above with Br replaced by H	(2) -43.6 (3) -123.2		503(b)
$C_6F_3HBr^+$		(1) -119.7 (3) -115.6 (4) +26.3 (6) -14.9	(1-3) 7 (1-4) 2 (1-5) 3 (1-6) 28 (3-4) 30 (3-5) 45 (3-6) 9 (4-5) 9 (4-6) 103 (5-6) 6	351(b)

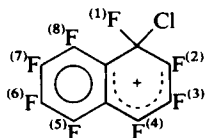
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_6F_5HCl^+$	As above with Br replaced by Cl	(1) -124.1 (4) +32.2	(3) -122.6 (6) -16.5	(1-3) 6.5 (1-5) 3 (3-4) 28.5 (3-6) 9 (4-6) 104.5	(1-4) 1.5 (1-6) 27.5 (3-5) 4.5 (4-5) 9.5 (5-6) 6	351(b)
$C_6F_5H_2^+$	As above with Br replaced by H	(1) -131.3 (4) +44.8	(3) -119.8 (6) +0.2	(1-6) 23	(4-6) 136-137	351(b)
$C_6F_5H_2^+$		(2) -45.5 (4) -31.5	(3) -144.6			503(b)
$C_6F_6H^+$		(1) -133.8 (3) -144.0 (6) -23.9	(2) -71.9 (4) +48.7	(1-3) 7 (3-4) 28	(1-6) 26 (4-6) 102	351(b)
$C_7FH_4^+$				(1-2) 9.0		350
$C_7FH_4O^+$		-69.3		(F-C1) 285 (F-C3) 14.8	(F-C2) 24.6 (F-C5) <1	502
$C_7FH_4O^+$		-102.2		(F-C1) 259.2 (F-C3) 10.5 (F-C5) 9.1 (F-C7) 3.8	(F-C2) 28.0 (F-C4) 3.0 (F-C6) 21.0	502
$C_7FH_5N^+$		-88.2		(F-C1) 268.4 (F-C3) 11.9	(F-C2) 23.7 (F-C4) 2.9	502

$C_7FH_5N^+$		-107.2	(F-C1) 253.8 (F-C2) 26.9 (F-C3) 9.0 (F-C5) 8.6 (F-C6) 21.0	502
$C_7FH_6^+$		+21.0	(F-C1) 375.5 (F-C2) <1 (F-C3) 14.2 (F-C5) 8.4 (F-C7) 23.5	502
$C_7FH_6O^+$		-71.8	(F-C1) 278.1 (F-C2) 24.8 (F-C3) 13.4 (F-C4) <1 (F-C5) 13.6 (F-C6) 24.8	502
$C_7FH_6O^+$		-106.6	(F-C1) 254.5 (F-C2) 22.9 (F-C3) 8.8 (F-C5) 7.8 (F-C6) 22.6 (F-C7) 2.3	502
$C_7FH_6O_2^+$		-89.2	(F-C1) 268.8 (F-C2) 23.2 (F-C3) 11.7	502
$C_7FH_6O_2^+$		-108.4	(F-C1) 251.4 (F-C2) 24.7 (F-C3) 8.7 (F-C5) 7.2 (F-C6) 21.7	502
$C_7FH_{10}^+$			$^1J(\text{F-C})$ 424.1	456
$C_7F_2H_5^+$		+12.2	(F-C1) 361.7 (F-C2) 8.2 (F-C3) 10.7 (F-C4) <1 (F-C5) 5.7	502

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_3H_4^+$		(1) -47.6 (2) -7.3	(1-C1) 301 (1-C2) 22.8 (1-C3) 18.8 (1-C4) <1 (1-C5) 5.6 (2-C1) 6.0 (2-C2) <1 (2-C3) 9.9 (2-C4) 9.8 (2-C5) 358.0	502
$C_7F_4H_5^+$		(2) -56.7 (3) -129.1		503(b)
$C_7F_4H_5^+$		(3) -112.4 (4) -32.9 (5) -142.7 (6) -43.0		351(b)
$C_7F_5H_3Br^+$		(1) -121.5 (3) -116.6 (4) +21.8 (6) -20.5	(1-3) 6.5 (1-6) 29 (3-4) 28 (3-6) 12 (4-6) 112 (5-6) 2.5	351(b)
$C_7F_5H_3Cl^+$	As above with Br replaced by Cl	(1) -125.5 (3) -123.4 (4) +27.1 (6) -22.1	(1-3) 6.5 (1-6) 28 (3-4) 27 (3-6) 11.5 (4-6) 112.5 (5-6) 2.5	351(b)
$C_7F_5H_4^+$		(1) -130.7 (3) -129.0 (4) +31.5 (6) -20.1	(1-3) 7.5 (1-4) 1.5 (1-5) 3 (1-6) 28 (2-4) 3 (2-6) 3 (3-4) 28 (3-6) 9 (4-5) 9 (4-6) 112 (5-6) 6	351(b)

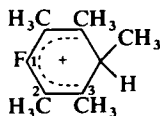
$C_7F_6H_3^+$		(1) -134.1 (2) -78.2 (3) -145.0 (4) +43.4 (6) -30.1	(1-6) 25 (3-4) 26 (4-6) 113	351(b)
$C_7F_7CsN_2$	$[(CF_3^{(1)})_2C \equiv CF^{(2)} \equiv C(CN)_2]^- Cs^+$	(1) -53.5 (2) -81.1		172(a)
$C_8FH_8^+$			(F-C1) 303.6 (F-C2) 24.5 (F-C3) 18.8 (F-C4) <1 (F-C5) 19.4 (F-C6) 24.5 (F-C7) 6.5	502
$C_8F_5H_6^+$		(1) -131.2 (3) -128.7 (4) +26.7 (6) -25.5	(1-3) 6.5 (1-6) 26.5 (2-3) 3.5 (2-4) 3 (2-6) 3 (3-4) 25.5 (3-6) 12 (4-6) 117 (5-6) 2.5	351(b)
$C_9FH_{10}^+$		-60.9	(F-C1) 294.2 (F-C2) 22.1 (F-C3) 16.9 (F-C4) <1 (F-C5) 5.3	502
$C_9FH_{10}^+$		-106.8	(F-C1) 256.3 (F-C2) 22.0 (F-C3) ~8 (F-C5) 7.6 (F-C6) 21.6	502
$C_9FH_{12}O_5^+$	$CH_3CO_2 \cdot CH_2$  (HF solvent)		(F-C1) 230.9 (F-C2) 47.1 (F-C4) 2.9	504
$C_9F_7H_3CsNO_2$	$[(CF_3^{(1)})_2C \equiv CF^{(2)} \equiv C(CN)CO_2 \cdot CH_2 \cdot CH_3]^- Cs^+$	(1) -54.7, -57.2 (2) -72.7		172(a)
$C_9F_7H_6CsO_4$	$[(CF_3^{(1)})_2C \equiv CF^{(2)} \equiv C(CO_2 \cdot CH_3)_2]^- Cs^+$	(1) -57.2, -60.3 (2) -57.2		172(a)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}FH_8^+$		-24.2	(F-C1) 329.0 (F-C2) 21.0 (F-C3) 28.6 (F-C5) 18.2 (F-C6) 8.5 (F-C7) 15.4	502
$C_{10}FH_{13}BrO_5^+$	 R = O.CO.CH ₃ (HF solvent)		(1-2) 62 (1-3) 6.5	504
$C_{10}FH_{13}ClO_5^+$	As above with Br replaced by Cl		(1-2) 56 (1-3) ~5	504
$C_{10}F_4H_7O^+$		(1) -53.1 (2) -180.6 (3) -64.8 (4) -68.6	(1-2) 22 (1-3) 59 (1-4) 3 (2-3) 97.5 (2-4) 70.5 (3-4) 105.5	100
$C_{10}F_7HCl^+$		(2) -39.1 (3) -147.8 (4) -10.9 (5) -106.0 (6) -142.3 (7) -107.7 (8) -122.4	(2-4) 90 (4-5) 130-140	505(b)
$C_{10}F_7HCl^+$		(1) -137.9 (2) -28.9 (4) +21.6 (5) -100.9 (6) -140.4 (7) -106.7 (8) -121.7	(2-4) 100 (4-5) 120-150	505(b)



(1) -115.4 (2) -69.2
(3) -144.5 (4) +4.6
(5) -88.4 (6) -137.9
(7) -95.1 (8) -118.0

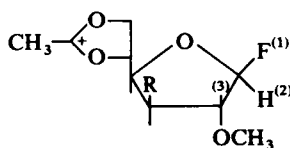
505(b)



-31.3

(F-C1) 313.3 (F-C2) 17.1
(F-C3) 23.1

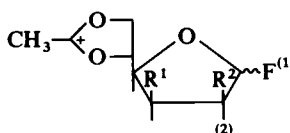
502



R = O·CO·CH₃

(1-2) 60 (1-3) 5.3

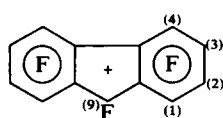
504



R¹ = O·CO·CH₃
R² = OCH₃

(1-2) 23

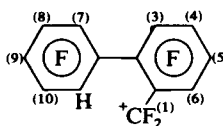
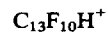
504



(1) -95.5 (2) -144.5
(3) -102.1 (4) -115.5
(9) -4.9

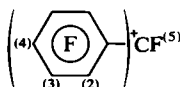
(1-2) 23 (1-3) 39
(1-9) 58

430(b)



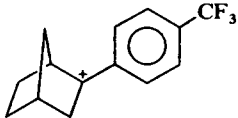
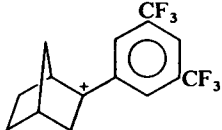
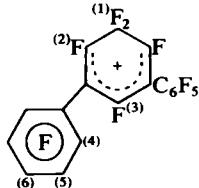
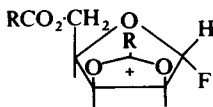
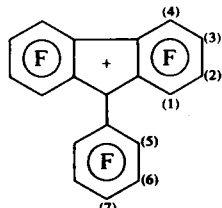
(1) -28.6 (3) -119.8
(4)(6) -88.2, -84.9
(5)(7)-(10) -135.8, -134.4,
-140.9, -145.6, -151.3

430(b)

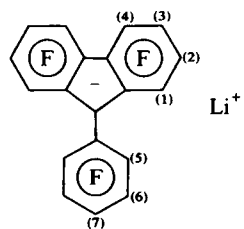


(2) -114.1 (3) -151.3
(4) -102.3 (5) -36.4

430(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{14}F_3H_{14}^+$			$^1J(F-C)$ 282 $^2J(F-C)$ 33.8	456
$C_{15}F_6H_{13}^+$			$^1J(F-C)$ 272.7 $^2J(F-C)$ 35.3	456
$C_{18}F_{15}^+$		(1) -133.2 (2) -28.4 (3) +65.6 (4) -133.2 (5) -157.1 (6) -141.7		386(b)
$C_{19}FH_{16}O_5^+$		R = C ₆ H ₅ (HF solvent)	$^1J(F-C)$ 229.4 $^2J(F-C)$ 47.1	504
$C_{19}F_{13}^+$		(1) -97.8 (2) -147.3 (3) -106.4 (4) -118.1 (5) -123.8 (6) -156.8 (7) -130.7		430

C₁₉F₁₃Li



(1) -160.0 (2) -169.5
 (3) -180.2 (4) -143.1
 (5) -139.7 (6) -166.7
 (7) -162.9

(1-2) 20.5 (1-3) 9
 (1-4) ~9 (2-3) 20.5
 (3-4) 19 (1-5) ~4
 (6-7) 21.5

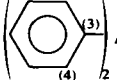
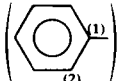
474(b)

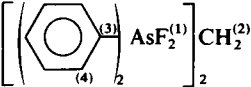
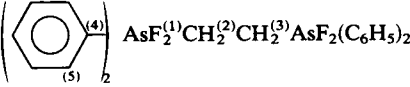
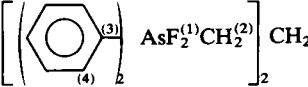
This Page Intentionally Left Blank

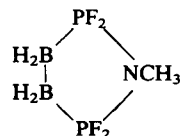
VIII. FLUORINE BONDED TO AN ELEMENT
OTHER THAN CARBON

This Page Intentionally Left Blank

Fluorine bonded to arsenic

Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
CF ₆ H ₃ As	CH ₃ F ⁽¹⁾ ·As·F ₄ ⁽²⁾ ·F ⁽³⁾	(1) -176 (3) -94	(2) -47	(1-2) 30 (2-3) 124		532
CF ₉ H ₃ AsNS	CH ₃ ⁽³⁾ ·NSF ₃ ⁽¹⁾⁺ AsF ₆ ⁽²⁾⁻	(1) +48·2 (3) -46	(2) -60·85 (4) -17	(1-3) 8·7 (2-3) 125	(As-2) ~900 (3-4) 50	552 537
CF ₁₃ H ₃ As ₂ OS	CH ₃ OSF ₂ ⁽¹⁾⁺ F ⁽²⁾ F ₄ ⁽³⁾ AsF ⁽⁴⁾ AsF ₄ F ⁻	(1) +21 (3) -46	(2) -83 (4) -17	(2-3) 125 (3-4) 50		537
C ₂ F ₈ AsO ₂ ⁻	F ⁽¹⁾ AsF ₄ ⁽²⁾ (CF ₃ CO ₂) ⁻	(1) -66·3 (3) -46	(2) -54·4 (4) -17	(1-2) 125 (2-3) 125		526 572
C ₄ F ₆ H ₆ AgAsN ₂	Ag(AsF ₆)·2CH ₃ CN	-72		(As-F) 948		572
C ₄ F ₆ H ₁₂ AsNO ₃ S	AsF ₄ ⁽²⁾ F ⁽¹⁾ (FSO ₂) ⁻ (CH ₃) ₄ N ⁺	(1) -69·6 (3) -46	(2) -47·5 (4) -17	(1-2) 100 (1-C2) 18·0	(1-3) 22·0	119 575
C ₅ F ₄ H ₁₄ As ₂	(CH ₃ ⁽²⁾) ₂ AsF ₂ ⁽¹⁾ ·CH ₂ ⁽³⁾ ·AsF ₂ (CH ₃) ₂	-46·3		(1-2) 124 (1-C2) 15·5	(1-C3) 22·5	119 575
C ₅ F ₈ H ₁₂ AsNO ₃ S	AsF ₄ ⁽¹⁾ F ⁽²⁾ (CF ₃ SO ₃) ⁻ (CH ₃) ₄ N ⁺	(1) -41·4 (3) -46	(2) -69·1 (4) -17	(1-2) 124 (As-F) 948		119 572
C ₆ F ₆ H ₉ AgAsN ₃	Ag(AsF ₆)·3CH ₃ CN	-72		(As-F) 948		572
C ₉ F ₄ H ₂₂ As ₂	[(CH ₃ CH ₂ ⁽²⁾) ₂ AsF ₂ ⁽¹⁾] ⁽³⁾ CH ₂	-72·2		(1-C2) 15·5 (1-C3) 22·5		575
C ₁₂ F ₃ H ₁₅ A ₆	(C ₆ H ₅) ₂ AsF ₃	-69·1				599
C ₁₃ F ₂ H ₁₃ As	 AsF ₂ ⁽¹⁾ ·CH ₃ ⁽²⁾	-78·0		(1-2) 7·0 (1-C3) 16·1	(1-C2) 21·5 (1-C4) 6·6	575
C ₁₃ F ₄ H ₃₀ As ₂	[(CH ₃) ₂ CH ⁽²⁾] ₂ AsF ₂ ⁽¹⁾] ⁽³⁾ CH ₂	-89·0		(1-C2) 15·0 (1-C3) 23·4		575
C ₁₆ F ₁₁ H ₃₆ As ₂ N	(CH ₃ ·CH ₂ ·CH ₂ ·CH ₂) ₄ N ⁺ F ⁽¹⁾ F ₄ ⁽²⁾ AsF ⁽³⁾ AsF ₄ F ⁻	(1) -85 (3) -21	(2) -48	(1-2) 127 (2-3) 51		537
C ₁₈ F ₂ H ₁₅ As	 AsF ₂	-88·9		(F-C1) 18·3 (F-C2) 7·0		575

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{25}F_4H_{22}As_2$		-74.1	(1-2) 10.0 (1-C2) 25.6 (1-C3) 15.5 (1-C4) 7.8	575
$C_{26}F_4H_{24}As_2$		-89.9	(1-C2) 20.7 (1-C3) 4.8 (1-C4) 16.0 (1-C5) 7.5	575
$C_{27}F_4H_{26}As_2$		-88.8	(1-2) 10.0 (1-C2) 18.0 (1-C3) 15.0 (1-C4) 6.0	575
F_7AsOS	$F_4^{(1)}AsF_2^{(2)} \cdot SOF_2^{(3)}$	(1) -32 (2) -88 (3) +43	(1-2) 130	537
F_8AsBr	$F_2^{(1)(2)}Br \cdots \underbrace{F-AsF_5}_{(3)} (-120^\circ \text{ in } SO_2ClF)$	(1) -61.0 (2) -68.3 (3) -56.3		531
$F_{10}AsIO$	$IOF^{(1)}F_4^{(2)} \cdot AsF_5^{(3)} (-110^\circ \text{ in } SO_2F_2)$	(1) +96.8 (2) +65.9 (3) -57.9	(1-2) 253	542
Fluorine bonded to gold				
F_6Au^-	AuF_6^-	-108.9		535

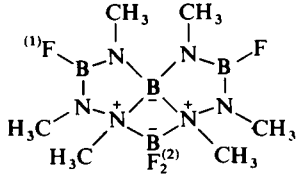
Fluorine bonded to boron

-77.3

(P-1) 1207 (1-2) 14

553

CF ₇ H ₃ BNS	CH ₃ ⁽³⁾ NSF ₃ ⁽¹⁾⁺ BF ₄ ⁽²⁾⁻	(1) +47·90	(2) -145·7	(1-3) 8·6	552
C ₂ F ₆ H ₆ BNS	[(CH ₃ ⁽³⁾) ₂ NSF ₂ ⁽¹⁾⁺ BF ₄ ⁽²⁾⁻	(1) +12·46	(2) -149·06	(1-3) 7·5	562
C ₂ F ₆ BO ₂ ⁻	¹¹ BF ₃ ⁽¹⁾ (CF ₃ CO ₂) ⁻ ¹⁰ BF ₃ (CF ₃ CO ₂) ⁻	-141·00 -140·98		(1-B) 4·5	526
C ₄ FH ₉ BCl	CH ₃ (CH ₂) ₃ BFCI	-33·3			569(a)
C ₄ F ₂ H ₉ B	CH ₃ (CH ₂) ₃ BF ₂	-75·2			569(a)
C ₄ F ₂ H ₁₂ BN ₃ O ₄	BF ₂ (NO ₂) ₂ ⁻ (CH ₃) ₄ N ⁺	-92		(F-B) 55·1	570
C ₄ F ₂ H ₁₂ BN ₇	BF ₂ (N ₃) ₂ ⁻ (CH ₃) ₄ N ⁺	-133·0		(F-B) 18	570
C ₄ F ₃ H ₁₂ BClO ₄	BF ₃ ClO ₄ ⁻ (CH ₃) ₄ N ⁺	-141·0			570
C ₄ F ₃ H ₁₂ BNO ₃	BF ₃ NO ₃ ⁻ (CH ₃) ₄ N ⁺	-141·7			570
C ₄ F ₃ H ₁₂ BN ₂ O ₂	BF ₃ NO ₂ ⁻ (CH ₃) ₄ N ⁺	-112·2		(F-B) 25·1	570
C ₄ F ₃ H ₁₂ BN ₄	BF ₃ N ₃ ⁻ (CH ₃) ₄ N ⁺	-133·0		(F-B) 18	570
C ₄ F ₄ H ₁₂ BN	BF ₄ ⁻ (CH ₃) ₄ N ⁺	-139·8			570
C ₄ F ₄ H ₁₂ BNO ₃ S	BF ₃ FSO ₄ ⁻ (CH ₃) ₄ N ⁺	-140·2			119
C ₄ F ₆ H ₈ BNOS	[CH ₂ ·CH ₂ ·O·CH ₂ ·CH ₂ NSF ₂ ⁽¹⁾⁺] BF ₄ ⁽²⁾⁻	(1) +109	(2) -148·7		562(j)
C ₄ F ₆ H ₁₀ BNS	(CH ₃ CH ₂ ⁽³⁾) ₂ NSF ₂ ⁽¹⁾⁺ BF ₄ ⁽²⁾⁻	(1) +13·27	(2) -149·05	(1-3) 6·0	562(j)
C ₄ F ₈ BO ₄ ⁻	BF ₂ (CF ₃ CO ₂) ₂ ⁻	-142·3		(F-B) 9	526
C ₅ FH ₇ BClO ₂	(CH ₃ ·CO·CH·CO·CH ₃)BFCI	-123·9			569(a)
C ₅ F ₂ H ₇ BO ₂	(CH ₃ ·CO·CH·CO·CH ₃)BF ₂	-138·2			569(a)
C ₅ F ₃ H ₁₂ BN ₂	BF ₃ CN ⁻ (CH ₃) ₄ N ⁺	-127·5		(F-B) 26·8	570
C ₅ F ₃ H ₁₂ BN ₂ O	BF ₃ OCN ⁻ (CH ₃) ₄ N ⁺	-136·9			570
C ₅ F ₃ H ₁₂ BN ₂ S	BF ₃ SCN ⁻ (CH ₃) ₄ N ⁺	-125·9			570
C ₅ F ₃ H ₁₃ BNO ₂	BF ₃ HCO ₂ ⁻ (CH ₃) ₄ N ⁺	-137·4		(F-B) 6	570
C ₅ F ₆ H ₁₀ BNS	CH ₂ ⁽³⁾ (CH ₂) ₄ NSF ₂ ⁽¹⁾⁺ BF ₄ ⁽²⁾⁻	(1) +13·17	(2) -148·7	(1-3) 5·5	562(j)
C ₅ F ₆ H ₁₂ BNO ₃ S	BF ₃ CF ₃ SO ₃ ⁻ (CH ₃) ₄ N ⁺	-137·5			119

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_2H_{12}BN_3$	$BF_2(CN)_2^- (CH_3)_4N^+$	-132.7		570
$C_6F_3H_{15}BNO_2$	$BF_3CH_3CO_2^- (CH_3)_4N^+$	-135.6	(F-B) 7.3	570
$C_6F_4H_{18}B_4N_6$		(1) -156.8 (2) -157.7	(2-B) 43	581
$C_6F_6H_{18}B_3P_3$	$[(CH_3)_2PBF_2]_3$	-158.0	(F-B) 96	584(b)
$C_8F_2H_{18}BNO_4$	$BF_2(CH_3CO_2)_2^- (CH_3)_4N^+$	-135.2	(F-B) 15.9	570
$C_8F_2H_{22}B_4N_6$	As $C_6F_4H_{18}B_4N_6$ above with two single F's replaced by CH_3	-158.4	(F-B) 43	581
$C_{10}FH_{21}BNO_6$	$BF(CH_3CO_2)_3^- (CH_3)_4N^+$	-136.9	(F-B) 22	570
$C_{10}F_3H_{17}BNS$	$BF_3SC_6H_5^- (CH_3)_4N^+$	-135.4		570
$C_{11}F_4H_5BO_4W$	<i>trans</i> - $BF_4(CO)_4WCC_6H_5$ (-50 °C)	-173.5		596(a)
$C_{16}F_6H_{34}BMgO_6$	$Mg(BF_3OH)_2(OCH_2\cdot CH_2\cdot CH_2\cdot CH_2)_4$	-156		602
F_4B^-	$^{11}BF_4^-$	-139.80		526
F_4B^-	$^{10}BF_4^-$	-139.75		526
F_5BBr	$F_2^{(1)(2)}Br \cdots \underbrace{FBF_3}_{(3)}$ (-120 °C in SO_2ClF)	(1) -60.6 (2) -70.5 (3) -127.7		531
Fluorine bonded to bismuth				
$C_{18}F_2H_{15}Bi$	$(C_6H_5)_3BiF_2$	-161.3		575

Fluorine bonded to bromine

FBrO ₂	FBrO ₂	+210		519
F ₃ BrO	F ₃ BrO	+152		519
F ₈ BrP	F ₂ ⁽¹⁾⁽²⁾ Br ··· $\underbrace{\text{F-PF}_5}_{(3)}$ (−130° in SO ₂ ClF)	(1) −59·9 (3) −64·5	(2) −69·2	531

Fluorine bonded to chlorine

FCl	FCl		(Cl-F) 770±10 (liquid) 855±10 (gas)	520
-----	-----	--	--	-----

Fluorine bonded to germanium

C ₂₂ F ₂ H ₃₄ Ge ₂ P ₂	[(CH ₃ ⁽²⁾) ₂ GeF ⁽¹⁾ (CH ₂) ₃ PC ₆ H ₅] ₂ (diastereoisomers)	[a] (1-2) 7·0 [b] (1-2) 6·2		115
---	---	--------------------------------	--	-----

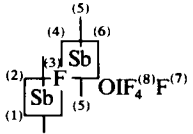
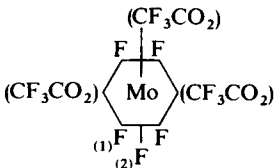
Fluorine bonded to hydrogen

C ₁₂ F ₂ H ₂₉ N	(CH ₃ ·CH ₂ CH ₂) ₄ N ⁺ HFH [−]	−149·4		522
--	--	--------	--	-----

Fluorine bonded to iodine

C ₂ F ₈ IO ₂ [−]	F ⁽¹⁾ IF ₄ ⁽²⁾ (CF ₃ CO ₂) [−]	(1) +53·6 (2) +4·1	(1-2) 68	526
C ₅ F ₅ H ₅ IN	IF ₅ ·C ₅ H ₅ N	+10·9		532

F bonded to Br

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_6I_2N_2$	$IF \cdot IF_5 \cdot (C_5H_5N)_2$	(-40 °C) +11.4, -102		532
$C_{25}F_{15}H_{25}I_5N_5$	$I_2 \cdot 3IF_5 \cdot (C_5H_5N)_5$	+4.9		532
F_5I	IF_5	+52.6, +3.4		533
$F_9I_3O_6$	$(IO_2F^{(1)}F_2^{(2)})_2$ (molten)	(1) +66.2 (2) +105.2	(1-2) 173	540
$F_{10}IOSb$	$IOF^{(1)}F_4^{(2)} \cdot Sb F^{(3)}F_4^{(4)}$ (-110° in SO_2F_2)	(1) +94.0 (2) +68.7 (3) -146.1 (4) -106.9	(1-2) 278 (3-4) 91	542
$F_{15}IOSb_2$		(1) -146.1 (2) -114.9 (3) -87.2 (4) -134.5 (5) -97.2 (6) -127.3 (7) +91.2 (8) +71.2	(1-2) 90 (2-3) 49 (7-8) 257	542
Fluorine bonded to iridium				
F_6Ir	IrF_6	-205		535
Fluorine bonded to molybdenum				
$C_2F_8MoO_2^-$	$F^{(1)}MoF_4^{(2)}(CF_3CO_2)^-$	(1) +134.8 (2) -60.3	(1-2) 54	526
$C_2F_{10}MoO_2^{2-}$	$F^{(2)}MoF_6^{(1)}(CF_3CO_2)^{2-}$	(1) +142.8 (2) -129.1	(1-2) 54 (Mo-1) 69	526
$C_6F_{14}MoO_6^{2-}$		(1) +134.8 (2) -60.3	(1-2) 54	526

Fluorine bonded to nitrogen

$C_4FH_{12}NO_3S$	$(CH_3)_4NFSO_3$	+41.2	119
$C_{12}FH_{28}N$	$(CH_3 \cdot CH_2 \cdot CH_2)_4NF$	-114.6	522

Fluorine bonded to niobium

$C_2F_8NbO_2^-$	$F_5Nb(CF_3CO_2)^-$	-128	526
$C_4F_6H_{12}NNbO_3S$	$F^{(1)}F_4^{(2)}Nb(FSO_3)^- (CH_3)_4N^+$	(1) +184 (2) +143.6	119
$C_5F_8H_{12}NNbO_3S$	$F^{(1)}F_4^{(2)}Nb(CF_3SO_3)^- (CH_3)_4N^+$	(1) +181.7 (2) +143.5	119
$C_{14}F_5H_{15}NbOPS$	$NbF^{(2)}F_4^{(1)}OP(C_6H_5)_2SCH_2 \cdot CH_3$	(1) +120.3 (2) +151.8 (1-2) 49	544(h)
$C_{14}F_5H_{15}NbO_2P$	As above with S replaced by O	(1) +121.8 (2) +156.2 (1-2) 49	544, 543(h)
$C_{14}F_5H_{15}NbPS_2$	As above with O replaced by S	(1) +120.3 (2) +151.8 (1-2) 49	543(h)
$C_{14}F_{10}H_{15}Nb_2OPS$	$ \begin{array}{c} {}^{(5)}F \quad F \\ \diagdown \quad \\ F^{(1)}F_4^{(2)}NbF^{(3)}Nb-F^{(6)} \\ \quad \diagup \\ {}^{(4)}F \quad L \end{array} $ $L = OP(C_6H_5)_2SCH_2 \cdot CH_3$	(1) +186.0 (2) +141.5 (1-2) 55 (1-3) 60 (3) -60.1 (4) +159.5 (3-6) 185 (5) +177.1 (6) +215.4	544(h)
$C_{14}F_{10}H_{15}Nb_2O_2P$	As above with S replaced by O	(1) +186.8 (2) +141.8 (1-2) 55 (1-3) 175 (3) -59.6 (4) +161.1 (3-6) 200 (5) +177.3 (6) +216.2	544, 543(h)
$C_{14}F_{10}H_{15}Nb_2PS_2$	As above with O replaced by S	(1) +186.0 (2) +141.5 (1-3) 160 (3-6) 185 (3) -60.1 (4) +159.5 (5) +177.1 (6) +215.4	543(h)
$C_{18}F_5H_{15}NbOP$	$NbF_4^{(1)}F^{(2)}OP(C_6H_5)_3$	(1) +117.8 (2) +148.3 (1-2) 49	544, 543(h)

F bonded to N

427

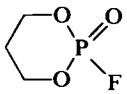
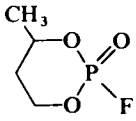
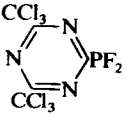
Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)		Ref.
$C_{18}F_{10}H_{15}Nb_2OP$	$ \begin{array}{c} {}^{(5)}F \quad F \\ \quad \quad \diagdown \quad \diagup \\ F^{(1)}F_4^{(2)}NbF^{(3)}Nb-F^{(6)} \\ \quad \quad \diagup \quad \diagdown \\ {}^{(4)}F \quad L \end{array} $ $L = OP(C_6H_5)_3$	(1) +185.5 (3) -60.9 (5) +176.3	(2) +141.3 (4) +155.8 (6) +213.8	(1-2) 55 (3-6) 200	(1-3) 200	544, 543(h)
$C_{28}F_4H_{30}NbO_2P_2S_2^+$	<i>trans</i> -[NbF ₂ ⁽¹⁾ F ₂ ⁽²⁾ (OP(C ₆ H ₅) ₂ SCH ₂ ·CH ₃) ₂] ⁺	(1) +143.3	(2) +171.8			544(h)
$C_{28}F_4H_{30}NbO_4P_2^+$	As above with S replaced by O	(1) +144.6	(2) +174.7	(1-2) 49		544, 543(h)
$C_{28}F_4H_{30}NbP_2S_4^+$	As above with O replaced by S	(1) +143.3	(2) +171.8	(1-2) 50		543(h)
$C_{36}F_4H_{30}NbO_2P_2^+$	<i>cis</i> [NbF ₂ ⁽¹⁾ F ₂ ⁽²⁾ (OP(C ₆ H ₅) ₃) ₂]	(1) +141.9	(2) +167.3	(1-2) 49		544, 543(h)
$F_{11}Nb_2^-$	$F^{(1)}F_4^{(2)}NbF^{(3)}NbF_4F^-$	(1) +183.6 (3) -64.1	(2) +137.9	(1-3) 200		543 544

Fluorine bonded to osmium

F_6Os	OsF_6	-88.7				535
---------	---------	-------	--	--	--	-----

Fluorine bonded to phosphorus

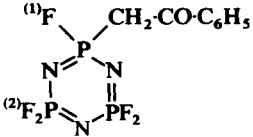
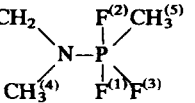
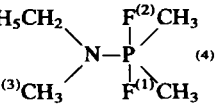
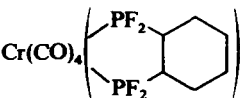
CF_2H_2BNPSe	$PF_2(NCSe)BH_2$	-62.2		(P-F) -1278 (F-H) 17.3		549
CF_2NPSe	$PF_2(NCSe)$	-56.05		(P-F) 1134		549
C_2F_2HOP	$HC\equiv CP(O)F_2$	-55.5		(P-F) 982		556
$C_2F_2H_3OP$	$CH_2=CHP(O)F_2$	-66.2		(P-F) 1078.5		556
$C_2F_2H_3PS$	$CH_2=CHP(S)F_2$	-47.3		(P-F) 1118.0		556
$C_2F_2H_{14}B_4N_2P$	$B_4H_8PF_2N(CH_3)_2$ (-125°C)			(P-F) 1080, 1150 (F-F) 40		557

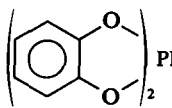
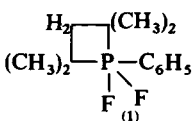
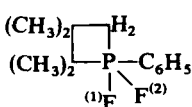
C ₂ F ₃ H ₆ P	(CH ₃ ⁽²⁾) ₂ PF ₃ ⁽¹⁾	(1a) -4·14 (1e) -86·2	(1a-2) 13·0 (1e-2) 3·3 (P-1a) 777 (P-1e) 966	558
C ₂ F ₄ HP	H ⁽²⁾ C≡CPF ₄ ⁽¹⁾	-40·1	(1-2) 1·9 (P-1) 950·4	556
C ₂ F ₄ H ₃ P	CH ₂ =CH·PF ₄	-54·8	(P-F) 941·3	556
C ₂ F ₈ O ₂ P ⁻	F ⁽¹⁾ PF ₄ ⁽²⁾ (CF ₃ CO ₂) ⁻	(1) -78·0 (2) -63·2	(1-2) 60 (P-1) 74·0 (P-2) 770	526
C ₃ FH ₆ O ₃ P		-69·7	(P-F) 998	563(a)
C ₃ F ₂ H ₉ P	(CH ₃ ⁽²⁾) ₃ PF ₂ ⁽¹⁾		(1-2) 11·4	558
C ₃ F ₈ H ₆ FeNP ₄ O	[CH ₃ N(PF ₂) ₂] ₂ FeCO		(P-F) 1250	567
C ₃ F ₁₂ H ₉ CrN ₃ P ₆	[CH ₃ N(PF ₂) ₂] ₃ Cr		(P-F) 1250	567, 568
C ₃ F ₁₂ H ₉ MoN ₃ P ₆	As above with Cr replaced by Mo		(P-F) 1220	567, 568
C ₃ F ₁₂ H ₉ N ₃ P ₆ W	As above with Cr replaced by W		(P-F) 1226	567, 568
C ₄ FH ₈ O ₃ P		(cis) -85·5 (trans) -69·8	(P-F) 990 (P-F) 1005	563(a)
C ₄ FH ₁₀ N ₂ OP	N(CH ₃)CH ₂ ·CH ₂ ·N(CH ₃)P(O)F	-75·6	(P-F) 933	140
C ₄ FH ₁₀ N ₂ P	N(CH ₃)CH ₂ ·CH ₂ ·N(CH ₃)PF	-67·0	(P-F) 1137	140
C ₄ FH ₁₀ O ₂ P	CH ₃ P(O)FOCH(CH ₃) ₂	-97·7	(P-F) 1045	563(a)
C ₄ F ₂ Cl ₆ N ₃ P		-48·9	(P-F) 995	95

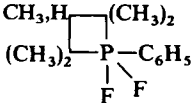
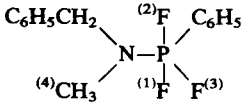
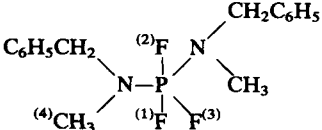
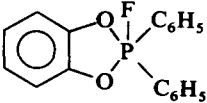
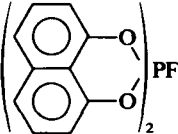
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_4F_3H_{10}N_2P$		(1) -112.6 (2) -48.7	(1-2) 47 (P-1) 711 (P-2) 1133	140
$C_4F_3H_{12}N_2P$	$[(CH_3^{(2)})_2N]_2PF_3^{(1)}$		(1a-2) 2.6 (1e-2) 1.6	558
$C_4F_4MoO_4P_4S$	$[Mo(PF_2^{(1)})(PF_2^{(2)})S(CO)_4]$	-27.6	(1-2) 7.5, 0.0 (P1-1) -1299 (P2-1) +47	523
$C_4F_4MoO_4P_2Se$	As above with S replaced by Se	-20.6	(1-2) 7.7, 0.7 (P1-1) -1292 (P2-1) +48 (Se-F) 27	523
$C_4F_6H_{12}NO_3PS$	$F^{(1)}PF_4^{(2)}(FSO_3)^- (CH_3)_4N^+$	(1) -82.6 (2) -60.1	(1-2) 63.0 (P-1) 750 (P-2) 791	119
$C_3FH_{12}O_2P$	$CH_3P(O)FOCH(CH_3)CH_2CH_3$ (diastereoisomers)	[a] -99.3 [b] -100.3	(P-F) 1012 (P-F) 1012	563(a)
$C_5F_2MnO_7P$	$Mn(CO)_4(PO_2F_2)$	-77.8	(P-F) 968	573(a)
$C_3F_4H_7O_2P$		(1) -49.7 (2) -73.1	(1-2) 59.0 (P-1) 823 (P-2) 762 (1-2) 3.0 (2-4) 3.0	574
$C_5F_8H_6N_2Ni_2O_3$	$[CH_3N(PF_2)_2]_2Ni_2(CO)_3$		$^1J(P-F)$ 1238 $^3J(P-F)$ 54	567
$C_3F_8H_7O_2P$	$CH_3^{(2)}PF_2^{(1)}(OCH_2^{(4)}CF_3^{(3)})_2$	(1) -30.1	(1-2) 12.7 (3-4) 7.9 (P-1) 819	220

$C_5F_8H_{12}NO_3PS$	$F^{(1)}PF_4^{(2)}(CF_3SO_3)^-(CH_3)_4N^+$	(1) -82.7 (2) -56.4	(1-2) 62.6 (P-1) 762 (P-2) 792	119
$C_6FH_{14}O_2P$	$CH_3P(O)FOCH(CH_2 \cdot CH_3)_2$	-100.5	(P-F) 1095	563(a)
$C_6F_3H_{13}NP$	$CH_2=CH \cdot PF_3^{(1)}N(CH_2CH_3)_2$	(1a) -47.3 (1e) -66.9	(1a-1e) 51.7 (P-1a) 796 (P-1e) 951	556
$C_6F_3H_{15}NP$	$CH_3 \cdot CH_2 \cdot CH^{(2)}(CH_3)PF_3^{(1)}N(CH_3)_2$	(1a) -45.6 (1a') -46.3 (1e) -85.0	(1a-1e) = (1a'-1e) 52 (1a-1a') 23 (1-2) 13 (P-1a) 818 (P-1a') 820 (P-1e) 994	580
$C_6F_4H_{18}Ga_2N_2O_2P_2$	$[(CH_3)_2GaN(CH_3)P(O)F_2]_2$	-79.45	(P-F) 1027	582(a)
$C_6F_5H_9N_3OP_3$		(1) -59.0 (2) -61.0	(P-1) 765 (P-2) 760	583
$C_6F_{24}H_{18}N_6Ni_2P_{12}$	$[CH_3N(PF_2)_2]_6Ni_2$		(P-F) ~1250, 1270	567
$C_7FH_{14}O_2P$	$CH_3P(O)FOC_6H_{11}$	-98.5	(P-F) 1040	563(a)
$C_7FH_{16}O_2P$	$CH_3P(O)FOCH(CH_3)C(CH_3)_3$ (diastereoisomers)	[a] -97.2 [b] -100.0	(P-F) 1045 (P-F) 1045	563(a)
$C_7F_2H_6ClFeNOP$	$C_5H_5 \cdot Fe(CO)[PF_2 \cdot NHCH_3]Cl$		(P-F) 1138	586
$C_7F_3H_{17}NP$	$CH_3CH_2 \cdot CH^{(2)}(CH_3)PF_3^{(1)}NHCH(CH_3)_2$	(1a) -47.0 (1a') -58.0 (1e) -76.6	(1a-1e) = (1a'-1e) 52 (1a-1a') 21 (1a'-2) 13 (P-1a) 792 (P-1a') 836 (P-1e) 1009	580
$C_7F_4H_8ClFeNOP_2$	$C_5H_5 \cdot Fe(CO)[(PF_2)_2NCH_3]Cl$		(P-F) 1168 (complexed P atom) 1258 (uncomplexed P atom)	586
$C_7F_4H_8MnNOP_2$	$C_5H_5 \cdot Mn(CO)(PF_2)_2NCH_3$		(P-F) ~1270	586

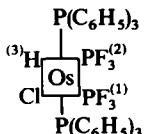
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_7F_4H_8MoN_2O_2P_2$	$C_5H_5 \cdot Mo(CO)(NO)(PF_2)_2NCH_3$		(P-F) 1156 (complexed P atom) 1255 (uncomplexed P atom)	586
$C_7F_4H_{18}P_2$	$(CH_3)_2PF_2(CH_2)_3PF_2(CH_3)_2$	-17.4	(P-F) 550	587
$C_7F_8H_{11}ClFeN_2P_4$	$C_5H_5 \cdot Fe[(PF_2)_2NCH_3]_2Cl$		(P-F) 1162 (complexed P atom) 1251 (uncomplexed P atom)	586
$C_7F_8H_{11}ClMoN_2P_4$	$C_5H_5 \cdot Mo[(PF_2)_2NCH_3]_2Cl$		(P-F) ~1170, ~1190, 1300, 1245	586
$C_7F_8H_{11}ClN_2P_4W$	$C_5H_5 \cdot W[(PF_2)_2NCH_3]_2Cl$		(P-F) ~1150, ~1270, ~1240	586
$C_7F_8H_{11}MnN_2P_4$	$C_5H_5 \cdot Mn[(PF_2)_2NCH_3]_2$		(P-F) ~1300 (complexed P atom) 1235 (uncomplexed P atom)	586
$C_7F_{10}H_9O_3P$	$CH_3^{(2)}PF^{(1)}(OCH_2^{(4)} \cdot CF_3^{(3)})_3$	(1) -34.0	(P-F) 813 (1-2) 14.2 (3-4) 7.9	220
$C_8FH_{10}FeN_2O_4P$	$CH_3N \cdot CH_2 \cdot CH_2 \cdot N(CH_3^{(2)})PF^{(1)} \cdot Fe(CO)_4$	+7.3	(P-F) 1165 (1-2) 2.5	588
$C_8FH_{12}FeN_2O_4P$	$[(CH_3^{(2)})_2N]_2PF^{(1)} \cdot Fe(CO)_4$	-48.7	(P-F) 1085 (1-2) 2.5	588
$C_8F_2H_9ClMoNO_2P$	$C_5H_5 \cdot Mo(CO)_2[PF_2NHCH_3]Cl$		(P-F) 1134	586
$C_8F_4H_3Fe_2NO_7P_2$	$CH_3N(PF_2)_2Fe_2(CO)_7$		$^3J(P-F)$ 16	567
$C_8F_4H_8ClMoNO_2P_2$	$C_5H_5 \cdot Mo(CO)_2[(PF_2)_2NCH_3]Cl$		(P-F) ~1150 (complexed P atom) 1247 (uncomplexed P atom)	586
$C_8F_4H_8ClNO_2P_2W$	$C_5H_5 \cdot W(CO)_2[(PF_2)_2NCH_3]Cl$		(P-F) 1136 (complexed P atom) 1260 (uncomplexed P atom)	586
$C_8F_4H_{10}NP$	$ \begin{array}{c} C_6H_5CH_2 \quad F^{(2)}F^{(4)} \\ \quad \quad \quad \quad / \\ \quad \quad \quad N \quad P \\ \quad \quad \quad \quad \\ CH_3^{(5)} \quad F^{(1)}F^{(3)} \end{array} $	(-80 °C) (1) -55.8 (2) -58.4 (3)(4) _{av} -72.2	(1-3,4) 64 (2-3,4) 72 (P-1) 779 (P-2) 779 (P-3,4) _{av} 915 (1,2,3,4-5) 2.5 (room temp.)	591

$C_8F_5H_7N_3OP_3$		(1) -58.0	(2) -62.0	(P-1) 778	(P-2) 788	583	
$C_8F_6H_{12}CuN_4P$	$CuPF_6 \cdot 4CH_3CN$	-77		(P-F) 735		572	
$C_8F_{12}Co_2O_8P_4Rh_2$	$Co_2Rh_2(Co)_8(PF_3^{(1)(2)})_4$	(1) -10.3	(2) -13.8	(P-1) 1410	(P-2) 1465	593	
$C_9FH_{11}ClO_4PW$	$C_5H_5 \cdot W(CO)_2[PF(OCH_3)_2]Cl$			(P-F) 1157		586	
$C_9F_3H_{13}NP$		(-60 °C)	(1) -26.9 (3) -65.5	(2) -29.2	(1-3) 49 (P-1) 808 (p-3) 957 (room temp.) { (1,2-4) 3.0 (3-4) 3.0	(2-3) 58 (P-2) 800 (1,2-5) 12.5 (3-5) 1.5	591
$C_9F_3H_{13}NP$	$C_6H_5PF_3^{(1)}NH^{(2)}CH(CH_3)_2$	(1a) -39.0 (1e) -70.7	(1a') -50	(1a-1e) = (1a'-1e) 52 (1a-1a') 12 (P-1a) 764 (P-1e) 980	(1a'-2) 27 (P-1a') 796	580	
$C_9F_6H_{27}I_2NiP_3$	$NiI_2[PF_2C(CH_3)_3]_3$	-75.1		(P-F) ~1180		594	
$C_{10}F_2H_9Cl_6O_2P$	$C_6H_5PF_2(OCH_2 \cdot CCl_3)_2$	-38.0		(P-F) 824		220	
$C_{10}F_2H_{11}Cl_4O_2P$	$C_6H_5PF_2(OCH_2 \cdot CHCl_2)_2$	-41.9		(P-F) 822		220	
$C_{10}F_2H_{16}NP$		(-120 °C)	(1) -14.4	(2) -16.8 (room temp.) {	(P-1) 655 (1,2-3) 3.0	(P-2) 652 (1,2-4) 13.0	591
$C_{10}F_4H_{10}CrO_4P_2$		[a] -63.6 [b] -64.7		(P-F) 1180 (P-F) 1158		595	
$C_{10}F_4H_{10}MoO_4P_2$	As above with Cr replaced by Mo	[a] -64.7 [b] -65.4		(P-F) 1170 (P-F) 1156		595	

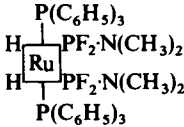
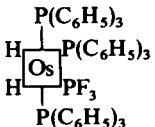
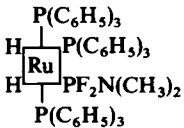
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{10}F_4H_{10}O_4P_2W$	As above with Cr replaced by W	[a] -66.1 [b] -66.1	(P-F) 1160 (P-F) 1141	595
$C_{10}F_6Co_2O_{10}P_2Rh_2$	$Co_2Rh_2(CO)_{10}(PF_3)_2$ (several isomers)	-40 °C [a] -10.5 [b] -11.5 [c] -13.6 [d] -14.7	(P-F) 1416 (P-F) 1431 (P-F) 1460 (P-F) 1460	593
$C_{10}F_8H_9O_2P$	$C_6H_5 \cdot PF_2^{(1)}(OCH_2^{(3)} \cdot CF_3^{(2)})_2$	(1) -43.2	(2-3) 8.3 (P-1) 808	220
$C_{10}F_{12}H_{15}CuN_5P_2$	$Cu(PF_6)_2 \cdot 5CH_3CN$	-78	(P-F) 729	572
$C_{11}FH_{15}ClO_4PW$	$C_5H_5 \cdot W(CO)_2[PF(OCH_2 \cdot CH_3)_2]Cl$		(P-F) 1155	586
$C_{12}FH_8O_4P$	 PF	-70.4	(P-F) 1011	597
$C_{12}F_2H_{13}N_2O_2P$	$C_6H_5PF_2[OCH(CH_3)CN]_2$	-39.8	(P-F) 826	220
$C_{12}F_{10}H_{11}O_3P$	$C_6H_5PF^{(1)}(OCH_2^{(3)} \cdot CF_3^{(2)})_3$	(1) -40.5	(2-3) 8.3 (P-1) 820	220
$C_{13}F_2H_8MnO_5P$	$Mn(CO)_3 \left(\text{pyridine-2,6-dimethylenediyl} \right) (PO_2F_2)$	-77.3	(P-F) 962	573(a)
$C_{13}F_2H_{19}P$		(-95 °C) -36.3 and -63.5 (ratio 1 : 2 : 4)	(P-F) 775 and 933	600
$C_{13}F_2H_{19}P$		(-80 °C) (1) -46 (2) -58.6	(1-2) 15.7 (P-1) 875 (P-2) 987	600

$C_{14}F_2H_{21}P$		(-90 °C) -58.8 and -71.2 (ratio 4 : 1)	(P-F) 922 and 900	600
$C_{14}F_3H_{15}NP$		(-50 °C) (1) -40.8 (2) -43.2 (3) -65.6	(1-2) 14 (1-3) 55 (2-3) 56 (P-1) 823 (P-2) 821 (P-3) 961 (room temp.) (1,2-5) 3.0 (3-5) 3.0	591
$C_{14}F_3H_{15}NP$	$C_6H_5PF_3^{(1)}NH^{(2)} \cdot CH(CH_3)C_6H_5$	(1a) -39.8 (1a') 47.6 (1e) -70.0	(1a-1e) = (1a'-1e) 52 (1a-1a') 15 (1a'-2) 23 (P-1a) 775 (P-1a') 806 (P-1e) 972	580
$C_{16}F_3H_{20}NP$		(1,2) _{av} -54.4 (3) -68.8	(1,2-3) _{av} 4.7 (1,2-4) 3.0 (3-4) 3.0 (P-1,2) 753 (P-3) 870	591
$C_{16}F_4H_{20}P_2$	$(C_6H_5)_2PF_2^{(1)}(CH_2)_2PF_2^{(2)}(CH_3)_2$	(1) -40.1 (2) -18.0	(P-1) 640 (P-2) 550	587
$C_{16}F_6H_{22}Cl_2OsP_4$	<i>cis</i> -[OsCl ₂ (PF ₃) ₂ [P(CH ₃) ₂ C ₆ H ₅] ₂]	-18.8	(P-F) 1230 ³ J(P-F) 5.0 ⁴ J(F-F) 4.0	601
$C_{16}F_6H_{22}Cl_2OsP_4$	<i>trans</i> -isomer of above	-32.9	(P-F) 1453	601
$C_{18}FH_{14}O_2P$		-26.2	(P-F) 772	597
$C_{19}F_2H_{15}OP$	$(C_6H_5)_2PF_2 \cdot CO \cdot C_6H_5$	-58.8	(P-F) 705	606
$C_{20}FH_{12}O_4P$		-85	(P-F) 1061	597

Molecular formula	Structure of compounds	Chemical shifts (pm)		Coupling constants (Hz)		Ref.
$C_{20}F_2H_{20}NP$		(-80 °C)	(1) -37.1	(2) -40.0 (room temp.)	(P-1) 708 (P-2) 711 {(1,2-3) 3.0}	591
$C_{22}FH_6OP$		-12			(P-F) 698	597
$C_{24}FH_{16}O_4P$		-72.1			(P-F) 736	597
$C_{24}FH_{18}O_2P$		-2.8			(P-F) 723	597
$C_{24}F_3H_{33}Cl_2OsP_4$	<i>cis</i> -[$OsCl_2(PF_3)[P(CH_3)_2C_6H_5]_3$]	-13.1			(P-F) 1208	601
$C_{24}F_3H_{33}Cl_2OsP_4$	<i>trans</i> -isomer of above	-25.4			(P-F) 1267	601
$C_{24}F_4H_{16}NiO_8P_4$		-14.0			$^1J(P-F) -1288.3$ $^3J(P-F) +29.2$ $^4J(F-F) +5.8$	607

$C_{24}F_4H_{16}O_8P_4Pt$	As above with Ni replaced by Pt	-13.8		$^1J(P-F) -1359.3$ $^3J(P-F) +34.2$ $^4J(F-F) +8.4$ $^2J(Pt-F) \pm 1445.1$	607
$C_{24}F_8H_{20}Br_2FeP_4$	$FeBr_2(PF_2C_6H_5)_4$	-55.1, -38.8			594
$C_{24}F_8H_{20}Cl_2FeP_4$	$FeCl_2(PF_2C_6H_5)_4$	-59.2, -40.3			594
$C_{24}F_8H_{20}CoIP_4$	$CoI(PF_2C_6H_5)_4$	-44.8		(P-F) ~1080	594
$C_{24}F_8H_{20}FeI_2P_4$	$FeI_2(PF_2C_6H_5)_4$	-46.6, -35.4			594
$C_{25}F_4H_{22}P_2$	$[(C_6H_5)_2PF_2]_2CH_2$	-27.8		(P-F) 656	587
$C_{26}F_4H_{24}P_2$	$[(C_6H_5)_2PF_2 \cdot CH_2]_2$	-40.8		(P-F) 648	587
$C_{27}F_4H_{26}P_2$	$[(C_6H_5)_2PF_2 \cdot CH_2]_2CH_2$	-40.0		(P-F) 644	587
$C_{27}F_4H_{28}P_2$	$[(C_6H_5)_2PF_2 \cdot CH_2 \cdot CH_2]_2$	-39.9		(P-F) 645	587
$C_{30}F_6H_{25}P_3$	$[(C_6H_5)_2PF_2^{(1)}]_2PF_2^{(2)}C_6H_5$	(1) -40.3 (2) -39.0		(P1-1) 650 (P2-2) 635	587
$C_{30}F_{10}H_{25}FeP_5$	$Fe(PF_2C_6H_5)_5$	-31.3		(P-F) ~1100	594
$C_{30}F_{10}Cl_5Fe_2P_5$	$[FeCl(C_6H_5PF_2)_5]FeCl_4$	-38		(P-F) ~1120	610
$C_{36}F_6H_{30}Cl_2OsP_4$	<i>cis</i> $[OsCl_2(PF_3)_2(P(C_6H_5)_3)_2]$	-18.4		(P-F) 1234	601
$C_{35}F_6H_{30}Cl_2P_4Ru$	As above with Os replaced by Ru	-12.4		$^1J(P-F) 1285$ $^3J(P-F) 3.5$	601
$C_{36}F_6H_{31}ClOsP_4$		(1) -7.4 (2) -10.3		(1-3) 37.0 (2-3) 15.9 (P1-1) 1346 (P2-2) 1254 (P1-2) 10.9 (P2-1) 3.5	288
$C_{36}F_6H_{31}ClP_4Ru$	As above with Os replaced by Ru	(1) -13.2 (2) -4.2		(1-3) 36.0 (2-3) 4.9 (P-1) 1350 (P2-2) 1244 (P1-2) 5.0 (P2-1) 1.0	288
$C_{36}F_6H_{32}OsP_4$	As above with Cl replaced by H	-8.3		$\frac{(P1-1)}{+(P1-2)}$ 1296 (P1-2) ~2	611

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{36}F_6H_{32}P_4Ru$	As above with Os replaced by Ru and Cl replaced by H	-6.8	$\begin{matrix} (P1-1) \\ + (P1-2) \end{matrix} \left. \vphantom{\begin{matrix} (P1-1) \\ + (P1-2) \end{matrix}} \right\} 1291$ (P1-2) 5.1	611
$C_{36}F_{10}H_{31}BP_4Ru$	As above with Os replaced by Ru and Cl replaced by BF_4	(1) -13.3 (2) -1.1	(1-3) 28.0 (2-3) 3.0 (P1-1) 1281 (P2-2) 1211 (P1-2) 2.0 (P-2) 1.0	288
$C_{37}F_3H_{30}Cl_2OP_3Ru$	$[RuCl_2(CO)(PF_3)(P(C_6H_5)_3)_2]$	-12.6	(P-F) 1305	
$C_{38}F_5H_{38}NP_4Ru$		(1) -6.2 (2) -27.9	(P1-1) 1300 (P2-2) 1096 (P2-1) 5.7	611
$C_{38}F_9H_{31}O_2P_4Ru$		(1) -13.4 (2) -1.8	(1-3) 36.0 (2-3) 10.5 (P-1) 1351 (P2-2) 1244	
$C_{39}F_2H_{30}MnO_5P_3$	$Mn(CO)_3[P(C_6H_5)_3]_2PO_2F_2$	-78.8	(P-F) 954	573(a)
$C_{39}F_2H_{30}MnO_{11}P_3$	$Mn(CO)_3[P(OC_6H_5)_3]_2PO_2F_2$	-80.0	(P-F) 957	573(a)
$C_{39}F_3H_{37}Cl_2NOP_3Ru$	<i>cis</i> - $[RuCl_2(H \cdot CO \cdot N(CH_3)_2)PF_3(P(C_6H_5)_3)_2]$	-4.7	(P-F) 1257	601
$C_{40}F_4H_{42}Cl_2N_2P_4Ru$	$[RuCl_2(PF_2 \cdot N(CH_3)_2)_2(P(C_6H_5)_3)_2]$	-21.8 -22.2 -32.3 -33.3	(P-F) 1079 $^4J(F-F)$ 21 (P-F) 1137 (P-F) 1131 (P-F) 1116	601

$C_{40}F_4H_{44}N_2P_4Ru$		-30.8		(P-F) 1082 $^3J(P-F)$ 6.0	611
$C_{42}F_8H_{42}P_4$	$[(C_6H_5)_2PF_2^{(1)}(CH_2)_2]_3PF_2^{(2)}$	(1) -42.7	(2) -41.8	(P1-1) 650 (P2-2) 620	587
$C_{48}F_4H_{40}NiO_8P_4$	$Ni[PF(OC_6H_5)_2]_4$	-29.9		$^1J(P-F)$ -1205.8 $^3J(P-F)$ +34.7 $^4J(F-F)$ +5.1	607
$C_{48}F_4H_{40}O_8P_4Pt$	$Pt[PF(OC_6H_5)_2]_4$	-22.5		$^1J(P-F)$ -1269.4 $^3J(P-F)$ +40.3 $^4J(F-F)$ +7.8 $^2J(Pt-F)$ ±1371.6	607
$C_{48}F_{16}H_{40}Co_2HgP_8$	$Hg[Co(PF_2C_6H_5)_4]_2$	-34.3		(P-F) ~1070	594
$C_{54}F_3H_{47}OsP_4$		-7.4		$^1J(P-F)$ 1248 $^3J(P-F)$ ~2	611
$C_{54}F_3H_{47}P_4Ru$	As above with Os replaced by Ru	-4.8		$^1J(P-F)$ 1285 $^3J(P-F)$ 5.8	611
$C_{54}F_4H_{45}Cl_4P_5Ru_2$	$[(C_6H_5)_3P)_2(PF_3^{(1)})RuCl_3RuCl(PF_3^{(2)})(P^{(3)}(C_6H_5)_3)]$				
	Isomer [a]	(1) -8.6	(2) -11.9	(P1-1) 1282 (P2-2) 1292	612
	[b]	(1) -8.2	(2) -10.0	(P1-1) 1283 (P2-2) 1287 (P3-2) 4.0	
	[c]	(1) -6.7	(2) -10.0	(P1-1) 1283 (P2-2) 1287 (P3-2) 4.0	
$C_{56}F_2H_{53}NP_4Ru$		-24.5		(P-F) 1094	611

F bonded to P

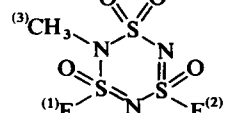
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{58}F_4H_{57}Cl_4N_2P_5Ru_2$	$[(C_6H_5)_3P]_2\{[(CH_3)_2N]PF_2^{(1)}\}RuCl_3RuCl\{PF_2^{(2)}(N(CH_3)_2)\}P(C_6H_5)_3]$			
	Isomer [a]	(1) -18.8 (2) -27.2	(P1-1) 1138 (P2-2) 1157 $^3J(P-F)$ 4.5	612
	[b]	(1) -23.8 (2) -29.1	(P1-1) 1153 (P2-2) 1193 $^3J(P-F)$ 4.5	
$C_{58}F_4H_{57}Cl_4N_2P_5Ru_2$	$[(C_6H_5)_3P]\{[(CH_3)_2N]PF_2\}_2RuCl_3RuCl(P(C_6H_5)_3)_2]$			
		-22.0	(P-F) 1111	612
		-23.7	1122	
		-27.2	1139	
		-31.5	1148	
		-34.2	1104	
		-36.7	1094	
$C_{72}F_3H_{60}Cl_4P_5Ru_2$	$[(C_6H_5)_3P]_2(PF_3)RuCl_3RuCl(P(C_6H_5)_3)_2]$	-8.5	(P-F) 1284	612
F_2ClPSe	$F_2P(Cl)(Se)$	-2.4	(P-F) 1304 (Se-F) 165	527
F_4H_2CsP	$F_4^{(1)}PH_2^{(2)-} Cs^+$	-49.1	(P-F) 798.7 (1-2) 120.8	530
F_4H_2KP	$F_4^{(1)}PH_2^{(2)-} K^+$	-47.6	(P-F) 799.3 (1-2) 119.8	530
$F_4H_3BP_2S$	$F_2^{(1)}P \cdot S \cdot PF_2^{(2)} \cdot BH_3^{(3)}$	(1) -57.8 (2) -56.6	(P1-1) 1336 (P-2-2) 1267 (2-3) 19	523
$F_4H_3BP_2Se$	$F_2^{(1)}P \cdot Se \cdot PF_2^{(2)} \cdot BH_3^{(3)}$	(1) -59.2 (2) -55.2	(P1-1) 1320 (P2-2) 1277 (2-3) 21	523
$F_4Cl_8O_4P_2Sb_2$	$[Cl_4Sb(O_2PF_2)]_2$		(P-F) 1013	527
F_7O_2PSe	$F_2^{(1)}P(O) \cdot OSeF_4^{(2)} \cdot F^{(3)}$	(1) -81.7 (2) +75.3 (3) +61.2	(P-1) 1078 (1-2) 2.0 (2-3) 219 (Se-2) 1407	539
$F_{12}HP_4Rh$	$H^{(2)}Rh(PF_3^{(1)})_4$	(-38 °C) -4.3 (-104 °C) -4.1, -8.0	(1-2) 16 (Rh-1) 16	545
$F_{12}ClIrP_4$	$IrCl(PF_3)_4$	-19.0		545
$F_{12}O_4P_2Sb_2$	$[F_4Sb(O_2PF_2)]_2$		(P-F) 1039	527

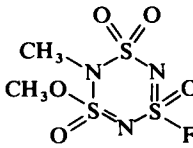
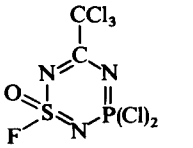
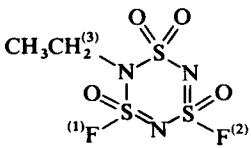
$\text{F}_{24}\text{P}_8\text{Ir}_2$	$\text{Ir}_2(\text{PF}_3)_8$	(24 °C) −4·4 (−20 °C) −4·0, −4·6	545
$\text{F}_{24}\text{P}_8\text{Rh}_2$	$\text{Rh}_2(\text{PF}_3)_8$	(24 °C) −2·8 (−20 °C) −0·5, −4·4	545

Fluorine bonded to platinum

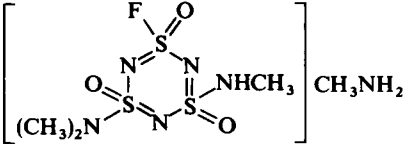
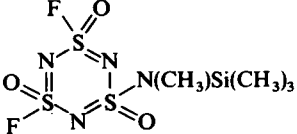
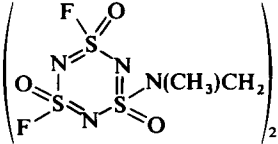
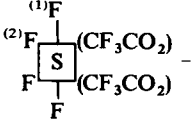
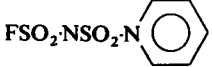
$\text{C}_6\text{F}_6\text{H}_{30}\text{O}_6\text{Pt}_3$	$[(\text{CH}_3)_3\text{Pt}(\text{H}_2\text{O})_3]_2\text{PtF}_6$	−364	585(h)
F_6Pt	PtF_6	+3966±50 (Pt-F) 1072	535

Fluorine bonded to sulphur

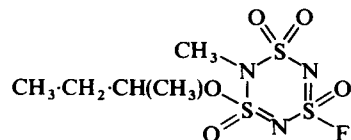
CFH ₃ ClO ₂ S	CH ₃ ⁺ OSOCIF [−]	+90·8					17
CF ₂ H ₃ N ₃ O ₄ S ₃		(1) +63·9	(2) +65·2	(1-2) 3·75	(1-3) 2·6		550
CF ₃ H ₃ S	CH ₃ ⁽²⁾ SF ₃ ⁽¹⁾	(−100 °C)	(1a) +58·0	(1e) −51·8	(1a-1e) 72·0	(1a-2) 15·5	10
					(1e-2) 12·2		
CF ₄ H ₂ S	CH ₂ ⁽³⁾⁽⁴⁾ =SF ₄ ⁽¹⁾⁽²⁾	(−150 °C)	(1) +59·0	(2) +53·6	(1-2) 154·4	(1-3) 10·5	551
					(2-3) 9·7	(2-4) 58·1	
CF ₄ H ₃ NS	CH ₃ ⁽²⁾ N=SF ₄ ⁽¹⁾	+71·3			(1-2) 3·65		552

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
CF ₅ H ₃ S	CH ₃ ⁽²⁾ SF ₄ ⁽¹⁾ F		(1-2) 9·8	551
CF ₆ S	CF ₃ ⁽²⁾ SF ₃ ⁽¹⁾	(-100 °C) (1a) +45·8 (1e) -47·7	(1a-1e) 62·7	10
CF ₆ S	As above	(1a) +52 (1e) -48	(1a-1e) 67 (1a-C) 11·8 (1e-C) 19·1 (2-C) 323·6	555
CF ₁₀ O ₅ S	(F ⁽¹⁾ SF ₄ ⁽²⁾ OO) ₂ CO	(1) +57·3 (2) +55·9	(1-2) 153·3	18
CF ₁₂ H ₃ ClO ₂ SSb ₂	CH ₃ OS(O)F ⁽¹⁾ Cl ⁺ F ⁽²⁾ F ₄ ⁽³⁾ SbF ⁽⁴⁾ SbF ₄ F ⁻	(1) +92 (2) -136 (3) -113 (4) -90	(2-3) 100 (3-4) 60	537
CF ₁₃ H ₃ OSSb ₂	CH ₃ OSF ₂ ⁽¹⁾⁺ F ⁽²⁾ F ₄ ⁽³⁾ SbF ⁽⁴⁾ SbF ₄ F ⁻	(1) +22 (2) -134 (3) -112 (4) -89	(2-3) 100 (3-4) 60	537
C ₂ FH ₆ N ₃ O ₅ S ₃		+64·0		550
C ₂ FCln ₃ N ₃ OPS		+76·9	(P-F) 7·0	30
C ₂ F ₂ H ₄ ClNOS	CH ₂ Cl·CH ₂ ·NS(O)F ₂	+46·3		33
C ₂ F ₂ H ₅ N ₃ O ₄ S ₃		(1) +69·86 (2) +65·0	(1-2) 4 (1-3) 3·5	550

$C_2F_2H_7N_3O_5S_3$		(1) +60.96	(2) +69.15	(1-2) 8.5	(1-3) 3.1	550
$C_2F_5H_3O_3S$	$F^{(1)}SF_4^{(2)} \cdot OO \cdot CO \cdot CH_3$	(1) +62.4	(2) +50.2	(1-2) 156.6		18
$C_2F_5H_4ClO_2S$	$F^{(1)}SF_4^{(2)} \cdot OOCH_2^{(3)} \cdot CH_2Cl$	(1) +63.6	(2) +45.6	(1-2) 155	(2-3) 0.9	561
$C_3FH_{10}N_5O_3S_3$		+78.8				525
$C_3F_2H_3N_3O_2S$		+50.8		(1-2) 0.9		37
$C_4FH_{10}N_5O_3S_3$		+76.05				525
$C_4FH_{12}N_5O_3S_3$		+72.2				525

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_4FH_{15}N_6O_3S_3$		+75.8		525
$C_4F_2H_{10}N_2S$	$(CH_3 \cdot CH_2)_2NSF_2N$	+84.9		414
$C_4F_2H_{12}N_4O_3S_3Si$		+73.8		525
$C_4F_2H_{12}N_4O_3S_3Sn$	As above with Si replaced by Sn	$\left\{ \begin{array}{cc} A+71.55 & B+71.90 \\ +73.87 \end{array} \right\}$ (two isomers)	(A-B) 19	525
$C_4F_3H_9N_2OSSi$	$(CH_3)_3SiN=SF_2^{(1)}=N-CF(O)$	(1) +63.3		37
$C_4F_4H_{10}N_8O_6S_6$		+73.95		525
$C_4F_{10}O_4S^-$		(1) +32.5 (2) +72.6	(1-2) 60	526
$C_5FH_5N_2O_4S_2$		+53.5		521 11

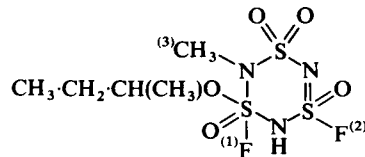
C₃FH₁₂N₃O₅S₃



+64·0, +64·17

550

C₅F₂H₁₃N₃O₅S₃

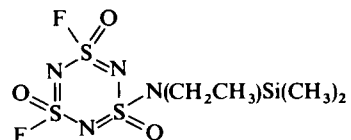


(1) +60·79, +60·74
(2) +69·74, +69·63

(1-2) 9·3 (1-3) 3·2

550

C₅F₂H₁₄N₄O₃S₃Si

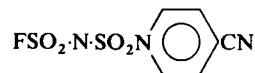


{ A +72·85 B +72·00 }
+73·10
(two isomers)

(A-B) 20

525

C₆FH₄N₃O₄S₂



+54·1

11

C₆FH₅OS

C₆H₅S(O)F

+6

576

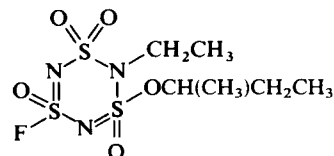
C₆FH₅O₂S

C₆H₅S(O)₂F

+65

576

C₆FH₁₄N₃O₅S₃



+64·13, +64·28

550

C₆F₂H₃NS

C₆H₅SF₂N

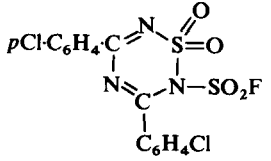
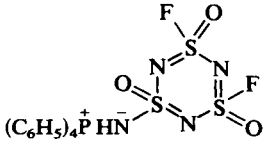
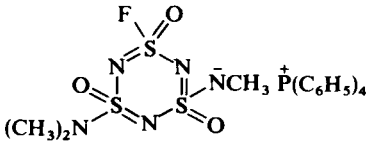
+97·5

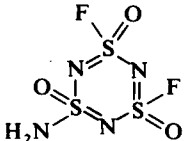
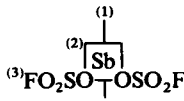
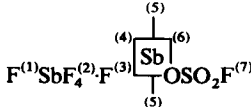
414

F bonded to S

445

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_6F_2H_{14}N_3O_5S_3$		(1) +65.91, +65.96 (2) +69.29, +69.44	(1-2) 9.4 (1-3) 3.0	550
$C_6F_2H_{18}N_2SSi_2$	$(CH_3)_3SiN=SF_2=NSi(CH_3)_2$	+77.4		37
$C_7FH_4ClN_2O_5S_2$		+64.9		11
$C_7F_2H_6N_4O_3S_3$		+73.7, +71.6		525
$C_7F_4H_{18}N_4OS_2Si_2$	$[(CH_3)_3SiNSF_2N]_2CO$	+64.2		37
$C_7F_5H_4NS$	$mCF_3 \cdot C_6H_4 \cdot SF_2^{(1)}N$	(1) +113.4		414
$C_8F_6H_{24}N_2O_6S_3$	$cis-SF_2^{(1)}F_2^{(2)}(FSO_3)_2^{2-}[(CH_3)_4N^+]_2$ (F(1) <i>trans</i> to FSO_3)	(1) +33.1 (2) +85.6	(1-2) 75	119
$C_9FH_{11}O_2S$		+67.74	(1-2) 1.90 (1-3) 0.96 (14) 0.68	413(n)
$C_9FH_{16}N_3O_3S$	$CH_2 \cdot CH_2OCH_2CH_2N \cdot SF=N \cdot CO \cdot NCH_2 \cdot CH_2OCH_2 \cdot CH_2$	+39.0		114(c)

$C_{10}F_{10}H_{24}N_2O_6S_3$	<i>cis</i> $SF_2^{(1)}F_2^{(2)}(CF_3SO_3)_2^-[(CH_3)_4N^+]_2$ (F(1) <i>trans</i> to CF_3SO_3)	(1) -39.3 (2) +13.3 (1-2) 74	119
$C_{11}FH_{20}N_3OS$	$\overline{CH_2(CH_2)_4NSF=N \cdot CO \cdot N(CH_2)_4CH_2}$	+39.0	114(c)
$C_{12}FH_{10}NS$	$(C_6H_5)_2SFN$	+102.9	414
$C_{14}FH_8Cl_2N_3O_4S_2$		+60.5	11
$C_{14}FH_8N_5O_8S_2$	As above with Cl replaced by NO_2	+62.0	11
$C_{14}FH_{10}N_3O_4S_2$	As above with Cl replaced by H	+59.9	11
$C_{16}FH_{14}N_3O_6S_2$	As above with Cl replaced by OCH_3	+58.9	11
$C_{24}FH_{20}N_3S$	$[(C_6H_5)_2N]_2SFN$	+92.9	414
$C_{24}F_2H_{21}N_4O_3PS_3$		+67.9	525
$C_{24}F_8H_{20}BN_4O_4PS_4$	$(C_6H_5)_4P[B(NSOF_2)_4]$	+51.8	609
$C_{27}FH_{29}N_5O_3PS_3$		+74.7	525
$C_{48}F_8H_{40}As_2CoN_4O_4S_4$	$[(C_6H_5)_4As]_2[Co(NSOF_2)_4]$	+24.5	613

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{48}F_8H_{40}As_2N_4O_4PdS_4$	$[(C_6H_5)_4As]_2[Pd(NSOF_2)_4]$	+69.9		613
$C_{48}F_8H_{40}HgN_4O_4P_2S_4$	$[(C_6H_5)_4P]_2[Hg(NSOF_2)_4]$	+67.1		613
$FN_5O_4S_6$	$FSO_2 \cdot NSO_2 \cdot N_4S_4$	+55.8		521 11
FO_3S^-	FSO_3^-	+44.9		19
$F_2H_2N_4O_3S_3$		+73.0		525
$F_2H_5N_5O_3S_3$	As above with NH_2 replaced by $NH_2 \cdot NH_3$	+66.7		525
$F_2O_6S_2$	$S_2O_6F_2$	+43		524
$F_4O_{12}S_4U$	$U(SO_3F)_4$	+39.6		528
$F_6O_3SSb^-$	$F^{(1)}SbF_4^{(2)} \cdot OSO_2F^{(3)-}$	(1) -136.3 (2) -97.6 (3) +46.5		19
$F_6O_6S_2Sb^-$		(1) -85.0 (2) -118.8 (3) +47.0	(1-2) 135	19
$F_{11}O_3SSb^-$		(1) -140.3 (2) -109.6 (3) -79.8 (4) -127.9 (5) -95.1 (6) -119.2 (7) +47.4	(1-2) 105 (2-3) 52 (3-4) 75 (3-5) 95 (4-6) 105	19

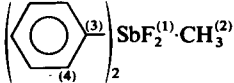
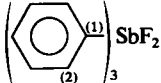
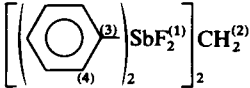
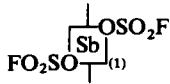
$F_{11}O_3SSb_2^-$		(1) -132.10 (2) -106.0 (3) +45.7	(1-2) 105	19
$F_{11}O_9S_3Sb_2^-$		(1) -88.6 (2) -120.3 (3) +48.2	(1-2) 135	19
$F_{16}O_3Sb_3S^-$		(1) +47.90 (5) -102.5		19
$F_{21}O_3Sb_4S^-$		(1) -139.2 (4) -123.3 (7) +48.1		19

F bonded to S

Fluorine bonded to antimony

CF_6H_3Sb	$CH_3F^{(1)} \cdot SbF_4^{(2)}F^{(3)}$	(1) -196 (2) -115 (3) -141	(1-2) 48 (2-3) 94	537
$C_2F_{10}O_6S_2Sb^-$		-87.3		19
$C_4F_{11}H_9Sb_2$	$(CH_3)_3C^+ F^{(1)} SbF_4^{(2)} \cdot F^{(3)} \cdot SbF_4F^-$	(1) -136 (2) -113 (3) -90	(1-2) 101 (2-3) 60	537

449

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}F_2H_5CrFeO_7Sb$	$C_5H_5(CO)_2FeSbF_2Cr(CO)_5$	-103.4		598
$C_{12}F_2H_5FeO_7SbW$	$C_5H_5(CO)_2FeSbF_2W(CO)_5$	-108.0		598
$C_{13}F_2H_5CrMoO_8Sb$	$C_5H_5(CO)_3MoSb_2 \cdot Cr(CO)_5$	-106.3		598
$C_{13}F_2H_5CrO_8SbW$	$C_5H_5(CO)_3WSbF_2Cr(CO)_5$	-108.8		598
$C_{13}F_2H_5MoO_8SbW$	$C_5H_5(CO)_3MoSbF_2W(CO)_5$	-110.4		598
$C_{13}F_2H_{13}Sb$		-137.1	(1-2) 5.0 (1-C2) 16.7 (1-C3) 13.8 (1-C4) 5.0	575
$C_{18}F_2H_{15}Sb$		-153.2	(F-C1) 15.4 (F-C2) 4.9	575
$C_{21}FH_{10}CrMo_2O_{11}Sb$	$[C_5H_5(CO)_3Mo]_2SbFCr(CO)_5$	-234.0		598
$C_{21}FH_{10}CrO_{11}SbW_2$	$[C_5H_5(CO)_3W]_2SbFCr(CO)_5$	-234.6		598
$C_{21}FH_{10}Mo_2O_{11}SbW$	$[C_5H_5(CO)_3Mo]_2SbFW(CO)_5$	-237.0		598
$C_{25}F_4H_{22}Sb_2$		-127.7	(1-C2) 21.0 (1-C3) 13.5 (1-C4) 4.5	575
$F_6O_6S_2Sb^-$		(1) -91.9		19
F_6Sb^-	SbF_6^-	-115.8		19

Fluorine bonded to selenium

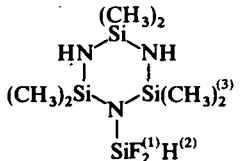
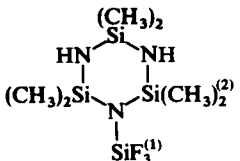
$F_{15}AsO_3Se_3$	$As(OSeF_4^{(1)}F^{(2)})_3$	(1) +78.2	(2) +74.3	(1-2) 230 (Se-2) 1351	(Se-1) 1316	539
-------------------	-----------------------------	-----------	-----------	--------------------------	-------------	-----

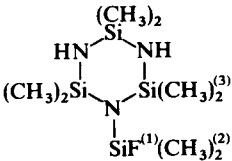
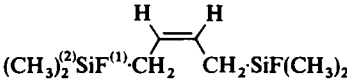
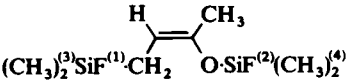
Fluorine bonded to silicon

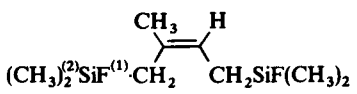
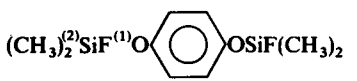
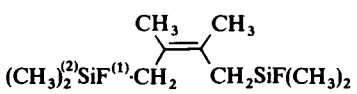
CF_3H_3Si	$CH_3 \cdot SiF_3$	-134.6				522
$CF_5H_3Si_2$	$CH_3^{(3)} \cdot SiF_2^{(2)} \cdot SiF_3^{(1)}$	(1) -125.7	(2) -131.9	(1-2) 14.80	(2-3) 7.41	554
$C_2F_4H_6Si_2$	$CH_3 \cdot SiF_2^{(1)(2)} \cdot SiF_2^{(3)} \cdot CH_3$			$\frac{1}{2}[J(Si-F) + J(Si-SiF)]$ 194		560
				(1-2) +4.8 or -5.9		
				(1-3) 17.1	(2-3) 9.2	
$C_2F_4H_6Si_2$	$(CH_3^{(3)})_2Si^{(5)}F^{(2)}Si^{(4)}F_3^{(1)}$	(1) -124.3	(2) -173.0	(1-2) 9.85	(2-3) 8.15	554
				(Si4-1) 368.4		
				(Si4-2) 56.6	(Si5-1) 42.4	
				(Si5-2) 296.2		
$C_2F_4H_6Si_2$	$CH_3^{(2)}Si^{(3)}F_2^{(1)} \cdot Si^{(4)}F_2^{(5)}CH_3$	-133.7		(1-2) 8.03	(1-5) 13.13	554
				(Si3-1) 334.2		
				(Si4-1) 53.8		
$C_3F_3H_9NPSi$	$(CH_3)_3P=NSiF_3$	-147.5				565
$C_3F_3H_9Si_2$	$(CH_3)_3SiSiF_3$	-122.7		(1-2) 0.9		554
$C_3F_3H_9Si_2$	$(CH_3)_2Si^{(4)}F^{(1)} \cdot Si^{(5)}F_2^{(2)}CH_3^{(3)}$	(1) -176.3	(2) -133.9	(1-2) 9.63	(2-3) 8.33	554
				(Si5-2) 341.8		
				(Si4-2) 39.4		
$C_4F_2H_{12}NPSi$	$(CH_3)_3P=NSiF_2^{(1)} \cdot CH_3^{(2)}$	-124.1		(1-2) 4.3	(P-F) 3.8	565
				(Si-F) 259		
$C_4F_2H_{12}Si_2$	$(CH_3)_3Si^{(4)} \cdot Si^{(3)}F_2^{(1)} \cdot CH_3^{(2)}$	-132.8		(1-2) 8.12		554
				(Si3-1) 347.7		
				(Si4-1) 29.6		
$C_4F_2H_{12}Si_2$	$(CH_3^{(3)})_2Si^{(4)}F^{(1)} \cdot Si^{(5)}F^{(2)} \cdot (CH_3)_2$	-176.0		(1-2) 9.02	(1-3) 8.93	554
				(Si4-1) 302.8	(Si5-1) 35.08	

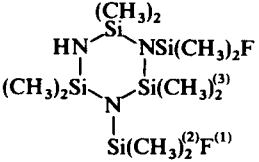
F bonded to Se

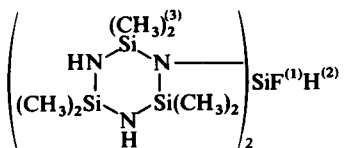
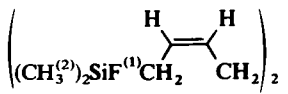
451

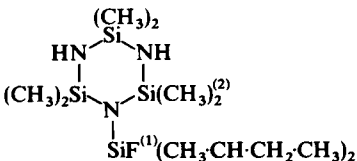
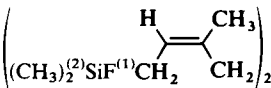
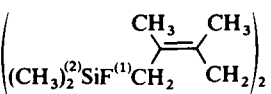
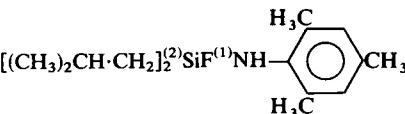
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_4F_3H_9Si$	$(CH_3)_3C \cdot SiF_3$	-106.7		571(b)
$C_4F_3H_{11}NPSi$	$CH_3 \cdot CH_2P(CH_3)_2 = NSiF_3$	-146.2	(P-F) 6.1	565
$C_5FH_{15}NPSi$	$(CH_3)_3P = NSiF^{(1)}(CH_3^{(2)})_2$	-128.8	(1-2) 5.85 (P-F) 3.35 (Si-F) 267.5	565
$C_5FH_{15}Si_2$	$(CH_3)_3Si^{(4)}Si^{(3)}F^{(1)}(CH_3^{(2)})_2$	-175.7	(1-2) 8.57 (Si3-1) 307.1 (Si4-1) 26.7	554
$C_6F_2H_{15}NSi$	$CH_3 \cdot CH_2 \cdot CH(CH_3)SiF_2^{(1)} \cdot N(CH_3^{(2)})_2$	-145.9	(1-2) 1.6	577(b)
$C_6F_2H_{16}N_2Si$	$CH_3 \cdot CH_2 \cdot CH(CH_3)SiF_2 \cdot NH \cdot N(CH_3)_2$	-146.63		578(b)
$C_6F_2H_{21}N_3Si_4$		-197.6	(1-2) 88.0 (1-3) 0.7	579(b)
$C_6F_3H_5Si$	$C_6H_5SiF_3$	-143.1		522
$C_6F_3H_{15}NPSi$	$(CH_3 \cdot CH_2)_3P = N \cdot SiF_3$	-146.7	(P-F) 5.2 (Si-F) 185	565
$C_6F_3H_{15}NPSi$	$(CH_3)_3CP(CH_3)_2 = NSiF_3$	-146.2	(P-F) 5.55 (Si-F) 184.5	565
$C_6F_3H_{20}N_3Si_4$		-187.1	(1-2) 1.0	579(b)
$C_6F_6H_{30}O_6Pt_2Si$	$[(CH_3)_3Pt(H_2O)_3]_2SiF_6$	-137		585(h)
$C_7F_2H_{18}NPSi$	$(CH_3 \cdot CH_2)_3P = N \cdot SiF_2^{(1)} \cdot CH_3^{(2)}$	-123.6	(1-2) 4.0 (P-F) 1.15 (Si-F) 257.5	565

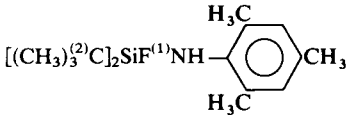
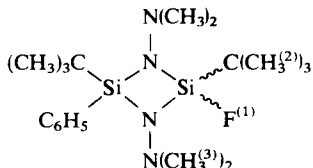
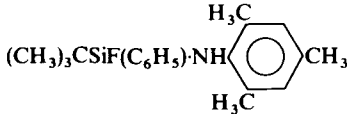
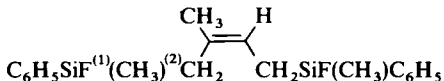
$C_7F_2H_{18}NPSi$	$(CH_3)_3CP(CH_3)_2=N \cdot SiF_2^{(1)} \cdot CH_3^{(2)}$	-123.5	(1-2) 4.2 (Si-F) 256.5	(P-F) 1.7	565
$C_8FH_{17}SSi$	$CH_3^{(2)}(CH_2)_2SiF^{(1)}(CH_3) \cdot CH^{(3)}(CH_3) \cdot SCH=CH_2$ (mixture of diastereoisomers)		(1-2) 7.3	(1-3) ~1.8	589
$C_8FH_{21}NPSi$	$(CH_3 \cdot CH_2)_3P=N \cdot SiF^{(1)}(CH_3^{(2)})_2$	-127.6	(1-2) 5.9 (Si-F) 267	(P-F) 0.3	565
$C_8FH_{21}NPSi$	$(CH_3)_3CP(CH_3)_2=N \cdot SiF^{(1)}(CH_3^{(2)})_2$	-127.7	(1-2) 5.9 (Si-F) 267	(P-F) 0.8	565
$C_8FH_{26}N_3Si_4$		-190.8	(1-2) 7.2	(1-3) 1.0	579(b)
$C_8F_2H_{18}$			(1-2) 7.5		590
$C_8F_2H_{18}OSi_2$			(1-3) 7.5	(2-4) 6.0	590
$C_8F_2H_{19}NSi$	$CH_3 \cdot CH_2 \cdot CH(CH_3)SiF_2^{(1)} \cdot NH^{(2)}C(CH_3^{(3)})_3$	-138.3	(1-2) 3.0 (Si-1) 277.7	(1-3) 0.5	577(b)
$C_8F_2H_{19}NSi$	$CH_3 \cdot CH_2 \cdot CH(CH_3)SiF_2^{(1)(2)} \cdot N(CH_2^{(3)}CH_3)_2$	-144.2	(1-2) 42 (Si-1,2) 286.9	(1,2-3) 1.5	577(b)
$C_8F_2H_{21}NSi_2$	$CH_3 \cdot CH_2 \cdot CH(CH_3) \cdot SiF_2^{(1)(2)} \cdot NCH_3^{(3)}[Si(CH_3^{(4)})_3]$	(1) -140.9 (2) -139.0	(1-2) 4.2 (1,2-4) 0.7 (Si-1,2) 288.7	(1,2-3) 1.2	577(b)
$C_8F_3H_{11}NPSi$	$C_6H_5P(CH_3)_2=N \cdot SiF_3$	-145.7			565
$C_8F_4H_{24}N_2Si_4$	$[CH_3^{(3)}SiF_2^{(1)(2)}N(Si(CH_3)_3)]_2$	(1)(2) -127.6, -128.8	(1-2) 39 (1-3)(2-3) 4.6 (Si-1)(Si-2) 274		592(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_9FH_{15}N_2Si$	$CH_3^{(2)}SiF^{(1)}(C_6H_5)NH^{(3)}N(CH_3)_2$	-143.83	(1-2) 5.6 (1-3) 1.2	578(b)
$C_9F_2H_{14}NPSi$	$C_6H_5P(CH_3)_2=N \cdot SiF_2^{(1)} \cdot CH_3^{(2)}$	-123.7	(1-2) 4.3 (P-F) 3.2 (Si-F) 258	565
$C_9F_2H_{20}Si_2$			(1-2) 7.5	590
$C_9F_3H_{21}NPS$	$(CH_3 \cdot CH_2 \cdot CH_2)_3P=N \cdot SiF_3$	-146.4	(P-F) 5.25 (Si-F) 184.5	565
$C_9F_3H_{21}NPS$	$[(CH_3)_2CH]_3P=N \cdot SiF_3$	-146.6	(P-F) 4.5 (Si-F) 180.5	565
$C_9F_3H_{21}NPS$	$[(CH_3)_3C]_2P(CH_3)=N \cdot SiF_3$	-146.7	(P-F) 4.8 (Si-F) 181	565
$C_{10}FH_{17}NPS$	$C_6H_5P(CH_3)_2=N \cdot SiF^{(1)} \cdot (CH_3^{(2)})_2$	-128.2	(1-2) 6.05 (P-F) 2.2 (Si-F) 267	565
$C_{10}FH_{25}N_2Si$	$(CH_3 \cdot CH_2 \cdot CH \cdot CH_3)_2SiF \cdot NHN(CH_3)_2$	-152.48, -152.68		578(b)
$C_{10}F_2H_{15}NSi$	$C_6H_5 \cdot SiF_2^{(1)} \cdot NH^{(2)}C(CH_3^{(3)})_3$	-136.5	(1-2) 3.5 (1-3) 0.5	571(b)
$C_{10}F_2H_{16}O_2Si_2$			(1-2) 6.0	590
$C_{10}F_2H_{22}Si_2$			(1-2) 7.5	590
$C_{10}F_2H_{24}NPSi$	$(CH_3 \cdot CH_2 \cdot CH_2)_3P=N \cdot SiF_2^{(1)} \cdot CH_3^{(2)}$	-123.7	(1-2) 4.2 (P-F) 0.85 (Si-F) 259	565
$C_{10}F_2H_{24}NPSi$	$[(CH_3)_2CH]_3P=N \cdot SiF_2^{(1)} \cdot CH_3^{(2)}$	-123.4	(1-2) 4.1 (P-F) 1.5 (Si-F) 256	565
$C_{10}F_2H_{24}NPSi$	$[(CH_3)_3C]_2P(CH_3)=N \cdot SiF_2^{(1)} \cdot CH_3^{(2)}$	-123.7	(1-2) 4.1 (P-F) 0.45 (Si-F) 256	565

$C_{10}F_2H_{25}NSi_2$	$CH_3 \cdot CH_2 \cdot CH(CH_3)SiF_2^{(1)(2)} \cdot NCH^{(4)}(CH_3)_2[Si(CH_3^{(5)})_3]$	(1) -133.3 (2) -129.9	(1-2) 40 (1,2-4) 2.5 (Si-1,2) 293.9	(1,2-3) 0.7 (1,2-5) 0.5	577(b)
$C_{10}F_2H_{27}NSi_3$	$CH_3 \cdot CH_2 \cdot CH(CH_3)SiF_2^{(1)(2)} \cdot N[Si(CH_3^{(3)(4)})_3]_2$	(1) -130.2 (2) -127.9	(1-2) 41 (2-4) 0.9	(1-3) 0.7 (Si-1,2) 291.7	577(b)
$C_{10}F_2H_{28}N_4Si_2$	$(CH_3)_2N \cdot NHSi(CH_3^{(2)})_2N[SiF_2^{(1)}C(CH_3^{(3)})_3]N(CH_3)_2$	-146.88	(1-2) 0.8	(1-3) 1.1	578(b)
$C_{10}F_2H_{31}N_3Si_5$		-191.2	(1-2) 5.7	(1-3) 0.6	579(b)
$C_{10}F_4H_{24}N_2Si_2$	$(CH_3^{(2)})_2N \cdot NSiF_2^{(1)}(CH_3 \cdot CH \cdot CH_2 \cdot CH_3)_2$	-139.35	(1-2) 0.9		578(b)
$C_{11}FH_{18}NSi$	$C_6H_5 \cdot SiF^{(1)}CH_3^{(2)} \cdot NHC(CH_3^{(3)})_3$	-139.6	(1-2) 5.4	(1-3) 0.3	571(b)
$C_{11}FH_{27}NPS$	$(CH_3 \cdot CH_2 \cdot CH_2)_3P=N \cdot SiF^{(1)}(CH_3^{(2)})_2$	-127.8	(1-2) 5.75 (Si-F) 267	(P-F) 0.2	565
$C_{11}FH_{27}NPS$	$[(CH_3)_2CH]_3 \cdot P=N \cdot SiF^{(1)}(CH_3^{(2)})_2$	-126.5	(1-2) 5.7 (Si-F) 264.5	(P-F) 3.25	565
$C_{11}FH_{27}NPS$	$[(CH_3)_3C]_2P(CH_3)=N \cdot SiF^{(1)}(CH_3^{(2)})_2$	-127.2	(1-2) 5.8 (Si-F) 264.5	(P-F) 2.15	565
$C_{11}F_2H_{17}NSi$	$CH_3 \cdot CH_2 \cdot CH(CH_3) \cdot SiF_2^{(1)(2)} \cdot N(CH_3^{(3)})C_6H_5$	(1) -141.1 (2) -139.9	(1-2) 41 (Si-1,2) 289.3	(1,2-3) 1.5	577(b)
$C_{11}F_2H_{27}NSi_2$	$(CH_3^{(4)})_3C \cdot SiF_2^{(1)} \cdot N[Si(CH_3^{(2)})_3]C(CH_3^{(3)})_3$	-126.6	(1-2) 1.4 (1-4) 0.8	(1-3) 1.4	571(b)
$C_{11}F_2H_{27}NSi_2$	$CH_3 \cdot CH_2 \cdot CH(CH_3) \cdot SiF_2^{(1)(2)} \cdot N[SiCH_3^{(3)}]_3C(CH_3^{(4)})_3$	(1) -128.8 (2) -126.3	(1-2) 40 (1,2-4) 0.7	(1,2-3) 1.1 (Si-1,2) 294.8	577(b)
$C_{12}FH_{21}N_2Si$	$(CH_3^{(2)})_3CSiF^{(1)}(C_6H_5)NHN(CH_3)_2$	-152.62	(1-2) 0.9		578(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{12}FH_{33}N_2Si_3$	$CH_3 \cdot CH_2CH(CH_3)SiF^{(1)}[NCH_3^{(2)}(Si(CH_3^{(3)}))_3]_2$	-139.9	(1-2) 1.6 (1-3) 1.0 (Si-1) 267.9	577(b)
$C_{12}FH_{41}N_6Si_7$		-200.90	(1-2) 73.0 (1-3) 0.7	579(b)
$C_{12}F_2H_{24}Si_2$			(1-2) 7.5	590
$C_{12}F_2H_{27}NSi$	$CH_3 \cdot CH_2 \cdot CH(CH_3)SiF_2^{(1)(2)} \cdot N[C(CH_3^{(3)(4)}))_3]_2$	(1) -126.0 (2) -123.4	(1-2) 40 (1-3) 0.8 (2-4) 0.7 (Si-1,2) 296.0	577(b)
$C_{12}F_3H_{27}NPSi$	$(CH_3 \cdot CH_2 \cdot CH_2 \cdot CH_2)_3P=N \cdot SiF_3$	-146.4	(P-F) 5.25 (Si-F) 185.5	565
$C_{12}F_3H_{27}NPSi$	$[(CH_3)_2CH \cdot CH_2]_3P=N \cdot SiF_3$	-146.3	(P-F) 4.8 (Si-F) 182.5	565
$C_{12}F_3H_{27}NPSi$	$[(CH_3)_3C]_3P=N \cdot SiF_3$	-146.9	(P-F) 3.7 (Si-F) 175	565
$C_{12}F_5H_{28}NSi$	$(CH_3 \cdot CH_2 \cdot CH_2)_4N^+SiF_5^-$	-137.4		522
$C_{13}F_2H_{23}NSi_2$	$C_6H_5 \cdot SiF_2^{(1)}N[Si(CH_3^{(2)})_3]C(CH_3^{(3)})_3$	-123.9	(1-2) 1.0 (1-3) 0.7	571(b)
$C_{13}F_2H_{30}NPSi$	$(CH_3 \cdot CH_2 \cdot CH_2 \cdot CH_2)_3P=N \cdot SiF_2^{(1)} \cdot CH_3^{(2)}$	-123.7	(1-2) 4.15 (P-F) 0.9 (Si-F) 259	565
$C_{13}F_2H_{30}NPSi$	$[(CH_3)_2CH \cdot CH_2]_3P=N \cdot SiF_2^{(1)}CH_3^{(2)}$	-123.7	(1-2) 4.2 (P-F) 0.9 (Si-F) 256.5	565
$C_{13}F_2H_{30}NPSi$	$[(CH_3)_3C]_3P=N \cdot SiF_2^{(1)}CH_3^{(2)}$	-122.7	(1-2) 4.05 (P-F) 3.2 (Si-F) 253	565
$C_{13}F_4H_{28}NSi$	$(CH_3 \cdot CH_2 \cdot CH_2)_4N^+CH_3SiF_4^-$	-111.0		522
$C_{14}FH_{24}NSi$	$C_6H_5 \cdot SiF^{(1)}(CH_2^{(2)} \cdot CH_2 \cdot CH_2 \cdot CH_3)NHC(CH_3^{(3)})_3$	-145.2	(1-2) 3.5 (1-3) 0.3	571(b)

$C_{14}FH_{33}NPSi$	$(CH_3 \cdot CH_2 \cdot CH_2 \cdot CH_2)_3P=N \cdot SiF^{(1)}(CH_3^{(2)})_2$	-127.8	(1-2) 5.85 (P-F) 0.2 (Si-F) 267.5	565
$C_{14}FH_{33}NPSi$	$[(CH_3)_2CH \cdot CH_2]_3P=N \cdot SiF^{(1)}(CH_3^{(2)})_2$	-127.4	(1-2) 5.7 (P-F) 2.15 (Si-F) 266.5	565
$C_{14}FH_{33}NPSi$	$[(CH_3)_3C]_3P=N \cdot SiF^{(1)}(CH_3^{(2)})_2$	-126.0	(1-2) 5.9 (P-F) 5.65 (Si-F) 264.5	565
$C_{14}FH_{38}N_3Si_4$		-166.4	(1-2) 1.3 $^1J(Si-F)$ 286.4 $^3J(Si-F)$ 7.3	579(b)
$C_{14}F_2H_{25}NSi_2$	$CH_3 \cdot CH_2 \cdot CH(CH_3) \cdot SiF_2^{(1)(2)} \cdot NSi(CH_3^{(3)})_3(C_6H_4CH_3p)$	(1) -137.5 (2) -136.5	(1-2) 41 (1,2-3) 0.8 (Si-1,2) 289.3	577(b)
$C_{14}F_2H_{28}Si_2$			(1-2) 7.5	590
$C_{14}F_4H_{16}N_2Si_2$	$[CH_3^{(3)}SiF_2^{(1)(2)}N(C_6H_5)]_2$	(1)(2) -131.6, -128.7	(1-2) 45 (Si-1,2) 281 (1-3)(2-3) 4.7	592(b)
$C_{14}F_4H_{16}N_2Si_2$	$(CH_3)_2N \cdot N(SiF_2C_6H_5)_2$	-145.20		578(b)
$C_{16}FH_{20}NSi$	$(C_6H_5)_2SiF^{(1)}NHC(CH_3^{(2)})_2$	-146.6	(1-2) 0.4	571(b)
$C_{16}F_2H_{32}Si_2$			(1-2) 7.5	590
$C_{17}FH_{30}NSi$		-132.2	(1-2) 5.3 (Si-F) 296.6	603(b)

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
C ₁₇ FH ₃₀ NSi		-152.2	(1-2) 1.0	603(b)
C ₁₇ FH ₃₆ N ₃ Si ₃	(CH ₃) ₃ CNHSi(CH ₃) ₂ NC ₆ H ₅ N[Si(CH ₃) ₃]SiF ⁽¹⁾ (CH ₃) ₂	-128.0	(1-2) 6.3	604(b)
C ₁₈ FH ₂₅ N ₂ Si	CH ₃ ·CH ₂ ·CH(CH ₃)SiF ⁽¹⁾ [NC ₆ H ₅ (CH ₃) ₂] ₂	-142.2	(1-2) 1.7 (Si-F) 277.7	577(b)
C ₁₈ FH ₃₅ N ₄ Si ₂	 (<i>cis</i> and <i>trans</i>)	[a] -129.70 [b] -129.30	(1-2) 1.2 (1-3) 0.5 (1-2) 1.0 (1-3) 0.6	578(b)
C ₁₈ F ₄ H ₂₈ N ₂ Si ₄	[C ₆ H ₅ SiF ₂ ⁽¹⁾⁽²⁾ N(Si(CH ₃) ₃)] ₂	(1)(2) -133.9, -137.7	(1-2) 39.5	592(b)
C ₁₈ F ₄ H ₃₃ NSi	(CH ₃ ·CH ₂ ·CH ₂) ₄ N ⁺ C ₆ H ₅ SiF ₄ ⁻	-118.7		522
C ₁₉ FH ₂₆ NSi		-150.0		603(b)
C ₁₉ F ₂ H ₂₄ Si ₂			(1-2) 7.5	590
C ₂₀ F ₂ H ₂₆ Si ₂	As above with H on double bond replaced by CH ₃		(1-2) 7.5	590
C ₂₀ F ₂ H ₂₈ N ₂ Si ₂	[C ₆ H ₅ SiFNC(CH ₃) ₃] ₂ (<i>cis</i> and <i>trans</i> isomers)	-137.8, -128.8		571(b)

$C_{20}F_3H_{28}NSi_2$	$C_6H_5SiF^{(1)}(CH_3)^{(3)}N \left(\begin{array}{c} H_3C \\ \\ -C_6H_4- \\ \\ H_3C \end{array} \right) CH_3 SiF_2^{(2)}C(CH_3)_3^{(4)}$	(1) -141.0 (2) -137.4	(1-3) 6.4 (2-4) 1.2 (Si1-1) 288.1 (Si2-2) 293.0	603(b)
$C_{20}F_3H_{28}NSi_2$	As above with C_6H_5 and $C(CH_3)_3$ groups interchanged	(1) -143.8 (2A) -133.9 (2B) -131.7	(1-2) 0.6 (1-3) 6.2 (1-4) 1.0 (2A-2B) 42 (Si1-1) 302.6 (Si-2A) 272.1	603(b)
$C_{20}F_3H_{28}NSi_2$	As above with CH_3 and $C(CH_3)_3$ groups interchanged	(1) -151.0 (2A) -128.0 (2B) -124.0	(2A-2B) 43 (2A-3) 5.1 (2B-3) 1.0 (2-5) 2.5	603(b)
$C_{21}F_2H_{45}N_3Si_5$	$\begin{array}{c} (CH_3)_2 \\ \\ HN-Si-N-SiF^{(2)}(C_6H_5)CH_3^{(3)} \\ \quad \\ (CH_3)_2Si-N-Si(CH_3)_2 \\ \\ SiF^{(1)}[CH_2CH(CH_3)_2]_2 \end{array}$	(1) -190.4 (2) -189.1	(2-3) 7.5	579(b)
$C_{21}F_3H_{38}NSi_2$	$[(CH_3)_2CHCH_2]_2SiF^{(1)}N \left(\begin{array}{c} H_3C \\ \\ -C_6H_4- \\ \\ H_3C \end{array} \right) CH_3 SiF_2^{(2)} \cdot CH_2CH(CH_3)_2$	(1) -135.9 (2) -128.8	(Si-1) 304.0 (Si-2) 284.4	603(b)
$C_{22}F_3H_{24}NSi_2$	$C_6H_5SiF^{(1)}(CH_3)^{(3)}N \left(\begin{array}{c} H_3C \\ \\ -C_6H_4- \\ \\ H_3C \end{array} \right) CH_3 SiF_2^{(2)} \cdot C_6H_5$	(1) -140.3 (2) -135.2	(1-2) 0.7 (1-3) 6.4 (Si-1) 287.5 (Si2-2) 269.8	603(b)
$C_{23}F_3H_{34}NSi_2$	$[(CH_3)_3C]_2SiF^{(1)}N \left(\begin{array}{c} H_3C \\ \\ -C_6H_4- \\ \\ H_3C \end{array} \right) CH_3 SiF_2^{(2)} \cdot C_6H_5$	(1) -153.4 (2) -129.4	(Si-1) 313.7 (Si-2) 274.5	603(b)

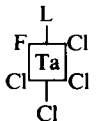
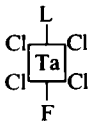
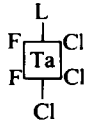
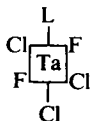
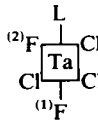
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_{23}F_3H_{34}NSi_2$	$(CH_3^{(3)})_3CSiF^{(1)}(C_6H_5)N \left(\begin{array}{c} H_3C \\ \\ \text{---} \text{C}_6\text{H}_4 \text{---} \\ \\ ^{(4)}H_3C \end{array} \right) CH_3 SiF_2^{(2)}CH_2CH(CH_3)_2$	(1) -150.7 (2A) -129.8 (2B) -126.0	(1-3) 1.1 (2A-2B) 44 (2-4) 3.0	603(b)
$C_{23}F_3H_{34}NSi_2$	As above with $CH_2CH(CH_3)_2$ replaced by $C(CH_3^{(5)})_3$	(1) -152.0 (2A) -135.4 (2b) -134.4	(1-3) 1.0 (2-5) 1.0 (2A-2B) 39.0	603(b)
$C_{24}F_2H_{32}Si_2$	$\left(\begin{array}{c} H \\ \\ CH_3^{(2)}SiF^{(1)}(C_6H_5)CH_2 \\ \\ CH_2 \end{array} \right)_2$		(1-2) 7.5	590
$C_{24}F_6H_{56}N_2Si$	$[(CH_3 \cdot CH_2 \cdot CH_2)_4N]_2SiF_6$	-127.4		522
$C_{25}F_3H_{30}NSi_2$	$(CH_3^{(3)})_3C \cdot SiF^{(1)}(C_6H_5)N \left(\begin{array}{c} H_3C \\ \\ \text{---} \text{C}_6\text{H}_4 \text{---} \\ \\ H_3C \end{array} \right) CH_3 SiF_2^{(2)} \cdot C_6H_5$	(1) -149.7 (2A) -135.3 (2B) -133.3	(1-3) 1.0 (2A-2B) 42	603(b)
$C_{30}F_5H_{61}N_2Si$	$[(CH_3 \cdot CH_2 \cdot CH_2)_4N]_2[C_6H_5SiF^{(1)}F_4^{(2)}]$	(1) -121.3 (2) -114.5	(1-2) 11.0	608
F_4Si	SiF_4	-162		522
F_9Si_3P	$P(^{28}SiF_3)_3$	-113.7	(P-F) 1.8	541
F_9Si_3P	$F_3^{29}SiP(^{28}SiF_3)_2$		(Si-F) 308.0 (F-F) 2.6	541

Fluorine bonded to tin

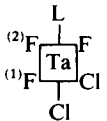
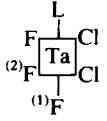
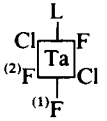
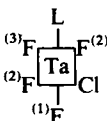
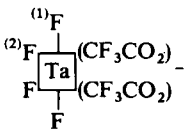
$C_8F_6H_{24}N_2O_6S_2Sn$	<i>cis</i> - $SnF_2^{(1)}F_2^{(2)}(FSO_3)^{2-} [(CH_3)_4N^+]_2$ (F(1) <i>trans</i> to FSO_3)	(1) -142.6 (2) -134.9	(1-2) 54	119
$C_8F_6H_{24}N_2O_6S_2Sn$	<i>trans</i> - $SnF_4(FSO_3)^{2-} [(CH_3)_4N^+]_2$	-134.7		119

$C_{10}F_{10}H_{24}N_2O_6S_2Sn$	<i>cis</i> - $SnF_2^{(1)}F_2^{(2)}(CF_3SO_3)_2^- [(CH_3)_4N^+]_2$ (F(1) <i>trans</i> to CF_3SO_3)	(1) -139.1	(2) -129.8	(1-2) 55	119
$C_{10}F_{10}H_{24}N_2O_6S_2Sn$	<i>trans</i> - $SnF_4(CF_3SO_3)_2^- [(CH_3)_4N^+]_2$	-130.3			119

Fluorine bonded to tantalum

$C_2F_8O_2Ta^-$	$F^{(1)}TaF_4^{(2)}(CF_3CO_2)^-$	(1) +73.7	(2) +56.6	(1-2) 27.5	526
$C_3FH_9Cl_4O_4PTa$	 $L = (CH_3O)_3PO$	+115.44			564
$C_3FH_9Cl_4O_4PTa$	 $L = (CH_3O)_3PO$	+179.52		(P-F) 9	564
$C_3F_2H_9Cl_3O_4PTa$	 $L = (CH_3O)_3PO$	+103.06			564
$C_3F_2H_9Cl_3O_4PTa$	 $L = (CH_3O)_3PO$	+123.88			564
$C_3F_2H_9Cl_3O_4PTa$	 $L = (CH_3O)_3PO$	(1) +161.86	(2) +99.4	(1-2) 33 (P-1) 9	564

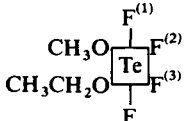
F bonded to Sn

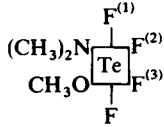
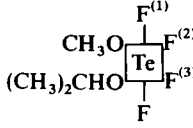
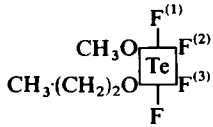
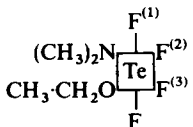
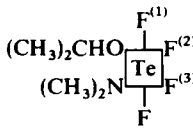
Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$C_3F_3H_9Cl_2O_4PTa$	 $L = (CH_3O)_3PO$	(1) +105.94 (2) +87.20	(1-2) 36	564
$C_3F_3H_9Cl_2O_4PTa$	 $L = (CH_3O)_3PO$	(1) +142.85 (2) +86.70	(1-2) 36 (P-1) 9	564
$C_3F_3H_9Cl_2O_4PTa$	 $L = (CH_3O)_3PO$	(1) +140.85 (2) +103.83	(1-2) 25 (P-1) 9	564
$C_3F_4H_9ClO_4PTa$	$TaF_4Cl(CH_3O)_3PO$	+85.44		564
$C_3F_4H_9ClO_4PTa$		(1) +120.1 (2) +84.37 (3) +70.16	(1-2) 27 (1-3) 35 (2-3) 35 (P-1) 9	564
$C_3F_5H_9O_4PTa$	$TaF_4^{(1)F^{(2)}}(CH_3O)_3PO$	(1) +62.36 (2) +96.70	(1-2) 25 (P-1) 7	564
$C_4F_6H_{12}NO_3STa$	$TaF_4^{(1)F^{(2)}}(FSO_2)^-(CH_3)_4N^+$	(1) +70.4 (2) +103.9	(1-2) 24	119
$C_4F_{10}O_4Ta^-$		(1) +86.9 (2) +70.3	(1-2) 26.5	526
$C_5F_8H_{12}NO_3STa$	$TaF_4^{(1)F^{(2)}}(CF_3SO_2)^-(CH_3)_4N^+$	(1) +70.4 (2) +103.5	(1-2) 25	119
$C_6F_{10}H_{12}NO_6S_2Ta$	<i>cis</i> $TaF_2^{(1)F^{(2)}}(CF_3SO_2)_2^-(CH_3)_4N^+$	(1) +92.5 (2) +127.2	(1-2) 26	119

$C_8F_6H_{12}CuN_4Ta$	$Cu(TaF_6) \cdot 4CH_3CN$	+37				572
$C_{10}F_{12}H_{15}CuN_5Ta_2$	$Cu(TaF_6)_2 \cdot 5CH_3CN$	+37				572
$C_{14}F_5H_{15}OPSTa$	$TaF^{(2)}F_4^{(1)} \cdot OP(C_6H_5)_2SCH_2 \cdot CH_3$	(1) +50.7	(2) +77.5	(1-2) 28		544, 543(h)
$C_{14}F_5H_{15}O_2PTa$	As above with S replaced by O	(1) +51.2	(2) +79.4	(1-2) 28		544(h)
$C_{14}F_5H_{15}PS_2Ta$	As above with O replaced by S	(1) +50.7	(2) +77.5	(1-2) 28		543(h)
$C_{14}F_{10}H_{15}OPSTa$	$ \begin{array}{c} {}^{(5)}F \quad F \\ \diagdown \quad \\ F^{(1)}F_4^{(2)}TaF^{(3)}Ta-F^{(6)} \\ \diagup \quad \\ {}^{(4)}F \quad L \end{array} $ L = $OP(C_6H_5)_2SCH_2 \cdot CH_3$	(1) +111.7 (3) -86.4 (5) +102.3	(2) +65.9 (4) +79.5 (6) +122.7	(1-2) 30 (3-6) 175	(1-3) 160	544(h)
$C_{14}F_{10}H_{15}O_2PTa_2$	As above with S replaced by O	(1) +112.0 (3) -86.1 (5) +103.4	(2) +66.2 (4) +80.3 (6) +123.7	(1-2) 30 (3-6) 180	(1-3) 140	544, 543(h)
$C_{14}F_{10}H_{15}PS_2Ta_2$	As above with O replaced by S	(1) +111.7 (3) -86.4 (5) +102.3	(2) +65.9 (4) +79.5 (6) +122.7	(1-3) 160 (3-6) 175		543(h)
$C_{18}F_5H_{12}Cl_3OPTa$	$TaF^{(1)}F_4^{(2)}(pClC_6H_4)_3PO$	(1) +52.3	(2) +79.6	(1-2) 26		605(h)
$C_{18}F_5H_{12}Cl_3OPTa$	As above with <i>m</i> -isomer	(1) +54.0	(2) +83.2	(1-2) 26		605(h)
$C_{18}F_5H_{12}N_3O_7PTa$	As above with <i>p</i> Cl replaced by <i>m</i> NO ₂	(1) +54.6	(2) +87.5	(1-2) 26		605(h)
$C_{18}F_5H_{15}OPTa$	$TaF_4^{(1)}F^{(2)}OP(C_6H_5)_3$	(1) +49.1	(2) +73.9	(1-2) 29		544, 543, 605(h)
$C_{18}F_{10}H_{15}OPTa_2$	$ \begin{array}{c} {}^{(5)}F \quad F \\ \diagdown \quad \\ F^{(1)}F_4^{(2)}TaF^{(3)}Ta-F^{(6)} \\ \diagup \quad \\ {}^{(4)}F \quad L \end{array} $ L = $OP(C_6H_5)_3$	(1) +110.9 (3) -87.0 (5) +99.9	(2) +65.5 (4) +77.9 (6) +119.5	(1-2) 30 (3-6) 200	(1-3) 165	544, 543(h)
$C_{21}F_5H_{21}OPTa$	$TaF^{(1)}F_4^{(2)}(pCH_3C_6H_4)_3PO$	(1) +47.4	(2) +70.2	(1-2) 28		605(h)
$C_{21}F_5H_{21}OPTa$	<i>m</i> -isomer of above	(1) +48.0	(2) +72.1	(1-2) 28		605(h)

Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)	Ref.
$C_{21}F_5H_{21}O_4PTa$	$TaF^{(1)}F_4^{(2)}(pCH_3OC_6H_4)_3PO$	(1) +46.0	(2) +60.9	(1-2) 26	605(h)
$C_{28}F_4H_{30}O_2P_2S_2Ta^+$	$cis-[TaF_2^{(1)}F_2^{(2)}(OP(C_6H_5)_2SCH_2CH_3)_2]^+$	(1) +67.5	(2) +93.3	(1-2) 28	544(h)
$C_{28}F_4H_{30}O_2P_2S_2Ta^+$	<i>trans</i> -isomer of above	+66.5			544(h)
$C_{28}F_4H_{30}O_4P_2Ta^+$	$cis-[TaF_2^{(1)}F_2^{(2)}(OP(C_6H_5)_2OCH_2CH_3)_2]^+$	(1) +69.4	(2) +96.5	(1-2) 28	544, 5543(h)
$C_{28}F_4H_{30}O_4P_2Ta^+$	<i>trans</i> -isomer of above	+68.5			544, 543(h)
$C_{28}F_4H_{30}P_2S_4Ta^+$	$cis-[TaF_2^{(1)}F_2^{(2)}(SP(C_6H_5)_2SCH_2CH_3)_2]^+$	(1) +67.5	(2) +93.3	(1-2) 28	543(h)
$C_{28}F_4H_{30}P_2S_4Ta^+$	<i>trans</i> -isomer of above	+66.5			543(h)
$C_{36}F_4H_{30}O_2P_2Ta^+$	$cis-[TaF_2^{(1)}F_2^{(2)}(OP(C_6H_5)_3)_2]^+$	(1) +66.1	(2) +85.9	(1-2) 28	544, 543(h)
$F_{11}Ta_2^-$	$[F^{(1)}F_4^{(2)}TaF_4F]^-$	(1) +108.0 (3) -88.8	(2) +63.8	(1-3) 165	543 544

Fluorine bonded to tellurium

$C_2F_4H_6O_2Te$	<i>trans</i> - $TeF_4(OCH_3)_2$	-61.6			559
$C_3F_3H_9O_3Te$	<i>fac</i> - $TeF_3(OCH_3)_3$	-58.1			559
$C_3F_3H_9O_3Te$	<i>mer</i> - $TeF^{(1)}F_2^{(2)}(OCH_3)_3$	(1) -59.7	(2) -67.9	(1-2) 134	559
$C_3F_4H_8O_2Te$		(1) -58.72 (2)(3) -48.39, -50.57		(1-2)(1-3) 152.7, 155.4 (2-3) 171.3	566

$C_3F_4H_9NOTe$		(1) -67.02 (2)(3) -47.49, -53.59	(1-2)(1-3) 145.7, 137.9 (2-3) 162.9	566
$C_4F_2H_{12}O_4Te$	<i>cis</i> - $TeF_2(OCH_3)_4$	-66.8	(Te-F) 3300	559
$C_4F_2H_{12}O_4Te$	<i>trans</i> - $TeF_2(OCH_3)_4$	-72.3		559
$C_4F_4H_8Br_2O_2Te$	<i>cis</i> - $TeF_4^{(1)(2)}(OCH_2 \cdot CH_2Br)_2$	(1) -45.1 (2) -52.5	(1-2) 159	559
$C_4F_4H_{10}O_2Te$		(1) -55.83 (2)(3) -45.89, -49.51	(1-2)(1-3) 153.3, 154.6 (2-3) 174.2	566
$C_4F_4H_{10}O_2Te$		(1) -59.89 (2)(3) -48.80, -50.62	(1-2)(1-3) 154.6, 153.0 (2-3) 173.9	566
$C_4F_4H_{11}NOTe$		(1) -64.36 (2)(3) -44.37, -53.00	(1-2)(1-3) 148.5, 138.9 (2-3) 162.6	566
$C_3FH_{15}O_5Te$	$TeF(OCH_3)_5$	-75.8	(Te-F) 3240	559
$C_5F_4H_{13}NOTe$		(1) -61.64 (2)(3) -41.90, -52.07	(1-2)(1-3) 147.9, 137.0 (2-3) 162.8	566
$C_6F_3H_{12}Br_3O_3Te$	<i>mer</i> - $TeF_2^{(1)}F^{(2)}(OCH_2 \cdot CH_2Br)_3$	(1) -54.3 (2) -49.9	(1-2) 136	559
$C_6F_3H_{12}Br_3O_3Te$	<i>fac</i> - $TeF_3(OCH_2 \cdot CH_2Br)_3$	-50.9		559

Molecular formula	Structure of compounds	Chemical shifts (ppm)		Coupling constants (Hz)	Ref.
C ₆ F ₃ H ₁₂ Cl ₃ O ₃ Te	<i>mer</i> -TeF ₂ ⁽¹⁾ F ⁽²⁾ (OCH ₂ ·CH ₂ Cl) ₃	(1) -55.9	(2) -51.1	(1-2) 137	559
C ₆ F ₃ H ₁₂ Cl ₃ O ₃ Te	<i>fac</i> -TeF ₃ (OCH ₂ ·CH ₂ Cl) ₃	-49.6			559
C ₆ F ₄ H ₁₄ O ₂ Te	<i>trans</i> -TeF ₄ [OCH(CH ₃) ₂] ₂	-38.3			559
C ₆ F ₄ H ₁₄ O ₂ Te	<i>cis</i> -TeF ₂ ⁽¹⁾ F ⁽²⁾ (OCH ₂ CH ₂ CH ₂ CH ₃) ₂	(1) -48.0	(2) -56.9	(1-2) 158	559
C ₆ F ₁₂ H ₆ O ₃ Te	<i>mer</i> -TeF ₂ ⁽¹⁾ F ⁽²⁾ (OCH ₂ ·CF ₃) ₃	(1) -58.9	(2) -56.5	(1-2) 143	559
C ₆ F ₁₂ H ₆ O ₃ Te	<i>fac</i> -TeF ₃ (OCH ₂ ·CF ₃) ₃	-54.7			559
C ₈ F ₂ H ₁₆ Br ₄ O ₄ Te	<i>cis</i> -TeF ₂ (OCH ₂ ·CH ₂ Br) ₄	-54.1			559
C ₈ F ₂ H ₁₆ Br ₄ O ₄ Te	<i>trans</i> -TeF ₂ (OCH ₂ ·CH ₂ Br) ₄	-58.9			559
C ₈ F ₂ H ₁₆ Cl ₄ O ₄ Te	<i>cis</i> -TeF ₂ (OCH ₂ ·CH ₂ Cl) ₄	-54.1			559
C ₈ F ₂ H ₁₆ Cl ₄ O ₄ Te	<i>trans</i> -TeF ₂ (OCH ₂ ·CH ₂ Cl) ₄	-59.1			559
C ₈ F ₁₄ H ₈ O ₄ Te	<i>cis</i> -TeF ₂ (OCH ₂ ·CF ₃) ₄	-58.7			559
C ₈ F ₁₄ H ₈ O ₄ Te	<i>trans</i> -TeF ₂ (OCH ₂ ·CF ₃) ₄	-60.6			559
C ₉ F ₃ H ₂₁ O ₃ Te	<i>fac</i> -TeF ₃ [OCH(CH ₃) ₂] ₃	-46.3			559
C ₉ F ₃ H ₂₁ O ₆ Te	<i>fac</i> -TeF ₃ (OCH ₂ ·CH ₂ OCH ₃) ₃	-53.4			559
C ₁₀ FH ₂₀ Br ₅ O ₅ Te	TeF(OCH ₂ ·CH ₂ Br) ₅	-58.4			559
C ₁₀ FH ₂₀ Cl ₅ O ₅ Te	TeF(OCH ₂ ·CH ₂ Cl) ₅	-60.5			559
C ₁₂ F ₂ H ₂₈ O ₄ Te	<i>fac</i> -TeF ₂ [OCH(CH ₃) ₂] ₄	-49.9			559
C ₁₂ F ₂ H ₂₈ O ₈ Te	<i>cis</i> -TeF ₂ (OCH ₂ ·CH ₂ OCH ₃) ₄	-56.4			559
C ₁₂ F ₂ H ₂₈ O ₈ Te	<i>trans</i> -TeF ₂ (OCH ₂ ·CH ₂ OCH ₃) ₄	-64.0			559
C ₁₅ FH ₃₅ O ₁₀ Te	TeF(OCH ₂ ·CH ₂ OCH ₃) ₅	-60.5			559
F ₁₄ O ₂ Te ₃	<i>cis</i> -F ⁽³⁾ F ⁽⁴⁾ ₄ TeO·TeF ⁽¹⁾ ₂ F ⁽²⁾ ₂ ·OTeF ₄ F	(1) -27.7 (3) -51.8	(2) -35.9 (4) -40.7	(1-2) 174 (2-3) 181	546, 547

$F_{14}O_2Te_3$	$trans\text{-}F^{(2)}F_4^{(3)}TeO\cdot TeF_4^{(1)}\cdot OTeF_4F$	(1) -20.7 (3) -40.3	(2) -51.7	(2-3) 182		546, 547
$F_{15}AsO_3Te_3$	$As(OTeF_4^{(1)}F^{(2)})_3$	(1) -37.5	(2) -47.6	(1-2) 193 (Te-2) 3568	(Te-1) 3643	539
$F_{20}O_4Te_4Xe$	$Xe(OTeF_4^{(1)}F^{(2)})_4$	(1) -39.4	(2) -47.2	(1-2) 188	(Xe-1,2) 61±1	548
$F_{20}O_4Te_5$	$Te(OTeF_4^{(1)}F^{(2)})_4$	(1) -34.3	(2) -44.1	(1-2) 183		547
$F_{22}O_4Te_5$	$cis\text{-}F_2^{(5)}Te(OTeF^{(1)}F_4^{(2)})_2(OTeF^{(3)}F_4^{(4)})_2$	(1) -51.5 (2)(4) -39.7	(3) -51.6 (5) -3.3	(1-2) 174	(3-4) 174	546, 547
$F_{22}O_4Te_5$	$trans\text{-}F_2^{(3)}Te(OTeF^{(1)}F_4^{(2)})_4$	(1) -50.6 (3) +3.9	(2) -38.2	(1-2) 177	(Te-1,2) 3587	539, 547
$F_{26}O_5Te_6$	$F^{(5)}Te(OTeF^{(1)}F_4^{(2)})(OTeF^{(3)}F_4^{(4)})_4$	(1) -48.0 (2)(4) -36.9	(3) -48.1 (5) +14.2			546, 547
$F_{30}O_6Te_6U$	$U(OTeF^{(1)}F_4^{(2)})_6$	(1)(2) -48.5		(Te-1,2) 3715		536
$F_{30}O_6Te_7$	$Te(OTeF^{(1)}F_4^{(2)})_6$	(1) -49.4	(2) -37.3	(1-2) 189		547

Fluorine bonded to titanium

$C_6F_6H_{30}O_6Pt_2Ti$	$[(CH_3)_3Pt(H_2O)_3]_2TiF_6$	+121				585(h)
$C_8F_6H_{24}N_2O_6S_2Ti$	$trans\text{-}TiF_4(FSO_3)_2^{2-}[(CH_3)_4N^+]_2$	+277.5				119
$C_{10}F_{10}H_{24}N_2O_6S_2Ti$	$trans\text{-}TiF_4^{(1)}(CF_3SO_3)_2^{2-}[(CH_3)_4N^+]_2$	(1) +276.9				119

Fluorine bonded to uranium

F_6U	UF_6	+764				536, 524
F_6U	$^{238}UF_6$ $^{235}UF_6$	+764±5 +764±5				535
$F_{10}OTeU$	$F^{(1)}UF_4^{(2)}\cdot OTeF_5$	(1) +759.5	(2) +785.7	(1-2) 9		536

Molecular formula	Structure of compounds	Chemical shifts (ppm)	Coupling constants (Hz)	Ref.
$F_{14}O_2Te_2U$	<i>cis</i> - $F_2^{(1)}UF_2^{(2)}(OTeF_5)_2$	(1) +772.5 (2) +796.4	(1-2) 14	536
$F_{14}O_2Te_2U$	<i>trans</i> - $F_4U(OTeF_5)_2$	+802.4		536
$F_{18}O_3Te_3U$	<i>sym</i> - $F_3^{(1)}U(OTeF_5)_3$	(1) +778.3		536
$F_{18}O_3Te_3U$	<i>asym</i> - $F^{(1)}UF_2^{(2)}(OTeF_5)_3$	(1) +783.7 (2) +806.2	(1-2) 22	536
$F_{22}O_4Te_4U$	<i>cis</i> - $F_2U(OTeF_5)_4$	+784.5		536
$F_{22}O_4Te_4U$	<i>trans</i> - $F_2U(OTeF_5)_4$	+811.4		536
$F_{26}O_5Te_5U$	$FU(OTeF_5)_5$	+787.0		536
Fluorine bonded to vanadium				
$C_4F_{11}H_{12}NO_3SV_2$	$F^{(1)}VF_4^{(2)} \cdot F^{(3)} \cdot VF_4^{(4)}(FSO_3)^- (CH_3)_4N^+$	(1) +489 (2) +421 (4) +416	(1-2) 110 (3-4) 120	119
$C_4F_{15}O_4V_2^-$		(1) +492 (2) +421 (3) -14 (4) +254 (5) +236	(1-2) 90	526
$C_3F_{13}H_{12}NO_3SV_2$	$F^{(1)}VF_4^{(2)} \cdot F^{(3)} \cdot VF_4^{(4)}(CF_3SO_3)^- (CH_3)_4N^+$	(1) +492 (2) +422.2 (4) +416.8	(1-2) 120 (3-4) 120	119
Fluorine bonded to tungsten				
$C_2F_{10}O_2W^{2-}$	$F^{(2)}WF_6^{(1)}(CF_3CO_2)^{2-}$	(1) +65.6 (2) -145.0	(1-2) 57.3 (W-1) 72 (W-2) 48	526
$C_6F_{14}O_6W^{2-}$		(1) +57.3 (2) -67.1	(1-2) 51 (W-1) 73 (W-2) 60	526

$C_{10}F_7H_{15}Cu_5W$	$[Cu(CH_3CN)_5]WF_7$	+142		538
F_4SeW	$WF_4(Se)$		(W-F) 36	529
$F_5H_3O_2W$	$WOF^{(1)}F_4^{(2)-}H_3O^+$		(1-2) 70·2 (W-F) 48·8	534
F_7MW	$M^+ WF_7^-$ (M = Li, Na, K, Rb, Cs)	+134 ± 2		538
F_7TiW	$TeWF_7$	+144		538
F_8W^{2-}	WF_8^{2-}	+63·8	(W-F) 68	526

Fluorine bonded to xenon

F_6Xe	XeF_6	+109·3	(Xe-F) 325	535
$F_{11}Xe_2^+$	$Xe_2F_{11}^+$	+74·0		535

F bonded to W

Miscellaneous studies

F-19 NMR has been used to characterize various organometallic species in solution. Thus the reaction of KrF_2 with AsF_5 , SbF_5 , and PtF_6 gives salts of the form $\text{KrF}^+\text{MF}_6^-$, $\text{KrF}^+\text{Sb}_2\text{F}_{11}^-$, $\text{Kr}_2\text{F}_3^+\text{MF}_6^-$, and $\text{Kr}_2\text{F}_3^+ \cdot x\text{KrF}_2 \cdot \text{MF}_6^-$. (614) The exchange products between GeF_6^{2-} and $\text{Ge}(\text{NCS})_4$, in various ratios, have been studied, (615) as have the gaseous products formed during the exposure of SF_6 to a high-current arc. (616) Equilibrium studies have been made of the covalent and ionic forms of various alkoxyphosphoranes (617) and the investigation of HF and neutral organophosphorus compounds has shown the formation of solvates with subsequent dissociation and association of HF in the organic phase. (618) Structural studies have shown that $-\text{OSeF}_5$ and $-\text{OTeF}_5$ groups have higher electronegativities than F. (619)

Thermodynamic parameters, determined by F-19 NMR, for the exchange of the fluoro ligands of BeF_4^{2-} (620) and methyl tetrafluorosilicate anions (621) have been reported, as have the equilibrium constants for the inter-conversion of HF, HF_2^- , and H_2F_2 . (622)

Rotational and vibrational motion in SiF_4 , BF_3 , and SF_6 have been examined by the observation in the change of the F-19 chemical shifts with temperature and pressure. (294, 295) Contributions of $^7\text{Li}^+$ to F-19 relaxation rates in $^7\text{LiBF}_4$ and $^6\text{LiBF}_4$ solutions in D_2O have been evaluated. (623)

IX. FLUORINE NMR STUDIES IN THE SOLID STATE, EXCLUDING POLYMERS

This Page Intentionally Left Blank

Molecular formula	Structure	Ref.
$C_2F_3AgO_2$	CF_3CO_2Ag	723
$C_2F_4S_2$	$S_2C_2F_4$	724
C_6F_6	C_6F_6	725
$C_6F_6H_{20}N_2Sn$	$(Me_3NH)_2SnF_6$	726
$C_8F_6H_{24}As_2Si$	$(Me_4As)_2SiF_6$	727
$C_8F_6H_{24}N_2Si$	$(Me_4N)_2SiF_6$	727
$C_8F_6H_{24}P_2Si$	$(Me_4P)_2SiF_6$	727
$C_8F_6H_{24}Sb_2Si$	$(Me_4Sb)_2SiF_6$	727
$C_{12}F_2H_8$	$pF \cdot C_6H_4 \cdot C_6H_4 \cdot Fp$	728
$FCa_5O_{12}P_3$	$Ca_5(PO_4)_3F$	691
FH_2AlO_2	$AlF(OH)_2$	696
FH_2O_2Si	$SiF(OH)_2$	696
FK	KF	692
FLi	LiF	693, 694
FNa	NaF	692
FOYb	YbOF	695
F_2HAlO	AlF_2OH	696
F_2HCs	$CsHF_2$	702, 703
F_2HK	KHF_2	704, 703
F_2HLi	$LiHF_2$	703
F_2HNa	$NaHF_2$	703
F_2HOCs	CsF_2OH	696
F_2HRb	$RbHF_2$	703
F_2HTl	$TlHF_2$	703
$F_2H_8O_4Zn$	$ZnF_2 \cdot 4H_2O$	705
F_2Ba	BaF_2	697, 692
F_2Ca	CaF_2	698, 699, 692
F_2Cd	CdF_2	692
F_2Kr	KrF_2	700
F_2Pb	PbF_2	701
F_2Sr	SrF_2	692
$F_{1+2x}Na_{1-x}Y_x$ ($0.5 < x < 0.62$)	$Na_{1-x}Y_xF_{1+2x}$	719
F_3HBa	$BaFHF_2$	703
F_3HSr	$SrFHF_2$	703
F_3Al	AlF_3	696
F_3CoK	$KCoF_3$	706
F_3CsNi	$CsNiF_3$	707
F_3CuK	$KCuF_3$	708
F_3Er	ErF_3	709
F_3Ho	HoF_3	710
F_3KMn	$KMnF_3$	711
F_3La	LaF_3	712
F_3MnRb	$RbMnF_3$	713
F_3NiK	$KNiF_3$	714
F_3Sc	ScF_3	696
F_3Tb	TbF_3	710
F_3Ti	TiF_3	692
F_4BK	KBF_4	692

Molecular formula	Structure	Ref.
F ₄ BLi	LiBF ₄	692
F ₄ BNa	NaBF ₄	692
F ₄ BRb	RbBF ₄	692
F ₄ CuK ₂	K ₂ CuF ₄	715
F ₄ Si	SiF ₄	716
F ₅ CsOU	CsUOF ₅	717
F ₆ AsHg _{2.86}	Hg _{2.86} AsF ₆	718
F ₆ BaSi	BaSiF ₆	692
F ₆ KP	KPF ₆	692
F ₆ KSi	KSiF ₆	692
F ₆ LiP	LiPF ₆	692
F ₆ S	SF ₆	720
F ₆ U	UF ₆	721
F ₆ W	WF ₆	729
F ₁₀ H ₄ Cs ₂ Ge	Cs ₂ GeF ₆ ·4HF	722
F ₁₀ H ₄ Cs ₂ Sn	Cs ₂ SnF ₆ ·4HF	722
F ₁₀ H ₄ K ₂ Sn	K ₂ SnF ₆ ·4HF	722

X. FLUORINE NMR STUDIES OF POLYMERS

The table is ordered according to the monomers forming the polymer, with the lowest molecular weight monomer appearing first.

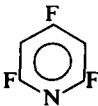
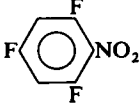
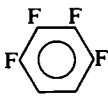
This Page Intentionally Left Blank

Molecular formula of monomer components	Structure of monomer components of polymer	Ref.
C_2FH_3/C_3F_6	$CH_2=CHF/CF_3\cdot CF=CF_2$	730
$C_2FH_3/C_4F_6H_2$	$CH_2=CHF/(CF_3)_2C=CH_2$	731
C_2FH_3	$CH_2=CHF$	732, 733
C_2F_4/C_3F_6	$CF_2=CF_2/CF_3\cdot CF=CF_2$	734
C_2F_4/C_3F_6	$CF_2=CF_2/CF_3\cdot CF=CF_2$	735
C_2F_4/C_3H_6	$CF_2=CF_2/CH_3\cdot CH=CH_2$	736
C_4F_6	$CF_2=CF\cdot CF=CF_2$	737
C_8FH_7	$pF\cdot C_6H_4\cdot CH=CH_2$	738
C_8FH_7	$mF\cdot C_6H_4\cdot CH=CH_2$	738
<i>Miscellaneous studies</i> : various fluorinated polymers have been studied		756

This Page Intentionally Left Blank

XI. FLUORINE NMR STUDIES USING LIQUID CRYSTAL SOLUTIONS

This Page Intentionally Left Blank

Molecular formula	Structure	Ref.
CFH ₃	CFH ₃	740
CF ₂ H ₂	CF ₂ H ₂	740
CF ₃ H	CF ₃ H	740
CF ₃ Br	CF ₃ Br	741
CF ₃ Cl	CF ₃ Cl	741
CF ₃ I	CF ₃ I	741
C ₂ F ₃ Cl ₃	CF ₃ CCl ₃	741
C ₂ F ₄ S ₂	S ₂ C ₂ F ₄	724
C ₅ F ₂ H ₃ N		742
C ₅ F ₃ H ₂ N		742
C ₆ FH ₄ Br	<i>p</i> Br·C ₆ H ₄ ·F	743
C ₆ FH ₄ Cl	<i>p</i> Cl·C ₆ H ₄ ·F	743
C ₆ FH ₄ I	<i>p</i> I·C ₆ H ₄ ·F	743
C ₆ FH ₄ NO ₂	<i>p</i> NO ₂ ·C ₆ H ₄ ·F	743, 744
C ₆ FH ₅	PhF	745, 746
C ₆ FH ₆ N	<i>p</i> NH ₂ ·C ₆ H ₄ ·F	743
C ₆ F ₂ H ₄	<i>o</i> F·C ₆ H ₄ ·F	746
C ₆ F ₂ H ₄	<i>m</i> F·C ₆ H ₄ ·F	746
C ₆ F ₂ H ₄	<i>p</i> F·C ₆ H ₄ ·F	746
C ₆ F ₃ H ₂ ·NO ₂		748
C ₆ F ₃ Cl ₃		747
C ₆ F ₄ H ₂		
C ₆ F ₄ H ₂		746, 749
		746
C ₆ F ₅ H	C ₆ F ₅ H	746, 749
C ₆ F ₆	C ₆ F ₆	746, 749

Molecular formula	Structure	Ref.
C ₇ FH ₄ DO	<i>p</i> F·C ₆ H ₄ ·CDO	750
C ₇ FH ₅ O	<i>p</i> F·C ₆ H ₄ ·CHO	750
C ₇ FH ₅ O	C ₆ H ₅ ·COF	751
C ₇ F ₃ H ₄ Cl	<i>p</i> CF ₃ ·C ₆ H ₄ ·Cl	752
C ₈ F ₁₅ CsO ₂	CF ₃ ·(CF ₂) ₆ ·CO ₂ Cs	753
C ₈ F ₁₅ LiO ₂	CF ₃ ·(CF ₂) ₆ ·CO ₂ Li	754
C ₈ F ₁₅ H ₄ NO ₂	CF ₃ ·(CF ₂) ₆ ·CO ₂ NH ₄	753
F ₃ PS	P(S)F ₃	739

XII. BIOLOGICAL APPLICATIONS OF F-19 NMR

F-19 NMR continues to be a useful method of studying various biological phenomena. In the present section we summarize the papers pertaining to these studies. Limitations of space prevent any detailed review and only the larger biological molecules and their reactions are mentioned. Data for the smaller molecules, such as the fluoro carbohydrates and steroids, are found in the preceding tables.

The incorporation of fluoroamino acids into various proteins has proved useful in the study of their structures and interactions. Thus 6-fluorotryptophan and 3-fluorotyrosine have been incorporated into alkaline phosphatase (625, 626) and dihydrofolate reductase. (627–629) The latter amino acid has also been incorporated into the lipoprotein of *E. coli* in order to study the *in situ* structure of its membrane. (630) Fluorotyrosyl derivatives of M13 coat protein have been incorporated into small phospholipid vesicles in order to study the motional properties within these membranes. (667) 5-Fluorouridine has been used to replace all the uridine and uridine-derived residues of the valine-specific tRNA_I of *E. coli* (631) and 5-fluorouracil has also been incorporated into RNA. (632)

3-Bromo-1,1,1-trifluoropropanone has been shown to react specifically with SH groups and as such allows quantitation and characterization of cysteine residues. (633) This particular reagent has been used to trifluoroacetylate the cysteine residues of aspartate transaminase, (634) parvalbumin from carp, (635) haemoglobin, (636) glyceraldehyde phosphate dehydrogenase, (637) coenzyme A, (638) and several others mentioned below, and as such has allowed F-19 NMR investigations of their structures to be made. Trifluoroacetyl derivatives of neurotoxin II have been used to analyse partially the backbone folding in solution. (639) S-Trifluoromethylmercuric trifluoroacetate has been used to derivatize papain, (640) while the structure of the iron-sulphur proteins, *Clostridium pasteurianum* F_{dox} and *spinach* F_{dox}, has been investigated by the preparation of paramagnetic complexes containing the *p*CF₃·C₆H₄ group. (641) The lysine residues of human serum albumin showed differential rates of reaction with 2,6-dinitro-4-trifluoromethylbenzene sulphonate. (642)

As in previous years, F-19 NMR has proved an invaluable tool for the investigation of the binding of smaller molecular weight compounds with proteins. Thus the following have been studied: binding of *p*-fluorocinnamate, (643) derivatives of tryptophan and phenylalanine, (668) ring-fluorinated derivatives of *N*-trifluoroacetylphenylalanine, (669) and *N*-acetyl derivatives of fluorophenylalanines (670) with chymotrypsin; difluorooxaloacetate (644, 645) and phosphopyridoxyl trifluoroethylamine (646, 647) with aspartate transaminase; fluoride ion with oxyhaemocyanin;

(648) various trifluoroacetylpeptide chloromethyl ketones with porcine pancreatic elastase; (649, 650) various trifluoromethyl analogues of nitrophenylhapten with mouse myeloma immunoglobins; (651, 652) cytochrome c derivatives, having single lysine residues modified by trifluoroacetylation or trifluoromethylphenyl carbamylation, with cytochrome oxidase and succinate-cytochrome c reductase; (653, 654) trifluoroacetylated derivatives of histone H3 with histone H4 and other histones; (655) Ca^{2+} and Mg^{2+} with trifluoromethylacetylated troponin C; (656) oligonucleotide with gene 5 of filamentous bacteriophages; (657, 658) the ternary complex between 5-fluoro-2'-deoxyuridylate, 5,10-methylene tetrahydrofolate, and native thymidylate synthetase; (659) and 5-fluorodeoxyuridylate-thymidylate synthetase binary complex. (666)

Phospholipid membranes have received a certain amount of attention. Halothane interaction with dipalmitoylphosphatidylcholine liposomes (660, 661, 665) has been studied. 8,8-Difluoromyristate has been incorporated into the phospholipids of *E. coli* and has been used to study intramembrane interactions, (662, 663) while motion within the lipid bilayer of synthetic 1-palmitoyl-2-(8,8-difluoropalmitoyl)-*sn*-glycero-3-phosphorylcholine (664) has been monitored.

Theoretical calculations of fluorine-proton nuclear Overhauser effects in fluorine-labelled macromolecular systems have been reported. (671)

Zeumatography, using F-19 NMR, has been shown to be possible by a variant of the sensitive point method. (755)

XIII. THEORETICAL STUDIES

Various reviews dealing with theoretical work have appeared. These deal with nuclear shielding, (763) semi-empirical calculations of the chemical shifts of nuclei other than protons, (672) calculation of coupling constants, (673) and coupling between carbon and first-row nuclei. (674) During the review period no new aspects of basic theory have been propounded. Indeed most of the work reported involving fluorine has used existing methods in an attempt to assess their applicability to predict chemical shifts and coupling constants. As such this has meant that the greatest use has been made of semi-empirical molecular orbital methods which allow comparative calculations to be made in quite large molecules in reasonable computation times.

Ab initio calculations of nuclear shielding have been restricted to small molecules. In particular the H-1 and F-19 nuclear shielding in HF has received considerable attention. (675, 757-760) Other small molecules that have been investigated are F^- , F_2 , FOH, (760, 762) and FHF^- . (675, 760, 761) Semi-empirical calculations, using INDO MO theory, have shown that

the diamagnetic component of the nuclear screening constant for F-19 remains roughly constant as its chemical environment changes and that chemical shift differences are mainly due to the paramagnetic component. (676) Consequently the latter contribution to screening constants and their anisotropies has been calculated, employing INDO and CNDO/2 methods, for 50 fluorine molecular environments including binary fluorides and fluorocarbons. (677) Although agreement with experimental data for the former is reasonably good the presence of electric field effects in the latter is not accounted for. A more limited series of calculations (SCF LCGO MO) have been performed on fluoroethane, ethylene, and acetylene. (678)

Ab initio SCF perturbation theory of nuclear spin-spin coupling constants continues to be available only for small molecules. The difficulties encountered in the calculation of the coupling in HF, even by the most sophisticated methods, emphasize the limitations of our present knowledge. (679, 680) Calculations in larger molecules invariably employ semi-empirical methods. Many calculations have been performed using SCF perturbation theory, working within the INDO MO framework, to calculate the Fermi contact, spin dipolar, and orbital contributions to J , by the calculational procedures of Blizzard and Santry. (681) The latter two contributions are particularly important for $J(\text{FF})$ and $J(\text{FC})$ calculations. Thus the variation of $J(\text{FF})$ and $J(\text{FH})$ in all the fluorobenzenes (402) and the effects of substituents upon $J(\text{FF})$ in the pentafluorobenzenes (417) and fluoro-substituted furans (682) were successfully reproduced. More difficulty has been experienced in the calculation of $J(\text{FC})$ values in aromatic systems. (402, 417, 683, 684) Although the importance of the spin-dipolar and orbital contributions was evident the trends in the couplings were poorly reproduced. More success has been achieved with the long-range $J(\text{FC})$ and $J(\text{FH})$ values in aliphatic molecules. Thus substituent trends upon $^3J(\text{FC})$, $^4J(\text{FC})$, and $^5J(\text{FH})$ in various deoxyfluoropyranoses were well reproduced. (510) A Karplus type angular dependence for $^3J(\text{FC})$ was predicted (510) and found experimentally. (685, 686) Substituent effects upon *trans* $^3J(\text{FC})$ were shown to be complex and analogous to those of $^3J(\text{CH})$ and $^3J(\text{CC})$. (687)

Many papers have appeared dealing with the transmission of substituent effects through non-conjugated and conjugated systems using the fluorine nucleus as a probe of the electronic environment. All the data in these papers are included in the tables, and their contents will not be considered further here. The relationship between NMR substituent chemical shift data and substituent reactivity parameters has been reconsidered in terms of a general theory for structure-property relationships. (688) Literature data for F-19, C-13, and H-1 chemical shifts in aromatic and olefinic systems have been correlated using a multi-parameter equation, (689) and a generalized attempt has been made to explain γ -effects upon F-19, C-13, and P-31 chemical shifts in terms of bond angle changes. (690)

XIV. INDEX OF REVIEWS

The following comprises a list of reviews, containing F-19 NMR data, that have appeared during 1976–1978.

General

- (a) L. Cavalli, *Annual Reports NMR Spectroscopy*, 1976, **6B**, 43. Fluorine-19 NMR spectroscopy.
- (b) R. Fields, *Annual Reports NMR Spectroscopy*, 1977, **7**, 1. Fluorine-19 NMR spectroscopy of fluoroalkyl and fluoroaryl derivatives of transition metals.
- (c) P. W. Hickmott, M. Cais, and A. Modiano, *Annual Reports NMR Spectroscopy*, 1976, **6C**, 1. NMR data on organic-metal carbonyl complexes (1965–1971).
- (d) *Pure Appl. Chem.*, 1976, **45**, 217. Recommendation for the presentation of NMR data for publication in chemical journals. B. Conventions relating to spectra from other nuclei.
- (e) R. K. Harris (ed.), *Nuclear Magnetic Resonance* (Specialist Periodical Reports), The Chemical Society, London, 1976, Vol. 5.
- (f) R. J. Abraham (ed.), *Nuclear Magnetic Resonance* (Specialist Periodical Reports), The Chemical Society, London, 1977, Vol. 6.
- (g) R. J. Abraham (ed.), *Nuclear Magnetic Resonance* (Specialist Periodical Reports), The Chemical Society, London, 1978, Vol. 7.
- (h) J. R. Wasson and D. R. Lorenz, *Analyt. Chem.*, 1976, **48**, 246R. NMR spectroscopy.

Fluorine Chemical Shifts

- (a) K. A. K. Ebraheem and G. A. Webb, *Progr. Nucl. Magn. Res. Spectroscopy*, 1977, **11**, 149. Semi-empirical calculations of the chemical shifts of nuclei other than protons.
- (b) W. J. Hehre, R. W. Taft, and R. D. Topsom, *Progr. Phys. Org. Chem.*, 1976, **12**, 159. *Ab initio* calculations of charge distributions in monosubstituted benzenes and in *meta*- and *para*-substituted fluorobenzenes. Comparison with H-1, C-13, and F-19 NMR substituent shifts.

Fluorine Coupling Constants

- (a) J. W. Emsley, L. Phillips, and V. Wray, *Progr. Nucl. Magn. Res. Spectroscopy*, 1976, **10**, 85. Fluorine coupling constant.
- (b) J. Hilton and L. H. Sutcliffe, *Progr. Nucl. Magn. Res. Spectroscopy*, 1976, **10**, 27. The “through-space” mechanism in spin-spin coupling.

- (c) J. Kowalewski, *Progr. Nucl. Magn. Res. Spectroscopy*, 1977, **11**, 1. Calculations of nuclear spin-spin coupling constants.
- (d) R. E. Wasylshen, *Annual Reports NMR Spectroscopy*, 1977, **7**, 1. Spin-spin coupling between carbon-13 and the first-row nuclei.

Other Aspects

- (a) U. Haeblerlen, *Adv. Magn. Resonance*, 1976, **8**, Suppl. 1. High resolution NMR in solids: selective averaging.
- (b) C. L. Khetrpal and A. C. Kunwar, *Adv. Magn. Resonance*, 1977, **9**, 301. NMR studies of molecules oriented in thermotropic liquid crystals.
- (c) B. Unterhalt and U. Pindur, *Dtsch. Apoth.-Ztg.*, 1977, **117**, 1317. The importance of fluoro organic compounds for drugs.
- (d) J. S. Cohen, *CRC Crit. Rev. Biochem.*, 1978, **5**, 25. Nuclear magnetic resonance studies of biologically relevant isotopes other than hydrogen.
- (e) P. Bouchet, C. Coquelet, and J. Elguero, *Bull. Soc. Chim. France*, 1977, 171. The fluoroazoles: synthesis, properties and applications.
- (f) S. V. Zemskov and S. P. Gabuda, *J. Struct. Chem. (U.S.S.R.)*, 1976, **17**, 772. Characteristic structural features of the hexafluoro complexes of noble metals.
- (g) J. T. Gerig, *Biol. Magn., Resonance*, 1978, **1**, 139. Fluorine magnetic resonance in biochemistry.

This Page Intentionally Left Blank

REFERENCES

1. S-G. Huang and M. T. Rogers, *J. Chem. Phys.*, 1978, **68**, 5601.
2. W. Gombler, *Z. Anorg. Chem.*, 1978, **439**, 193.
3. N. D. Volkov, V. P. Nazaretyan and L. M. Yagupolskii, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1655.
4. G. A. Wheaton and D. J. Burton, *J. Org. Chem.*, 1978, **43**, 2643.
5. W. Gombler, *J. Fluorine Chem.*, 1977, **9**, 233.
6. A. B. Burg, *Inorg. Nuclear Chem. Letters*, 1977, **13**, 199.
7. J. A. Morrison, L. L. Gerchman, R. Eujen and R. L. Lagow, *J. Fluorine Chem.*, 1977, **10**, 333.
8. R. G. Cavell, J. A. Gibson and K. I. The, *J. Amer. Chem. Soc.*, 1977, **99**, 7841.
9. D. Naumann, H. H. Heinsen and E. Lehmann, *J. Fluorine Chem.*, 1976, **8**, 243.
10. W. Gombler and R. Budenz, *J. Fluorine Chem.*, 1976, **7**, 115.
11. H. W. Roesky, M. Aramaki and L. Schoenfelder, *Z. Naturforsch.*, 1978, **33b**, 1072.
12. K. Seppelt, *Angew. Chem.*, 1977, **89**, 325.
13. G. Kloeter, W. Lutz, K. Seppelt and W. Sundermeyer, *Angew. Chem.*, 1977, **89**, 754.
14. R. G. Cavell and L. F. Doty, *Inorg. Chem.*, 1978, **17**, 3086.
15. W. Gombler, *Z. Anorg. Chem.*, 1978, **439**, 207.
16. E. Lehmann, *J. Chem. Res.*, 1978, 42.
17. G. A. Olah and D. J. Donovan, *J. Amer. Chem. Soc.*, 1978, **100**, 5163.
18. M. J. Hopkinson and D. D. Desmarteau, *J. Fluorine Chem.*, 1976, **7**, 501.
19. D. Brunel, A. Germain and A. Commeyras, *Nouveau J. Chim.*, 1978, **2**, 275.
20. L. A. Shimp and R. J. Lagow, *Inorg. Chem.*, 1977, **16**, 2974.
21. K. G. Sharp, S. Li and R. B. Johannesen, *Inorg. Chem.*, 1976, **15**, 2295.
22. C. A. Udovich, M. A. Krevalis and R. J. Clark, *Inorg. Chem.*, 1976, **15**, 900.
23. K. Okuhara, *J. Org. Chem.*, 1978, **43**, 2745.
24. L. S. Boguslavskaya, N. B. Malnikova, A. P. Voronin and V. R. Kartashov, *J. Fluorine Chem.*, 1978, **12**, 257.
25. J. Burdon, J. R. Knights, I. W. Parsons and J. C. Tatlow, *J. Chem. Soc., Perkin I*, 1976, 1930.
26. J. Schraml, J. Vcelak, G. Engelhardt and V. Chvalovsky, *Coll. Czech. Chem. Comm.*, 1976, **41**, 3758.
27. T. Spoormaker and M. J. A. de Bie, *Rec. Trav. Chim.*, 1978, **97**, 135.
28. A. Haas and M. Lieb, *Chem. Ber.*, 1978, **111**, 2891.
29. A. Haas and M. Yazdanbakhsh, *Chem. Ber.*, 1976, **109**, 1976.
30. W. Heider and O. Glemser, *Chem. Ber.*, 1978, **111**, 731.
31. J. Lee and B. G. Willoughby, *Spectrochim. Acta*, 1977, **33A**, 395.
32. R. E. Banks, R. N. Haszeldine and P. E. Jackson, *J. Fluorine Chem.*, 1978, **12**, 153.
33. H. Kluever and O. Glemser, *Chem. Ber.*, 1977, **110**, 1597.
34. R. Eujen and R. J. Lagow, *J. Chem. Soc., Dalton*, 1978, 541.
35. S. Ng, *J. Magn. Resonance*, 1978, **30**, 139.
36. B. Haley, R. N. Haszeldine, B. Hewitson and A. E. Tipping, *J. Chem. Soc., Perkin I*, 1976, 525.
37. I. Stahl, R. Mews and O. Glemser, *J. Fluorine Chem.*, 1978, **11**, 455.
38. I. Stahl, R. Mews and O. Glemser, *Z. Naturforsch.*, 1978, **33b**, 1417.
39. E. R. Falardeau and D. D. DesMarteau, *J. Amer. Chem. Soc.*, 1976, **98**, 3529.
40. G. H. Sprenger and A. H. Cowley, *J. Fluorine Chem.*, 1976, **7**, 333.
41. A. Haas, M. Haerberlein and C. Krueger, *Chem. Ber.*, 1976, **109**, 1769.
42. I. Stahl, R. Mews and O. Glemser, *J. Fluorine Chem.*, 1976, **7**, 55.

43. S. D. Morse and J. M. Shreeve, *Inorg. Chem.*, 1978, **17**, 2169.
44. A. B. Burg, *Inorg. Chem.*, 1978, **17**, 2322.
45. E. K. S. Liu and R. J. Lagow, *J. Organometal. Chem.*, 1978, **145**, 167.
46. W. Gombler, *Angew. Chem.*, 1977, **89**, 740.
47. T. Yasumura and R. J. Lagow, *Inorg. Chem.*, 1978, **17**, 3108.
48. S. D. Morse and J. M. Shreeve, *Inorg. Chem.*, 1977, **16**, 33.
49. R. N. Haszeldine, C. R. Pool, A. E. Tipping, and R. O'B. Watts, *J. Chem. Soc., Perkin I*, 1976, 513.
50. K. G. Sharp and I. Schwager, *Inorg. Chem.*, 1976, **15**, 1697.
51. A. D. Berry and W. B. Fox, *J. Org. Chem.*, 1978, **43**, 365.
52. R. A. De Marco and W. B. Fox, *J. Fluorine Chem.*, 1978, **12**, 137.
53. R. E. Nettle and W. B. Fox, *J. Fluorine Chem.*, 1977, **9**, 219.
54. R. J. Abraham and P. Loftus, *J. Chem. Soc., Perkin II*, 1976, 1142.
55. U-H. Dolling, A. W. Douglas, E. J. J. Grabowski, E. F. Schoenewaldt, P. Sohar and M. Sletzing, *J. Org. Chem.*, 1978, **43**, 1634.
56. R. N. Haszeldine, I. Mir and A. E. Tipping, *J. Chem. Soc., Perkin I*, 1976, 2349.
57. N. McMurray, J. M. Tedder, L. L. T. Vertommen and J. C. Walton, *J. Chem. Soc., Perkin II*, 1976, 63.
58. J. Kollonitsch and L. Barash, *J. Amer. Chem. Soc.*, 1976, **98**, 5591.
59. B. W. Bycroft, A. Chowdhury and M. F. A. Dove, *J. Fluorine Chem.*, 1978, **12**, 45.
60. R. J. Abraham, P. Loftus and W. A. Thomas, *Tetrahedron*, 1977, **33**, 1227.
61. K. A. Koehler, E. E. Stone, R. A. Shelton, F. Jarnagin, L. S. Koehler and E. T. Fossel, *J. Magn. Resonance*, 1978, **30**, 75.
62. N. P. Aktaev, G. F. Ilin, G. A. Sokolskii and I. L. Knunyants, *Izv. Akad. Nauk S.S.S.R., Ser. Khim.*, 1977, 1112.
63. W. Lidy and W. Sundermeyer, *Chem. Ber.*, 1976, **109**, 1491.
64. R. D. Bagnall, W. Bell and K. Pearson, *J. Fluorine Chem.*, 1978, **11**, 93.
65. A. Haas, D. Kortmann and W. Voemel, *J. Fluorine Chem.*, 1978, **11**, 337.
66. C. A. Burton and J. M. Shreeve, *J. Amer. Chem. Soc.*, 1976, **98**, 6545.
67. C. G. Krespan, *J. Fluorine Chem.*, 1976, **8**, 105.
68. R. E. Banks, W. D. Davies, R. N. Haszeldine and D. R. Taylor, *J. Fluorine Chem.*, 1977, **10**, 487.
69. R. N. Haszeldine, B. Hewitson and A. E. Tipping, *J. Chem. Soc., Perkin I*, 1976, 1178.
70. G. Didernich, A. Haas and M. Yazdanbakhsh, *Chem. Ber.*, 1977, **110**, 916.
71. A. V. Fokin, G. I. Drozd and M. A. Landau, *J. Struct. Chem. (U.S.S.R.)*, 1976, **17**, 336.
72. D. Dakternieks, G-V. Roeschenthaler and R. Schmutzler, *Z. Naturforsch.*, 1978, **33b**, 507.
73. G-V. Roeschenthaler, *Z. Naturforsch.*, 1978, **33b**, 311.
74. C. L. Bumgardner and J. C. Wozny, *J. Fluorine Chem.*, 1978, **11**, 527.
75. O. A. Radchenko, A. Ya. Ilchenko, L. N. Markovskii and L. M. Yagupolskii, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 251.
76. M. G. Barlow and K. W. Cheung, *J. Fluorine Chem.*, 1977, **10**, 191.
77. C. J. Marsden and L. S. Bartell, *J. Chem. Soc., Dalton*, 1977, 1582.
78. T. Kitazume and J. M. Shreeve, *J. Amer. Chem. Soc.*, 1978, **100**, 492.
79. A. E. Kemppainen, M. J. Thomas and P. J. Wagner, *J. Org. Chem.*, 1976, **41**, 1294.
80. K. Baum, D. A. Lerdal and J. S. Horn, *J. Org. Chem.*, 1978, **43**, 203.
81. M. Karnel, W. Kimpenhaus and J. Buddrus, *Chem. Ber.*, 1976, **109**, 2351.
82. T. Schaefer and W. J. E. Parr, *Canad. J. Chem.*, 1977, **55**, 2835.
83. J. Burdon, T. N. Huckerby and R. Stephens, *J. Fluorine Chem.*, 1977, **10**, 523.
84. R. N. Haszeldine, D. L. Hobson and D. R. Taylor, *J. Fluorine Chem.*, 1976, **8**, 115.
85. M. G. Barlow and K. W. Cheung, *J. Fluorine Chem.*, 1977, **10**, 111.
86. S. Trofimenko, R. W. Johnson and J. K. Doty, *J. Org. Chem.*, 1978, **43**, 43.

87. M. G. Barlow, R. N. Haszeldine and J. A. Pickett, *J. Chem. Soc., Perkin I*, 1978, 378.
88. K. J. Klabunde, T. Groshens, M. Brezinski and W. Kennelly, *J. Amer. Chem. Soc.*, 1978, **100**, 4437.
89. G. Didernich, A. Haas and M. Yazdanbakhsh, *Chem. Ber.*, 1977, **110**, 916.
90. W. Clegg and S. Morton, *J. Chem. Soc., Dalton*, 1978, 1452.
91. T. R. Fernandes, R. N. Haszeldine and A. E. Tipping, *J. Chem. Soc., Dalton*, 1978, 1024.
92. E. K. Liu and R. J. Lagow, *J. Chem. Soc., Chem. Comm.*, 1977, 450.
93. H. Blancou, P. Moreau and A. Commeyras, *Tetrahedron*, 1977, **33**, 2061.
94. P. Calas, P. Moreau and A. Commeyras, *J. Fluorine Chem.*, 1978, **12**, 67.
95. G. Schoening and O. Glemser, *Chem. Ber.*, 1977, **110**, 1148.
96. J. T. Hill, *J. Fluorine Chem.*, 1977, **9**, 97.
97. A. Ya. Zapevalov, I. P. Kolenko, V. S. Plashkin and P. G. Neifeld, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 239.
98. B. Modarai, *J. Org. Chem.*, 1976, **41**, 1980.
99. T. Omori, S. Nagase, H. Baba, K. Kodaira and T. Abe, *J. Fluorine Chem.*, 1977, **9**, 279.
100. R. D. Chambers, A. Parkin and R. S. Matthews, *J. Chem. Soc., Perkin I*, 1976, 2107.
101. T. Kitazume and J. M. Shreeve, *Inorg. Chem.*, 1977, **16**, 1818.
102. K. I. The and R. G. Cavell, *Inorg. Chem.*, 1977, **16**, 2887.
103. R. G. Cavell, J. A. Gibson and K. I. The, *Inorg. Chem.*, 1978, **17**, 2880.
104. G. E. Gerhardt and R. J. Lagow, *J. Org. Chem.*, 1978, **43**, 4505.
105. S.-L. Yu and D. D. DesMarteau, *Inorg. Chem.*, 1978, **17**, 2484.
106. K. I. The and R. G. Cavell, *Inorg. Chem.*, 1976, **15**, 2518.
107. T. Kitazume and J. M. Shreeve, *Inorg. Chem.*, 1978, **17**, 2173.
108. A. B. Burg, *Inorg. Chem.*, 1978, **17**, 593.
109. A. Haas and U. Niemann, *Chem. Ber.*, 1977, **110**, 67.
110. E. Elzik, R. Dahan and A. Parlier, *Compt. Rend. (C)*, 1978, **286**, 353.
111. V. Dedek, F. Liska and L. Cvak, *Coll. Czech. Chem. Comm.*, 1978, **43**, 985.
112. J. Kollonitsch, S. Marburg and L. M. Perkins, *J. Org. Chem.*, 1976, **41**, 3107.
113. K. I. Pashkevich, V. I. Saloutin and I. Ya. Postovskii, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 45.
114. L. M. Markovskii, V. E. Pashinnik and N. A. Kirsanova, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 963.
115. J. Escudie, C. Couret and J. Satge, *Bull. Soc. Chim. France II*, 1978, 361.
116. V. Dedek, F. Liska and L. Cvak, *Coll. Czech. Chem. Comm.*, 1978, **43**, 2649.
117. C. J. Brookes, P. L. Coe, A. E. Pedler and J. C. Tatlow, *J. Chem. Soc., Perkin I*, 1978, 202.
118. T. Kitazume and J. M. Shreeve, *J. Fluorine Chem.*, 1977, **9**, 175.
119. S. Brownstein, J. Bornais and G. Latremouille, *Canad. J. Chem.*, 1978, **56**, 1419.
120. R. D. Bagnall, W. Bell and K. Pearson, *J. Fluorine Chem.*, 1977, **9**, 359.
121. R. V. Grigorash, V. V. Lyalin, L. A. Alekseva and L. M. Yagupolskii, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 782.
122. N. Pellissier, *Org. Magn. Resonance*, 1977, **9**, 563.
123. Yu. Ya. Sudenkor, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 1240.
124. G.-V. Roeschenthaler, *Z. Naturforsch.*, 1978, **33b**, 131.
125. T. Abe and S. Nagase, *J. Fluorine Chem.*, 1978, **12**, 359.
126. R. N. Haszeldine, I. Mir and A. E. Tipping, *J. Chem. Soc., Perkin I*, 1976, 556.
127. C. J. Schack and K. O. Christe, *J. Fluorine Chem.*, 1978, **12**, 325.
128. T. Martini and C. Schumann, *J. Fluorine Chem.*, 1976, **8**, 535.
129. R. N. Haszeldine, I. Mir, A. E. Tipping and A. G. Wilson, *J. Chem. Soc., Perkin I*, 1976, 1170.
130. S. D. Morse, K. A. Laurence, G. H. Sprenger and J. M. Shreeve, *J. Fluorine Chem.*, 1978, **11**, 327.
131. M. O. Riley, Y. K. Kim and O. R. Pierce, *J. Fluorine Chem.*, 1977, **10**, 85.

132. J. A. Young and M. H. Bennett, *J. Org. Chem.*, 1977, **42**, 4055.
133. M. G. Barlow and K. W. Cheung, *J. Fluorine Chem.*, 1978, **12**, 35.
134. J. Leroy and C. Wakselman, *J. Chem. Soc., Perkin I*, 1978, 1224.
135. J. Grobe and R. Rau, *J. Fluorine Chem.*, 1978, **11**, 265, 291.
136. A. Haas and U. Niemann, *J. Fluorine Chem.*, 1978, **11**, 509.
137. D. Cantacuzene, C. Wakselman and R. Dorme, *J. Chem. Soc., Perkin I*, 1977, 1365.
138. V. P. Shendrik, J. Svoboda, O. Paleta and V. Dedek, *J. Fluorine Chem.*, 1978, **11**, 485.
139. H. W. Roesky, M. Diehl, H. Fuess and J. W. Bats, *Angew. Chem.*, 1978, **90**, 73.
140. T. Kitazume and J. M. Shreeve, *J. Fluorine Chem.*, 1976, **8**, 437.
141. H.-B. Eikmeier, K. C. Hodges, O. Stelzer and R. Schmutzler, *Chem. Ber.*, 1978, **111**, 2077.
142. P. L. Coe and A. G. Holton, *J. Fluorine Chem.*, 1977, **10**, 553.
143. C. G. Krespan, *J. Org. Chem.*, 1978, **43**, 637.
144. M. Le Blanc, G. Santini, F. Jeanneaux and J. G. Riess, *J. Fluorine Chem.*, 1976, **8**, 525.
145. L. L. Gervits, K. N. Makarov, Yu. A. Cheburkov and I. L. Knunyants, *J. Fluorine Chem.*, 1977, **9**, 45.
146. P. W. L. Bosbury, R. Fields and R. N. Haszeldine, *J. Chem. Soc., Perkin I*, 1978, 422.
147. T. S. Croft, *J. Fluorine Chem.*, 1976, **7**, 433.
148. T. B. Zapevalova, A. Ya. Zapevalov, V. P. Sass, I. P. Kolenko, V. S. Plashkin and S. V. Sokolov, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 451.
149. N. Ishikawa and M. Maruta, *Nippon Kagaku Kaishi*, 1977, 1411.
150. A. Battais, B. Boutevin and P. Moreau, *J. Fluorine Chem.*, 1978, **12**, 481.
151. T. B. Zapevalova, V. S. Plashkin, A. Ya. Zapevalov and S. V. Sokolov, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 900.
152. T. Abe, K. Kodaira, H. Baba and S. Nagase, *J. Fluorine Chem.*, 1978, **12**, 1.
153. S.-L. Yu and D. D. DesMarteau, *J. Fluorine Chem.*, 1978, **12**, 315.
154. H. W. Roesky, M. Diehl and M. Banek, *Chem. Ber.*, 1978, **111**, 1503.
155. G.-V. Roeschenthaler, J. A. Gibson and R. Schmutzler, *Chem. Ber.*, 1977, **110**, 611.
156. J. A. Gibson, G.-V. Roeschenthaler, R. Schmutzler and R. Starke, *J. Chem. Soc., Dalton*, 1977, 450.
157. D. Dakternieks, G.-V. Roeschenthaler and R. Schmutzler, *J. Fluorine Chem.*, 1978, **11**, 387.
158. C. Beguin and R. Dupeyre, *Mol. Phys.*, 1976, **32**, 699.
159. T. Schaefer, J. B. Rowbotham, W. J. E. Parr, K. Marat and A. F. Janzen, *Canad. J. Chem.*, 1976, **54**, 1322.
160. J. Bromilow, R. T. C. Brownlee and A. V. Page, *Tetrahedron Letters*, 1976, 3055.
161. M. Zupan, *J. Fluorine Chem.*, 1976, **8**, 305.
162. M. J. Van Hamme, D. J. Burton and P. E. Greenlimb, *Org. Magn. Resonance*, 1978, **11**, 275.
163. T. Schaefer, K. Marat, K. Chum and A. F. Janzen, *Canad. J. Chem.*, 1977, **55**, 3936.
164. D. A. Burgess and I. D. Rae, *Austral. J. Chem.*, 1977, **30**, 1611.
165. F. E. Herkes, *J. Fluorine Chem.*, 1977, **9**, 113.
166. R. A. Newmark and J. R. Hill, *Org. Magn. Resonance*, 1977, **9**, 589.
167. T. Schaefer and J. B. Rowbotham, *Canad. J. Chem.*, 1976, **54**, 2243.
168. H. W. Roesky and K. Ambrosius, *Z. Naturforsch.*, 1978, **33b**, 759.
169. R. E. Banks, R. N. Haszeldine and J. M. Robinson, *J. Chem. Soc., Perkin I*, 1976, 1226.
170. T. Nguyen, M. Rubinstein and C. Wakselman, *J. Fluorine Chem.*, 1978, **11**, 573.
171. R. A. Bekker, Sh. O. Badanyan, G. G. Melikyan and I. L. Knunyants, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1461.
172. I. L. Knunyants, L. A. Rozov, Yu. V. Zeifman and Yu. A. Cheburkov, *J. Fluorine Chem.*, 1977, **10**, 351.
173. M. S. Raasch, *J. Org. Chem.*, 1978, **43**, 2500.

174. Yu. G. Shermolovich, O. M. Polumbrik, L. M. Markovskii, E. P. Saenko, G. G. Furin and G. G. Yakobson, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 2404.
175. R. E. Banks, M. G. Barlow and T. J. Noakes, *J. Fluorine Chem.*, 1977, **10**, 45.
176. R. N. Haszeldine, A. J. Mitchinson, R. Rowland and A. E. Tipping, *J. Chem. Soc., Perkin I*, 1976, 517.
177. R. D. Chambers, A. A. Lindley and H. C. Fielding, *J. Fluorine Chem.*, 1978, **12**, 337.
178. K. Burger, S. Tremmel and H. Schickaneder, *J. Fluorine Chem.*, 1976, **7**, 471.
179. R. E. Banks and C. Oppenheim, *J. Fluorine Chem.*, 1978, **12**, 27.
180. D. Dakternieks, G-V. Roeschenthaler and R. Schmutzler, *J. Fluorine Chem.*, 1978, **12**, 413.
181. J. Varwig and R. Mews, *Angew. Chem.*, 1977, **89**, 675.
182. G. Cal, J. M. Chemerda, D. F. Reinhold and R. M. Purick, *J. Org. Chem.*, 1977, **42**, 143.
183. E. Elkik and H. Assadifar, *Bull. Soc. Chim. France II*, 1978, 129.
184. T. Nakai, K. Tanaka and N. Ishikawa, *J. Fluorine Chem.*, 1977, **9**, 89.
185. M. Zupan and A. Pollak, *J. Org. Chem.*, 1976, **41**, 4002.
186. T. Schaefer, W. Danchura and W. Niemczura, *Canad. J. Chem.*, 1978, **56**, 36.
187. R. N. Haszeldine, D. R. Taylor and E. W. White, *J. Fluorine Chem.*, 1977, **10**, 27.
188. E. S. Turbanov, N. A. Orlova and A. A. Petrov, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 1089.
189. T. Nakai, K. Tanaka and N. Ishikawa, *Nakai Nippon Kagaku Kaishi*, 1976, 1888.
190. W. A. Konig and U. Hess, *Annalen*, 1977, 1087.
191. K. V. Werner, *J. Fluorine Chem.*, 1976, **8**, 451.
192. T. Kitazume and J. M. Shreeve, *J. Amer. Chem. Soc.*, 1978, **100**, 985.
193. J. Leroy and C. Wakselman, *Canad. J. Chem.*, 1976, **54**, 218.
194. B. E. Maryanoff and D. F. McComsey, *J. Org. Chem.*, 1978, **43**, 2733.
195. K. V. Dvornikova, V. E. Platonov and G. G. Yakobson, *J. Fluorine Chem.*, 1978, **11**, 1.
196. C. J. Boriack, E. D. Laganis and D. M. Lernal, *Tetrahedron Letters*, 1978, 1015.
197. R. D. Chambers, A. A. Lindley, P. D. Philpot, H. C. Fielding, J. Hutchinson and G. Whittaker, *J. Chem. Soc., Chem. Comm.*, 1978, 431.
198. Y. Kobayashi, S. Fujino, H. Hamana, I. Kumadaki and Y. Hanzawa, *J. Amer. Chem. Soc.*, 1977, **99**, 8511.
199. J. A. Gibson, G-V. Roeschenthaler and V. Wray, *J. Chem. Soc., Dalton*, 1977, 1491; J. A. Gibson, G-V. Roeschenthaler and R. Schmutzler, *Z. Naturforsch.*, 1977, **32b**, 599.
200. T. B. Zapevalova, V. S. Plashkin, L. N. Senyushov, V. P. Sass and S. V. Sokolov, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 909.
201. T. Kitazume and J. M. Shreeve, *J. Chem. Soc., Chem. Comm.*, 1978, 154.
202. T. Schaefer, K. Chum, K. Marat and R. E. Wasylshen, *Canad. J. Chem.*, 1976, **54**, 800.
203. H. Kimoto, H. Muramatsu and K. Inukai, *Nippon Kagaku Kaishi*, 1977, 684.
204. N. Ishikawa and S. Sasaki, *Bull. Chem. Soc. Japan*, 1977, **50**, 2164.
205. E. R. Bissell, *J. Fluorine Chem.*, 1977, **9**, 5.
206. K. Burger, J. Albanbauer, F. Kaefig and S. Penninger, *Annalen*, 1977, 624.
207. H-O. Harada and N. Ishikawa, *J. Fluorine Chem.*, 1978, **11**, 87.
208. G. Santini, M. Le Blance and J. G. Riess, *J. Organometal. Chem.*, 1977, **140**, 1.
209. V. G. Lukmanov, L. A. Alekseeva and L. M. Yagupolskii, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1979.
210. A. R. L. Bursics, E. Bursics-Szekeres, M. Murray and F. G. A. Stone, *J. Fluorine Chem.*, 1976, **7**, 619.
211. V. N. Boiko, G. M. Shchupak and L. M. Yagupolskii, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 621.
212. R. D. Chambers, J. Hutchinson and P. D. Philpot, *J. Fluorine Chem.*, 1977, **9**, 15.
213. C. Wakselman and J. Leroy, *J. Fluorine Chem.*, 1978, **12**, 101.
214. K. Burger, R. Ottlinger, A. Frank and U. Schubert, *Angew. Chem.*, 1978, **90**, 824.
215. D. W. Hensley and R. P. Stewart, *Inorg. Chem.*, 1976, **15**, 2418.

216. J. Hilton and L. H. Sutcliffe, *Spectrochim. Acta*, 1976, **32A**, 201.
217. M. M. Dhingra and K. R. Tatta, *Org. Magn. Resonance*, 1977, **9**, 23.
218. S. A. Lopyreva, T. I. Ryabtseva, V. P. Sass, A. V. Tumanova and S. V. Sokolov, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1032.
219. R. J. De Pasquale, *J. Fluorine Chem.*, 1976, **8**, 311.
220. D. D. Poulin, C. Demay and J. G. Riess, *Inorg. Chem.*, 1977, **16**, 2278.
221. F. Cavagna and C. Schumann, *J. Magn. Resonance*, 1976, **22**, 333.
222. L. A. Shimp and R. J. Lagow, *J. Org. Chem.*, 1977, **42**, 3437.
223. G. Alvernhe, E. Kozłowska, S. Lacombe-bar and A. Laurent, *Tetrahedron Letters*, 1978, 5203.
224. K. C. Joshi, V. N. Pathak and S. Bhargava, *J. Inorg. Nuclear Chem.*, 1977, **39**, 803.
225. J. Grobe and W. Mohr, *J. Fluorine Chem.*, 1976, **8**, 341.
226. H. Harada, S. Mizutaki, S. Hayashi and N. Ishikawa, *J. Fluorine Chem.*, 1978, **12**, 211.
227. R. L. Amey and J. C. Martin, *J. Amer. Chem. Soc.*, 1978, **100**, 300.
228. K. Burger, A. Meffert and S. Bauer, *J. Fluorine Chem.*, 1977, **10**, 57.
229. K. Burger, W-D. Roth and K. Neumayr, *Chem. Ber.*, 1976, **109**, 1984.
230. T. W. Bastock, A. E. Pedler and J. C. Tatlow, *J. Fluorine Chem.*, 1976, **8**, 11.
231. W. T. Feast, R. R. Hughes and W. K. R. Musgrave, *J. Fluorine Chem.*, 1977, **9**, 271.
232. C. Wakselman and T. Nguyen, *J. Org. Chem.*, 1977, **42**, 565.
233. W. J. Feast, R. R. Hughes and W. K. R. Musgrave, *J. Fluorine Chem.*, 1977, **10**, 585.
234. K. Burger, H. Schickaneder and M. Pinzel, *Annalen*, 1976, 30.
235. P. W. L. Bosbury, R. Fields, R. N. Haszeldine and D. Moran, *J. Chem. Soc., Perkin I*, 1976, 1173.
236. R. S. Matthews, *J. Magn. Resonance*, 1978, **30**, 537.
237. T. S. Croft and J. J. McBrady, *J. Org. Chem.*, 1976, **41**, 2256.
238. A. I. Ayi, R. Condom, P. C. Maria, T. N. Wade and R. Guedj, *Tetrahedron Letters*, 1978, 4507.
239. A. M. Aleksandrov, V. P. Kukhar, G. I. Danilenko and A. P. Krasnoshchek, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1505.
240. M. V. Buchanan, D. F. Hillenbrand and J. W. Taylor, *Analyt. Chem.*, 1977, **49**, 2146.
241. P. M. Hergenrother and M. Hudlicky, *J. Fluorine Chem.*, 1978, **12**, 439.
242. L. M. Yagupolskii, V. G. Lukmanov, V. N. Boiko and L. A. Alekseeva, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 2221.
243. M. Le Blanc, G. Santini, J. Gallucci and J. G. Riess, *Tetrahedron*, 1977, **33**, 1453.
244. D. L. Reger and M. D. Dukes, *J. Organometal. Chem.*, 1978, **153**, 67.
245. R. E. Moore and G. L. Driscoll, *J. Org. Chem.*, 1978, **43**, 4978.
246. A. F. Hadfield, L. Hough and A. C. Richardson, *Carbohydrate Res.*, 1978, **63**, 51.
247. N. A. Marron and J. E. Gano, *J. Amer. Chem. Soc.*, 1976, **98**, 4653.
248. K. Burger and R. Ottlinger, *J. Fluorine Chem.*, 1978, **11**, 29.
249. D. Bryce-Smith, A. Gilbert, B. H. Orger and P. J. Twitchett, *J. Chem. Soc., Perkin I*, 1978, 232.
250. G-V. Roeschenthaler, K. Sauerbrey and R. Schmutzler, *Chem. Ber.*, 1978, **111**, 3105.
251. P. Moreau, R. Albadri and A. Commeyras, *Nouveau J. Chim.*, 1977, **1**, 497.
252. K. C. Eapen and C. Tamborski, *J. Fluorine Chem.*, 1978, **12**, 271.
253. Y. Kobayashi, I. Kumadaki, A. Chsawa, Y. Sekine and A. Ando, *J. Chem. Soc., Perkin I*, 1977, 2355.
254. K. Burger, H. Schickaneder, W. Threnn, G. Ebner and C. Zettl, *Annalen*, 1976, 2156.
255. Y. Kobayashi, Y. Hanzawa and Y. Nakanishi, *Tetrahedron Letters*, 1977, 3371.
256. G. Robertson, E. K. S. Liu and R. J. Lagow, *J. Org. Chem.*, 1978, **43**, 4981.
257. P. R. Ortiz de Montellano and W. A. Vinson, *J. Org. Chem.*, 1977, **42**, 2013.
258. P. A. W. Dean and D. G. Ibbott, *Canad. J. Chem.*, 1976, **54**, 177.

259. L. A. Simonyan, E. A. Avetisyan, Z. V. Safronova and N. P. Gambaryan, *Bull Acad. Sci., U.S.S.R.*, 1977, **26**, 1906.
260. P. W. Kent and S. D. Dimitrijevic, *J. Fluorine Chem.*, 1977, **10**, 455.
261. J. Albanbauer, K. Burger, E. Burgis, D. Marquarding, L. Schabl and I. Ugi, *Annalen*, 1976, 36.
262. L. S. Kobrina, V. L. Salenko and G. G. Yakobson, *J. Org. Chem.*, 1977, **13**, 1565.
263. R. Koppang, *J. Fluorine Chem.*, 1978, **11**, 19.
264. S. Stauber and M. Zupan, *J. Fluorine Chem.*, 1977, **10**, 271.
265. M. Sharma and W. Korytnyk, *Tetrahedron Letters*, 1977, 573.
266. M. Laing, P. Sommerville, Z. Dawoodi, M. J. Mays and P. J. Wheatley, *J. Chem. Soc., Chem. Comm.*, 1978, 1035.
267. N. S. Mirzabekyants, Yu. A. Cheburkov and I. L. Knunyants, *Bull. Acad. Sci., U.S.S.R.*, 1978, **26**, 2332.
268. K. Burger, K. Einhellig, W-D. Roth and E. Daltrozso, *Chem. Ber.*, 1977, **110**, 605.
269. G. E. Ditchfield and A. E. Pedler, *J. Fluorine Chem.*, 1977, **10**, 447.
270. Y. Kobayashi, I. Kurnadaki, A. Ohsawa and A. Ando, *J. Amer. Chem. Soc.*, 1977, **99**, 7350.
271. T. I. Savchenko, T. D. Petrova, V. E. Platonov and G. G. Yakobson, *J. Fluorine Chem.*, 1977, **9**, 505.
272. V. M. Vlasov, O. V. Zakharova and G. G. Yakobson, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 2206.
273. V. M. Vlasov, O. V. Zakharova and G. G. Yakobson, *J. Fluorine Chem.*, 1977, **9**, 257.
274. R. D. Chambers and R. Middleton, *J. Chem. Soc., Perkin I*, 1977, 1500.
275. R. Anderson, K. B. Baucom, T. Psarras, C. E. Snyder and R. E. Cochy, *J. Fluorine Chem.*, 1976, **7**, 581.
276. K. Tada and F. Toda, *Tetrahedron Letters*, 1978, 563.
277. P. A. Corrigan, R. S. Dickson, G. D. Fallon, L. J. Michel and C. Mok, *Austral. J. Chem.*, 1978, **31**, 1937.
278. L. S. Kobrina, V. P. Sass, S. V. Sokolov and G. G. Yakobson, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1145.
279. H. C. Clark and K. J. Reimer, *Canad. J. Chem.*, 1976, **54**, 2077.
280. E. M. Hyde, J. D. Kennedy, B. L. Shaw and W. McFarlane, *J. Chem. Soc., Dalton*, 1977, 1571.
281. J. A. Gibson, G-V. Roeschenthaler, D. Schomburg and W. S. Sheldrick, *Chem. Ber.*, 1977, **110**, 1887.
282. B. Sket and M. Zupan, *Tetrahedron Letters*, 1978, 2607.
283. P. Wieland, *Helv. Chim. Acta*, 1976, **59**, 1027.
284. A. C. Jarvis and R. D. W. Kemmitt, *J. Organometal. Chem.*, 1977, **136**, 121.
285. D. H. R. Barton, L. J. Danks, A. K. Ganguly, R. H. Hesse, G. Tarzia and M. M. Pechet, *J. Chem. Soc., Perkin I*, 1976, 101.
286. W. Storzer and G-V. Roeschenthaler, *Z. Naturforsch.*, 1978, **33b**, 305.
287. J. Defaye, H. Driguez, B. Henrissat, J. Gelas and E. Bar-Guilloux, *Carbohydrate Res.*, 1978, **63**, 61.
288. A. R. Al-Ohaly, R. A. Head and J. F. Nixon, *J. Chem. Soc., Dalton*, 1978, 889.
289. A. P. Gulya, D. G. Batyr and E. V. Iorga, *Russ. J. Inorg. Chem.*, 1976, **21**, 567.
290. E. F. Perozzi and J. C. Martin, *J. Org. Chem.*, 1977, **42**, 3222.
291. G. Felsky, P. M. Fredericks and G. D. Meakin, *J. Chem. Soc., Perkin I*, 1978, 1529.
292. J. Burgess, J. G. Chambers, D. A. Clarke and R. D. W. Kemmitt, *J. Chem. Soc., Dalton*, 1977, 1906.
293. K. N. Makarov, L. L. Gervits and I. L. Knunyants, *J. Fluorine Chem.*, 1977, **10**, 157.
294. C. J. Jameson, A. K. Jameson and S. M. Cohen, *J. Chem. Phys.* 1977, **67**, 2771.
295. C. J. Jameson, *J. Chem. Phys.*, 1977, **67**, 2814.

296. C. J. Jameson, A. K. Jameson and H. Parker, *J. Chem. Phys.*, 1978, **69**, 1318.
297. C. J. Jameson and A. K. Jameson, *J. Chem. Phys.*, 1978, **69**, 1655.
298. R. J. Finney, M. Wolfe and J. Jonas, *J. Chem. Phys.*, 1977, **67**, 4004.
299. M. Wolfe, E. Arndt and J. Jonas, *J. Chem. Phys.*, 1977, **67**, 4012.
300. J. Angerer and W. Suchanski, *J. Magn. Resonance*, 1976, **21**, 57.
301. N. Muller, *J. Magn. Resonance*, 1976, **22**, 479.
302. N. Muller, *J. Magn. Resonance*, 1976, **23**, 327.
303. N. Muller, *J. Magn. Resonance*, 1977, **28**, 203.
304. N. N. Kalibabchuk, V. Ya. Semenii, V. S. Kuts, L. K. Dyachek and V. N. Zavatskii, *Teor. Eksp. Khim.*, 1977, **13**, 774.
305. H. L. Yeager and H. Reid, *J. Phys. Chem.*, 1976, **80**, 850.
306. N. S. Golubev and N. A. Saforov, *React. Kinet. Catal. Lett.*, 1977, **7**, 451.
307. A. A. Samoilenko, A. I. Serebryanskaya, Yu. S. Bogachev, N. N. Shapetko and A. I. Shatenshtein, *Dokl. Akad. Nauk S.S.S.R.*, 1978, **239**, 388.
308. F. F.-L. Ho, *Talanta*, 1976, **23**, 734.
309. E. G. Sundholm, *Tetrahedron*, 1977, **33**, 991.
310. J. Svoboda, *Czech. J. Phys.*, 1978, **B28**, 476.
311. M. J. Thomas, P. J. Wagner, M. L. Manion-Schilling and H. D. Roth, *J. Amer. Chem. Soc.*, 1977, **99**, 3842.
312. M. J. Thomas and P. J. Wagner, *J. Amer. Chem. Soc.*, 1977, **99**, 3845.
313. M. L. Manion-Schilling, R. S. Hutton and H. D. Roth, *J. Amer. Chem. Soc.*, 1977, **99**, 7792.
314. A. Germain and A. Commeyras, *J. Chem. Soc., Chem. Comm.*, 1978, 118.
315. J. San Filippo, R. G. Nuzzo and L. J. Romano, *J. Org. Chem.*, 1976, **41**, 392.
316. A. J. Van de Wal, E. M. Merckx, G. L. Lemiere, T. A. Van Osselaer, J. A. Lepoivre and F. C. Alderweireldt, *Bull. Soc. Chim. Belges*, 1978, **87**, 545.
317. E. M. Merckx, A. J. Van de Wal, J. A. Lepoivre and F. C. Alderweireldt, *Bull. Soc. Chim. Belges*, 1978, **87**, 21.
318. R. L. Soulen and D. W. Paul, *J. Fluorine Chem.*, 1977, **10**, 261.
319. W. R. Dolbier and T. H. Fielder, *J. Amer. Chem. Soc.*, 1978, **100**, 577.
320. B. Al-Saleh, R. E. Banks, M. G. Barlow and J. C. Hornby, *J. Fluorine Chem.*, 1978, **12**, 341.
321. H.-G. Horn, R. Koentges, F. Kolkman and H. C. Marsmann, *Z. Naturforsch.*, 1978, **33b**, 1422.
322. R. D. Chambers, R. D. Hercliffe and R. Middleton, *J. Chem. Soc., Chem. Comm.*, 1978, 305.
323. H.-G. Horn and F. Kolkman, *Z. Naturforsch.*, 1978, **33b**, 1427.
324. A. P. Sevastyan, Yu. A. Fialkov, V. A. Khranovskii and L. M. Yagupolskii, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 191.
325. R. N. Haszeldine, D. R. Taylor and E. W. White, *J. Fluorine Chem.*, 1978, **11**, 441.
326. D. A. Burgess, I. D. Rae and J. D. Snell, *Austral. J. Chem.*, 1977, **30**, 543.
327. D. A. Burgess, I. D. Rae, L. K. Smith and J. D. Snell, *Austral. J. Chem.*, 1976, **29**, 1435.
328. L. M. Yagupolskii and Yu. L. Yagupolskii, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1853.
329. G. M. Brooke and D. H. Hall, *J. Fluorine Chem.*, 1977, **10**, 495.
330. J. Fornies, M. Green, A. Laguna, M. Murray, J. L. Spencer and F. G. A. Stone, *J. Chem. Soc., Dalton*, 1977, 1515.
331. Yu. A. Fialkov, A. P. Sevastyan, V. A. Khranovskii and L. M. Yagupolskii, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 939.
332. A. E. Asato, H. Matsumoto, M. Denny and R. S. H. Liu, *J. Amer. Chem. Soc.*, 1978, **100**, 5957.
333. L. M. Yakupolskii, M. M. Kremlev, Yu. A. Failkov, V. A. Khranovskii and V. M. Yurchenko, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1438.

334. V. A. Mukhedkar, B. J. Kavathekar and A. J. Mukhedkar, *J. Inorg. Nuclear Chem.*, 1977, **39**, 1003.
335. K. B. Wiberg, D. E. Barth and W. E. Pratt, *J. Amer. Chem. Soc.*, 1977, **99**, 4287.
336. H-J. Schneider, W. Gschwendtner, D. Heiske, V. Hoppen and F. Thomas, *Tetrahedron*, 1977, **33**, 1769; H-J. Schneider private communication.
337. R. A. Bekker, V. Ya. Popkova and I. L. Knunyants, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1956.
338. G. A. Wheaton and D. J. Burton, *J. Fluorine Chem.*, 1977, **9**, 25.
339. M. L. Greenlee, E. L. Fritzen and J. S. Swenton, *J. Org. Chem.*, 1978, **43**, 4512.
340. J. Bensoam and F. Mathey, *Tetrahedron Letters*, 1977, 2797.
341. D. J. Burton and J. L. Hahnfeld, *J. Org. Chem.*, 1977, **42**, 828.
342. N. E. Akhmetova and V. D. Shteingarts, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1172.
343. S. Stavber and M. Zupan, *J. Chem. Soc., Chem. Comm.*, 1978, 969.
344. M. Christl and W. Buchner, *Org. Magn. Resonance*, 1978, **11**, 461.
345. P. Mueller, R. Etienne, J. Pfyffer, N. Pineda and M. Schipoff, *Tetrahedron Letters*, 1978, 3151.
346. A. Gregorcic and M. Zupan, *Tetrahedron*, 1977, **33**, 3243.
347. A. Gregorcic and M. Zupan, *Coll. Czech. Chem. Comm.*, 1977, **42**, 3192.
348. T. Ishihara, T. Ando, T. Muranaka and K. Saito, *J. Org. Chem.*, 1977, **42**, 666.
349. H. J. J. Loozen, W. M. M. Robben and H. M. Buck, *Rec. Trav. Chim.*, 1976, **95**, 248.
350. P. Mueller, J. Pfyffer, E. Wentrup-Byrne and U. Burger, *Helv. Chim. Acta*, 1978, **61**, 2081.
351. A. A. Shtark and V. D. Shteingarts, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1536.
352. A. Gregorcic and M. Zupan, *J. Chem. Soc., Perkin I*, 1977, 1446.
353. G. Bauer and G. Haegeler, *Angew. Chem.*, 1977, **89**, 493.
354. M. Zupan, B. Sket and B. Pahor, *Tetrahedron Letters*, 1977, 4541.
355. R. G. Plevey and R. E. Talbot, *J. Fluorine Chem.*, 1977, **10**, 577.
356. R. G. Plevey and D. J. Sparrow, *J. Chem. Soc., Perkin I*, 1976, 573.
357. T. V. Senchenko, V. E. Platonov, and G. G. Yakobson, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1765.
358. J. Battersby, R. Stephens, and J. C. Tatlow, *J. Fluorine Chem.*, 1977, **9**, 127.
359. E. Elkik, A. Parlier, and R. Dahan, *Compt. Rend. (C)*, 1977, **284**, 141.
360. W. Adcock, B. D. Gupta and T-C. Khor, *Austral. J. Chem.*, 1976, **29**, 2571.
361. G. M. Brooke, D. H. Hall and H. M. M. Shearer, *J. Chem. Soc., Perkin I*, 1978, 780.
362. B. Sket and M. Zupan, *J. Chem. Soc., Perkin I*, 1977, 2169.
363. S. Stavber and M. Zupan, *J. Fluorine Chem.*, 1978, **12**, 307.
364. R. R. Perkins and R. E. Pincock, *Org. Magn. Resonance*, 1976, **8**, 165.
365. P. Metzger and E. Casadevall, *Bull. Soc. Chim. France II*, 1978, 109.
366. J. P. Zahra, B. Waegell, J. Reisse, G. Pouzard and J. Fournier, *Bull. Soc. Chim. France*, 1976, 1896.
367. H-J. Schneider and W. Ansorge, *Tetrahedron*, 1977, **33**, 265.
368. R. E. Banks, M. G. Barlow and M. M. Saleh, *J. Fluorine Chem.*, 1977, **10**, 81.
369. D. J. Burton, R. D. Howells, P. D. Van der Valk, *J. Amer. Chem. Soc.*, 1977, **99**, 4830.
370. R. G. Plevey, I. J. Sallomi, D. F. Thomas and J. C. Tatlow, *J. Chem. Soc., Perkin I*, 1976, 2270.
371. M. Zupan and B. Sket, *J. Org. Chem.*, 1978, **43**, 696.
372. M. Zupan, *J. Fluorine Chem.*, 1977, **9**, 177.
373. G. V. Thi and P. Margaretha, *Helv. Chim. Acta*, 1976, **59**, 2236.
374. V. T. Yue, C. J. Courson, M. R. Brinkman and P. P. Gaspar, *J. Org. Chem.*, 1978, **43**, 4873.
375. M. Zupan, B. Sket and B. Pahor, *J. Org. Chem.*, 1978, **43**, 2297.
376. L. S. Kobrina, V. N. Kovtonyuk and G. G. Yakobson, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1331.

377. W. J. Feast, R. R. Hughes and W. K. R. Musgrave, *J. Chem. Soc., Perkin I*, 1977, 152.
378. I. J. Hotchkiss, R. Stephens and J. C. Tatlow, *J. Fluorine Chem.*, 1976, **8**, 379.
379. R. D. Chambers, G. Taylor and R. L. Powell, *J. Chem. Soc., Chem. Comm.*, 1978, 433.
380. H. Kimoto, H. Muramatsu and K. Inukai, *J. Fluorine Chem.*, 1977, **9**, 417.
381. B. Sket and M. Zupan, *Tetrahedron Letters*, 1977, 2811.
382. W. Adcock and T-C. Khor, *J. Org. Chem.*, 1977, **42**, 218.
383. K. Hayano, Y. Ohfuné, H. Shirahama and T. Matsumoto, *Tetrahedron Letters*, 1978, 1991.
384. W. R. Cullen, M. Williams, F. W. B. Einstein and C-H. Huang, *J. Fluorine Chem.*, 1978, **11**, 365.
385. M. Sheves, B. Sialom and Y. Mazur, *J. Chem. Soc., Chem. Comm.*, 1978, 554.
386. Yu. V. Pozdnyakov and V. D. Shteingarts, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1772.
387. H. L. Holland, *Tetrahedron Letters*, 1978, 881.
388. R. E. Banks and B. E. Cross, *J. Chem. Soc., Perkin I*, 1977, 512.
389. B. E. Cross and A. Erasmuson, *J. Chem. Soc., Chem. Comm.*, 1978, 1013.
390. D. H. R. Barton, L. J. Danks, R. H. Hesse, M. M. Pechet and C. Wilshire, *Nouveau J. Chim.*, 1977, **1**, 315.
391. L. Toscano, G. Guadagnini and G. Grisanti, *Arzneim.-Forsch.*, 1977, **27**, 1636.
392. G. Cleve, E. Frost, G-A. Hoyer, D. Rosenberg and A. Seeger, *Arzneim.-Forsch.*, 1977, **27**, 2188.
393. D. H. R. Barton, R. H. Hesse, R. E. Markwell, M. M. Pechet and S. Rozen, *J. Amer. Chem. Soc.*, 1976, **98**, 3036.
394. R. D. Bates, B. E. Wagner and E. H. Poindexter, *J. Phys. Chem.*, 1976, **80**, 320.
395. J. Burdon, J. C. Hotchkiss and W. B. Jennings, *Tetrahedron*, 1977, **33**, 1745.
396. R. Bolton and J. P. B. Sandall, *J. Chem. Soc., Perkin II*, 1978, 137.
397. R. Bolton, S. M. Kazeroonian and J. P. B. Sandall, *J. Fluorine Chem.*, 1976, **8**, 471.
398. R. Bolton, S. M. Kazeroonian and J. P. B. Sandall, *J. Fluorine Chem.*, 1978, **11**, 9.
399. J. M. Angelelli, M. A. Delmas, J. C. Maire and J. P. Zahra, *J. Organometal. Chem.*, 1977, **128**, 313.
400. J. M. Angelelli, M. A. Delmas and J. C. Maire, *J. Organometal. Chem.*, 1978, **154**, 79.
401. F. J. Weigert and W. A. Sheppard, *J. Org. Chem.*, 1976, **41**, 4006.
402. V. Wray, L. Ernst and E. Lustig, *J. Magn. Resonance*, 1977, **27**, 1.
403. V. A. Chertkov and N. M. Sergeev, *J. Magn. Resonance*, 1976, **21**, 159.
404. P. E. Hansen, A. Berg, H. J. Jakobson, A. P. Manzara and J. Michl, *Org. Magn. Resonance*, 1977, **10**, 179.
405. W. Adcock, J. Alste, S. Q. A. Rizvi and M. Aurangzeb, *J. Amer. Chem. Soc.*, 1976, **98**, 1701.
406. E. Lustig, D. N. Lincoln and V. Wray, *J. Magn. Resonance*, 1976, **21**, 509.
407. R. Bolton and J. P. B. Sandall, *J. Chem. Soc., Perkin II*, 1978, 141.
408. S. L. Manatt and M. A. Cooper, *J. Amer. Chem. Soc.*, 1977, **99**, 4561.
409. L. Ernst, D. N. Lincoln and V. Wray, *J. Magn. Resonance*, 1976, **21**, 115.
410. R. Bolton and J. P. B. Sandall, *J. Chem. Soc., Perkin II*, 1978, 1288.
411. L. Ernst and V. Wray, *J. Magn. Resonance*, 1977, **28**, 373.
412. V. Wray and D. N. Lincoln, *J. Magn. Resonance*, 1975, **18**, 374.
413. W. J. E. Parr and T. Schaefer, *Canad. J. Chem.*, 1976, **54**, 3564.
414. A. F. Clifford, J. L. Howell and D. L. Wooton, *J. Fluorine Chem.*, 1978, **11**, 433.
415. P. Sartori and G. Bauer, *J. Fluorine Chem.*, 1978, **12**, 203.
416. O. I. Andreevskaya, G. G. Furin and G. G. Yakobson, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1558.
417. V. Wray, *J. Chem. Soc., Perkin II*, 1978, 855.
418. G. J. Den Otter and C. Maclean, *J. Mol. Struct.*, 1976, **31**, 47.
419. E. R. Falardeau and D. D. DesMarteau, *J. Chem. Eng. Data*, 1976, **21**, 386.

420. T. Schaefer, W. Danchura, W. Niemczura and J. Peeling, *Canad. J. Chem.*, 1978, **56**, 2442.
421. T. Schaefer, W. Danchura and W. Niemczura, *Canad. J. Chem.*, 1978, **56**, 2233.
422. A. Heesing and W. Schmaladt, *Chem. Ber.*, 1978, **111**, 320.
423. T. Schaefer and W. J. E. Parr, *J. Chem. Phys.*, 1976, **65**, 1197.
424. G. M. Brooke and D. H. Hall, *J. Chem. Soc., Perkin I*, 1976, 1464.
425. W. J. Frazee, M. E. Peach and J. R. Sweet, *J. Fluorine Chem.*, 1977, **9**, 377.
426. A. G. Budnik, S. L. Rusakov, E. I. Khomenko and G. G. Yakobson, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 1185.
427. B. C. Musial and M. E. Peach, *J. Fluorine Chem.*, 1976, **7**, 459.
428. T. N. Gerasimova, N. V. Semikolenova and E. P. Fokin, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 91.
429. E. R. Falardeau and D. D. DesMarteau, *J. Fluorine Chem.*, 1976, **7**, 409.
430. Yu. V. Pozdnyakovich and V. D. Shteingarts, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 559.
431. D. Doddrell, M. Barfield, W. Adcock, M. Aurangzeb and D. Jordan, *J. Chem. Soc., Perkin II*, 1976, 402.
432. K. L. Kirk, *J. Org. Chem.*, 1976, **41**, 2373.
433. R. Bolton and J. P. B. Sandall, *J. Chem. Soc., Perkin II*, 1978, 746.
434. B. Sket and M. Zupan, *J. Org. Chem.*, 1978, **43**, 835.
435. A. V. Varlamov, Z. Papouskova, J. Pola, P. Trska and V. Chvalovsky, *Coll. Czech. Chem. Comm.*, 1977, **42**, 489.
436. J. M. Angelelli, M. A. Delmas and J. C. Maire, *J. Organometal. Chem.*, 1977, **131**, 225.
437. P. Sartori, K. Ahlers and H-J. Frohn, *J. Fluorine Chem.*, 1976, **8**, 457.
438. J. C. Maire and Y. Limouzin, *J. Organometal. Chem.*, 1976, **122**, 21.
439. W. Girke and E. D. Bergmann, *Chem. Ber.*, 1976, **109**, 1038.
440. L. Ernst, *Z. Naturforsch.*, 1975, **30b**, 788.
441. W. Adcock, G. L. Aldous and W. Kitching, *Tetrahedron Letters*, 1978, 3387.
442. W. Adcock, D. P. Cox and W. Kitching, *J. Organometal. Chem.*, 1977, **133**, 393.
443. M. Rabinovitz, I. Agranat, H. Selig and C-H. Lin, *J. Fluorine Chem.*, 1977, **10**, 159.
444. V. M. Vlasov, O. V. Zakharova and G. G. Yakobson, *Izvest. Sibirsk. Otdel. Akad. Nauk, Ser. Khim. Nauk*, 1977, 127.
445. R. E. Banks, M. G. Barlow, R. N. Haszeldine and J. C. Massey, *J. Fluorine Chem.*, 1978, **12**, 53.
446. G. M. Brooke, R. S. Matthews and A. C. Young, *J. Chem. Soc., Perkin I*, 1977, 1411.
447. R. E. Banks and S. M. Hitchen, *J. Fluorine Chem.*, 1978, **12**, 159.
448. R. Bolton, J. P. B. Sandall and G. H. Williams, *J. Fluorine Chem.*, 1978, **11**, 591.
449. T. Hirohashi, S. Inabe and H. Yamamoto, *Bull. Chem. Soc. Japan*, 1976, **49**, 1395.
450. A. Agarwal, M. J. McGlinchey and T-S. Tan, *J. Organometal. Chem.*, 1977, **141**, 85.
451. E. P. Fokin, T. N. Gerasimova, T. V. Fomenko and N. V. Semikolenova, *J. Org. Chem. (U.S.S.R.)*, 1978, **14**, 772.
452. R. Koppang, *J. Fluorine Chem.*, 1977, **9**, 449.
453. G. B. Deacon, R. M. Slade and D. G. Vince, *J. Fluorine Chem.*, 1978, **11**, 57.
454. J. M. Gascoyne, P. J. Mitchell and L. Phillips, *J. Chem. Soc., Perkin II*, 1977, 1051.
455. A. Prakash, *J. Fluorine Chem.*, 1977, **9**, 81.
456. G. A. Olah, G. K. S. Prakash and G. Laing, *J. Amer. Chem. Soc.*, 1977, **99**, 5683.
457. W. Adcock and D. P. Cox, *Tetrahedron Letters*, 1976, 2719.
458. W. Adcock and T-C. Khor, *Tetrahedron Letters*, 1977, 3769.
459. W. Adcock and T-C. Khor, *J. Amer. Chem. Soc.*, 1978, **100**, 7799.
460. W. Adcock and T-C. Khor, *Tetrahedron Letters*, 1976, 3063.
461. L. M. Stock and M. R. Wasielewski, *J. Org. Chem.*, 1976, **41**, 1660.
462. G. M. Brooke and A. C. Young, *J. Fluorine Chem.*, 1976, **8**, 223.

463. H. Selig, M. Rabinovitz, I. Agranat, C.-H. Lin and L. Ebert, *J. Amer. Chem. Soc.*, 1976, **98**, 1601.
464. W. Adcock and T.-C. Khor, *J. Org. Chem.*, 1978, **43**, 1272.
465. M. Protiva, I. Cervena, M. Rajsner, J. Metysova and M. Hrubantova, *Coll. Czech. Chem. Comm.*, 1978, **43**, 2656.
466. T. Takemura and N. Mori, *Chem. Letters*, 1978, 857.
467. S. I. Pombrik, D. N. Kravtsov, B. A. Kvasov and E. I. Fedin, *J. Organometal. Chem.*, 1977, **136**, 185.
468. S. I. Pombrik, V. F. Ivanov, A. S. Peregudov, D. N. Kravtsov, A. A. Fedorov and E. I. Fedin, *J. Organometal. Chem.*, 1978, **153**, 319.
469. G. W. Parshall, private communication reported in D. R. Coulson, *J. Amer. Chem. Soc.*, 1976, **98**, 3111.
470. B. F. Malichenko, G. A. Chernjanskaja, A. E. Borodin and N. N. Kalibabchuk, *Dopovidi Akad. Nauk Ukrain. R.S.R.*, Ser. B, 1977, 809.
471. R. Koppang, *J. Fluorine Chem.*, 1976, **8**, 389.
472. C. Crocker, R. J. Goodfellow, J. Gimeno and R. Uson, *J. Chem. Soc., Dalton*, 1977, 1448.
473. G. B. Deacon and G. J. Farquharson, *J. Organometal. Chem.*, 1977, **135**, 73.
474. V. M. Vlasov, G. G. Yakobson, E. S. Petrov and A. I. Shatenshtein, *J. Fluorine Chem.*, 1977, **9**, 321.
475. A. W. Douglas, *Canad. J. Chem.*, 1978, **56**, 2129.
476. S. I. Pombrik, D. N. Kravtsov, A. S. Peregudov, E. I. Fedin and A. N. Nesmeyanov, *J. Organometal. Chem.*, 1977, **131**, 355.
477. I. Schuster, *Tetrahedron*, 1977, **33**, 1075.
478. G. B. Deacon, W. D. Raverty and D. G. Vince, *J. Organometal. Chem.*, 1977, **135**, 103.
479. R. Kosfeld and L. Henke, *Org. Magn. Resonance*, 1976, **8**, 141.
480. Y. Jayathirtha and V. Krishnam, *Z. Naturforsch.*, 1978, **33a**, 243.
481. J. W. Emsley, L. Phillips and V. Wray, *Progr. N.M.R. Spectroscopy*, 1977, **10**, 85.
482. V. Wray and D. N. Lincoln, *J. Chem. Soc., Perkin II*, 1976, 1307.
483. V. Wray and D. N. Lincoln, *Org. Mag. Resonance*, 1977, **9**, 155.
484. N. Muller, *J. Magn. Resonance*, 1977, **25**, 111.
485. R. Koppang, C. G. Russell and D. G. Lee, *Acta Chem. Scand.*, 1977, **B31**, 491.
486. R. E. Banks, R. N. Haszeldine and J. C. Massey, *J. Fluorine Chem.*, 1978, **12**, 331.
487. M. J. Robins, M. MacCoss, S. R. Naik and G. Ramani, *J. Amer. Chem. Soc.*, 1976, **98**, 7381.
488. Y. Takeuchi, K. L. Kirk and L. A. Cohen, *J. Org. Chem.*, 1978, **43**, 3570.
489. I. J. Hotchkiss, R. Stephens and J. C. Tatlow, *J. Fluorine Chem.*, 1977, **10**, 541.
490. R. E. Banks, R. N. Haszeldine and E. Phillips, *J. Fluorine Chem.*, 1977, **9**, 243.
491. P. Bouchet, A. Fruchier, G. Joncheray and J. Elguero, *Org. Magn. Resonance*, 1977, **9**, 716.
492. R. E. Banks, R. I. Higgons, A. Prakash, M. Rawstron and G. R. Sparkes, *J. Fluorine Chem.*, 1977, **9**, 327.
493. M. G. Barlow, D. E. Brown and R. N. Haszeldine, *J. Chem. Soc., Chem. Comm.*, 1977, 669.
494. M. G. Barlow, D. W. Brown and R. N. Haszeldine, *J. Chem. Soc., Perkin I*, 1978, 363.
495. D. H. R. Barton, R. H. Hesse, G. P. Jackman and M. M. Pechet, *J. Chem. Soc., Perkin I*, 1977, 2604.
496. R. S. Matthews, *J. Magn. Resonance*, 1978, **29**, 65.
497. R. S. Matthews, *Org. Magn. Resonance*, 1977, **9**, 318.
498. R. D. Chambers, R. D. Hercliffie and W. K. R. Musgrave, *J. Chem. Soc., Chem. Comm.*, 1978, 304.
499. R. S. Matthews, *Org. Magn. Resonance*, 1976, **8**, 240.
500. R. S. Matthews, *Org. Magn. Resonance*, 1976, **8**, 628.

501. J. T. Gerig and R. S. McLeod, *J. Amer. Chem. Soc.*, 1976, **98**, 3970.
502. R. J. Spear, D. A. Forsyth and G. A. Olah, *J. Amer. Chem. Soc.*, 1976, **98**, 2493.
503. A. A. Shtark, Yu. V. Pozdnyakovich and V. D. Shteingarts, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1544.
504. K. Bock and C. Pedersen, *Acta Chem. Scand.*, 1976, **B30**, 727.
505. P. N. Dobronravov and V. D. Shteingarts, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1553.
506. C. W. Gillies, *J. Amer. Chem. Soc.*, 1977, **99**, 7239.
507. T. Wehrung, H. Oberhammer, A. Haas, B. Koch and N. Welcman, *J. Mol. Struct.*, 1976, **35**, 253.
508. T. B. Zapevalova, V. S. Plashkin, B. N. Selishchev, K. N. Bildinov and M. S. Shcherbakova, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 2389.
509. J. Dolezalova, M. Cerny, T. Trnka and J. Pacak, *Coll. Czech. Chem. Comm.*, 1976, **41**, 1944.
510. V. Wray, *J. Chem. Soc., Perkin II*, 1976, 1598.
511. M. J. Hamor, T. A. Hamor, C. M. Jankins, R. Stephens and J. C. Tatlow, *J. Fluorine Chem.*, 1977, **10**, 605.
512. A. J. Brink, O. G. de Villiers and A. Jordan, *Carbohydrate Res.*, 1977, **54**, 285.
513. G. R. Owen, J. P. H. Verheyden and J. G. Moffat, *J. Org. Chem.*, 1976, **41**, 3010.
514. J. C. A. Boeyens, A. J. Brink and J. Jordan, *S. African J. Chem.*, 1978, **31**, 7.
515. I. D. Jenkins, J. P. H. Verheyden and J. G. Moffat, *J. Amer. Chem. Soc.*, 1976, **98**, 3346.
516. D. S. Genghof, C. F. Brewer and E. J. Hehre, *Carbohydrate Res.*, 1978, **61**, 291.
517. R. A. Kolinski, H. Plenkiewicz, H. Voellnagel-Neugebauer and R. Wozniacki, *Roczniki Chem.*, 1977, **51**, 1365.
518. R. W. Rendell and B. Wright, *Tetrahedron*, 1978, **34**, 197.
519. R. J. Gillespie and P. H. Spekkens, *J. Chem. Soc., Dalton*, 1977, 1539.
520. K. T. Gillen, D. C. Douglass and J. E. Griffiths, *J. Chem. Phys.*, 1978, **69**, 461.
521. H. W. Roesky and M. Aramaki, *Angew. Chem.*, 1978, **90**, 127.
522. R. K. Marat and A. F. Janzen, *Canad. J. Chem.*, 1977, **55**, 3845.
523. D. J. Arnold, E. R. Cromie and D. W. H. Rankin, *J. Chem. Soc., Dalton*, 1977, 1999.
524. W. W. Wilson, C. Naulin and R. Bougon, *Inorg. Chem.*, 1977, **16**, 2252.
525. H. Wagner, D.-L. Wagner and O. Glemser, *Chem. Ber.*, 1977, **110**, 683.
526. S. Brownstein, *Canad. J. Chem.*, 1978, **56**, 343.
527. N. Krueger, K. Dehnicke and A.-F. Shihada, *Z. Anorg. Chem.*, 1978, **438**, 169.
528. L. M. Emme and G. L. Gard, *J. Fluorine Chem.*, 1978, **12**, 77.
529. M. J. Atherton and J. H. Holloway, *Inorg. Nuclear Chem. Letters*, 1978, **14**, 121.
530. K. O. Christie, C. J. Schack and E. C. Curtis, *Inorg. Chem.*, 1976, **15**, 843.
531. T. Cyr and S. Brownstein, *J. Inorg. Nuclear Chem.*, 1977, **39**, 2143.
532. E. Lehmann, D. Naumann and M. Schmeisser, *J. Fluorine Chem.*, 1976, **7**, 135.
533. E. Lehmann, D. Naumann and M. Schmeisser, *J. Fluorine Chem.*, 1976, **7**, 33.
534. H. Selig, W. A. Sunder, F. C. Schilling and W. E. Falconer, *J. Fluorine Chem.*, 1978, **11**, 629.
535. K. Seppelt and N. Bartlett, *Z. Anorg. Chem.*, 1977, **436**, 122.
536. K. Seppelt, *Chem. Ber.*, 1976, **109**, 1046.
537. J.-Y. Calves and R. J. Gillespie, *J. Amer. Chem. Soc.*, 1977, **99**, 1788; *J. Chem. Soc., Chem. Soc., Chem. Comm.*, 1976, 506.
538. A. R. Grimmer, L. Friedrich and H. Meinert, *Z. Chem.*, 1977, **17**, 150.
539. K. Seppelt, *Chem. Ber.*, 1977, **110**, 1470.
540. R. J. Gillespie and J. P. Krasznai, *Inorg. Chem.*, 1976, **15**, 1251.
541. K. G. Sharp, *J. Chem. Soc. Chem. Comm.*, 1977, 564.
542. M. Brownstein, R. J. Gillespie and J. P. Krasznai, *Canad. J. Chem.*, 1978, **56**, 2253.
543. Yu. A. Buslayev, E. G. Ilyin, M. E. Ignatov, L. S. Buturina and T. A. Mastryukova, *J. Fluorine Chem.*, 1978, **12**, 381.

544. E. G. Ilin, M. E. Ignatov, L. S. Butorina, T. A. Mastryukova and Yu. A. Buslaev, *Dokl. Akad. Nauk S.S.S.R.*, 1978, 537.
545. M. A. Bennett, R. N. Johnson and T. W. Turney, *Inorg. Chem.*, 1976, **15**, 2938.
546. D. Lentz, H. Pritzkow and K. Seppelt, *Angew. Chem.*, 1977, **89**, 741.
547. D. Lentz, H. Pritzkow and K. Seppelt, *Inorg. Chem.*, 1978, **17**, 1926.
548. D. Lentz and K. Seppelt, *Angew. Chem.*, 1978, **90**, 391.
549. S. Craddock, E. A. V. Ebsworth, M. L. McConnell, D. W. H. Rankin and M. R. Todd, *J. Chem. Soc., Dalton*, 1977, 1925.
550. D-L. Wagner, H. Wagner and O. Glemser, *Z. Naturforsch.*, 1978, **33b**, 300.
551. G. Kleemann and K. Seppelt, *Angew. Chem.*, 1978, **90**, 547.
552. R. Mews, *Angew. Chem.*, 1978, **90**, 561.
553. R. T. Paine, *J. Amer. Chem. Soc.*, 1977, **99**, 3884.
554. K. G. Sharp and P. A. Sutor, *J. Amer. Chem. Soc.*, 1977, **99**, 8046.
555. R. W. Braun, A. H. Cowley, M. C. Cushner and R. J. Lagow, *Inorg. Chem.*, 1978, **17**, 1679.
556. W. Althoff, M. Fild, H-P. Rieck and R. Schmutzler, *Chem. Ber.*, 1978, **111**, 1845.
557. T. F. Moore, A. R. Garber and J. D. Odom, *Inorg. Nuclear Chem. Letters*, 1978, **14**, 45.
558. C. G. Moreland, G. O. Doak, L. B. Littlefield, N. S. Walker, J. W. Gilje, R. W. Braun and A. H. Cowley, *J. Amer. Chem. Soc.*, 1976, **98**, 2161.
559. G. W. Fraser and G. D. Meikle, *J. Chem. Soc., Dalton*, 1977, 1985.
560. K. G. Sharp and K. L. Servis, *J. Magn. Resonance*, 1977, **28**, 295.
561. M. J. Hopkinson, N. S. Walker and D. D. DesMarteau, *J. Org. Chem.*, 1976, **41**, 1407.
562. L. N. Markovskii, V. E. Pashinnik and E. P. Saenko, *J. Org. Chem. (U.S.S.R.)*, 1977, **13**, 1025.
563. Y. Margalit, G. Amitai and Y. Ashani, *Phosphorus and Sulfur*, 1977, **3**, 315.
564. C. M. P. Favez and A. E. Merbach, *J. Fluorine Chem.*, 1976, **8**, 409.
565. W. Wolfsberger, *J. Fluorine Chem.*, 1978, **11**, 159.
566. J. Dalton, W. McFarlane, C. R. Wickens, G. W. Fraser and J. B. Millar, *Org. Magn. Resonance*, 1976, **8**, 522.
567. R. B. King and J. Gimeno, *Inorg. Chem.*, 1978, **17**, 2390.
568. R. B. King and J. Gimeno, *J. Chem. Soc., Chem. Comm.*, 1977, 142.
569. J-P. Costes and G. Cros, *Compt. Rend. (C)*, 1978, **286**, 39.
570. S. Brownstein and G. Latremouille, *Canad. J. Chem.*, 1978, **56**, 2764.
571. U. Klingebiel and A. Meller, *Chem. Ber.*, 1976, **109**, 2430.
572. A. C. Baxter, J. H. Cameron, A. McAuley, F. M. McLaren and J. M. Winfield, *J. Fluorine Chem.*, 1977, **10**, 289.
573. F. L. Wimmer and M. R. Snow, *Austral. J. Chem.*, 1978, **31**, 267.
574. W. S. Sheldrick and M. J. C. Hewson, *Z. Naturforsch.*, 1978, **33b**, 834.
575. I. Ruppert and V. Bastian, *Angew. Chem.*, 1978, **90**, 226.
576. F. Seel, R. Budenz, R. D. Flaccus and R. Staab, *J. Fluorine Chem.*, 1978, **12**, 437.
577. U. Klingebiel, *Z. Anorg. Chem.*, 1978, **442**, 130.
578. U. Klingebiel, H. Hluchy and A. Meller, *Chem. Ber.*, 1978, **111**, 906.
579. U. Klingebiel, D. Enterling, L. Skoda and A. Meller, *J. Organometal. Chem.*, 1977, **135**, 167.
580. M. Sanchez and A. H. Cowley, *J. Chem. Soc., Chem. Comm.*, 1976, 690.
581. H. Füssstetter, H. Noth and W. Winterstein, *Chem. Ber.*, 1977, **110**, 1931.
582. H. Schrem and J. Weidlein, *J. Organometal Chem.*, 1977, **133**, C13.
583. J. G. Dupont and C. W. Allen, *Inorg. Chem.*, 1977, **16**, 2964.
584. M. A. Sens, J. D. Odom and M. H. Goodrow, *Inorg. Chem.*, 1976, **15**, 2825.
585. G. I. Zharkova and S. V. Zemskov, *Izvest. Sibirsk. Otdel. Akad. Nauk, Ser. Khim. Nauk*, 1977, 105.
586. R. B. King and J. Gimeno, *Inorg. Chem.*, 1978, **17**, 2396.

587. I. Ruppert and V. Bastian, *Angew. Chem.*, 1977, **89**, 763.
588. R. G. Montemayor, D. T. Sauer, S. Fleming, D. W. Bennett, M. G. Thomas and R. W. Parry, *J. Amer. Chem. Soc.*, 1978, **100**, 2231.
589. M. G. Voronkov, G. V. Dmitrieva, G. M. Gavrilova, V. V. Keiko, E. O. Tsetlina and B. A. Trofimov, *Bull. Acad. Sci. U.S.S.R.*, 1978, **26**, 2361.
590. K. Tamao, S. Okazaki and M. Kumada, *J. Organometal. Chem.*, 1978, **146**, 87.
591. S. C. Peake, M. J. C. Hewson, O. Schlak, R. Schmutzler, R. K. Harris and M. I. M. Wazeer, *Phosphorus and Sulfur*, 1978, **4**, 67.
592. U. Klingebiel, *Z. Naturforsch.*, 1977, **32b**, 1212.
593. H. Eshtiagh-Hosseini and J. F. Nixon, *J. Organometal. Chem.*, 1978, **150**, 129.
594. R. Bartsch, M. Hausard and O. Stelzer, *Chem. Ber.*, 1978, **111**, 1420.
595. D. L. Gallup and J. G. Morse, *Inorg. Chem.*, 1978, **17**, 1340.
596. E. O. Fisher, S. Walz, A. Ruhs and F. R. Kreissl, *Chem. Ber.*, 1978, **111**, 2765.
597. L. B. Littlefield and G. O. Doak, *Phosphorus and Sulfur*, 1977, **3**, 35.
598. P. Panster and W. Malisch, *Chem. Ber.*, 1976, **109**, 2112.
599. L. B. Littlefield and G. O. Doak, *J. Amer. Chem. Soc.*, 1976, **98**, 7881.
600. N. J. De'Ath, D. B. Denney, D. Z. Denney and Y. F. Hsu, *J. Amer. Chem. Soc.*, 1976, **98**, 768.
601. R. A. Head and J. F. Nixon, *J. Chem. Soc., Dalton*, 1978, 895.
602. P. Paetzold, W.-G. Druckenbrodt and A. Richter, *Chem. Ber.*, 1978, **111**, 189.
603. U. Klingebiel, J. Neemann and A. Meller, *J. Fluorine Chem.*, 1978, **12**, 427.
604. U. Klingebiel, G. Wendenberg and A. Meller, *Z. Naturforsch.*, 1977, **32b**, 1482.
605. M. E. Ignatov, E. G. Ilin, A. A. Shvets and Yu A. Buslaev, *Dokl. Akad. Nauk S.S.S.R.*, 1978, **240**, 865.
606. S. Neumann, D. Shomburg, G. Richtarsky and R. Schmutzler, *J. Chem. Soc., Chem. Comm.*, 1978, 946.
607. C. Crocker and R. J. Goodfellow, *J. Chem. Soc., Dalton*, 1977, 1687.
608. R. K. Marat and A. F. Janzen, *J. Chem. Soc., Chem. Comm.*, 1977, 671.
609. R. Eisenbarth and W. Sundermeyer, *Angew. Chem.*, 1978, **90**, 226.
610. W. S. Sheldrick and O. Stelzer, *Chem. Ber.*, 1977, **110**, 3421.
611. R. A. Head and J. F. Nixon, *J. Chem. Soc., Dalton*, 1978, 885.
612. R. A. Head and J. F. Nixon, *J. Chem. Soc., Dalton*, 1978, 901.
613. R. Eisenbarth and W. Sundermeyer, *Z. Naturforsch.*, 1978, **33b**, 1194.
614. R. J. Gillespie and G. J. Schrobilgen, *Inorg. Chem.*, 1976, **15**, 22.
615. Y. A. Buslaev, V. O. Gelmboldt and S. P. Petrosyants, *Dokl. Akad. Nauk S.S.S.R.*, 1977, **236**, 880.
616. K. Hirooka, H. Kuwahara, M. Noshiro and Y. Jitsugiri, *Asahi Garasu Kenkyu Hokoku*, 1977, **27**, 39.
617. F. Jeanneaux and J. G. Riess, *Tetrahedron Letters*, 1978, 4845.
618. L. S. Bulyanitsa, G. P. Savoskina, E. N. Sventitskii and O. V. Andreeva, *Russ. J. Inorg. Chem.*, 1976, **21**, 694.
619. D. Lentz and K. Seppelt, *Angew. Chem.*, 1978, **90**, 390.
620. S. F. Lincoln, A. C. Sandercock and D. R. Stranks, *Austral. J. Chem.*, 1977, **30**, 271.
621. R. K. Marat and A. F. Janzen, *Canad. J. Chem.*, 1977, **55**, 1167.
622. C. Coulombeau, *J. Fluorine Chem.*, 1977, **9**, 483.
623. M. Contreras and H. G. Hertz, *J. Solution Chem.*, 1978, **7**, 99.
624. M. J. McGlinchey and T.-S. Tan, *J. Amer. Chem. Soc.*, 1976, **98**, 2271.
625. D. T. Browne and J. D. Otvos, *Biochem. Biophys. Res. Comm.*, 1976, **68**, 907.
626. W. E. Hull and B. D. Sykes, *Biochemistry*, 1976, **15**, 1535.
627. J. B. Kimber, D. V. Griffiths, B. Birdsall, R. W. King, P. Scudder, J. Feeney, G. C. K. Roberts and A. S. V. Burgen, *Biochemistry*, 1977, **16**, 3492.

628. G. C. K. Roberts, J. Feeney, B. Birdsall, J. B. Kimber, D. V. Griffiths, R. W. King and A. S. V. Burgen, *NMR Biol.* (Proc. Br. Biophys. Soc. Spring Meet.) 1977, 95 (ed. R. A. Dwek, I. D. Campbell and R. E. Richards), Academic Press, London.
629. J. B. Kimber, J. Feeney, G. C. K. Roberts, B. Birdsall, D. V. Griffiths, A. S. V. Burgen and B. D. Sykes, *Nature*, 1978, **271**, 184.
630. N. Lee, M. Inouye and P. C. Lauterbur, *Biochem. Biophys. Res. Comm.*, 1977, **78**, 1211.
631. J. Horowitz, J. Ofengand, W. E. Daniel and M. Cohn, *J. Biol. Chem.*, 1977, **252**, 4418.
632. A. G. Marshall and J. L. Smith, *J. Amer. Chem. Soc.*, 1977, **99**, 635.
633. W. E. Brown and K. B. Seamon, *Analyt. Biochem.*, 1978, **87**, 211.
634. W. J. Critz and M. Martinez-Carrion, *Biochemistry*, 1977, **16**, 1559.
635. D. J. Nelson, *Inorg. Chim. Acta*, 1978, **27**, L71.
636. W. H. Huestis and M. A. Raftery, *Biochem. Biophys. Res. Comm.*, 1978, **81**, 892.
637. J. W. Long and F. W. Dahlquist, *Biochemistry*, 1977, **16**, 3792.
638. K. K. Yabusaki and C. E. Ballou, *Biochemistry*, 1978, **17**, 875.
639. V. F. Bystrov, A. S. Arseniev and Yu. D. Gavrilov, *J. Magn. Resonance*, 1978, **30**, 151.
640. R. M. Bendall and G. Lowe, *FEBS Letters*, 1976, **72**, 231.
641. D. M. Kurtz, G. B. Wong and R. H. Holm, *J. Amer. Chem. Soc.*, 1978, **100**, 6777.
642. J. T. Gerig, K. E. Katz and J. D. Reinheier, *Biochim. Biophys. Acta*, 1978, **534**, 196.
643. J. T. Gerig, B. A. Halley and C. E. Ortiz, *J. Amer. Chem. Soc.*, 1977, **99**, 6219.
644. P. A. Briley, R. Eisingthal, R. Harrison and G. D. Smith, *Biochem. J.*, 1977, **163**, 325.
645. G. D. Smith, R. Eisingthal and R. Harrison, *Biochem. Soc. Trans.*, 1978, **6**, 190.
646. J. C. Slebe and M. Martinez-Carrion, *J. Biol. Chem.*, 1978, **253**, 2093.
647. M. Martinez-Carrion, J. C. Slebe, B. Boettcher and A. M. Relimpio, *J. Biol. Chem.*, 1976, **251**, 1853.
648. J.-M. Guo, C. Chang, N. C. Li and K. T. Douglas, *Biochemistry*, 1978, **17**, 452.
649. J.-L. Dimicoli, J. Bieth and J.-M. Lhoste, *Biochemistry*, 1976, **15**, 2230.
650. A. Renaud, J.-L. Dimicoli, P. Lestienne and J. Bieth, *J. Amer. Chem. Soc.*, 1978, **100**, 1005.
651. R. R. Hardy and J. H. Richards, *Biochemistry*, 1978, **17**, 3866.
652. D. A. Kooistra and J. H. Richards, *Biochemistry*, 1978, **17**, 345.
653. N. Saudenmayer, M. B. Smith, H. T. Smith, F. K. Spies and F. Millett, *Biochemistry*, 1976, **15**, 3198.
654. F. Millett, A. Ahmed, H. Smith, M. Smith, S. Ng and N. Staudenmayer, *Biomol. Struct. Funct.* (Symp.), 1977, 423 (ed. P. F. Agris), Academic Press, New York.
655. P. Puigdomenech, J. R. Daban, J. Palau, F. Podo, L. Guidoni and P. A. Temussi, *Biochim. Biophys. Acta*, 1977, **492**, 12.
656. K. B. Seamon, D. J. Hartshorne and A. A. Bothner-By, *Biochemistry*, 1977, **16**, 4039.
657. J. E. Coleman, R. A. Anderson, R. G. Ratcliffe and I. M. Armitage, *Biochemistry*, 1976, **15**, 5419.
658. J. E. Coleman and I. M. Armitage, *NMR Biol.* (Proc. Br. Biophys. Soc. Spring Meet.), 1977, 171 (ed. R. A. Dwek, I. D. Campbell and R. E. Richards), Academic Press, London.
659. R. A. Byrd, W. H. Dawson, P. D. Ellis and R. B. Dunlap, *J. Amer. Chem. Soc.*, 1977, **99**, 6139.
660. L. S. Koehler, E. T. Fossel and K. A. Koehler, *Biochemistry*, 1977, **16**, 3700.
661. L. S. Koehler, W. Curley and K. A. Koehler, *Mol. Pharm.*, 1977, **13**, 113.
662. M. P. N. Gent, P. F. Cottam and C. Ho, *Proc. Nat. Acad. Sci. USA*, 1978, **75**, 630.
663. M. P. N. Gent and C. Ho, *Biochemistry*, 1978, **17**, 3023.
664. M. P. N. Gent, I. M. Armitage and J. H. Prestegard, *J. Amer. Chem. Soc.*, 1977, **99**, 3749.
665. K. A. Koehler, M. K. Jain, E. E. Stone, E. T. Fossel and L. S. Koehler, *Biochim. Biophys. Acta*, 1978, **510**, 177.

666. C. A. Lewis, P. D. Ellis and R. B. Dunlap, *Biochem. Biophys. Res. Comm.*, 1978, **83**, 1509.
667. D. S. Hagen, J. H. Weiner and B. D. Sykes, *Biochemistry*, 1978, **17**, 3860.
668. B. C. Nicholson and T. M. Spotswood, *Austral. J. Chem.*, 1978, **31**, 2167.
669. M. Tsavalos, B. C. Nicholson and T. M. Spotswood, *Austral. J. Chem.*, 1978, **31**, 2179.
670. H. J. Knight, E. H. Williams and T. M. Spotswood, *Austral. J. Chem.*, 1978, **31**, 2187.
671. J. T. Gerig, *J. Amer. Chem. Soc.*, 1977, **99**, 1721.
672. K. A. K. Ebraheem and G. A. Webb, *Progr. NMR Spectroscopy*, 1977, **11**, 149.
673. J. W. Emsley, L. Phillips and V. Wray, *Progr. NMR Spectroscopy*, 1976, **10**, 83; J. Kowalewski, *ibid.*, 1977, **11**, 1.
674. R. E. Wasylshen, *Ann. Rep. NMR*, 1977, **7**, 245.
675. R. Ditchfield, *Chem. Phys. Lett.*, 1976, **40**, 53.
676. K. A. K. Ebraheem, G. A. Webb and M. Witanowski, *Org. Magn. Resonance*, 1976, **8**, 317.
677. K. A. K. Ebraheem and G. A. Webb, *Org. Magn. Resonance*, 1977, **10**, 70.
678. H. Sterk, W. Fabian, J. J. Suschnigg and R. Janoschek, *Org. Magn. Resonance*, 1977, **9**, 389.
679. M. F. Guest, V. R. Saunders and R. E. Overill, *Mol. Phys.*, 1978, **35**, 427.
680. H. Fukui, A. Sanyoshi and K. Miura, *J. Chem. Phys.*, 1978, **69**, 943.
681. A. C. Blizzard and D. P. Santry, *J. Chem. Phys.*, 1971, **55**, 950.
682. I. Brown and D. W. Davies, *Chem. Phys. Lett.*, 1976, **37**, 132.
683. P. E. Hansen, A. Berg, H. J. Jakobsen, A. P. Manzara and J. Michl, *Org. Magn. Resonance*, 1977, **10**, 179.
684. D. Doddrell, M. Barfield, W. Adcock, M. Aurangzeb and D. Jordan, *J. Chem. Soc., Perkin II*, 1976, 402.
685. V. Wray, unpublished data.
686. H. J. Schneider and W. Ansorge, *Tetrahedron*, 1977, **33**, 265.
687. M. Barfield, J. L. Marshall, E. D. Canada and M. R. Willcott, *J. Amer. Chem. Soc.*, 1978, **100**, 7075.
688. M. Godfrey, *J. Chem. Soc., Perkin II*, 1977, 769.
689. W. B. Smith and T. W. Proulx, *Org. Magn. Resonance*, 1976, **8**, 567.
690. D. G. Gorenstein, *J. Amer. Chem. Soc.*, 1977, **99**, 2254.
691. A. M. Vakhrameev, S. P. Gabuda and R. G. Knuborets, *Zh. Strukt. Khim.*, 1978, **19**, 298.
692. D. P. Burum, D. D. Elleman and W-K. Rhim, *J. Chem. Phys.*, 1978, **68**, 1164, see comment R. E. J. Sears, *J. Chem. Phys.*, 1978, **69**, 4321.
693. H. T. Stokes and D. C. Ailion, *Phys. Rev. (B)*, 1977, **16**, 4746.
694. V. M. Buznik, Y. N. Ivanov, A. Kriger and Y. N. Mosvich, *Tezisy Dokl. - Vses. Soveshch. Radiats. Fiz. Khim. Ionnykh Krist.*, III, 1975, **2**, 225.
695. Y. N. Moskvich, V. M. Buznik, P. P. Fedorov and B. P. Sobolev, *Kristallografiya*, 1978, **23**, 416.
696. G. A. Kirakosyan, L. N. Komissarova, L. A. Nesterova, G. Ya. Pushkina, V. F. Chubaev and V. M. Shatskii, *Russ. J. Inorg. Chem.*, 1977, **22**, 795.
697. D. R. Figueroa and J. H. Strange, *J. Phys. (C)*, 1976, **9**, L203.
698. R. J. Booth, M. R. Mustafa and B. R. McGarvey, *Phys. Rev. (B)*, 1978, **17**, 4150.
699. K. F. Lau and R. W. Vaughan, *J. Chem. Phys.*, 1976, **65**, 4825.
700. H. Selig, P. K. Gallagher and L. B. Ebert, *Inorg. Nuclear Chem. Lett.*, 1977, **13**, 427.
701. J. B. Boyce, J. Mikkelsen and M. O'Keefe, *Solid State Commun.*, 1977, **21**, 955.
702. B. D. Zilberman, T. D. Fedotova and S. P. Gabuda, *J. Struct. Chem. (U.S.S.R.)*, 1976, **17**, 238.
703. C. J. Ludman, T. C. Waddington, E. K. C. Pang and J. A. S. Smith, *J. Chem. Soc., Faraday II*, 1977, **73**, 1003.

704. P. van Hecke, H. W. Spiess, U. Haeberlen and S. Haussuhl, *J. Magn. Resonance*, 1976, **22**, 93.
705. L. M. Avkhutskii, G. A. Oktyabrskii and Y. V. Gagarinskii, *Proc. Acad. Sci. U.S.S.R., Phys. Chem. Sect.*, 1977, **234**, 522.
706. R. D. Hogg, S. P. Vernon and V. Jaccarino, *Solid State Commun.*, 1977, **23**, 781.
707. M. Hamashima, Y. Ajiro, T. Kubo and Y. Yamaguchi, *J. Phys. Soc. Japan*, 1978, **45**, 1057.
708. H. Kubo, F. Tanake, N. Kaneshima and K. Hirakawa, *Solid State Commun.*, 1976, **18**, 79.
709. M. R. Mustafa, B. R. McGarvey and E. Banks, *J. Magn. Resonance*, 1977, **25**, 341.
710. A. Reuveni and B. R. McGarvey, *J. Magn. Resonance*, 1978, **29**, 21.
711. T. Horiguchi and T. Morita, *Phys. Letters (A)*, 1978, **66**, 223.
712. E. Ildstad, I. Svare and T. Fjeldly, *Phys. Stat. Sol. (A)*, 1977, **43**, K65.
713. T. Horiguchi and T. Morita, *J. Phys. Soc. Japan*, 1977, **43**, 1485.
714. Y. Onoda and S. Minomura, *J. Phys. Soc. Japan*, 1976, **40**, 463.
715. L. D. Khoi and P. Veillet, *Phys. Rev. (B)*, 1976, **13**, 1919.
716. J. Makela and L. Niemela, *Phys. Letters (A)*, 1976, **56**, 307.
717. P. Joubert, J.-M. Weulersse, R. Bougon and B. Gaudreau, *Canad. J. Chem.*, 1978, **56**, 2546.
718. E. Ehrenfreund, P. R. Newman, A. J. Heeger, N. D. Miro and A. G. MacDiarmid, *Phys. Rev. (B)*, 1977, **16**, 1781.
719. Y. Chabre and C. Berthier, *Colloq. Metall.*, 1976, **19**, 897.
720. S. K. Garg, *J. Chem. Phys.*, 1977, **66**, 2517.
721. I. Ursu, D. E. Demco, V. Simplaceanu and N. Valcu, *Rev. Roum. Phys.*, 1977, **22**, 705.
722. N. K. Moroz, A. M. Panich and S. P. Gabuda, *Zhur. Strukt. Khim.*, 1978, **19**, 293.
723. M. Punkkinen and L. P. Ingman, *Chem. Phys. Letters*, 1976, **38**, 138.
724. H. Raber and M. Mehring, *Chem. Phys. Letters*, 1977, **49**, 498.
725. J. A. Ripmeester, D. A. Wright, C. A. Fyfe and R. K. Boyd, *J. Chem. Soc., Faraday II*, 1978, **74**, 1164.
726. V. A. Varnek, S. P. Gabuda and L. I. Obmoin, *J. Struct. Chem. (U.S.S.R.)*, 1977, **18**, 213.
727. H. Rager and A. Weiss, *Ber. Bunsenges. Phys. Chem.*, 1978, **82**, 535.
728. T. K. Halstead, H. W. Spiess and U. Haeberlen, *Mol. Phys.*, 1976, **31**, 1569.
729. S. A. Seryshev, A. G. Lundin and M. L. Afanasev, *Phys. Stat. Sol. (B)*, 1978, **90**, K35.
730. W. W. Schmiegel, *Kautsch. Gummi, Kunstst.*, 1978, **31**, 137.
731. M. F. Froix, A. O. Goedde and J. M. Pochan, *Macromolecules*, 1977, **10**, 778.
732. V. J. McBrierty, D. C. Douglass and T. A. Webber, *J. Polymer Sci. A2*, 1976, **14**, 1271.
733. D. C. Douglass and V. J. McBrierty, *Macromolecules*, 1978, **11**, 766.
734. K. Yamaguchi, N. Hayakawa and J. Okamoto, *J. Appl. Polymer Sci.*, 1978, **22**, 2653.
735. Y. Akatsuka, M. Noshiro and Y. Jitsugiri, *Asahi Garasu Kenkyu Hokoku*, 1976, **26**, 111.
736. G. Kojima, H. Wachi, K. Ishigure and Y. Tabata, *J. Polymer Sci. A1*, 1976, **14**, 1317.
737. M. S. Toy and R. S. Stingham, *J. Polymer Sci., Polymer Letters*, 1976, **14**, 717.
738. K. Matsuo, K. F. Kuhlmann, H. W. H. Yang, F. Geny, W. H. Stockmayer and A. A. Jones, *J. Polymer Sci., Polym. Phys. Ed.*, 1977, **15**, 1347.
739. A. J. Montana and N. Zumbulyadis, *J. Chem. Phys.*, 1976, **65**, 4756.
740. A. J. Montana and B. P. Dailey, *J. Chem. Phys.*, 1977, **66**, 989.
741. A. J. Montana and B. R. Appleman, *J. Chem. Phys.*, 1977, **66**, 1850.
742. M. Duchene, J. W. Emsley, J. C. Lindon, D. S. Stephenson and S. R. Salman, *J. Magn. Resonance*, 1976, **22**, 207.
743. S. A. Spearman and J. H. Goldstein, *J. Magn. Resonance*, 1978, **30**, 59.
744. J. Van der Giessen and C. A. De Lange, *Chem. Phys. Lett.*, 1978, **56**, 121.
745. A. S. Tracey, *Mol. Phys.*, 1977, **33**, 339.
746. S. A. Spearman and J. H. Goldstein, *J. Magn. Resonance*, 1977, **26**, 237.
747. J. W. Emsley and J. C. Lindon, *J. Chem. Soc., Faraday II*, 1976, **72**, 1436.

- 748. G. J. Den Otter, J. Bulthuis, C. A. De Lange and C. Maclean, *Chem. Phys. Lett.*, 1977, **45**, 603.
- 749. G. J. Den Otter and C. Maclean, *J. Mol. Struct.* 1976, **81**, 47.
- 750. M. Duchene, J. W. Emsley, J. C. Lindon, J. Overstall, D. S. Stephenson and S. R. Salman, *Mol. Phys.*, 1977, **33**, 281.
- 751. T. C. Wong and E. E. Burnell, *J. Magn. Resonance*, 1977, **22**, 227.
- 752. Y. Sugino and K. Takahashi, *Bull. Chem. Soc. Japan*, 1978, **51**, 2527.
- 753. A. Jasinski, P. G. Morris and P. Mansfield, *Mol. Cryst. Liq. Cryst.*, 1978, **45**, 183.
- 754. G. J. T. Tiddy, *J. Chem. Soc., Faraday*, 1977, **73**, 1731.
- 755. G. N. Holland, P. A. Bottomley and W. S. Hinshaw, *J. Magn. Resonance*, 1977, **28**, 133.
- 756. K. Ishigure, H. Ohashi, Y. Tabata and K. Oshima, *Macromolecules*, 1977, **10**, 567.
- 757. A. J. Sadlej, *Acta Phys. Polon.*, 1976, **A49**, 667.
- 758. B. Day and A. D. Buckingham, *Mol. Phys.*, 1976, **32**, 343.
- 759. P. Swannstrom, W. P. Kraemer and G. H. F. Diercksen, *Theor. Chim. Acta*, 1977, **44**, 109.
- 760. M. Zaucer, D. Pumpernik, M. Hladnik and A. Azman, *Z. Naturforsch.*, 1977, **32a**, 411.
- 761. P. Van Hecke, H. W. Spiess and U. Haeberlen, *J. Magn. Resonance*, 1976, **22**, 93, 103.
- 762. M. Zaucer, D. Pumpernik, M. Hladnik and A. Azman, *Z. Naturforsch.*, 1976, **31a**, 1727.
- 763. W. T. Raynes, in "Nuclear Magnetic Resonance (Specialist Periodical Reports)", R. J. Abraham (ed.), The Chemical Society, London, 1978, Ch. 1.

This Page Intentionally Left Blank

Cumulative index of authors who have contributed to Volumes 1–10

- Akitt, J. W. **5A**, 466
 Ando, I., **10A**, 1
 Asakura, T., **10A**, 81
- Bedford, G. R., **4**, 1
 Bishop, E. O., **1**, 91
 Bladon, P., **2**, 1
 Buhler, **4**, 237
- Cargioli, J. D., **9**, 221
 Cavalli, L., **6B**, 43
 Crabb, T. A., **6A**, 250, **8**, 1
 Cudby, M. E. A., **4**, 363
- Dwek, R. A., **2**, 293
- Ewing, D. F., **6A**, 148
- Fields, R., **5A**, 99, **7**, 1
- Gillies, D. G., **5A**, 560
 Gryff-Keller, A., **8**, 228
- Haigh, C. W., **4**, 311
 Hall, G. E., **1**, 227
 Haque, R., **4**, 237
 Henderson, W. G., **2**, 219
 Huckerby, T. N., **3**, **2**, **5A**, 1
- Inch, T. D., **2**, 35, **5A**, 305
- Jones, K., **3**, 261, **4**, 391
 Jones, R. A. Y., **1**, 1
- Kidd, R. G., **10A**, 1
 Krow, G., **6A**, 148
- Lini, D. C., **6A**, 148
- McFarlane, W., **1**, 135, **5A**, 353, **9**, 320
 Mavel, G., **5B**, 1
- Mooney, E. F., **1**, 244, **2**, 125, 153, 219,
3, 261, **4**, 391
- Nixon, J. F., **2**, 346
- O'Neill, I. K., **10A**, 134
 Orell, K. G., **9**, 1
- Page, J. E., **3**, 149
 Pidcock, A., **2**, 346
- Ramey, K. C., **6A**, 148
 Richards, C. P. **10A**, 134
 Richards, R. E., **2**, 293
 Ronayne, J., **2**, 83
 Rycroft, D. S., **9**, 320
- Sewell, P. R., **1**, 165
 Shaw, D., **5A**, 560
 Smith, P. J., **8**, 292
 Smith, W. B., **8**, 199
 Stefaniak, L., **7**, 118
 Sutherland, I. O., **4**, 71
 Szymański, S., **8**, 228
- Taylor, D., **2**, 293
 Thomas, W. A. **1**, 44
 Thomas, W. A., **3**, 92
 Thomas, W. A., **6B**, 1
 Tupčiauskas, A. P., **8**, 292
- Wasylishen, R. E., **7**, 246
 Webb, G. A., **3**, 211, **5A**, 395, **6A**, 1, **7**,
 118
 Wehrli, F. W., **9**, 126
 Williams, D. H., **2**, 83
 Williams, E. A., **9**, 221
 Willis, H. A., **4**, 363
 Winson, P. H., **1**, 244, **2**, 125, 153
 Witanowski, M., **7**, 118, **5A**, 395, **8**, 228
 Wray, V., **10B**, 1

Cumulative index of topics covered in Volumes 1–10 of this series

- Alkaloids, NMR of, **6A**, 250, **8**, 1
Aluminium and gallium, NMR spectroscopy of liquids containing compounds of, **5A**, 466
Amino acids, peptides and proteins, NMR and conformations of, **6B**, 1
Biological ^{31}P NMR spectroscopy, **10A**, 134
Boron-11 NMR spectroscopy, **2**, 219
Carbohydrates and related compounds, NMR spectroscopy in the study of, **2**, 35, **5A**, 305
Carbon-13 NMR spectroscopy: Carbon-13 chemical shifts and coupling constants, **2**, 153
 of steroids, **8**, 199
Carbon-13 and first row nuclei, spin-spin coupling between, **7**, 246
Carbon-13 and proteins two bond couplings between **6A**, 389
Conformational analysis, NMR spectroscopy in, **1**, 44
 NMR spectroscopy as an aid in, **3**, 92
Conformational changes by NMR spectroscopy, the investigation of the kinetics of, **4**, 71
Conformations of amino acids, peptides and proteins, NMR and, **6B**, 1
Dynamic NMR spectra, problems in the theory and analysis of, **8**, 228
Fluorine-19 NMR spectroscopy, **1**, 244, **3**, 261, **4**, 391, **5A**, 99, **6B**, 53
 of fluoroalkyl and fluoroaryl derivatives of transition metals, **7**, 1
 (1976–1978), **10B**, 1
General review of NMR, **6A**, 148
Heteronuclear magnetic double resonance, **1**, 135, **5A**, 353
High-resolution NMR spectra, the interpretation of, **1**, 91
High resolution NMR spectroscopy, the application of fourier transformation to, **5A**, 560
Iterative computer programmes in the analysis of NMR spectra, a simple guide to the use of, **4**, 311
Kinetics of conformational changes by NMR spectroscopy, the investigation, of, **4**, 71
Liquids containing compounds of aluminium and gallium, NMR spectroscopy in, **5A**, 466
Magnetic multiple resonance, **9**, 320
Nitrogen NMR spectroscopy, **2**, 125, **5A**, 395, **7**, 118
Nuclear electron double resonance in liquids, **2**, 293
Nuclear shielding of the transition metals, **10A**, 1
Paramagnetic species, NMR spectroscopy of, **3**, 211, **6A**, 1, **9**, 1
Pesticide chemistry, NMR spectroscopy in, **4**, 237
Phosphorus compounds, NMR studies of, (1965–1969), **5B**, 1
Phosphorus-31 NMR spectra of co-ordination compounds, **2**, 346
 biological aspects of, **10A**, 134

- Polymers, the NMR spectra of, **1**, 165, **4**, 363
 NMR chemical shift calculations and stereochemical structures of, **10A**, 81
Protons and carbon-13, two-bond couplings between, **6A**, 389
Proton magnetic resonance, general reviews of, **1**, **1**, **2**, **1**, **3**, **1**, **4**, **1**, **5A**, **1**
Quadrupolar nuclei, NMR of the less common, **9**, 126
Signal-to-noise enhancement of NMR spectra, **1**, 227
Silicon-29 NMR spectroscopy, **9**, 221
Solvent effects in proton magnetic resonance spectroscopy, **2**, 83
Spin-Spin coupling between carbon-13 and the first row nuclei, **7**, 246
Steroids, NMR spectra of, **3**, 149
 ¹³C NMR spectroscopy of, **8**, 199
Synthetic polymers, NMR chemical shift calculations and stereochemical structures of, **10A**, 81
Tin-119 nuclei in organotin compounds, chemical shifts of, **8**, 292
Transition metals, nuclear shielding of, **10A**, **1**

This Page Intentionally Left Blank